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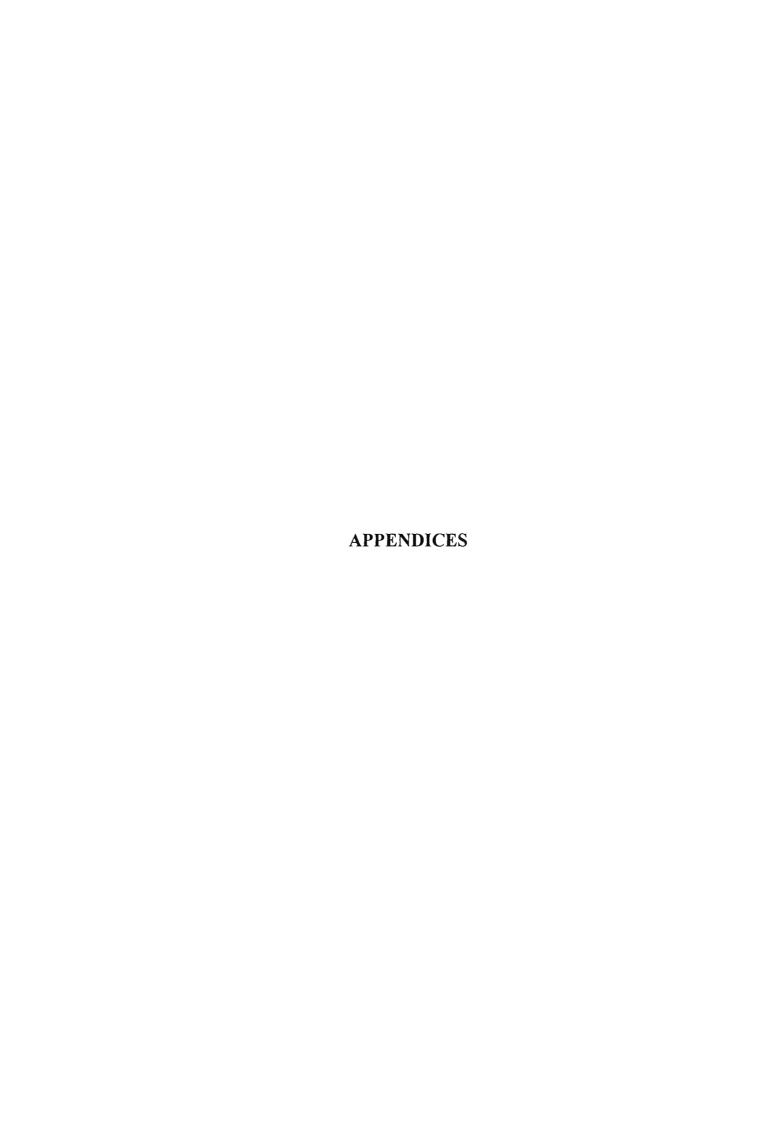
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#### APPENDIX A

# PURIFICATION OF THERMOSTABLE CHITINASE PRODUCED BY Bacillus licheniformis SK-1

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#### **ABSTRACT**

Chitinase was partially purified from the culture medium of *Bacillus licheniformis* SK-1 by colloidal chitin affinity adsorption followed by DEAE-cellulose column chromatography. The partial purified enzyme showed a single protein band on native polyacrylamide gel electrophoresis. The isoelectric point of the major component in the partial purified chitinase was 4.62. The partial purified chitinase showed a major band with MW 72 kDa and 2 minor bands with MW 58, 70 kDa on SDS-PAGE, respectively. In addition, the partial purified chitinase showed a glycoprotein nature on native polyacrylamide gel when stained with Schiff's reagent. The partial purified chitinase revealed two activity optima at pH is 6 and 8 when colloidal chitin was used as substrate. The enzyme exhibited a broad activity temperatures ranging between 40 to 70 °C, with optimum at 55 °C. The K<sub>m</sub> and V<sub>max</sub> of the partial chitinase was 0.23 mg colloidal chitin ml<sup>-1</sup> and 0.45 U ml<sup>-1</sup>.

#### INTRODUCTION

Chitinases, a group of enzymes capable of degrading chitin to low-molecularweight products, have been shown to be produced by a number of microorganisms. The production of inexpensive chitinolytic enzymes is an important element in the utilization of shellfish wastes that not only solves environmental problems but also promotes the economic value of the marine products [1]. Thus, chitinolytic enzymes have been purified from many microorganism, and their enzymatic properties have been investigated. Bacillus spp. is regarded as a group of bacteria particularly efficient in the breakdown of chitin. [2]. Previously, thermostable endochitinases, which are useful for the preparation of chitobiose [3], have been isolated from B. licheniformis X-7u [4], and Streptomyces thermoviolaceus OPC-520 [5], while thermostable exochitinase from B. staerothermophilus CH-4 [6] was characterized. B. licheniformis SK-1 produces thermostable chitinase when cultured in a medium containing colloidal chitin as a carbon source. We were able to use crude chitinase from this microorganism to prepare GlcNAc from crystalline chitin [7]. This paper describes the purification and some properties of chitinases from the culture medium of B. licheniformis SK-1.

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#### EXPERIMENTAL PROCEDURE

Microorganism and culture. SK-1 was isolated from soil in Angthong Province, Thailand. SK-1 was cultured at 50 °C on 0.02% colloidal chitin minimum medium (CCMM) agar plate, 0.05% yeast extract, 0.1% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, 0.03% MgSO<sub>4</sub>, 0.6% KH<sub>2</sub>SO<sub>4</sub> and 1.0% K<sub>2</sub>HPO<sub>4</sub>, pH 7.5 and 2% agar. For liquid culture, the medium containing 0.1% CCMM. The microorganism was incubated with shaking at 50 °C in CCMM for 4-5 days.

Purification of chitinase. SK-1 was cultured with shaking at 50 °C in 0.1% CCMM (2L) for 4-5 days. The cells were removed by centrifugation (8,000 X g; 20 min) to obtain culture fluid. The culture fluid was stirred gently with fresh colloidal chitin (10 mg/mg of protein) overnight at 0 °C for affinity adsorption [8]. The colloidal chitin was then washed three times with 10 mM potassium phosphate buffer (KPB, pH 6.0) and collected by centrifugation. The precipitated colloidal chitin was resuspended in 20 ml of KPB and incubated at 50 °C overnight to digest the colloidal chitin. The digested solution was dialyzed against 25 mM Tris-HCl buffer, pH 7.5. The dialyzed enzyme was applied to a DEAE-cellulose column, previously equilibrated with 25 mM Tris-HCl buffer, pH 7.5, and eluted with 0-1.0 gradient of NaCl.

Enzyme assay. Chitinase activity was assayed in 1.5 ml of a reaction mixture containing 0.1% colloidal chitin in 0.1 M phosphate buffer pH 6.0 and 0.1 ml of enzyme solution. After incubation at 50 °C for 10 min, the reaction was stopped by boiling, then centrifuged. Reducing sugar produced in the supernatant was measured by the modified Schales method [9]. One unit of chitinase activity was defined as the amount of enzyme that liberate reducing sugar corresponding to one μmole of *N*-acetyl-D-glucosamine per minute.

*Protein measurement*. Protein measurement was performed by the method of Bradford et al.[10] using bovine serum albumin as the standard. For chromatographic profile, the protein concentration was estimated by measuring the absorbance at 280 nm.

Determination of the isoelectric point (pl) of the enzyme. Isoelectric focusing was performed according to Robertson et al. [11] with ampholine carrier ampholytes pH 3-10 at final concentration of 1%. After electrofocusing, the gel was stained with Coomassie brilliant blue R-250.

Glycoprotein in native protein and activity stain on PAGE. The enzyme preparations were analyzed by nondenaturing PAGE using 7.5% gel, according to Laemmli [12]. Three sets of gel with identical samples were electrophoresis. After electrophoresis, one gel was stained for protein with Coomassie blue R-250, the second gel was stained for activity, using glycol chitin as substrate [14]. The glycoprotein nature of the enzymes was detected in the third gel (7-15% gradient) using Schiff's reagent as a staining agent according to the methods described by Kaschnitz et al.[13].

Determination of molecular weight and activity staining. The molecular weights of the enzyme was determined by 10% SDS-PAGE. In electrophoresis, standard proteins were used. The activity of chitinase in SDS-PAGE was detected according to the method of Trudel and Asselin [14].

Optimal pH of chitinase. The chitinase activity was measured at various pHs by a colorimetric method, using colloidal chitin as a substrate. The enzyme was preincubated in 0.1 M citrate buffer at pH 3-6, 0.1 M phosphate buffer at pH 6-8 and 0.1 M Tris-HCl buffer at pH 8-10, at 50 °C for 30 min. After which, the enzyme activity was assay.

Optimal temperature of chitinase. The enzyme was preincubated in 0.1 M phosphate buffer pH 6.0 at different temperatures from 40 to 70 °C for 30 min and assayed for chitinase activity.

#### RESULTS AND DISCUSSION

The partial purified enzyme gave a single protein band on native polyacrylamide gel electrophoresis and isoelectric focusing gel (Fig.1). The isoelectric point was 4.62 by isoelectric focusing. The partial purified enzyme gave a three bands on SDS-PAGE (Fig.2). The molecular weight was 58, 70 and 72 kDa, as judged by SDS-PAGE.

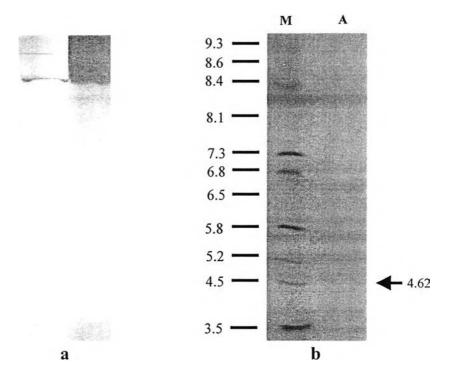


Figure. 1 Native gel of the partial purified chitinase from *B. licheniformis* SK-1(a); lane A: partial purified chitinase, lane B: chitinase activity). IEF pattern of chitinase (b); lane M: IEF standard marker, lane A: partial purified chitinase.



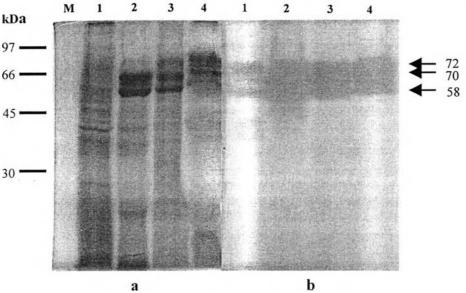


Figure 2. SDS-PAGE of the partial purified chitinases, protein stain (a) and activity stain (b). Lane M: standard marker, lane 1: crude chitinases, lane 2: affinity adsorption, lane 3: bound column peak 1, lane 4: bound column peak 2.

It has been shown earlier that the thermophilic strain of *B. licheniformis* X-7u produced four forms of chitinase [4]. Two of the chitinases had the molecular masses of 89 and 76 kDa with different N-terminal sequence, whereas the other two have formed

presumably owing to limited proteolysis of the chitinase 76 C-terminal region Hence, it appears that strain used in our work produced a different array of chitinase.

Multiplicity of chitinase produced by bacterial cells might be considered as a rather common phenomenon. Serratia marcescens [16, 17, 18, 19], S. liquefaciens [20], Streptomyces Olivaceoviridis [21, 22], Streptomyces pliccatus [23], B. circulans [24, 25, 26], and B. licheniformis X 7-u [4] have been shown to produce several forms of chitinase. These forms reflect the multiplicity of structural genes coding for chitinases or a result of post-translational modifications, among which limited proteolysis plays a prominent role.

Hence, B. licheniformis appears to produce an array of chitinase that might be useful for adaptation of these bacteria to utilize the sources of chitin encountered in their habitat [15]

Polyacrylamide gel electrophoresis of the purified enzyme sample stained positive with Schiff's reagent, suggesting the glycoprotein nature of the enzymes (Fig. 3).

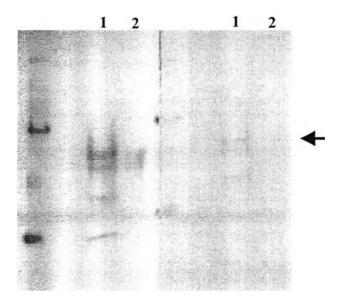


Figure 3. Glycoprotein staining. Lane M: positive control (transferrin, BSA), lane1 : crude chitinases, lane 2. partial chitinases.

The enzyme showed optimal activity at 2 pHs: 6 and 8 (Fig.4). *B. licheniformis* was also shown to possess two pH optima of activity against glycol and colloidal chitin [4, 15], i.e., in both acidic and alkaline solutions, whereas only one optimum activity was observed at pH 5 towards low molecular mass substrates. This unusual property, peculiar to some bacterial chitinases, seems to reflect a difference in pH influence on the activity of an enzyme and the structure of the high molecular mass substrate [15].

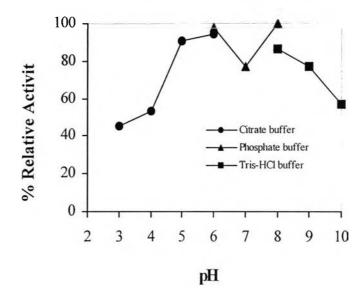


Figure 4. Effect of pH of B. licheniformis SK-1 chitinase activity.

The chitinase had a high optimum temperature at 55 °C (Fig. 5) compared with the enzymes from other mesophile bacteria [27] and fungi [28]; with optimum temperature around 45 °C.

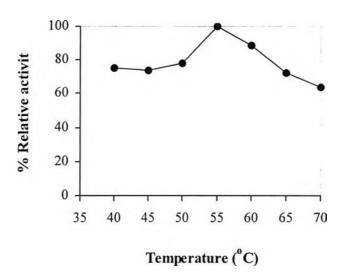
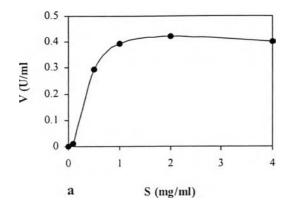


Figure 5. Effect of temperature on B. licheniformis SK-1 chitinase activity.

When the activity of the partial purified chitinase was followed at various substrate concentrations using colloidal chitin as the substrate, the profile followed Michaelis-Menten kinetics. The  $K_m$  value calculated from Lineweaver-Burk plots (Fig. 6) was 0.23 mg/ml for colloidal chitin and the  $V_{max}$  was 0.45 U ml<sup>-1</sup>.



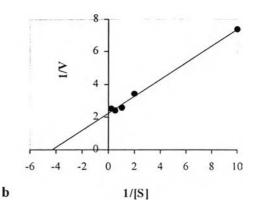


Figure 6. Dependence of the reaction rate of chitinase on the substrate concentration.

- a) Saturation curve
- b) Lineweaver-Burk plot

#### **CONCLUSIONS**

Chitinase was partially purified from B. licheniformis SK-1 with molecular weight of 58, 70 and 72 kDa on SDS-PAGE. It stained positive for glycoprotein and its isoelectric was 4.62. The optimum pH and temperature of the chitinase from B. licheniformis SK-1 was at pH 6 and 8, and 55 °C. The  $K_m$  of the chitinase was 0.23 mg colloidal chitin ml<sup>-1</sup> and the  $V_{max}$  was 0.45 U ml<sup>-1</sup>.

#### **ACKNOWLEDGMENT**

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# Quantitative production of 2-Acetamido-2-deoxy-D-glucose from crystalline chitin by bacterial chitinase

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#### **Abstract**

Fine powdered  $\alpha$ — and  $\beta$ -chitin can be completely hydrolyzed with chitinase (EC 3.2.1.14) and  $\beta$ -N-acetylhexosaminidase (EC 3.2.1.52) for the production of 2-acetamido-2-deoxy-D-glucose (GlcNAc). Crude chitinase from *Burkholderia cepacia* TU09 and *Bacillus licheniformis* SK-1 were used to digest  $\alpha$ — and  $\beta$ —chitin powder. Chitinase from *Burkholderia cepacia* TU09 produced GlcNAc over 85% yield from  $\beta$ -chitin and  $\alpha$ -chitin within 1 day and 7 days, respectively. *Bacillus licheniformis* SK-1 chitinase completely hydrolyzed  $\beta$ -chitin within 6 days, giving the final GlcNAc yield of 75% along with 20% of chitobiose. However, only a 41% yield of GlcNAc was achieved from digesting  $\alpha$ -chitin with *Bacillus licheniformis* SK-1 chitinase.

Key words: Chitin, Chitinase, N-Acetyl-D-glucosamine, 2-Acetamido-2-deoxy-D-glucose, Bacillus licheniformis, Burkholderia cepacia

2-Acetamido-2-deoxy-D-glucose (*N*-Acetyl-D-glucosamine, GlcNAc) and 2-amino-2-deoxy-D-glucose (D-glucosamine, GlcN) have recently been promoted for treatment or as nutriceutical agents for patients with osteoarthritis and inflammatory bowel disease. In contrast to GlcN hydrochloride or sulfate, both of which have a bitter taste, GlcNAc has sweet taste which can be conveniently used in daily consumption. However, GlcNAc has not been widely commercialized mainly due to the lack of an economical process for production of this compound that is acceptable for food and medicine. The current acid hydrolysis of chitin using concentrated HCl is inefficient, and poses environmental and technical concerns. On the other hand, hydrolysis of chitin with enzymes can produce GlcNAc under mild and environmentally friendly conditions. An approach whereby commercially available crude enzymes were used to hydrolyze amorphous chitin substrate was carried out.

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Unfortunately, this method added an additional substrate preparation step into the production of GlcNAc. The work on commercially available crude enzymes was also extended to a production of GlcNAc by direct hydrolysis of  $\beta$ -chitin powder.<sup>5, 6</sup> These reports have shown that enzymatic hydrolysis of chitin can produce GlcNAc in relatively higher yields than the acid hydrolysis. Nevertheless, the remaining major impediment of an enzymatic hydrolysis process is the extremely low hydrolytic susceptibility of the natural chitin substrate, due to its high crystallinity. We would thus like to show herein for the first time that crystalline chitin in both  $\alpha$ - and  $\beta$ -forms could be cleanly hydrolyzed, producing GlcNAc in virtually quantitative yield.

Powdered α-chitin (14 μm in size) from crab shells and β-chitin (3 μm in size) from squid pens were used as substrates for digestion by crude bacterial chitinase from *Burkholderia cepacia* TU09 and *Bacillus licheniformis* SK-1. A typical reaction contains100 mU/mL (1 unit = the amount of enzyme that produces 1 μmole of GlcNAc per minute from colloidal chitin) of the enzyme and 10-40 mg/mL of the substrate, unless indicated otherwise. Digestion reactions were carried out in 3-5 mL of 0.1 M citrate-phosphate buffer, pH 6.0, in 10 mL glass vials. The reactions were incubated in a shaking water bath, with moderate shaking, at 37°C and 50°C when the enzyme from *Burkholderia cepacia* TU09 and *Bacillus licheniformis* SK-1 was used, respectively. At each time point, a portion of the reaction mixture was withdrew, diluted with H<sub>2</sub>O then mixed with CH<sub>3</sub>CN (at the ratio 31/69), filtered, and analyzed by HPLC (column: Shodex Asahipak NH2P-50; flow rate: 1 mL/min; mobile phase: 31/69 H<sub>2</sub>O-CH<sub>3</sub>CN; detection: UV at 210 nm). The amount of GlcNAc in the reaction mixture was determined from a calibration curve of GlcNAc standard.

The percent yield of GlcNAc production increased with the reduction of substrate/enzyme ratio. Although, at the substrate/enzyme ratio of 100 mg/U, chitinase from *Bacillus licheniformis* SK-1 completely hydrolyzed  $\beta$ -chitin, it gave a mixture of GlcNAc and N,N'-diacetylchitobiose [(GlcNAc)<sub>2</sub>] (**Table 1**). The gradual increase of the GlcNAc/(GlcNAc)<sub>2</sub> product ratio with incubation time implied the presence of low  $\beta$ -N-acetylhexosaminidase (EC 3.2.1.52) activity in the crude enzyme from *Bacillus licheniformis* SK-1 under the reaction conditions. On the other hand, hydrolysis of  $\beta$ -chitin with chitinase from *Burkholderia cepacia* TU09 gave mostly GlcNAc with a trace amount of chitotriose. At the substrate/enzyme ratio of 100 mg/U, a 90% yield of GlcNAc was obtained within one day, and a quantitative yield was realized upon prolonged incubation (**Table 2**).

The tightly packed chitin strands of  $\alpha$ -chitin are known to have low susceptibility to enzymatic hydrolysis. We found that when chitinase from *Bacillus licheniformis* SK-1 was used, it was unable to completely hydrolyze  $\alpha$ -chitin. Only 41% of  $\alpha$ -chitin was hydrolyzed in 6 days, even when the concentration of enzyme used in the reaction was 10-fold of the amount that was used to completely hydrolyze  $\beta$ -chitin (Table 3). We speculate that the crystalline domains in  $\alpha$ -chitin were completely resistant to digestion by chitinase from *Bacillus licheniformis* SK-1. The GlcNAc produced was probably liberated from amorphous regions of the substrate. Chitinase from *Burkholderia cepacia* TU09 showed superior characteristic in hydrolyzing  $\alpha$ -chitin as 85% yield of GlcNAc was achieved after 7 days of incubation (Table 4). It is worth noting that the hydrolysis of  $\alpha$ -chitin with chitinase *from Burkholderia cepacia* TU09 consists of two steps. First, a rapid hydrolysis step in the first 24 hours, where we believe that the amorphous portion ( $\sim$ 40%) of the chitin

particle is hydrolyzed. The second step is a slower step, where the remaining tightly packed chitin is slowly hydrolyzed. Because of this slower degradation rate, 300 mU/mL of enzyme was used to ensure sufficient amount of active enzyme present throughout the hydrolysis. The isolation and characterization of the enzymes used here will be published elsewhere.

Table 1. Production of GlcNAc from  $\beta$ -chitin by chitinase from *Bacillus licheniformis* SK-1.

β-chitin/enzyme <sup>a</sup> (mg/U)	Digestion time (day)	% Yield <sup>b</sup>		
	• • •	GlcNAc	(GlcNAc)	Total
400	1	9	22	31
	3	18	27	45
	6	25	22	47
200	1	16	18	34
	3	34	38	72
	6	46	29	75
100	1	28	50	78
	3	53	40	93
	6	75	20	95

 $<sup>^{</sup>a}$  [E] = 0.1 U/mL in 0.1 M citrate-phosphate buffer, pH 6.0.

<sup>b</sup> HPLC yield.

**Table 2**. Production of GlcNAc from  $\beta$ -chitin by chitinase from *Burkholderia cepacia* TU09.

β-chitin/enzyme <sup>a</sup> (mg/U)	Digestio time (day)	n % Yield <sup>b</sup> GlcNAc
400	1	31
	3	57
	6	65
200	1	62
	3	81
	6	84
100	1	90
	3	96
	6	100

 $<sup>^{</sup>a}$ [E] = 0.1 U/mL in 0.1 M citrate-phosphate buffer, pH 6.0.

b HPLC yield.

**Table 3**. Production of GlcNAc from  $\alpha$ -chitin by chitinase from *Bacillus licheniformis* SK-1.

α - Chitin/Enzyme (mg/U)	Digestion time (day)	% Yield GlcNAc <sup>a</sup>
	1	32
10	3	40
	6	41

 $<sup>{}^{</sup>a}[E] = 1$  U/mL in 0.1 M citrate-phosphate buffer, pH 6.0.

<sup>b</sup> HPLC yield.

**Table 4**. Production of GlcNAc from  $\alpha$ -chitin by chitinase from *Burkholderia cepacia* TU09.

α - Chitin/Enzyme <sup>a</sup> (mg/U)	Digestion time (day)	% Yield <sup>b</sup> GlcNAc
100	1	37
	3	54
	7	57
33	1	41
	3	57
	7	85

<sup>&</sup>lt;sup>a</sup>[E] = 0.1 U/mL in 0.1 M citrate-phosphate buffer, pH 6.0. b HPLC yield.

We have demonstrated here for the first time that chitinase from certain bacteria can completely hydrolyze both powdered  $\alpha$ - and  $\beta$ -chitin to give GlcNAc in very high to quantitative yield. The cleanliness of the reaction, mild conditions, ease of substrate preparation, and high production yield undeniably render the approach of using enzyme more attractive than the current acid hydrolysis process for the production of GlcNAc. Despite all these beneficial factors in using bacterial chitinase, care must be taken in further development to ensure food safety and enhance cost efficiency for industrial production of GlcNAc.

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# APPENDIX B

Table B. Biochemical characteristics of the bacteria strain SK-1.

Characteristics	Reaction	Characteristics	Reaction
Gram reaction	+ve	Fermentative production of acid from	
Fermentative production of acid from		Salicine	+
Glycerol	+	Cellobiose	+
Erythritol	-	Maltose	+
D-arabinose	-	Lactose	+
L-arabinose	+	Melibiose	-
Ribose	+	Sucrose	+
D-xylose	+	Trehalose	+
L-xylose	-	Inuline	+
Adonitol	-	Melezitose	-
] □β-methyl –D-xylose	-	D-raffinose	-
Galactose	+	Starch	+
D-glucose	+	Glycogen	+
D-fructose	+	Xylitol	+
D-mannose	+	□β-gentiobiose	+
L-sorbose	-	D-turanose	+
Rhamnose	-	D-lyxose	-
Dulcitol	-	D-tagatose	+
Inoitol	-	D-fucose	-
Manntal	+	L-fucose	-
Sorbitol	+	D-arabitol	
α-methyl-D-mannoside	-	L-arabitol	-
α-methyl-D-glucoside	+	Gluconate	+
N-acetyl-glucosamine	+	2-keto-gluconate	-
Amygdaline	+	5-keto-gluconate	-
Arbutine	+	Esculine	+

Remark: +ve = Gram positive bacteria; + = Positive reaction; - = Negative reaction

#### APPENDIX C

# Preparation for non-denaturing polyacrylamide gel electrophoresis Stock reagents

## 30% Acrylamide, 0.8% bis-acrylamide, 100 ml

acrylamide 29.2 g

N, N'-dimethylene-bis-acrylamide 0.8 g

Adjust volume to 100 ml with distilled water

### 1.5 M Tris-HCl pH 8.8

Tris (hydoxymethyl)-aminomethane 18.17 g

Adjust pH to 8.8 with 1 M HCl and adjusted volume to 100 ml with distilled water

# 2 M Tris-HCl pH 8.8

Tris (hydoxymethyl)-aminomethane 24.0 g

Adjust pH to 8.8 with 1 M HCl and adjusted volume to 100 ml with distilled water

## 0.5 M Tris-HCl pH 6.8

Tris (hydoxymethyl)-aminomethane 6.06 g

Adjust pH to 6.8 with 1 M HCl and adjusted volume to 100 ml with distilled water

## 1 M Tris-HCl pH 8.8

Tris (hydoxymethyl)-aminomethane 12.0 g

Adjust pH to 8.8 with 1 M HCl and adjusted volume to 100 ml with distilled water

#### Solution B (1.5 M Tris-HCl pH 8.8)

2 M Tris-HCl pH 8.8 75 ml

Distilled water 25ml

## Solution B-SDS (1.5 M Tris-HCl pH 8.8)

2 M Tris-HCl pH 8.8 75 ml

10% SDS 4 ml

Distilled water 21 ml

#### Solution C (0.5 M Tris-HCl pH 6.8)

Distilled water

50 ml 1 M Tris-HCl pH 8.8 50 ml

Solution B-SDS (1.5 M Tris-HCl pH 6.8)

50 ml 1 M Tris-HCl pH 6.8

10% SDS 4 ml

46 ml Distilled water

5x Sample buffer

3.1 ml 1 M Tris-HCl pH 6.8

5 ml Glycerol

1% Bromophenol blue 0.5 ml

1.4 ml Distilled water

# Non-denaturing electrophoresis buffer, 1 liter (25 mM Tris, 192 mM glycine)

Tris (hydroxymethyl)-aminomethane 3.03 g

Glycine 14.40 g

Dissolved in distilled water to 1 liter without pH adjustment (final pH should be 8.3).

# SDS electrophoresis buffer, 1 liter (25 mM Tris, 192 glycine, 0.1% SDS)

Tris (hydroxymethyl)-aminomethane 3.03 g

Glycine 14.40 g

**SDS** 1 g

Dissolved in distilled water to 1 liter without pH adjustment (final pH should be 8.3).

## Preparation of non-denaturing PAGE

## 10% Separating gel

30% Acrylamide solution 3.3 ml

2.5 ml Solution B

Distilled water 4.2 ml

50 ml  $10\% (NH_4)_2S_2O_8$ 

10 ml **TEMED** 

#### 3% Separating gel

30% Acrylamide solution 0.4 ml

Solution C 1.0 ml

Distilled water	2.6 ml
10% (NH <sub>4</sub> ) <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	30 μl
TEMED	5 u

# Preparation of SDS-PAGE

# 10% Separating gel

Prepare as described for non-denaturing gel but using with solution B containing SDS instead solution B.

# 3% Separating gel

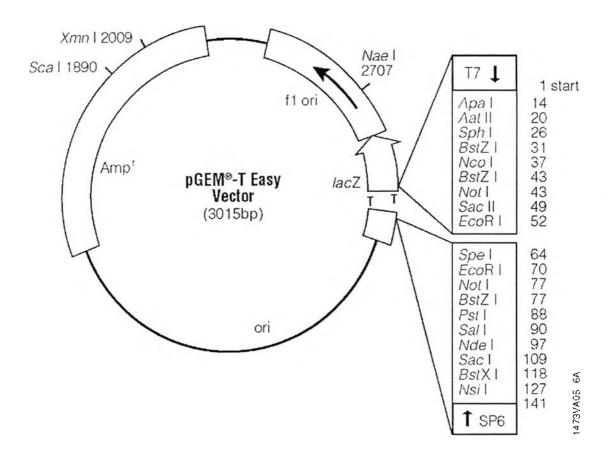
Prepare as described for non-denaturing gel but using with solution C containing SDS instead solution C.

# Preparation of gel system IEF

30% Acrylamide solution	3.3 ml
Distilled water	4.2 ml
Ampholyte solution pH 3-10	288 μΙ
Mixed well and degassed before added:	
$10\% (NH_4)_2S_2O_8$	50 μl
TEMED	20 µl

# APPENDIX D

# Restriction map of pGEM-T easy



# **APPENDIX E**

# Standard curve of N-acetyl-D-glucosamine for chitinolytic enzyme assay by colorimetric method.

Standard curve for N-acetyl-D-glucosamine (GlcNAc) was made by determining the absorbance value at 420 nm of standard N-acetyl-D-glucosamine according to the method of Schale.

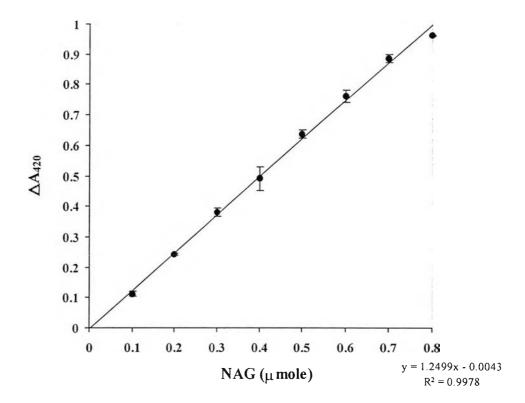


Figure E1. Correlation between final concentration of standard N-acetyl-D-glucosamine and optical density (absorbance) at 420 nm.

# Standard curve of p-nitrophenol for chitinolytic enzyme assay by colorimetric method.

Standard curve for p-nitrophenol was made by determining the absorbance value at 420 nm of standard p-nitrophenol according to the colorimetric method.

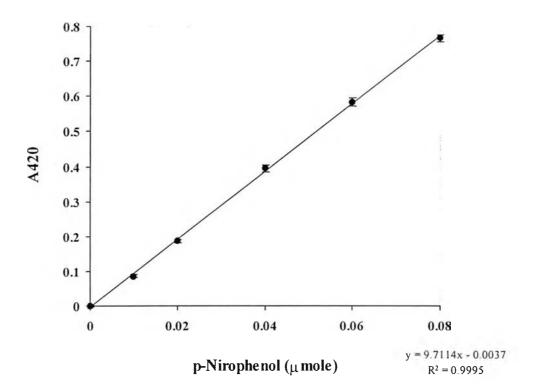


Figure E2. Correlation between final concentration of standard p-nitrophenol and optical density (absorbance) at 420 nm.

## **APPENDIX F**

# Standard curve of protein concentration by Bradford's colorimetric method.

Standard curve for bovine serum albumin (BSA) was made by determining the absorbance value at 595 nm of BSA according to the method of Bradford.

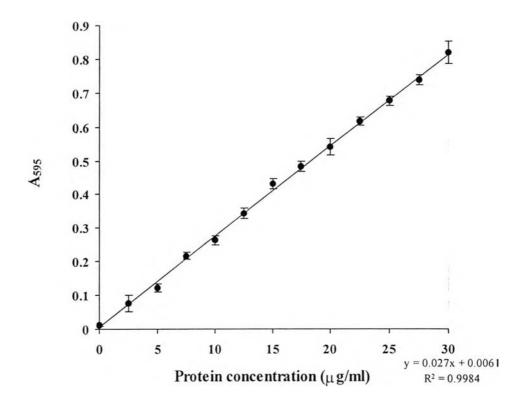


Figure F Relationship between standard protein (BSA) concentration and optical density (absorbance) at 595 nm.

# APPENDIX G

# Calibration curve of standard pI markers

The migration distance from cathode of pI standard markers w measured and plotted against pI's.

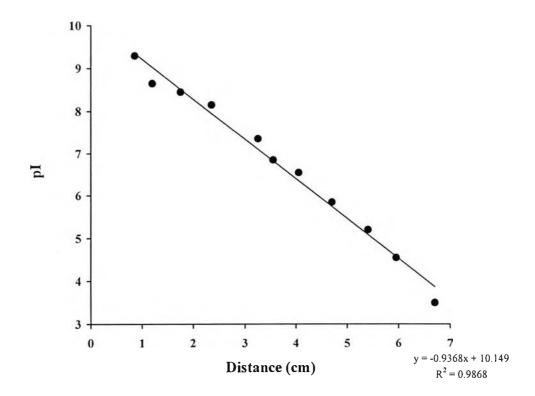


Figure G. Calibration curve for standard pI markers.

#### APPENDIX H



#### LIST OF PUBLICATIONS

#### Abstract

- Kudan S., Achakulwisut, E., Weeratien, K., Tertdee, A., Chisawatana, O. and
   Pichyangkura, R. 2000. Properties and Characterization of Chitinase from Bacillus licheniformis Strains SK-1 and RN001. Poster presented at the 2<sup>nd</sup>
   Joint Seminar on Development of Thermotolerant Microbial Resources and their Applications, 21-25 November, Yamagushi University, Yamaguchi, Japan. Abstracts book. p.133(P3-15).
- Kudan, S. and Pichyangkura, R. 2001. Identification and characterization of thermophilic bacterium which produce chitinase isolate SK-1. Poster presented at the 27<sup>th</sup> Congress on Science and Technology of Thailand, 16-18 October, Lee Gardens Plaza Hotel, Hat Yai, Songkla, Thailand. Abstracts book. p.547 (17-16P-47).

### **Proceeding**

Kudan, S., and Pichyangkura, R. Purification of thermostable chitinase produced by Bacillus licheniformis SK-1. Proceeding of the Fifth Asia Paciffic Chitin-Chitosan Symposium and Exhibition, Bangkok, Thailand, March 13-15, 2002. in press.

### Paper

Pichyangkura, R., <u>Kudan, S.</u>, Kuttiyawong, K., Sukwathanasininitt, M., and Aiba, S. 2002. Quantitative production of 2-acetamido-2-deoxy-D-glucose from crystalline chitin by bacterial chitinase. *Carbohydr. Res.* 337: 557-559.

### Nucleotide sequence

GenBank accession no., AF411341



# **BIOGRAPHY**

Mr. Sanya Kudan was born on June 4<sup>th</sup> 1974 in Sukhothai. After He finished Mattayom VI in 1991 from Mattayom Wat Nongkhaem School, he was enrolled in the Biotechnology, Ramkhamhaeng University and graduated with a B.S. in 1997. After which, he worked in Associate Professor Dr.Jerapan Krungkai's laboratory for 2 years. He entered the graduate program for M.Sc. in Biochemistry at Chulalongkorn University in 1999.