

REFERENCES

- Alexander, M. 1994. **Biodegradation and Bioremediation**. U.S.A: Academic Press.
- Ambujom, S., and Manilal, V.B. 1995. Phenol degradation by a stable aerobic consortium and its bacterial isolates. **Biotechnol. Lett.** 17 : 443-448.
- Appanna, V.D., Gazso, L.G., Huang, J. and Pierre, M.St. 1996. Mechanism of chromium detoxification in *Pseudomonas fluorescens* is dependent on iron. **Bull. Environ. Contam. Toxicol.** 57 : 875-880.
- Arslan, P., Beltrame, M. and Tomasi, A. 1987. Intracellular chromium reduction. **Biochim. et Biophys. Acta.** 931 : 10-15.
- Bayly, R.C., and Wigmore, G.J. 1973. Metabolism of phenol and cresols by mutants of *Pseudomonas putida*. **J. Bacteriol.** 113 : 1112-1120.
- Bird, N. P., Chambers, J. G., Leech, R. W., and Cummins, D. 1985. A note on the use of metal species in microbiological tests involving growth media. **J. Appl. Bacteriol.** 59: 353-355.
- Bopp, L.H., Chakrabarty, A.M. and Ehrlich, H.L. 1983. Chromate resistance plasmid in *Pseudomonas fluorescens*. **J. Bacteriol.** 155 : 1105-1109.
- Bossert I., and Young L.Y. 1986. Anaerobic oxidation of p-cresol by a denitrifying bacterium. **Appl. Environ. Microbiol.** 52 : 1117-1122.
- Boyd, S.A., Shelton, D.R., Berry, D., Tiedje, J.M. 1983. Anaerobic biodegradation of phenolic compounds in digested sludge. **Appl. Environ. Microbiol.** 46 : 50-54.
- Bruijn, F.J. de, Lupski, J.R. and Weinstock, G.M. 1998. **Bacterial genomes**. U.S.A: Chapman & Hall.

- Cervantes, C., Ohtake, H., Chu, L., Misra, T.K. and Silver, S. 1990. Cloning, nucleotide sequence, and expression of the chromate resistance determinant of *Pseudomonas aeruginosa* plasmid pUM505. **J. Bacteriol.** 172 : 287-291.
- Chakrabarty, A.M., Mylroie, J.R., Friella, D.A. and Vacca, J.G. 1975. Transformation of *Pseudomonas putida* and *Escherichia coli* with plasmid-linked drug-resistance factor DNA. **Proc. Nat Acad. Sci.** 72 : 3647-3651.
- Chang, S.Y., Li, C.T., Hiang, S.Y. and Chang, M.C. 1995. Intraspecific protoplast fusion of *Candida tropicalis* for enhancing phenol degradation. **Appl. Microbiol. Biotechnol.** 43 : 534-538.
- Cheremisinoff, P.N. 1995. **Handbook of water and wastewater treatment technology.** U.S.A: Marcel Dekker.
- Chitra, S., Sekaran, G. and Chandrakasan, G. 1996. Immobilized mutant strain of *Pseudomonas pictorum* for the degradation of phenol in wastewater. **J. Gen. Appl. Microbiol.** 42 : 355-361.
- Corn, M. 1993. **Handbook of hazardous materials.** U.S.A: Academic Press.
- Dagley, S., and Gibson, D.T. 1965. The bacterial degradation of catechol. **Biochem. J.** 95 : 466-474.
- Das, S., and Chandra, A.L. 1990 Chromate reduction in Streptomycers. **Experientia.** 46 : 731-733.
- Dwyer, D.F., Krumme, M.L., Boyd, S.A. and Tiedje, J.M. Kinetics of phenol biodegradation by an immobilized methanogenic consortium. **Appl. Environ. Microbiol.** 52 : 345-351.
- Evans, W.C. 1977. Biochemistry of the bacterial catabolism of aromatic compounds in anaerobic environments. **Nature.** 270 : 17-22.
- Evans, W.C., and Fuch, G. 1988. Anaerobic degradation of aromatic compounds. **Annu. Rev. Microbiol.** 42 : 289-317.

- Feist, C.F., and Hegeman, G.D. 1969. Phenol and benzoate metabolism by *Pseudomonas putida* : regulation of tangential pathways. **J. Bacteriol.** 100 : 869-877.
- Greenberg, A.E., Clesceri L.S. and Eaton A.D. 1992. **Standard methods for the examination of water and wastewater.** U.S.A: American Public Health Association.
- Horitsu, H., Futo, S., Miyazawa, Y., Ogai, S. and Kawai, K. 1987. Enzymatic reduction of hexavalent chromium by hexavalent tolerant *Pseudomonas ambigua* G-1. **Agric. Biol. Chem.** 51 : 2417-2420.
- Ishibashi, Y., Cervantes, C. and Silver, S. 1990. Chromium reduction in *Pseudomonas putida*. **Appl. Environ. Microbiol.** 56 : 2268-2270.
- Kananidhinan L. 1996. **The uses of aquatic plant in constructed wetlands for chromium treatment from electroplating industrial wastewater.** Thailand : Science Thesis. Chulalongkorn University.
- Kasak, L., Horak, R., Nurk, A., Talvik, K. and Kivisaar, M. 1993. Regulation of the catechol 1,2-dioxygenase- and phenol monooxygenase-encoding *pheBA* operon in *Pseudomonas putida* PaW85. **J. Bacteriol.** 175 : 8038-8042.
- Keith, L.H., and Telliard, W.A. 1979. Priority pollutants I. A perspective view. **Environ Sci. Technol.** 13 : 416-423.
- Kim, S.U., and Dhurjati, P. 1987. Analysis of two interaction bacterial populations with opposite substrate preferences. **Biotechnol. Bioeng.** 29 : 1015-1023.
- Kim, Y., Ayoubi, P. and Harker, A.R. 1996. Constitutive expression of the cloned phenol hydroxylase gene(s) from *Alcaligenes eutrophus* JMP134 and concomitant trichloroethylene oxidation. **Appl. Environ. Microbiol.** 62 : 3227-3233.

- Kitchainukool K. and Nakkeaw P. 1994. **The use of anion-exchange method in chromium removal from chrome plating effluent.** Thailand : Science Project. Chulalongkorn University.
- Llovera, S., Bonet, R., Simon-Pujol, M.D. and Congregado. F. 1993. Chromate reduction by resting cells of *Agrobacterium radiobacter* **EPS-916. Appl. Environ. Microbiol. 59 : 3516-3518.**
- Lovley, D.R. 1993. Dissimilatory metal reduction. **Annu. Rev. Microbiol. 47 : 263-290.**
- Lovley, D.R., and Lonagan, D.J. 1990. Anaerobic oxidation of toluene, phenol, and p-cresol by the dissimilatory iron-reducing organism, **GS-15. Appl. Environ. Microbiol. 56 : 1858-1864.**
- Lovley, D.R., and Phillips, E.J.P. 1994. Reduction of chromium by *Desulfovibrio vulgaris* and its C₃ cytochrome. **Appl. Environ. Microbiol. 60 : 726-728.**
- Luli, G.W., Talnagi, J.W., Strohl, W.R. and Pfister, R.M. 1983. Hexavalent chromium-resistant bacteria isolates from river sediments. **Appl. Environ. Microbiol. 46 : 846-854.**
- MacLeod, R. A., Kuo, S. C., and Gelinas, R. 1967. Metabolic injury to bacteria II Metabolic injury induced by distilled water of copper in the plating diluent. **J. Bacteriol. 93 : 317-324.**
- Masscheleyn, P.H., Pardue, J.H., Delaune, R.D. and Patrick, W.H. 1992. Chromium redox chemistry in a lower mississippi valley bottomland hardwood wetland. **Environ. Sci. Technol. 26 : 1217-1226.**
- Murakami, Y., and Alexander, M, 1989. Destruction and formation of toxins by one bacterial species affect biodegradation by a second species. **Biotechnol. Bioeng. 33 : 832-838.**

- Ohtake, H., Cervantes, C. and Silver, S. 1987. Decreased chromate by *Pseudomonas fluorescens* carrying a chromate resistance plasmid. **J. Bacteriol.** 169 : 3853-3856.
- Ohtake, H., Fujii, E. and Yoda, K. 1990. A survey of effective electron donors for reduction of toxic hexavalent chromium by *Enterobacter cloacae* (strain HO1). **J. Gen. Appl. Microbiol.** 36 : 203-208.
- Panikov, N.S. 1995. **Microbial growth kinetics**. U.K: Chapman&Hall.
- Pansawas T., and Chavalparit A. 1994. **Chromium recovery from tanning industrial wastewater**. Thailand : Industrial factory department. 97 p.
- Peter, M., Heinaru, E., Talpsep, E., Wand, H., Stottmeister, U., Heinaru, A. and Nurk, A. 1997. Acquisition of a deliberately introduced phenol degradation operon, *pheBA*, by different indigenous *Pseudomonas* species. **Appl. Environ. Microbiol.** 63 : 4899-4906.
- Petrilli, F.L., and Flora, S.De. 1977. Toxicity and mutagenicity of hexavalent chromium on *Salmonella typhimurium*. **Appl Environ. Microbiol.** 33 : 805-809.
- Ramos, J.L., Marques, S. and Timmis, K.N. 1997. Transcriptional control of the *Pseudomonas* TOL plasmid catabolic operons is achieved through an interplay of host factors and plasmid-encoded regulators. **Annu. Rev. Microbiol.** 51 : 341-373.
- Sawyer C.N., McCarty P.L. and Parkin, E.F. 1994. **Chemistry for Environmental Engineering**. U.S.A: McGraw-Hill.
- Shen H., Pritchard, P.H. and Sewell, G.W. 1996. Kinetic of chromate reduction during naphthalene degradation in a mixed culture. **Biotechnol. Bioeng.** 52 : 357-363.

- Shen H., and Wang, Y.T. 1993. Characterization of enzymatic reduction of hexavalent chromium by *Escherichia coli* ATCC 33456. **Appl. Environ. Microbiol.** 59 : 3771-3777.
- Shen H., and Wang, Y.T. 1994a. Modelling hexavalent chromium reduction in *Escherichia coli* ATCC 33456. **Biotechnol. Bioeng.** 43 : 293-300.
- Shen H., and Wang, Y.T. 1994b. Biological chromium reduction by *E. coli*. **J. Environ. Eng.** 120 : 560-572.
- Shen H., and Wang, Y.T. 1995a. Simultaneous chromium reduction and phenol degradation in a coculture of *Escherichia coli* ATCC 33456 and *Pseudomonas putida* DMP-1. **Appl. Environ. Microbiol.** 61 : 2754-2758.
- Shen H., and Wang, Y.T. 1995b. Modeling simultaneous hexavalent chromium reduction and phenol degradation by a defined coculture of bacteria. **Biotechnol. Bioeng.** 48 : 606-613.
- Silver S., and Phung, L. T. 1996. Bacterial heavy metal resistance: new surprises. **Annu. Rev. Microbiol.** 14: 381-386, 753-789.
- Smith, M.R., Ewing, M., and Rutledge, C. 1991. The interactions of various aromatic substrates degraded by *Pseudomonas sp.* NCIB10643: synergistic inhibition of growth by two compounds that serve as growth substrates. **Appl. Microbiol. Biotechnol.** 34 : 536-538.
- Smithson, G. R. 1971. **An investigation of techniques for the removal of chromium from electroplating wastes.** U.S.A. : U.S. E.P.A. Press.
- Stanley, R.A. 1974. Toxicity of heavy metals and salts to Eurasian water milfoil (*Myriophyllum spicatum* L.). **Arch. Environ. Contam. Toxicol.** 2 : 331-341.

- Subba-Rao. R.V., Rubin, H.E., and Alexander, M. 1982. Kinetics and extent of mineralization of organic chemicals at trace levels in freshwater and sewage. **Appl. Environ. Microbiol.** 43 : 1139-1150.
- Suzuki, T., Miyata, N., Horitsu, H., Kawai, K., Takamizawa, K., Tai, Y. and Okazaki, M. 1992. NAD(P)H-dependent chromium(VI) reductase of *Pseudomonas ambigua* G-1 : a Vr(V) intermediate is formed during the reduction of Cr(VI) to Cr(III). **J. Bacteriol.** 174 : 5340-5345.
- Summers, A.O., and Jacoby, G.A. 1978. Plasmid-determined resistance to boron and chromium compounds in *Pseudomonas aeruginosa*. **Antimicrobial Agents and Chemotherapy.** 13 : 637-640.
- Tabak, H.H., Chambers, C.W. and Kabler, P.W. 1964. Microbial metabolism of aromatic compounds. **J. Bacteriol.** 87 : 910-919.
- US. EPA. 1978. **Reviews of the environmental effects of pollutants. III. Chromium.** Washington DC. US. EPA. 285 p.
- Voordouw, G., Pollock, W.B.R., Bruschi, M., Guerlesquin, F., Rapp-Giles, B.J. and Wall, J.D. 1990. Functional expression of *Desulfovibrio vulgaris* hildenborough cytochrome c_3 in *Desulfovibrio desulfuricans* G200 after conjugational gene transfer from *Escherichia coli*. **J. Bacteriol.** 172 : 6122-6126.
- Wang, P., Mori, T., Toda, K. and Ohtake, H. 1990. Membrane – associated chromate reductase activity from *Enterobacter cloacae*. **J. Bacteriol.** 172 : 1670-1672.
- Wang Y.T., Suidan, M.T., Pfeffer, J.T. and Najam, I. 1989. The effect of concentration of phenols on their batch methanogenesis. **Biotechnol. Bioeng.** 33 : 1353-1357
- Wentz, C. A. 1995. **Hazardous Waste Management.** Second edition. Singapore: McGraw – Hill.

WHO. 1988. **Chromium**. Geneva. Switzerland: WHO.

WHO. 1994. **Phenol**. Geneva Switzerland: WHO.

Wiggins, B.A., and Alexander M. 1988. Role of chemical concentration and second carbon sources in acclimation of microbial communities for biodegradation. **Appl. Environ. Microbiol.** 54 : 2803-2807.

Wiggins, B.A., Jones, S.H., and Alexander, M. 1987. Explanations for the acclimation period preceding the mineralization of organic chemicals in aquatic environments. **Appl. Environ. Microbiol.** 53 : 791-796.

Wilkinson, J. F. 1958. Extracellular bacterial polysaccharides. **Bacteriological Reviews.** 22: 46-69. macromolecules in soil 2. Characterization of the maximum binding ability of the macromolecules. **Soil Sci.** 123: 188.



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APPENDICES

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APPENDIX A

BACTERIAL SOURCES

Fifty samples were chosen as bacterial source. They were collected during December 1997 to November 1998, and were categorized into various groups as follows :

Soil

The soil samples were collected from:

Sampling Sites	Date of Collection
Chromium plating, area I, Bangkok	December 6, 1997
Chromium plating, area II, Bangkok	December 6, 1997
Printing machinery, area II, Bangkok	December 7, 1997
Wood, area I, Bangkok	February 3, 1998
Metal plating, area I, Bangkok	February 9, 1998
Painting, area II, Bangkok	February 10, 1998
Klong Taweewattana, area I, Bangkok	April 8, 1998
Klong Bangbua, area IX, Nonthaburi	April 8, 1998
Oil refinery, area VI, Samutprakarn	April 9, 1998
Duck farm (1), area VII, Nakornprathom	April 10, 1998
Duck farm (2), area VII, Nakornprathom	April 10, 1998
Chicken farm (1), area VII, Nakornprathom	April 10, 1998
Chicken farm (2), area VII, Nakornprathom	April 10, 1998
Chicken farm (3), area VII, Nakornprathom	April 10, 1998
Pig farm (1), area VII, Nakornprathom	April 10, 1998
Pig farm (2), area VII, Nakornprathom	April 10, 1998
Pig farm (3), area VII, Nakornprathom	April 10, 1998

Sampling Sites	Date of Collection
Wastewater treatment, area VI, Samutprakarn	April 10, 1998
Chemical production, area V, Samutprakarn	May 9, 1998
Can production industry, area VI, Samutprakarn	May 9, 1998
Industrial sector, area V, Samutprakarn	May 9, 1998
Rice field, area VIII, Pathumtani	June 8, 1998
Orange orchard, area III, Bangkok	June 8, 1998
Treatment sector, area IV, Bangkok	June 8, 1998
Area (1), Chulalongkorn University	October 3, 1998
Garbage heap, area (1), Chulalongkorn University	October 3, 1998
Chaopraya River (1), area II, Bangkok	October 3, 1998
Chaopraya River (2), area II, Bangkok	October 3, 1998
ferric company (1), area X, Karnchanaburi	November 3, 1998
ferric company (2), area X, Karnchanaburi	November 3, 1998

Wastewater

The samples were collected from:

Sampling Sites	Date of Collection
Chromium plating, area I, Bangkok	December 6, 1997
Chromium plating, area II, Bangkok	December 6, 1997
Printing machinery, area II, Bangkok	December 7, 1997
Wood, area I, Bangkok	February 3, 1998
Metal plating, area I, Bangkok	February 9, 1998
Painting, area II, Bangkok	February 10, 1998
Oil refinery, area VI, Samutprakarn	April 9, 1998
Wastewater treatment, area VI, Samutprakarn	April 10, 1998
Chemical production, area V, Samutprakarn	May 9, 1998
Industrial sector, area V, Samutprakarn	May 9, 1998
Treatment sector, area IV, Bangkok	June 8, 1998

Sludge

The samples were collected from:

Sampling Sites	Date of Collection
Chromium plating, area I, Bangkok	December 6, 1997
Wood, area I, Bangkok	February 3, 1998
Painting, area II, Bangkok	February 10, 1998
Oil refinery, area VI, Samutprakarn	April 9, 1998
Chemical production, area V, Samutprakarn	May 9, 1998

Natural Water

The samples were collected from:

Sampling sites	Date of Collection
Klong Taweewattana, area I, Bangkok	April 8, 1998
Klong Bangbua, area IX, Nonthaburi	April 8, 1998
Chaopraya River (1), area II, Bangkok	October 3, 1998
Chaopraya River (2), area II, Bangkok	October 3, 1998

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APPENDIX B

CULTURE MEDIA

1. Nutrient Broth (NB) (Difco laboratories, Detroit, Michigan, U.S.A.)
(Shen and Wang, 1995a)

Formula in gram per 1 liter

- Beef extract	3
- Bacto peptone	5

Final pH 7.1

2. Nutrient Agar (NA) (Difco)

Formula in gram per 1 liter

- Beef extract	3
- Bacto peptone	5
- Agar	15

Final pH 7.1

The NB medium was prepared by suspending 8 gram of NB in 1 L of distilled water and added 15 gram of agar when prepare NA medium and boil to dissolve completely by microwave. Then, the medium was autoclaved at 121°C, 1 atmosphere for 15 min. All media were dispended in plates and before used, plate was incubated over night for checking of sterilization.

3. Pseudomonas Selective Isolation Agar (PSIA)

(adapted from Krulger and Sheikh, 1986)

Formula in milliliter and gram per 1 liter

- Nitrofurantoin (5% solution)	7
- Crystal violet (0.1 % solution)	2
- TSB	30
- Agar	15
- Distilled Water	990

Pseudomonas selective isolation agar (PSIA) was prepared as follows. A stock solution of 5 % (wt/vol) nitrofurantoin (Sigma, Steinheim, Germany), was prepared in N,N-dimethylformamide (Merck, Darmsatadt, Germany). A stock solution of 0.1 % (wt/vol) crystal violet (Merck, Darmsatadt, Germany) was prepared in deionized water. The stock solution were stored at room temperature, and nitrofurantoin solution was protected from exposure to light. The medium (PSIA) was prepared by suspending 30 gram of TSB and 15 gram of agar in 990 ml distilled water and added 2 ml of crystal violet stock solution. After the mixture was autoclaved at 121 ° C for 15 min and then cooled to 50°C 7 ml of nitrofurantoin stock solution were added (adapted from Krulger and Sheikh, 1986). All media were dispended in plates and before used plates was incubated over night for checking of sterilization.

4. MacConkey-inositol-potassium tellurite (MCIK) agar
(adapted from Toman, Cirerana and Jofre, 1986)

Formula in milliliter and gram per 1 liter

- MacConkey Agar	40
- Myo-inositol	10 mM
- Potassium tellurite	0.003

The medium (MCIK) was prepared by suspending 40 gram of MacCongey agar in 1 L of distilled water. After the mixture was autoclaved at 121 ° C for 15 min and then cooled to 50 ° C, myo-inositol (Fluka, Messerschmittstr, Switzerland), final concentration 10 mM. and potassium tellulite (Merck, Darmsatadt, Germany), final concentration, 3 µg/ml were added (adapted from Toman, Cirerana and Jofre, 1986). All media were dispended in plates and before used plates was incubated over night.

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5. Shigella and Salmonella (SS) Agar (Difco)

Formula in gram per 1 liter

- Bacto Beef Extract	5
- Bacto Proteose Peptone	5
- Bacto Lactose	10
- Bacto Bile Salt No. 3	8.5
- Sodium Citrate	8.5
- Sodium Thiosulfate	8.5
- Ferric Citrate	1
- Bacto Agar	13.5
- Brilliant Green	0.33 mg
- Neutral Red	0.025

Final pH 7.0 ± 0.2 at 25°C

Suspend 60 gram in 1 liter distilled or deionized water and boil carefully for no more than 2-3 minutes to dissolve completely. Avoid overheating. Do not autoclave.

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6. MacConkey Agar (Difco)

Formula in gram per 1 liter

- Bacto Peptone	17
- Bacto Proteose Peptone	3
- Bacto Lactose	10
- Bacto Bile Salts No. 3	1.5
- Sodium Chloride	5
- Bacto Agar	13.5
- Neutral Red	0.03
- Bacto Crystal Violet	0.001

The medium was prepared by suspending 50 gram of MacConkey agar in 1 L of distilled water and boil to dissolve completely. After, the medium was autoclaved at 121 °C for 15 min. Avoid overheating. All media were dispensed in plates and before used plates was incubated over night.

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Selective Medium

Strains	MCIK	PSIA	MacConkey	EMB	SS	EC	BHI	SRB	King A	S
CrR-2	+ / pink	-	white	+ / pink	-	+ / growth	+	-	- / yellow	-
CrR-14	-	+ / pink	pink	-	-	-	+	+	- / yellow	-
CrR-15	-	-	-	-	-	+ / growth	+	+	- / yellow	-
PhR-26	+ / pink	-	-	-	-	-	+	++	- / yellow	-
PhR-33	-	+ / pink	pink	-	-	-	+	+	- / yellow	-
PhR-64	-	-	white	+ / pink	-	+ / growth	+	+	- / yellow	-
CPR-4	+ / pink	+ / pink	pink	+ / pink	colorless	+ / growth	+	+	- / yellow	+ / green
CPR-16	+ / pink	+ / pink	pink	+ / pink	pink	+ / growth	+	++	+ / green	+ / green
CPR-17	+ / pink	-	white	+ / pink	colorless	+ / growth	+	++	- / yellow	+ / green

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APPENDIX C

FORMULAR AND PREPARATION OF SOME BIOCHEMICAL TESTS AND RESULT OF SOME SELECTED BACTERIAL STRAINS

1. Motility test medium

Formula in gram per 1 liter

- Beef extract	3
- Peptone	10
- NaCl	5
- Agar	4

Final pH 7.3

2. MR/VP broth

Formula in gram per 1 liter

- Polypeptone	7
- Glucose	5
- Dipotassium phosphate	5

Final pH 6.9

3. Simmons Citrate Agar

Formula in gram per 1 liter

- Magnesium Sulfate	0.2
- Ammonium Dihydrogen Phosphate	1
- Dipotassium Phosphate	1
- Sodium Citrate	2
- Sodium Chloride	5
- Bacto Agar	15
- Bacto Brom Thymol Blue	0.08

Final pH 6.8 at 25 °C

Preparation : To rehydrate the medium, suspend 24.2 grams in 1L, cold freshly distilled water and heat to boiling to dissolve the medium completely. Sterilize in the autoclave for 15 minutes at 15 pounds pressure (121 °C).

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4. Three Sugar Iron Agar (TSI)

Formula in gram per 1 liter

- Bacto Beef Extract	3	
- Bacto Yeast Extract	3	
- Bacto Peptone	15	
- Proteose Peptone	5	
- Bacto Dextrose	1	
- Bacto Lactose	10	
- Saccharose	10	
- Ferrous Sulfate	0.2	
- Sodium Sulfate	5	
- Sodium Thiosulfate	0.3	
- Bacto Agar	12	
- Bacto Phenol Red	24	mg

Final pH 7.4 at 25 °C

Preparation : To rehydrate the medium, suspend 65 grams in 1000 ml, cold freshly distilled water and heat to boiling to dissolve the medium completely. Sterilize in the autoclave for 15 minutes at 15 pounds pressure (121 °C). Allow the tubes to solidify in a slanting position in a manner which will give a generous butt.

Biochemical Tests

Strains	motile	nitrate	citrate	TSI	indole	gelatin	oxidase	P	urease	LB	KCN	MR	VP
CrR-2	+	+	-	KK+-	+	-	-	-	-	+	-	+	-
CrR-14	-	++	+	AA--	+	+	+	+	-	-	+	-	-
CrR-15	+	+	+	AA--	-	-	-	-	-	+	-	-	+
PhR-26	+	++	+	KK--	-	-	-	-	+	-	+	-	+
PhR-33	-	++	+	AA--	+	+	+	+	-	-	+	-	-
PhR-64	+	+	-	KK+-	+	-	-	-	-	+	-	+	-
CPR-4	-	++	+	AA--	+	+	+	+	-	-	+	-	-
CPR-16	+	++	+	KK++	+	-	-	+	+	-	+	+	+
CPR-17	+	+	-	KK+-	+	-	-	-	-	+	-	+	-

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APPENDIX D

STANDARD CURVE OF CHROMIUM AND PHENOL

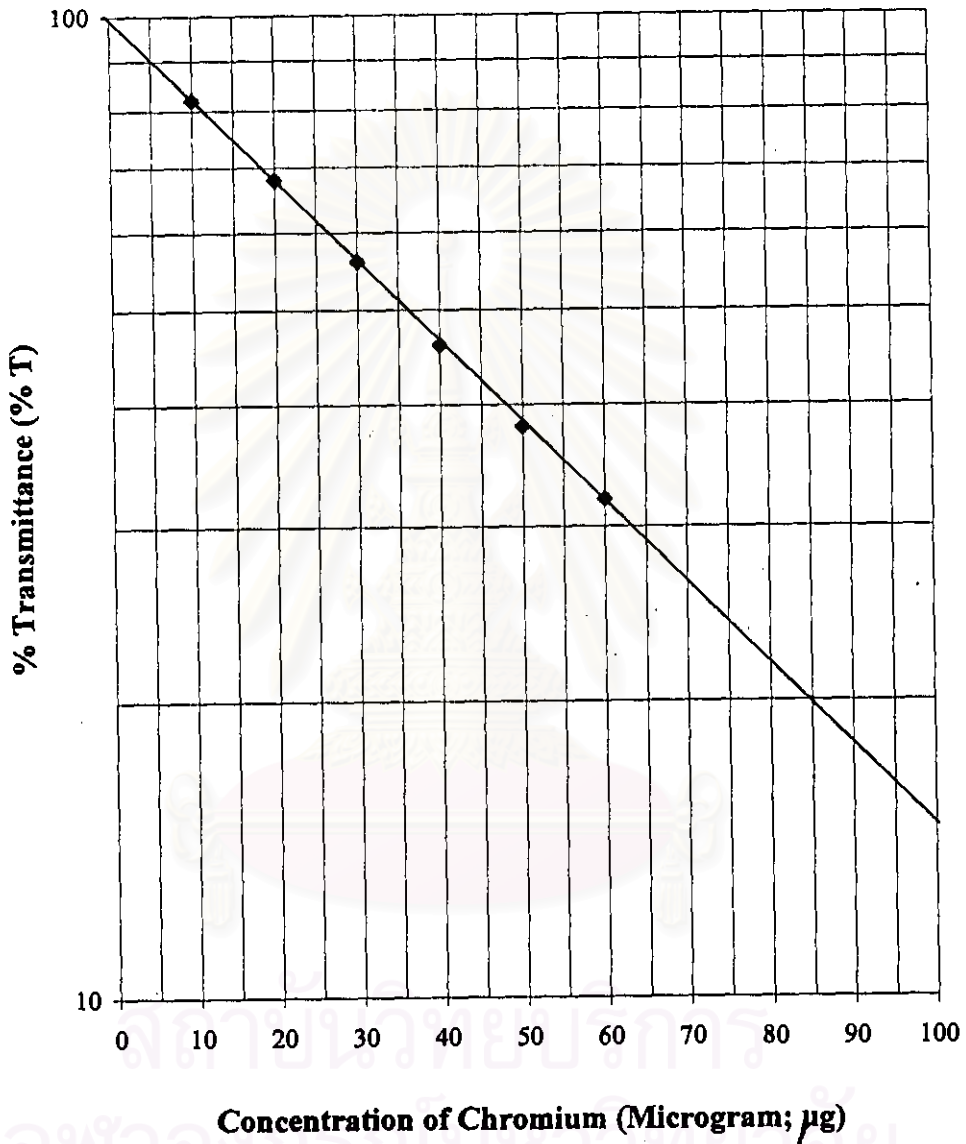


Figure D.1 Standard curve of chromium by colorimetric method (Greenberg, A.E., et.al., 1992)

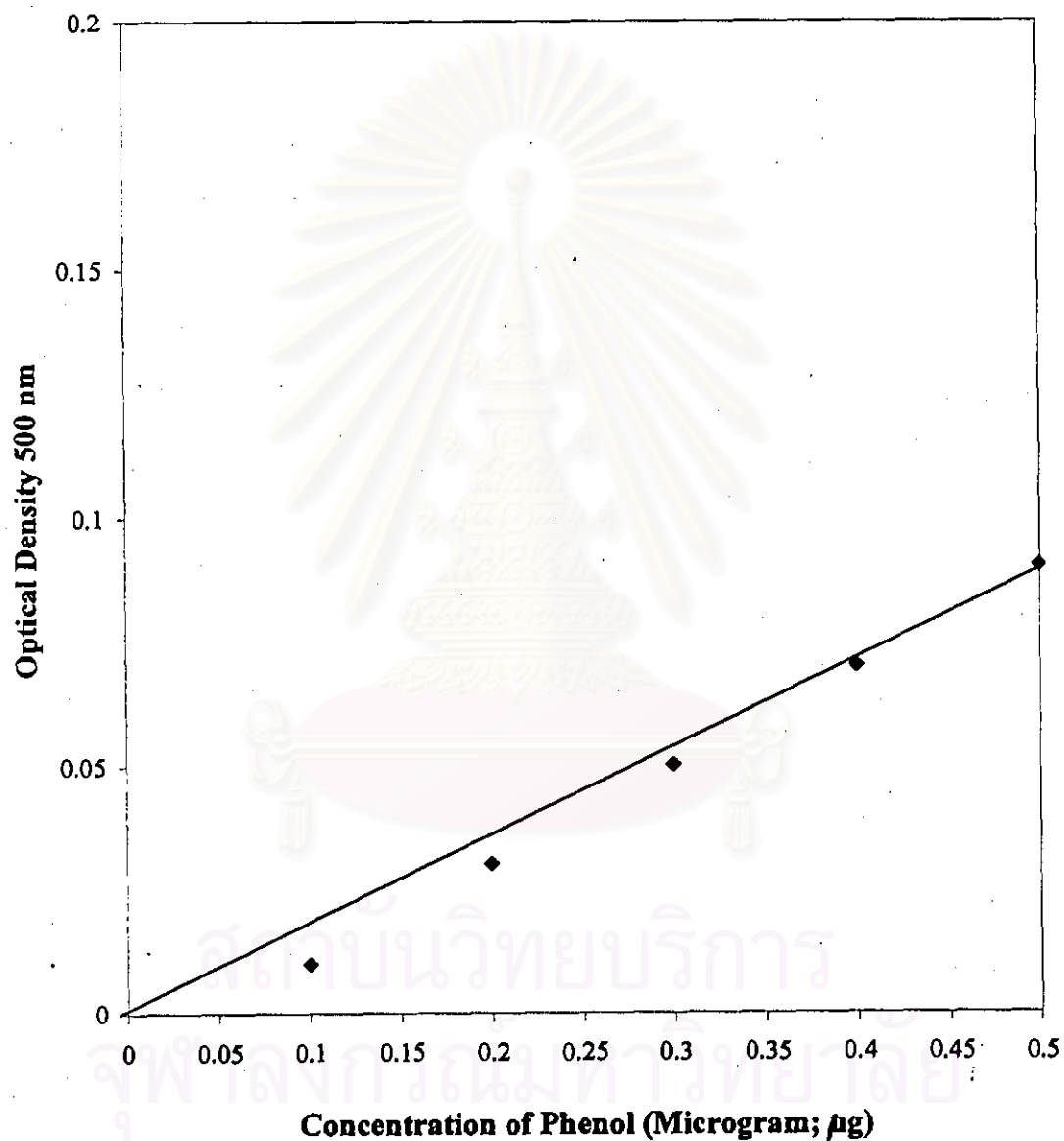


Figure D.2 Standard curve of phenol by direct photometric method (Greenberg A.E., et.al., 1992)

APPENDIX E

**SOME CHARACTERISTICS OF THE
SELECTED BACTERIAL ISOLATES**



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Strains	Identified as	Cr Resistance ($\mu\text{g/ml}$)	Ph Resistance ($\mu\text{g/ml}$)	As Resistance ($\mu\text{g/ml}$)	Zn Resistance ($\mu\text{g/ml}$)	Stability (times) ¹	pH	Temperature (°C)	Efficiency of Degradation (%) ²		
									p- Cresol	p-Chloro- phenol	p-Nitro- phenol
CrR-2	<i>Escherichia sp.</i>	2400	1000	200	100	18	7	37	-	-	-
CrR-14	<i>Pseudomonas sp.</i>	2400	1000	200	100	18	7	37	-	-	-
CrR-15	<i>Enterobacter sp.</i>	2400	1000	200	100	18	7	37	-	-	-
PhR-26	<i>Klebsiella sp.</i>	500	2000	200	100	15	7	37	100	16.0	30.0
PhR-33	<i>Pseudomonas sp.</i>	500	2000	200	100	15	7	37	100	22.0	24.0
PhR-64	<i>Escherichia sp.</i>	500	2000	200	100	15	7	37	100	26.0	26.0
CPR-4	<i>Pseudomonas sp.</i>	1200	1200	200	100	18	7	37	100 ³	22.0	26.0
CPR-16	<i>Proteus sp.</i>	1200	1200	200	100	18	7	37	100 ³	28.0	36.0
CPR-17	<i>Escherichia sp.</i>	1200	1200	200	100	18	7	37	100 ³	34.0	36.0

¹ 20 times of repeated subculturing

² Initial concentration 50 $\mu\text{g/ml}$, on the third week

³ Initial concentration 50 $\mu\text{g/ml}$, on the second week

Strains	Efficiency of Cr(VI) detoxification (%)				Efficiency of Cr(III) production (%)				Efficiency of phenol degradation (%)			
	100*	200	300	400	100	200	300	400	100	200	300	400
CrR-2+PhR-26	86.0	83.0	80.3	85.5	5.7	3.0	5.7	2.3	92.0	94.0	88.3	94.5
CrR-2+PhR-33	86.0	85.0	75.0	86.3	0.3	2.0	0.3	1.8	92.0	94.0	89.0	94.8
CrR-2+PhR-64	85.0	84.5	79.3	86.0	2.3	1.0	2.3	1.3	93.0	94.5	91.3	94.0
CrR-14+PhR-26	80.0	77.5	79.7	79.8	2.7	2.0	2.7	1.5	99.0	95.5	94.7	99.8
CrR-14+PhR-33	75.0	76.5	91.3	79.0	4.0	2.0	4.0	1.0	99.0	98.0	96.7	96.3
CrR-14+PhR-64	78.0	77.0	79.7	79.8	4.3	1.0	4.3	1.5	99.0	97.5	87.7	96.3
CrR-15+PhR-26	77.0	77.0	87.0	76.8	4.3	1.0	4.3	0.8	98.0	94.0	97.0	97.0
CrR-15+PhR-33	76.0	78.5	76.3	77.8	1.0	1.0	1.0	0.5	99.0	96.0	85.3	94.0
CrR-15+PhR-64	76.0	77.0	78.0	77.5	2.0	1.0	2.0	2.3	98.0	94.5	97.3	96.8
CPR-4	85.0	82.5	80.3	83.3	4.3	3.0	4.3	1.0	92.0	98.0	90.3	97.5
CPR-16	49.0	79.5	80.3	88.0	6.0	3.0	6.0	2.5	53.0	87.0	91.3	85.3
CPR-17	84.0	83.0	82.0	84.0	3.7	2.0	3.7	2.3	95.0	98.5	87.3	98.3

* Concentration ($\mu\text{g/ml}$)

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Strains	Efficiency of Cr(VI) detoxification (%)				Efficiency of Cr(III) production (%)				Efficiency of phenol degradation (%)			
	500*	1000	1500	2000	500	1000	1500	2000	500	1000	1500	2000
CrR-2+PhR-26	78.0	92.0	95.3	97.0	20.0	8.0	5.3	3.5	92.2	78.3	58.9	38.9
CrR-14+PhR-33	78.0	90.0	94.0	96.0	20.0	10.0	6.0	4.5	98.8	76.7	57.8	41.9
CrR-15+PhR-64	80.0	91.0	94.7	97.5	18.0	9.0	5.3	5.0	91.2	91.1	53.3	47.8
CPR-4	44.0	86.0	92.0	94.5	24.0	15.0	4.7	3.0	67.6	49.3	44.1	35.5
CPR-16	48.0	82.0	90.0	94.5	4.0	4.0	3.3	2.0	80.0	61.8	47.1	35.3
CPR-17	38.0	81.0	90.7	95.0	8.0	5.0	1.3	1.5	96.0	88.9	44.5	37.8

* Concentration ($\mu\text{g/ml}$)

BIOGRAPHY

Miss Siriphon Thaweephongathikun was born in Bangkok on the 3 October 1974. She entered King Mongkut's Institute of Technology Thonburi in June 1992 and graduated a Bachelor of Science (Microbiology) in March 1996. She furthered her education at the Interdepartment of Environmental Science, Graduate School of Chulalongkorn University, in 1996.



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