

คาบิเร่เมร์ สารกลุ่มทรีโซอกซ่าโซลมาโคโรไลด์ที่ยึดเหนี่ยวเอกตินจากฟองน้ำทะเล  
*Pachastrissa nux* ของไทย และอนุพันธุ์กึ่งสัมเคราะห์ชนิดเดิมของแสงของคาบิเร่เมร์ คือ

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KABIRAMIDES, THE ACTIN-BINDING TRISOXAZOLE MACROLIDES FROM  
THE THAI MARINE SPONGE *PACHASTR/SSA NUX*, AND SEMISYNTHETIC FLUORESCENT  
DERIVATIVES OF KABIRAMIDE C

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ชุติมา เพ็ชรประยูร : คบิเรไมด์ สารกลุ่มทริโซอกซ่าไซโลมาโครไรล์ที่ยึดเหนี่ยวแกกตินจากฟองน้ำทะเล *Pachastrissa nux* ของไทย และอนุพันธุ์ของสังเคราะห์ชนิดเรืองแสงของคบิเรไมด์ ซี. (KABIRAMIDES, THE ACTIN-BINDING TRISOXAZOLE MACROLIDES FROM THE THAI MARINE SPONGE *PACHASTRISSA NUX*, AND SEMISYNTHETIC FLUORESCENT DERIVATIVES OF KABIRAMIDE C)

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จากการศึกษาทางเคมีของฟองน้ำทะเล *Pachastrissa nux* ที่พบในบริเวณเกาะสีชัง อ่าวไทย สามารถแยกสารกลุ่มทริโซอกซ่าไซโลมาโครไรล์ที่เคยพบแล้วได้ 2 ชนิด คือ kabiramide C และ D และสารใหม่ 1 ชนิด คือ kabiramide F การพิสูจน์โครงสร้างทางเคมีของสารที่แยกได้เหล่านี้ ทำได้โดยการวิเคราะห์ข้อมูลทางスペกตรสโคปีจาก MS UV IR และ NMR ร่วมกับการเบรียบเทียบข้อมูลกับเอกสารต่างๆ จากนั้นได้นำสาร kabiramide C (KabC) ซึ่งเป็นสารที่แยกได้ในปริมาณสูงมาทำการเปลี่ยนแปลงโครงสร้างเป็น 7-(4-aminomethyl-1*H*-1,2,3-triazol-1-yl)kabiramide C (AMT-KabC) โดยใช้ปฏิกิริยา Mitsunobu และ 1,3 dipolar cycloaddition เพื่อใช้เป็นสารมัธยันต์ในการสังเคราะห์อนุพันธุ์เรืองแสงของ kabiramide C จำนวน 5 ชนิด คือ อนุพันธุ์ tetramethylrhodamine (TMR-KabC), rhodol green (RG-KabC), IC5 (IC5-KabC), dapoxyl (DAP-KabC), and fluorescein diester (FDE-KabC) จากปฏิกิริยาของสาร AMT-KabC กับอนุพันธุ์ *N*-succinimidyl ester ของสารเรืองแสง

ได้ทำการศึกษาคุณสมบัติของ kabiramide ที่แยกได้ และอนุพันธุ์เรืองแสงที่สังเคราะห์ ในการยึดเหนี่ยวกับแกกติน โดยใช้เทคนิคต่างๆ ทางฟลูออเรสเซนต์ ได้แก่ การวิเคราะห์สเปกตัมการเรืองแสง การวัด fluorescence resonance energy transfer (FRET) การวัด fluorescence anisotropy (FA) และ การวัด iodide quenching พบว่า อนุพันธุ์ kabiramide ทุกชนิด ยึดเหนี่ยวกับแกกตินเป็นสารประกอบเชิงช้อนในอัตราส่วน 1:1 ได้ทดสอบอนุพันธุ์เรืองแสงที่สังเคราะห์กับเซลล์ NIH 3T3 พบว่า เฉพาะอนุพันธุ์ TMR-KabC และ FDE-KabC เท่านั้นที่สามารถผ่านเข้าสู่เซลล์ และแสดงการเรืองแสงที่ชัดเจนในบริเวณขอบเซลล์ที่มีการเคลื่อนที่ นอกจากนี้ สาร kabiramide C, D และ F แสดงฤทธิ์ต้านเชื้อ *Candida albicans* โดยให้บริเวณยับยั้งการเจริญของเชื้อขนาดเดินผ่านศูนย์กลาง 22 มิลลิเมตร ที่ความเข้มข้น 100 ไมโครกรัมต่อเดซิลิตร

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KEY WORDS: KABIRAMIDE / TRISOXAZOLE MACROLIDE / ACTIN-BINDING / SEMISYNTHETIC FLUORESCENCE / PACHASTRISSA NUX SPONGE

CHUTIMA PETCHPRAYOON: KABIRAMIDES, THE ACTIN-BINDING TRISOXAZOLE MACROLIDES FROM THE THAI MARINE SPONGE *Pachastrissa nux*, AND SEMISYNTHETIC FLUORESCENT DERIVATIVES OF KABIRAMIDE C. THESIS ADVISOR: KHANIT SUWANBORIRUX, Ph.D., 188 pp. ISBN 974-53-2125-7.

Three trisoxazole macrolides, including two known kabiramides C and D, and a new kabiramide F were isolated from the marine sponge *Pachastrissa nux*, collected from Sichang Island in the Gulf of Thailand. Their identification and structure elucidation were achieved by analyses of MS, UV, IR, 1D-NMR, and 2D-NMR spectral data as well as comparison with the literatures. The major compound, kabiramide C (KabC), was structurally modified to give the key intermediate 7-(4-aminomethyl-1*H*-1,2,3-triazol-1-yl)kabiramide C (AMT-KabC) by using Mitsunobu reaction and 1,3-dipolar cycloaddition reaction. Furthermore, five fluorescent conjugates of kabiramide C, including derivatives of tetramethylrhodamine (TMR-KabC), rhodol green (RG-KabC), IC5 (IC5-KabC), dapoxyl (DAP-KabC), and fluorescein diester (FDE-KabC) were synthesized by coupling AMT-KabC with *N*-succinimidyl esters of the fluorescence dyes.

The steady state fluorescence techniques, including fluorescence emission spectral analysis, fluorescence resonance energy transfer (FRET), fluorescence anisotropy (FA), and iodide quenching were used to determine the actin binding properties of these kabiramide derivatives. All of the isolated and semisynthetic fluorescent kabiramide derivatives bound stoichiometrically to G-actin in 1:1 complex. Five fluorescent kabiramides were further separately introduced to living NIH 3T3 cells. Only TMR-KabC and FDE-KabC showed good permeability through plasma membrane of the cells and expressed intense fluorescence at the protrusion sites. Additionally, kabiramides C, D, and F exhibited equal antifungal activity against *Candida albicans* with inhibition zone of 22 mm at the concentration of 100 µg/disc.

Field of Study Pharmaceutical Chemistry Student's signature.....

and Natural Products Advisor's signature .....

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## ABBREVIATIONS

%	=	percent or part per hundred
$\delta$	=	chemical shift
$\tau$	=	excited-state lifetime
$\epsilon$	=	molar absorptivity or molar extinction coefficient
$\lambda_{\max}$	=	wave length at maximum absorption
$\nu_{\max}$	=	wave number at maximum absorption
$[\alpha]^{23}_D$	=	specific rotation at 23 °C and sodium D line (589 nm)
°C	=	degree Celsius
$\mu\text{g}$	=	microgram
$\mu\text{l}$	=	micro liter
$\mu\text{M}$	=	micro molar
$\mu\text{m}$	=	micrometer
$^{13}\text{C}$ NMR	=	carbon-13 nuclear magnetic resonance
$^1\text{H}$ NMR	=	proton nuclear magnetic resonance
2D NMR	=	two dimensional nuclear magnetic resonance
Å	=	angstrom
ADP	=	adenosine diphosphate
$\text{Al}_2\text{O}_3$	=	aluminum oxide
Ala	=	alanine
ATCC	=	American Type Culture Collection (Maryland, USA)
ATP	=	adenosine triphosphate
br s	=	broad singlet
c	=	concentration
Ca	=	calcium
$\text{CaH}_2$	=	calcium hydride
cald	=	calculated
$\text{CDCl}_3$	=	deuterated chloroform
$\text{CH}_2\text{Cl}_2$	=	dichloromethane

CHCl <sub>3</sub>	=	chloroform
cm	=	centimeter
CO <sub>2</sub>	=	carbon dioxide
Cys	=	cysteine
d	=	doublet
dd	=	doublet of doublets
ddd	=	doublet of doublets of doublets
DEPT	=	distortionless enhancement by polarization transfer
DMEM	=	Dulbecco's Modified Eagle medium
DMSO	=	dimethyl sulphoxide
DMSO- <i>d</i> <sub>6</sub>	=	deuterated dimethyl sulphoxide
dt	=	doublet of triplets
DTT	=	dithiothreitol
EI	=	electron impact ionization
em	=	emission
ESI-TOF	=	electrospray ionization-time of flight
Et <sub>3</sub> N	=	triethylamine
EtOAc	=	ethyl acetate
EtOH	=	ethanol
ex	=	excitation
FA or r	=	fluorescence anisotropy
FBS	=	fetal bovine serum
Fmoc	=	<i>N</i> -(9 <i>H</i> -fluoren-9-yl-methoxycarbonyl)
FRET	=	fluorescence resonance energy transfer
g	=	gram or earth's gravitational field
Gly	=	glycine
H,H COSY	=	homonuclear (proton-proton) correlation spectroscopy
H <sub>2</sub> O	=	water
HeLa	=	human cervix carcinoma cells
HMBC	=	proton-detected heteronuclear multiple bond correlation
HMQC	=	proton-detected heteronuclear multiple quantum coherence

$\text{HN}_3$	=	hydrazoic acid
Hz	=	hertz
I or FI	=	fluorescence intensity
$\text{IC}_{50}$	=	50% inhibition concentration
Ile	=	isoleucine
IR	=	infrared
$J$	=	coupling constant
$\text{K}_2\text{HPO}_4$	=	potassium phosphate dibasic
KCl	=	potassium chloride
$\text{KH}_2\text{PO}_4$	=	potassium phosphate monobasic
KI	=	potassium iodide
km	=	kilometer
L	=	liter
Leu	=	leucine
M	=	molar
m	=	multiplet
$m/z$	=	mass to charge ratio
$\text{M}^+$	=	molecular ion
MeOH	=	methanol
Met	=	methionine
mg	=	milligram
$\text{MgCl}_2$	=	magnesium chloride
$\text{MgSO}_4$	=	magnesium sulphate
MHz	=	megahertz
MIC	=	minimum inhibitory concentration
min	=	minute
mL	=	milliliter
mm	=	millimeter
MS	=	mass spectroscopy
NaCl	=	sodium chloride
$\text{NaHCO}_3$	=	sodium bicarbonate

NaN <sub>3</sub>	=	sodium azide
ng	=	nanogram
NIH 3T3	=	mouse fibroblast cells
nM	=	nano molar
nm	=	nanometer
NMR	=	nuclear magnetic resonance
NSS	=	normal saline solution
Phe	=	phenylalanine
PPh <sub>3</sub>	=	triphenylphosphine
ppm	=	part per million
R <sup>2</sup>	=	correlation coefficient
RP18	=	reversed phase carbon eighteen
rt	=	room temperature
s	=	singlet
SDA	=	Sabouraud dextrose agar
Ser	=	serine
Si gel	=	silica gel
sp.	=	species
t	=	triplet
THF	=	tetrahydrofuran
Thr	=	threonine
TLC	=	thin layer chromatography
Tyr	=	tyrosine
UV	=	ultraviolet
w or wt	=	weight