



## REFERENCES

- Baird, R.S., Bunge, A.L. and Noble, R.D. 1987. Batch extraction of amines using emulsion liquid membranes. Importance of reaction reversibility. *American Institute of Chemical Engineers Journal* 33 (1) : 43-53.
- Boey, S.C. Garcia Del Carro, M.C. and Pyle, D.L. 1987. Extraction of citric acid by liquid membrane extraction. *Chemical Engineering Research and Design* 65: 218-223.
- Boey, S.C. 1990. *Citric acid extraction by liquid membranes*. Ph.D.'s thesis of Department of Chemical Engineering, Imperial College of Science, Technology and Medicine, University of London.
- Bunge, A.L. and Noble, R.D., 1984. A diffusion model for reversible consumption in emulsion liquid membranes. *Journal of Membrane Science* 22 : 55.
- Cahn, R.P. and Li, N. N., 1974. Separation of phenol from waste water by liquid membrane technique. *Separation Science* 9(6) : 505-519.
- Chan, C.C. and Lee, C.J 1984. Mechanistic models of mass transfer across a liquid membrane. *Journal of Membrane Science* 20 : 1-2.
- Chaudhuri J.B. 1990. *Kinetic studies on the emulsion liquid membrane extraction of lactic acid*. Ph.D.'s thesis, Department of Food Science and Technology, University of Reading.
- Cussler, E.L. 1986. *Diffusion, mass transfer in fluid systems*. Cambridge University Press, New York.
- Danesi, P.R. 1984-1985. Separation of metal species by supported liquid membranes. *Separation Science and Technology* 19 (11-12) : 857-894.
- \_\_\_\_\_, P.R. and Rickert, P.G. 1986. Some observations on the performance of hollow-fibre supported liquid membranes for Co-Ni separation. *Solvent Extraction and Ion Exchanges* 4(1) : 149-164.
- \_\_\_\_\_, P.R., Yinger, L.R., Rickert, P.G. 1987. Lifetime of supported liquid membranes. *Journal of Membrane Science* 31 (2) : 117-146.

- Dadgar, A.M. and Fontch, G. L. 1985. Evaluation of solvents for the recovery of clostridium fermentation products by liquid-liquid extraction. *Bio. & Biotech. Symp.* 15 : 611-620.
- Datta S., Mukhopadhyay, A. and Sanya, S.K.1993. Facilitated transport through a liquid surfactant membrane with continuous phase resistance : Role of drop-size distribution. *Separation Science and Technology* 28 (6) : 1327-1340.
- Deblay P., Minier, M. and Renon, H. 1990. Separation of L-valine from fermentation broths using a supported liquid membrane. *Biotechnology and Bioengineering* 35 : 123-131.
- Draxler, J. and Marr, R. 1986. Emulsion liquid membrane. Part 1 : Phenomenon and industrial application. *Chemical Engineering and Processing* 20 : 319-329.
- Eyal A.M. and Bressler, E. 1993. Mini-review: Industrial separation of carboxylic and amino acids by liquid membranes: Applicability process considerations, and potential advantages. *Biotechnology and Bioengineering* 41: 287.
- Gladek, L., Slemaszek, J. and Szust, J. 1982. Modeling of mass transport with a very fast reaction through liquid membrane. *Journal of Membrane Science* 12:153.
- Haro, T., Ohtake, T., Matsumoto, M., Ogawa, S. and Hori, F. 1990. Extraction of penicillin with liquid surfactant membrane. *Journal of Chemical Engineering of Japan* 23 : 772.
- Ho, W.S., Hatton, T.A., Lightfoot, E.N. and Li, N. N., 1982. Batch extraction with liquid surfactant membranes: a diffusion controlled model *American Institute of Chemical Engineers Journal* 28 (4) : 662-670.
- Itoh, H. Thien, M.P. Hatton, T.A. and Wang, D.I.C. 1990a. A liquid emulsion membrane process for separation of amino acids. *Biotechnology and Bioengineering* 35 : 853.
- \_\_\_\_\_, H. Thien, M.P. Hatton, T.A. and Wang, D.I.C. 1990b. Water transport mechanism in liquid emulsion membrane process for the separation of amino acid. *Journal of Membrane Science* 51 : 309-322.
- Kim. K., Choi, S., and Ihm, S. 1983. Simulation of phenol from waste water by liquid membrane emulsion. *Ind. Eng. Chem. Fundam.* 22 : 167.
- Lee, K.H. Evans D.F. and Cussler E.L. 1978. Selective copper recovery with two types of liquid membranes. *American*

- Institute of Chemical Engineers Journal* 24 (5) : 860-868.
- Li, N. N. 1968. *U.S. Patent* 3 410 : 794.
- Likidis, Z. and Schirgerl, K. 1987. Recovery of penicillin by reactive extraction in centrifugal extractors. *Biotechnology and Bioengineering* 30 : 1032-1040.
- Lorbach, D.M. and Hatton, T.A. 1988. Polydispersity and back mixing effects in diffusion controlled mass transfer with irreversible chemical reaction an analysis of liquid emulsion membrane process. *Chemical Engineering Science* 43 (3) : 405-418.
- \_\_\_\_\_, D. M. and Marr, R. 1987. Emulsion liquid membranes. Part 2. Modeling mass transfer of zinc with bis (2- ethylhexyl) dithiophosphoric acid. *Chemical Engineering and Processing* 21 : 83-93.
- Marr, R. and Kopp, A. 1982. Liquid membrane technology - a survey of phenomena, mechanisms, and models. *International Chemical Engineering* 22 : 44-60.
- Matulevicins E.S. and Li, N. N. 1975. Facilitated transport through liquid membrane. *Separation Purification Methods* 4 : 73.
- Molinari, R., Bartolo, L. D., and Drioli, E. 1992. Coupled transport of amino acid through a supported liquid membrane I. Experimental optimization. *Journal of Membrane Science* 73 : 203-215.
- Reisinger, H. and Marr, R. 1992. Multicomponent liquid membrane permeation of organic acids. *Chem. Eng. Technol* 15 : 363.
- Scholler, C. Chandhuri, J.B. and Pyle, D.L. 1993. Emulsion liquid membrane extraction of lactic acid from aqueous solutions and fermentation broth. *Biotechnology and Bioengineering* 42 : 50-58.
- Sengupta, A., Basu, R. and Sirkar, K.K. 1988. Separation of solutes from aqueous solutions by contained liquid membranes. *American Institute of Chemical Engineers Journal* 34(10):1698-1708.
- Stroevé, P. and Varanasi, P.P. 1984. Extraction with double emulsion in a batch reactor effect of continuous phase resistance. *American Institute of Chemical Engineers Journal* 30 : 1007.
- Sugiura, M. and Yamaguchi, J. 1983. Coupled transport of picrate anion through liquid membrane supported by a microporous

- polymer film. *Journal of colloid and Interface Science* 96 (2): 454-459.
- Takeuchi, H., Takahashi, K. and Goto, W. 1987. Some observations on the stability of supported liquid membranes. *Journal of Membrane Science* 34 : 19-31.
- Teramoto, M., Takihana, H., Shibutani, M., Yuasa T. and Hara, N. 1983. Extraction of phenol and cresol by liquid surfactant membrane. *Sep. Sci. Technol.* 18 : 397.
- \_\_\_\_\_, M. Yamashiro, T., Inoue, A., Yamamoto, A., Matsuyama, H. and Miyake, Y. 1991. Extraction of amino acids by emulsion liquid membranes containing di (2-ethylhexyl) phosphoric acids as a carrier biotechnology; coupled, facilitated transport; diffusion. *Journal of Membrane Science* 58 : 11.
- Thien, M.P. and Hatton, T.A. 1987. *Liquid emulsion membranes and their applications in biochemical processing*. Mimeo, Department of Chemical Engineering, M.I.T.
- \_\_\_\_\_, M.P., Hatton, T.A. and Wang, D.I.C. 1986. Liquid emulsion membranes and their application in biochemical separations. In separation, recovery, and purification in biotechnology. eds' Asenjo, J.A. and Hong, J. *American Chemical Society Symposium Series* 314 : 67-77.
- \_\_\_\_\_, M.P. Hatton, T.A. and Wang, D.I.C. 1988. Separation and concentration of amino acids using liquid emulsion membranes. *Biotechnology and Bioengineering* 32 : 604.
- Yan N., Shi, Y, and Su, Y.F. 1992. A mass transport model for Type I Facilitated Transport in liquid membranes. *Chemical Engineering Science* 47 (17/18): 4365-4371.
- Yoshikawa, M. Kishida, M. Tanigaki, M. and Eguchi, W. 1989. Novel liquid membrane transport system for tryptophan. *Journal of Membrane Science* 47 : 53.

## APPENDIX

### EXPERIMENTAL DATA

Table 1. Experimental Data on ELM Extraction of 0.006 M Phenylalamine (Phe) Solution.

Time (min.)	Concentration of Phe in the Feed Phase (M)		
	pH 2	pH 3	pH 5
1	$5.25 \cdot 10^{-3}$	$4.75 \cdot 10^{-3}$	$4.15 \cdot 10^{-3}$
2	$5.05 \cdot 10^{-3}$	$4.13 \cdot 10^{-3}$	$3.75 \cdot 10^{-3}$
4	$4.40 \cdot 10^{-3}$	$3.47 \cdot 10^{-3}$	$3.25 \cdot 10^{-3}$
8	$4.10 \cdot 10^{-3}$	$0.30 \cdot 10^{-3}$	$2.90 \cdot 10^{-3}$
15	$3.75 \cdot 10^{-3}$	$0.28 \cdot 10^{-3}$	$2.60 \cdot 10^{-3}$
30	$3.25 \cdot 10^{-3}$	$0.23 \cdot 10^{-3}$	$2.15 \cdot 10^{-3}$

Table 2. Experimental Data on ELM Extration of 0.006 M Tryptophan (Trp) Solution.

Time (min.)	Concentration of Trp in the Feed Phase (M)		
	pH 2	pH 3	pH 5
1	$5.00 \cdot 10^{-3}$	$3.80 \cdot 10^{-3}$	$2.75 \cdot 10^{-3}$
2	$4.65 \cdot 10^{-3}$	$3.35 \cdot 10^{-3}$	$2.55 \cdot 10^{-3}$
4	$3.95 \cdot 10^{-3}$	$2.75 \cdot 10^{-3}$	$2.05 \cdot 10^{-3}$
8	$3.40 \cdot 10^{-3}$	$2.30 \cdot 10^{-3}$	$1.80 \cdot 10^{-3}$
15	$2.80 \cdot 10^{-3}$	$2.10 \cdot 10^{-3}$	$1.65 \cdot 10^{-3}$
30	$2.70 \cdot 10^{-3}$	$1.95 \cdot 10^{-3}$	$1.50 \cdot 10^{-3}$

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Table 2. Experimental Data on ELM Extration of 0.006 M Tryptophan (Trp) Solution.

Time (min.)	Concentration of Trp in the Feed Phase (M)		
	pH 2	pH 3	pH 5
1	$5.00 \cdot 10^{-3}$	$3.80 \cdot 10^{-3}$	$2.75 \cdot 10^{-3}$
2	$4.65 \cdot 10^{-3}$	$3.35 \cdot 10^{-3}$	$2.55 \cdot 10^{-3}$
4	$3.95 \cdot 10^{-3}$	$2.75 \cdot 10^{-3}$	$2.05 \cdot 10^{-3}$
8	$3.40 \cdot 10^{-3}$	$2.30 \cdot 10^{-3}$	$1.80 \cdot 10^{-3}$
15	$2.80 \cdot 10^{-3}$	$2.10 \cdot 10^{-3}$	$1.65 \cdot 10^{-3}$
30	$2.70 \cdot 10^{-3}$	$1.95 \cdot 10^{-3}$	$1.50 \cdot 10^{-3}$

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Table 3. Experimental Data on ELM Extraction of 0.001 M Tryptophan (Trp) Solution.

Time (min.)	Concentration of Trp in the Feed Phase (M)		
	pH 2	pH 3	pH 5
1	$9.00 \cdot 10^{-4}$	$6.40 \cdot 10^{-4}$	$6.00 \cdot 10^{-4}$
2	$8.30 \cdot 10^{-4}$	$5.90 \cdot 10^{-4}$	$4.80 \cdot 10^{-4}$
4	$7.30 \cdot 10^{-4}$	$5.00 \cdot 10^{-4}$	$3.90 \cdot 10^{-4}$
8	$6.80 \cdot 10^{-4}$	$4.20 \cdot 10^{-4}$	$3.60 \cdot 10^{-4}$
15	$5.80 \cdot 10^{-4}$	$3.60 \cdot 10^{-4}$	$3.00 \cdot 10^{-4}$
30	$5.20 \cdot 10^{-4}$	$2.60 \cdot 10^{-4}$	$2.60 \cdot 10^{-4}$

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Table 4. Experimental Data on ELM Extraction of 0.006 M Phe + 0.006 M Trp Solution.

Time (min.)	Conc. of amino acid in the Feed Phase (M)					
	pH 2		pH 3		pH 5	
	Phe	Trp	Phe	Trp	Phe	Trp
1	5.60*10 <sup>-3</sup>	5.60*10 <sup>-3</sup>	3.68*10 <sup>-3</sup>	3.48*10 <sup>-3</sup>	3.70*10 <sup>-3</sup>	3.80*10 <sup>-3</sup>
2	5.40*10 <sup>-3</sup>	5.40*10 <sup>-3</sup>	3.48*10 <sup>-3</sup>	3.20*10 <sup>-3</sup>	3.33*10 <sup>-3</sup>	3.33*10 <sup>-3</sup>
4	4.90*10 <sup>-3</sup>	4.60*10 <sup>-3</sup>	3.28*10 <sup>-3</sup>	3.07*10 <sup>-3</sup>	3.16*10 <sup>-3</sup>	2.80*10 <sup>-3</sup>
8	4.40*10 <sup>-3</sup>	3.90*10 <sup>-3</sup>	2.85*10 <sup>-3</sup>	2.52*10 <sup>-3</sup>	2.59*10 <sup>-3</sup>	2.47*10 <sup>-3</sup>
15	4.20*10 <sup>-3</sup>	3.75*10 <sup>-3</sup>	2.70*10 <sup>-3</sup>	2.46*10 <sup>-3</sup>	2.28*10 <sup>-3</sup>	2.17*10 <sup>-3</sup>
30	3.90*10 <sup>-3</sup>	3.40*10 <sup>-3</sup>	2.64*10 <sup>-3</sup>	2.22*10 <sup>-3</sup>	2.04*10 <sup>-3</sup>	1.99*10 <sup>-3</sup>

Table 5. Experimental Data on ELM Extraction of 0.006 M Phe + 0.001 M Trp Solution.

Time (min.)	Conc. of amino acid in the Feed Phase (M)					
	pH 2		pH 3		pH 5	
	Phe	Trp	Phe	Trp	Phe	Trp
1	$5.25 \cdot 10^{-3}$	$8.75 \cdot 10^{-4}$	$4.95 \cdot 10^{-3}$	$8.00 \cdot 10^{-4}$	$3.90 \cdot 10^{-3}$	$7.00 \cdot 10^{-4}$
2	$4.95 \cdot 10^{-3}$	$8.13 \cdot 10^{-4}$	$4.05 \cdot 10^{-3}$	$6.37 \cdot 10^{-4}$	$3.60 \cdot 10^{-3}$	$6.00 \cdot 10^{-4}$
4	$4.44 \cdot 10^{-3}$	$7.35 \cdot 10^{-4}$	$3.98 \cdot 10^{-3}$	$6.13 \cdot 10^{-4}$	$3.00 \cdot 10^{-3}$	$5.60 \cdot 10^{-4}$
8	$3.96 \cdot 10^{-3}$	$6.35 \cdot 10^{-4}$	$3.30 \cdot 10^{-3}$	$4.75 \cdot 10^{-4}$	$2.70 \cdot 10^{-3}$	$4.50 \cdot 10^{-4}$
15	$3.75 \cdot 10^{-3}$	$5.50 \cdot 10^{-4}$	$3.00 \cdot 10^{-3}$	$4.00 \cdot 10^{-4}$	$2.20 \cdot 10^{-3}$	$3.70 \cdot 10^{-4}$
30	$3.39 \cdot 10^{-3}$	$5.50 \cdot 10^{-4}$	$2.55 \cdot 10^{-3}$	$3.75 \cdot 10^{-4}$	$1.90 \cdot 10^{-3}$	$3.00 \cdot 10^{-4}$

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