

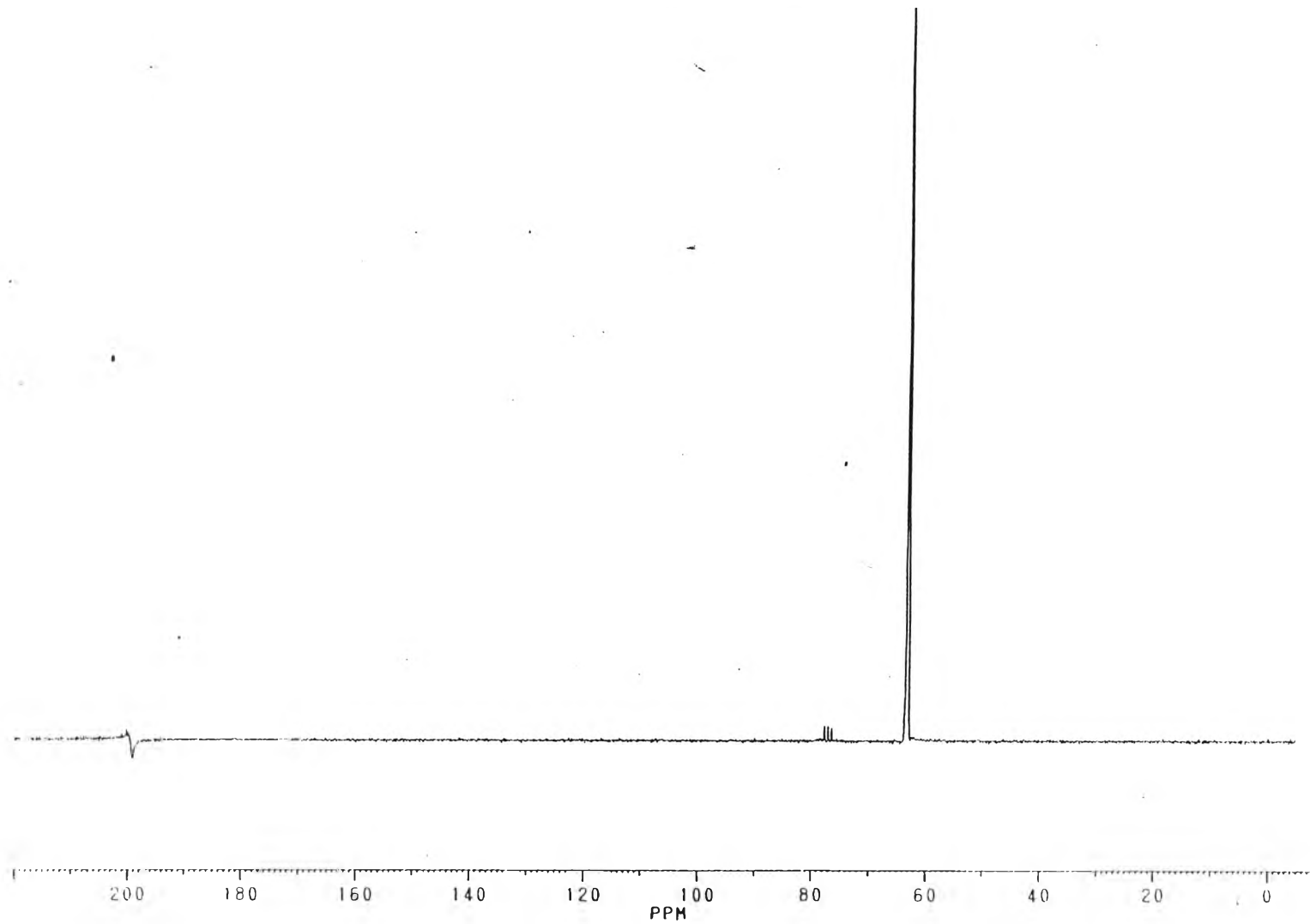
## REFERENCES

1. Phatanaphakdee, Kawin. Synthesis of Lubricating Base Oils from Palm Oil. Master 's Thesis, Multidisciplinary of Petrochemistry - Polymer, Graduate School, Chulalongkorn University, (1995).
2. Tubthim, Darunee. Synthesis Diester Lubricating Base Oils from Palm Oil. Master 's Thesis, Multidisciplinary of Petrochemistry - Polymer, Graduate School, Chulalongkorn University, (1996).
3. Mcketta, J.J. Lubricating oil. Encyclopedia of Chemical Processing and Design 28 (1987) : 378 - 393, 435 - 449.
4. Mark, H.F., and Gaylord, N.G. Lubricants. Encyclopedia of Polymer Science and Technology 8 (1971) : 325 - 337.
5. Mark, H.F., Bikales, N.M., Overberger, C.G., and Menges, G. Waxes. Encyclopedia of polymer science and Engineering 17 (1989) : 784 - 795.
6. Mark, H.F., and Gaylord, N.G. Waxes. Encyclopedia of Polymer Science and Technology 14 (1971) : 768 - 779.
7. Hollinshead, C., and Springs, S.F. Friction Reducing Petroleum Mixtures and Method of Making Same. US 3,849,323 (1974).
8. Salamone J.C. Antifoaming agents. Polymeric Materials Encyclopedia 1 (1996) : 285 - 296.
9. Kirk, R.E., and Othmer, D.F. Waxes. Encyclopedia of Chemical Technology 15 (1956) : 1 - 17.
10. Kirk, R.E., and Othmer, D.F. Waxes. Encyclopedia of Chemical Technology 24 (1984) : 466 - 497.
11. Albin, W.H. The Chemistry and Technology of Waxes. 2<sup>nd</sup> ed. New York : The Guinn Co., Inc., (1960).

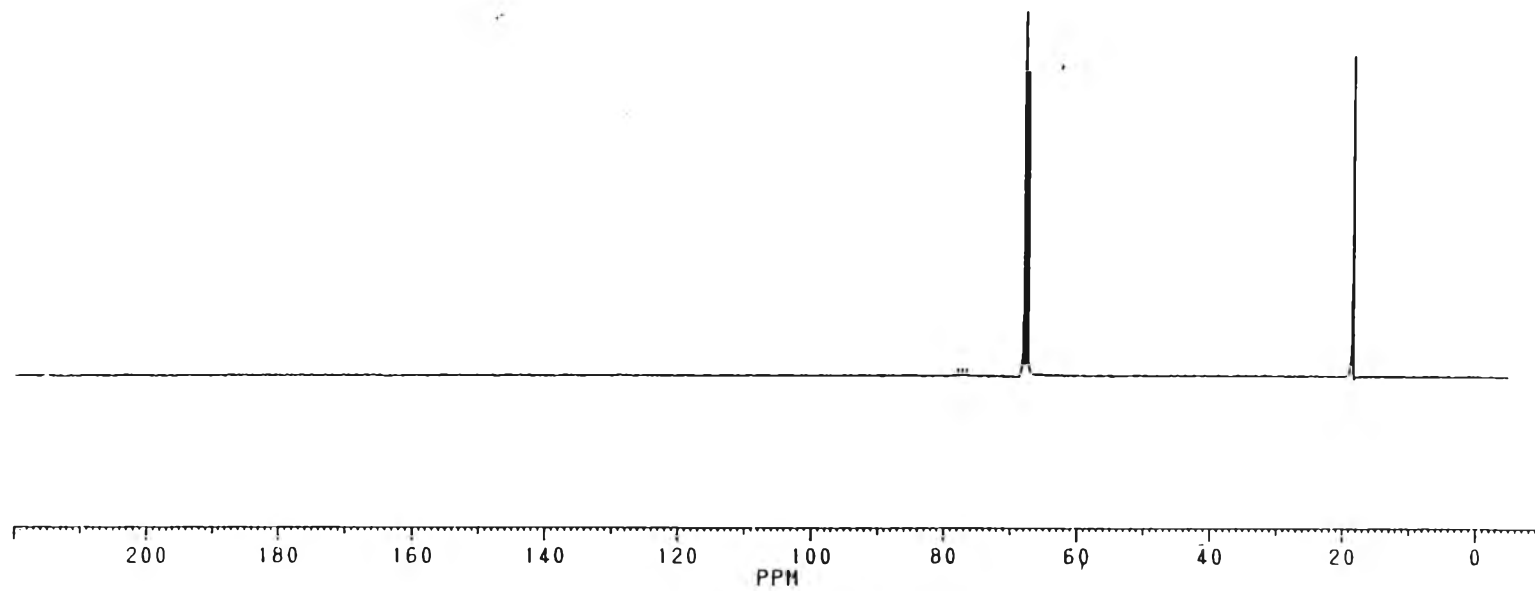
12. Wills, J.G. Lubrication Fundamental. New York : Marcel Dekker Inc., (1980). 34 - 40, 75 - 87.
13. Aboul E.I. Naga, H.H., Salem, A.E.M. Base Oil Thermooxidation. Lubrication Engineering. 42 (April 1989) ; 210 - 217.
14. Hobson, G.D. Modern Petroleum Technology. 5<sup>th</sup> e.d., London : John Willey & Sons, (1984) ; part 1- 2, 16 - 25.
15. McConnell, R.L., Joyner, F.B., and Trotter, J.R. Branched Polyester Waxes. US 4,481,351 (1984).
16. McConnell, R.L., Trotter, J.R., and Joyner, F.B. Polyester Waxes Based on Mixed Aliphatic Dibasic Acids. US 4,481,352 (1984).
17. Trotter, J.R., Joyner, B.F., and McConnell R.L. Polyester Waxes Based on 1,12-Dodecanedioic Acid. US 4,487,919 (1984).
18. Arrandeau, J.P., and Patraud J. Cosmetic Make - Up Composition. US 4,820,510 (1989).
19. Brawn, E.A., Cockett M.A., Hexell, J., and Cockett, M.A. Transparent Hot Melt Jet Ink. US 5,185,035 (1993).
20. Hall, M.J., and Others. Extrudable Thermoplastic Particulated. US 5,236,649 (1993).
21. Craig, M.S. Solid Marking Composition Containing Giltter. US 5,261,952 (1993).
22. Craig, M.S. Solid Marking Composition Containing Giltter. US 5,383,954 (1995).
23. Mark, H.F., Bikales, N.M., and Gaylord, N.G. Antifoaming Agents. Encyclopedia of Polymer Science and Technology 2 (1965) : 164- 165.
24. Mcketta, J.J., and Cunningham, W.A. Foam. Encyclopedia of Chemical Processing and Design 23 (1985) : 318 - 329.

## **APPENDICES**

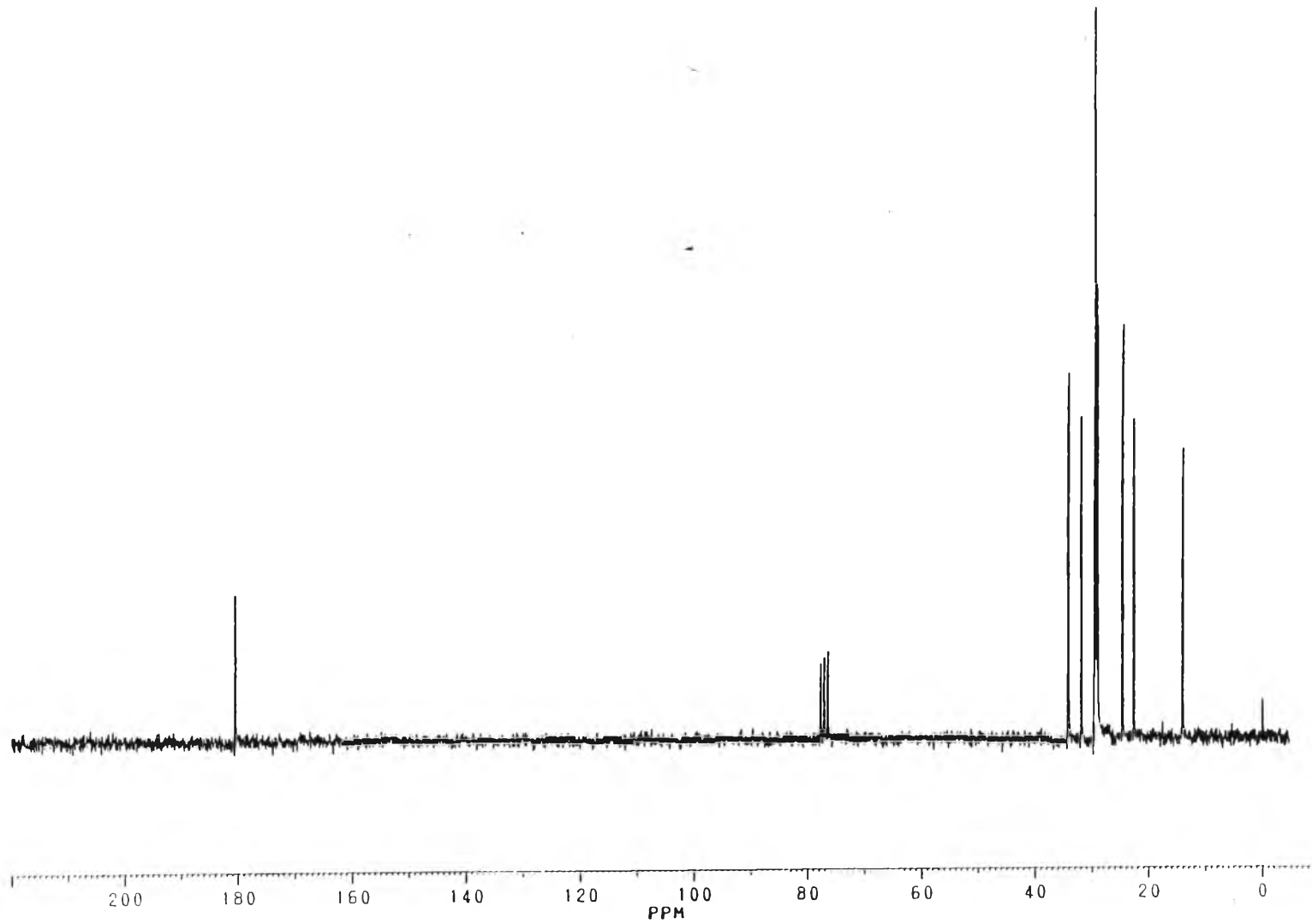
**APPENDIX A**



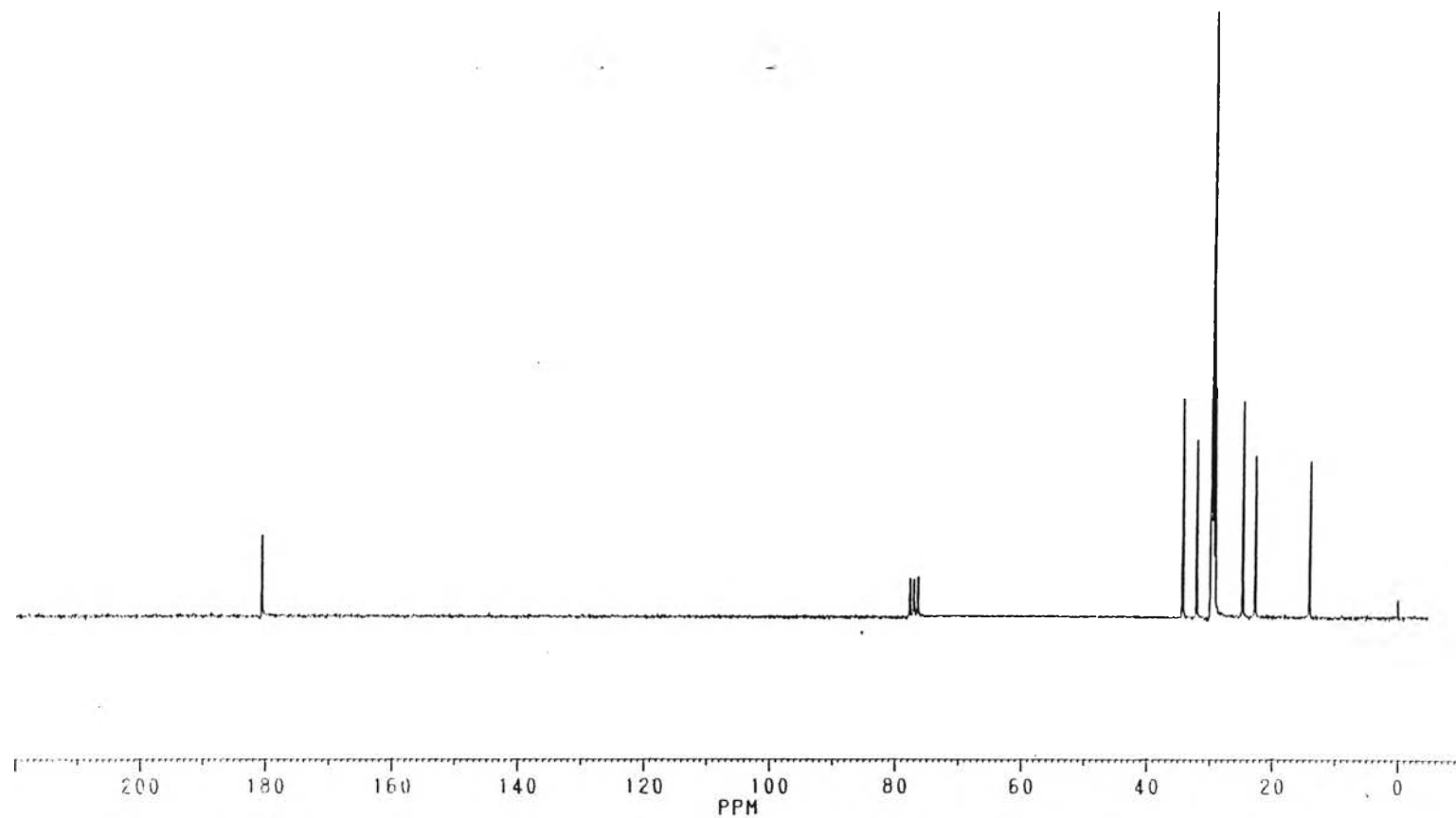
**Figure A1**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-ethanediol



**Figure A2**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-propanediol

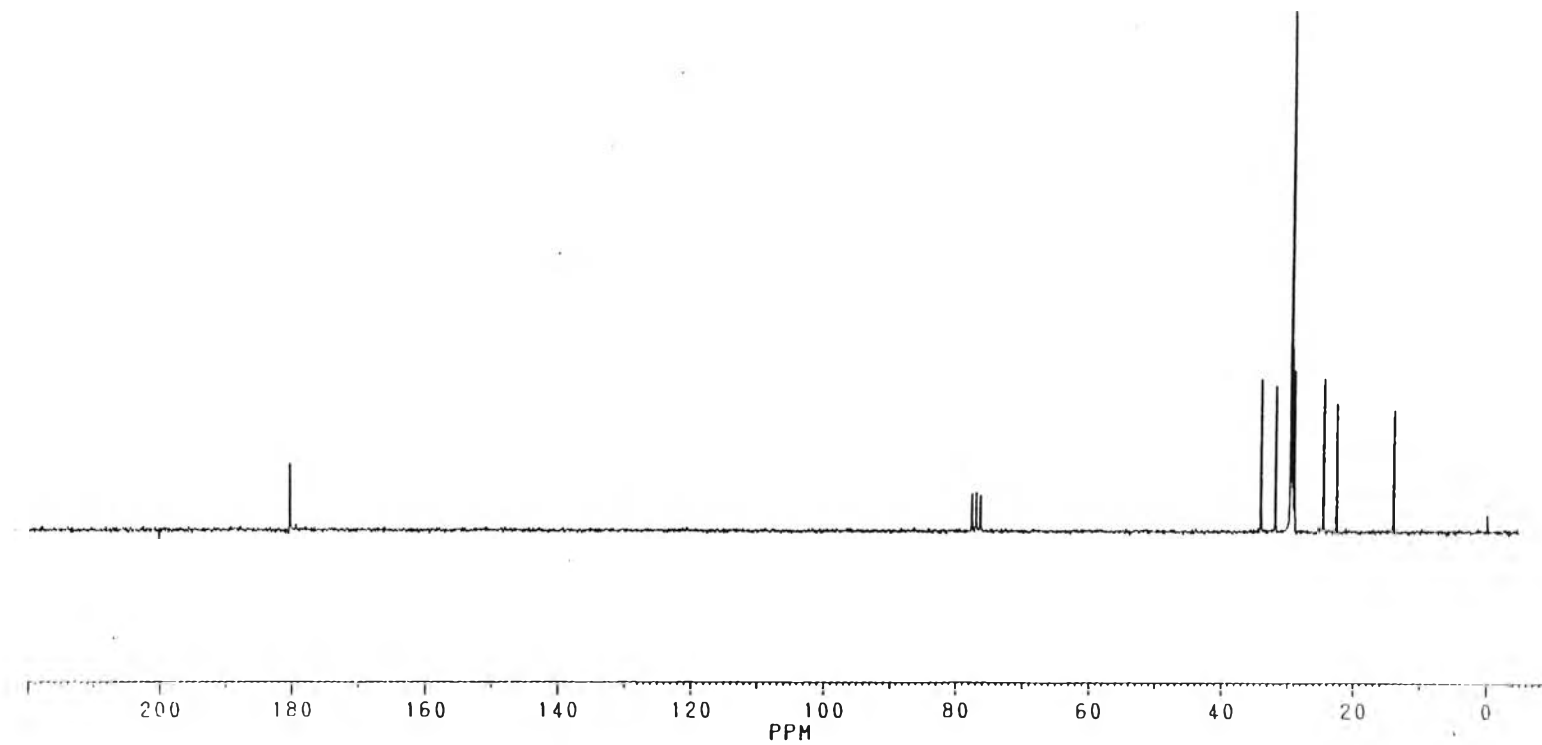


**Figure A3**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of lauric acid

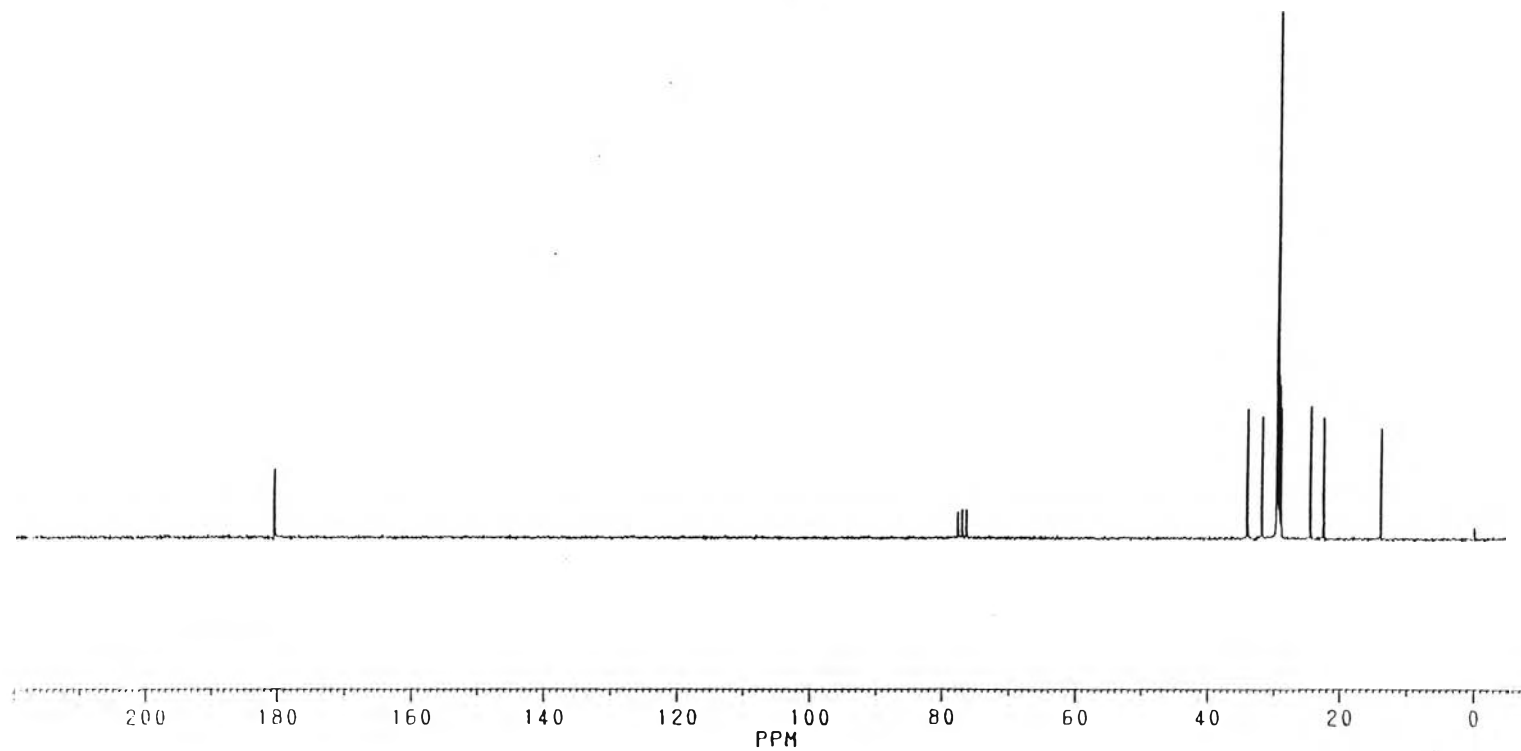


**Figure A4**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of myristic acid

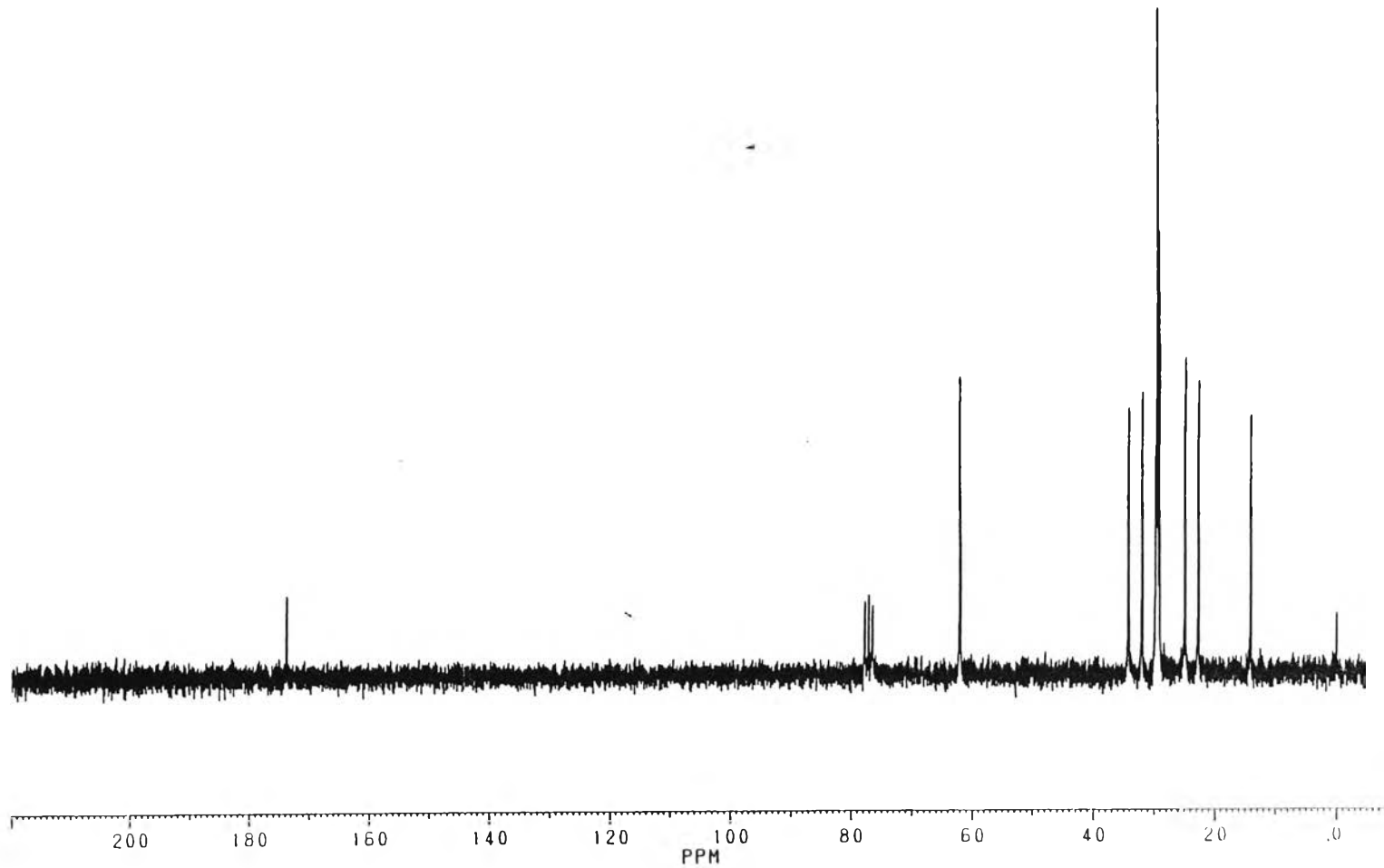




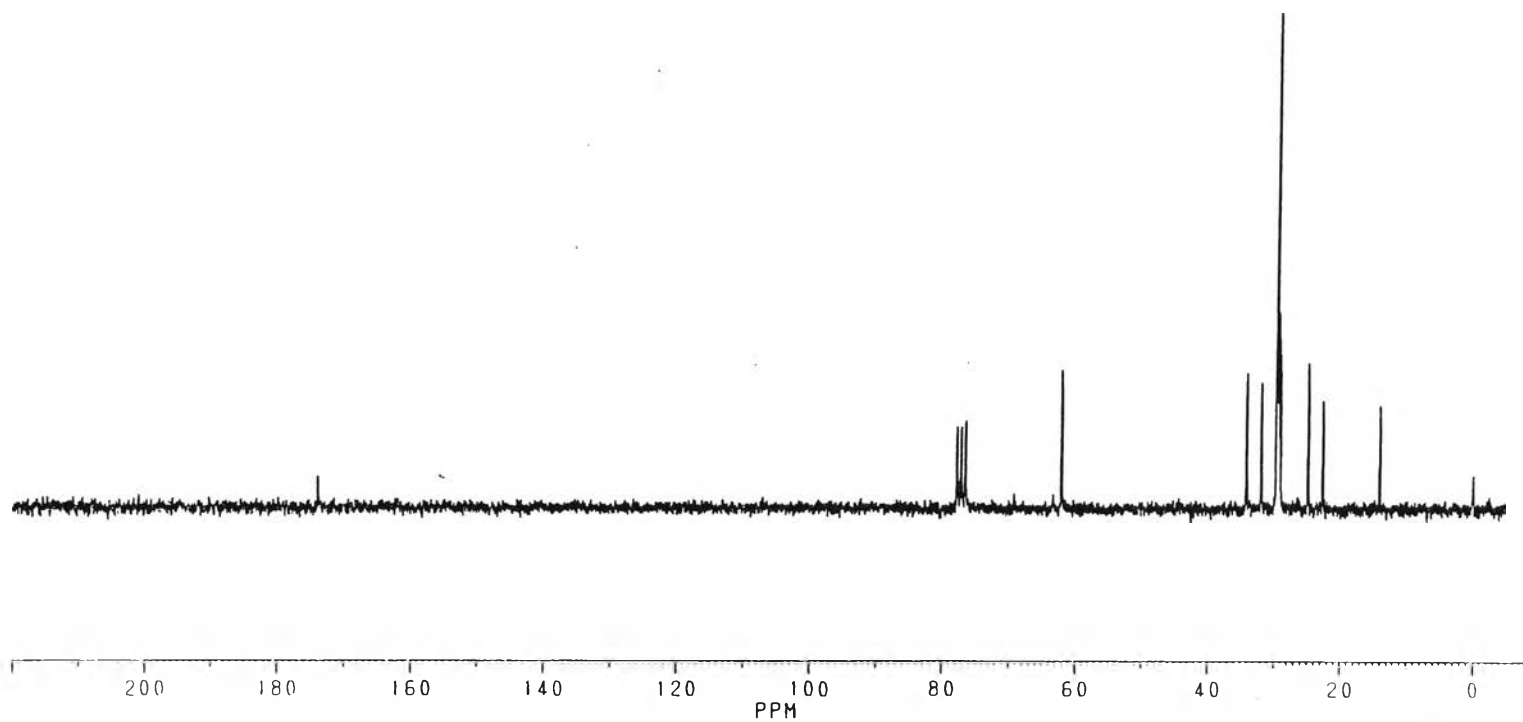
**Figure A5**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of palmitic acid



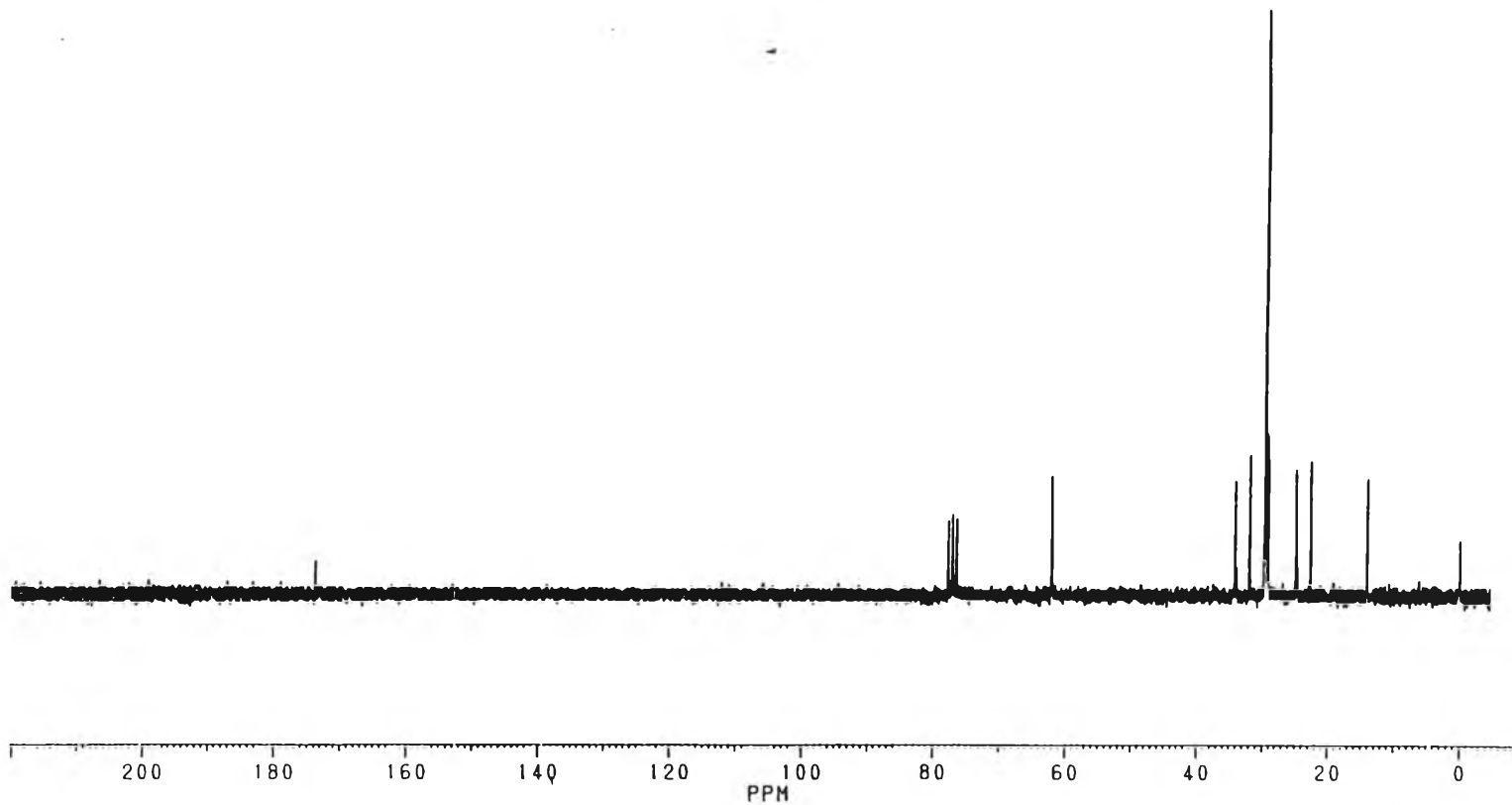
**Figure A6**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of stearic acid



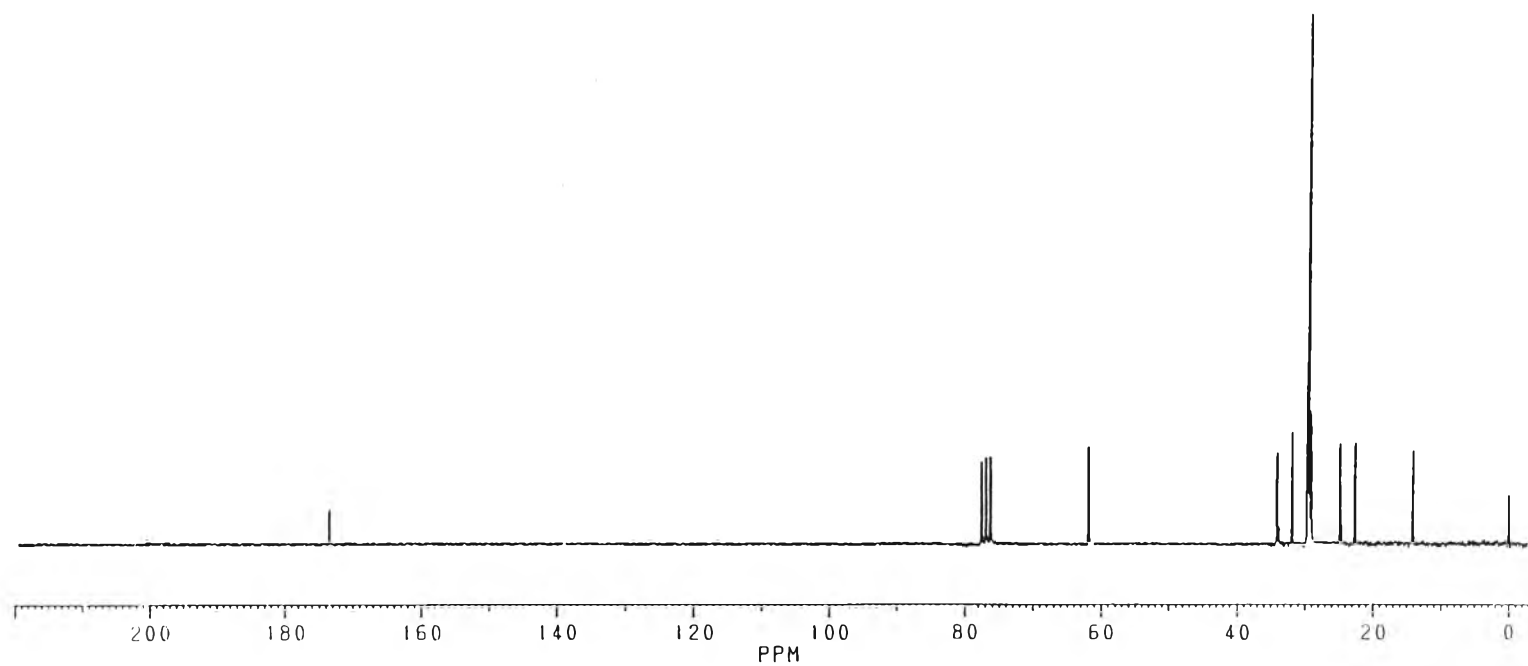
**Figure A7**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-ethanedilaurate



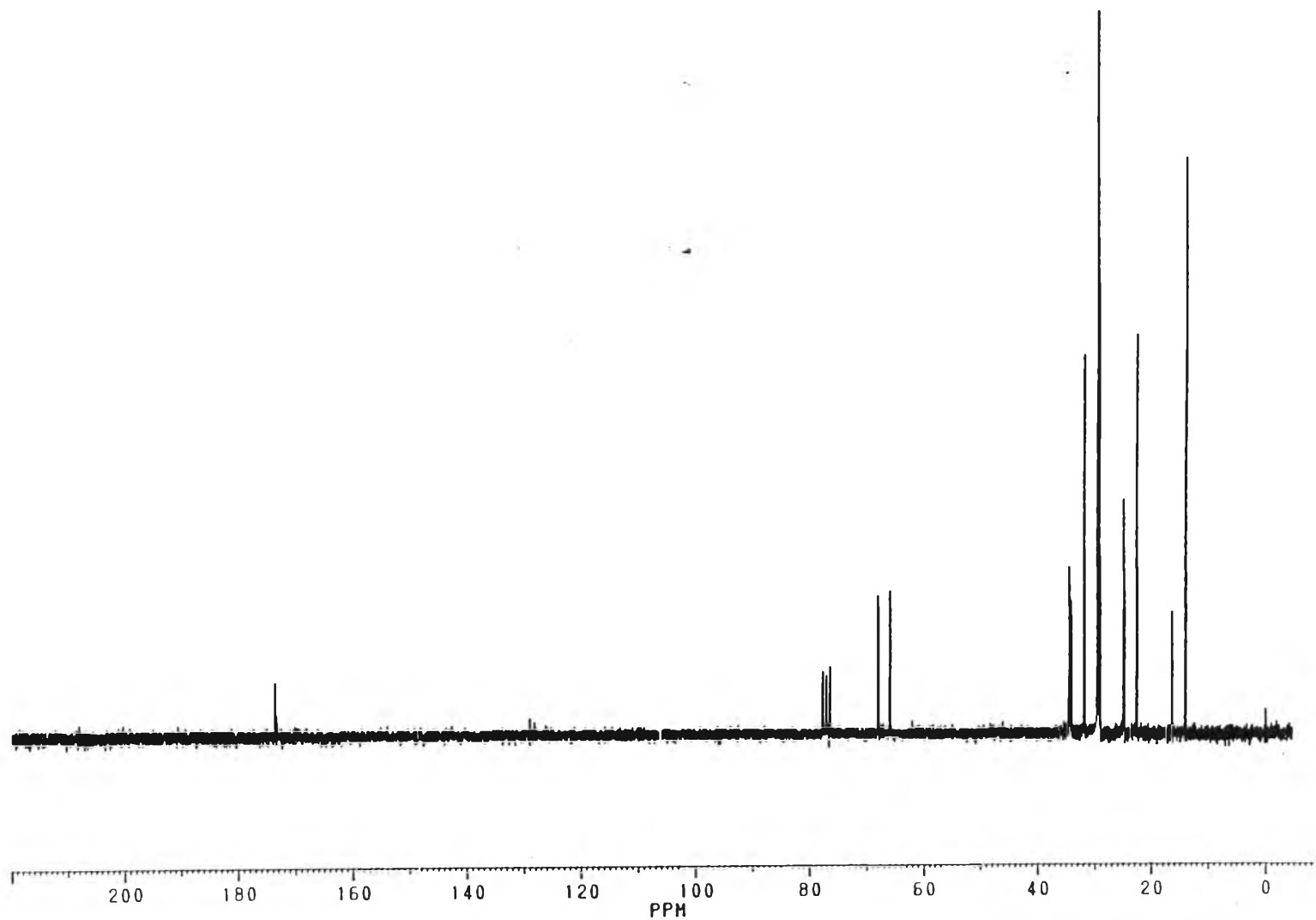
**Figure A8**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-ethanedimyristate



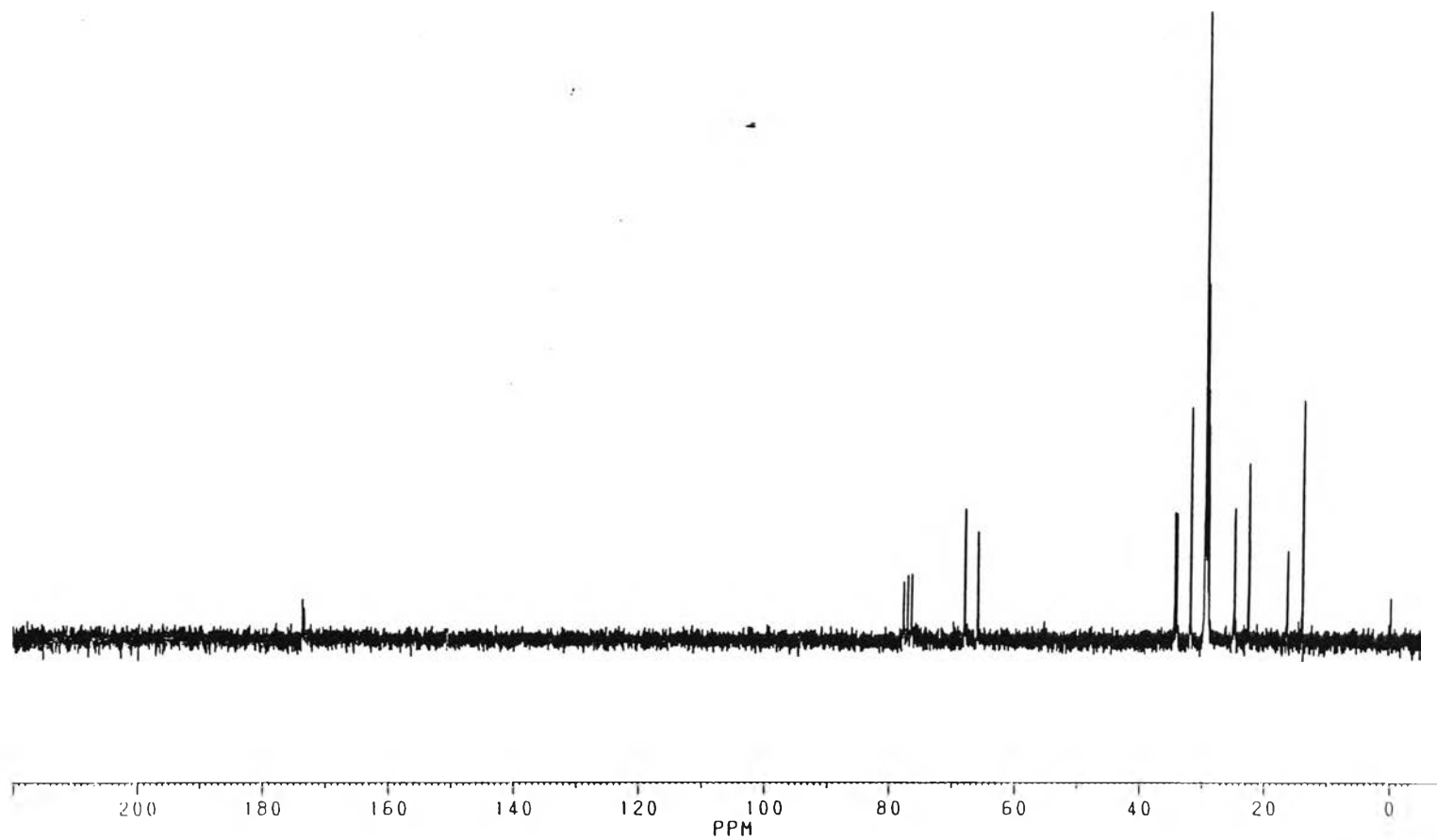
**Figure A9**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-ethanedipalmitate



**Figure A10**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-ethanedistearate

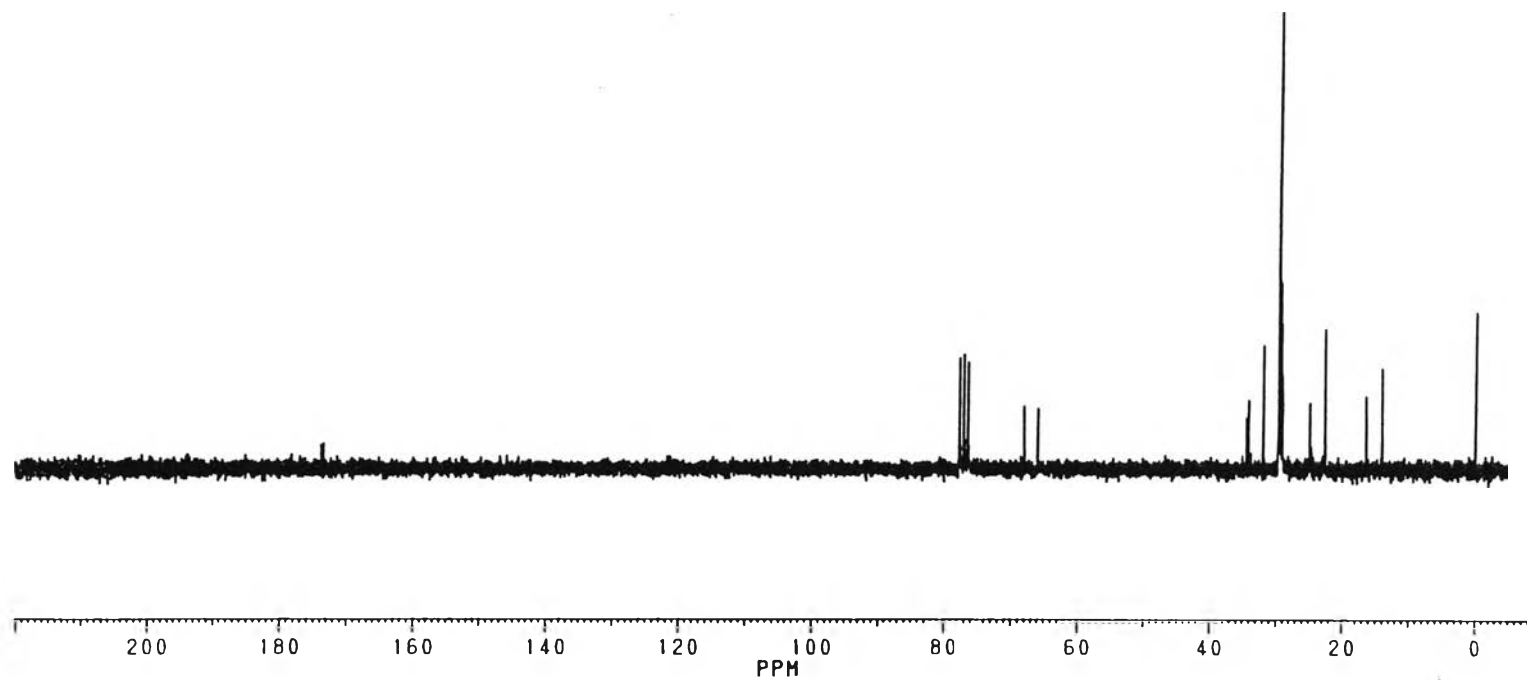


**Figure A11**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-propanedilaurate

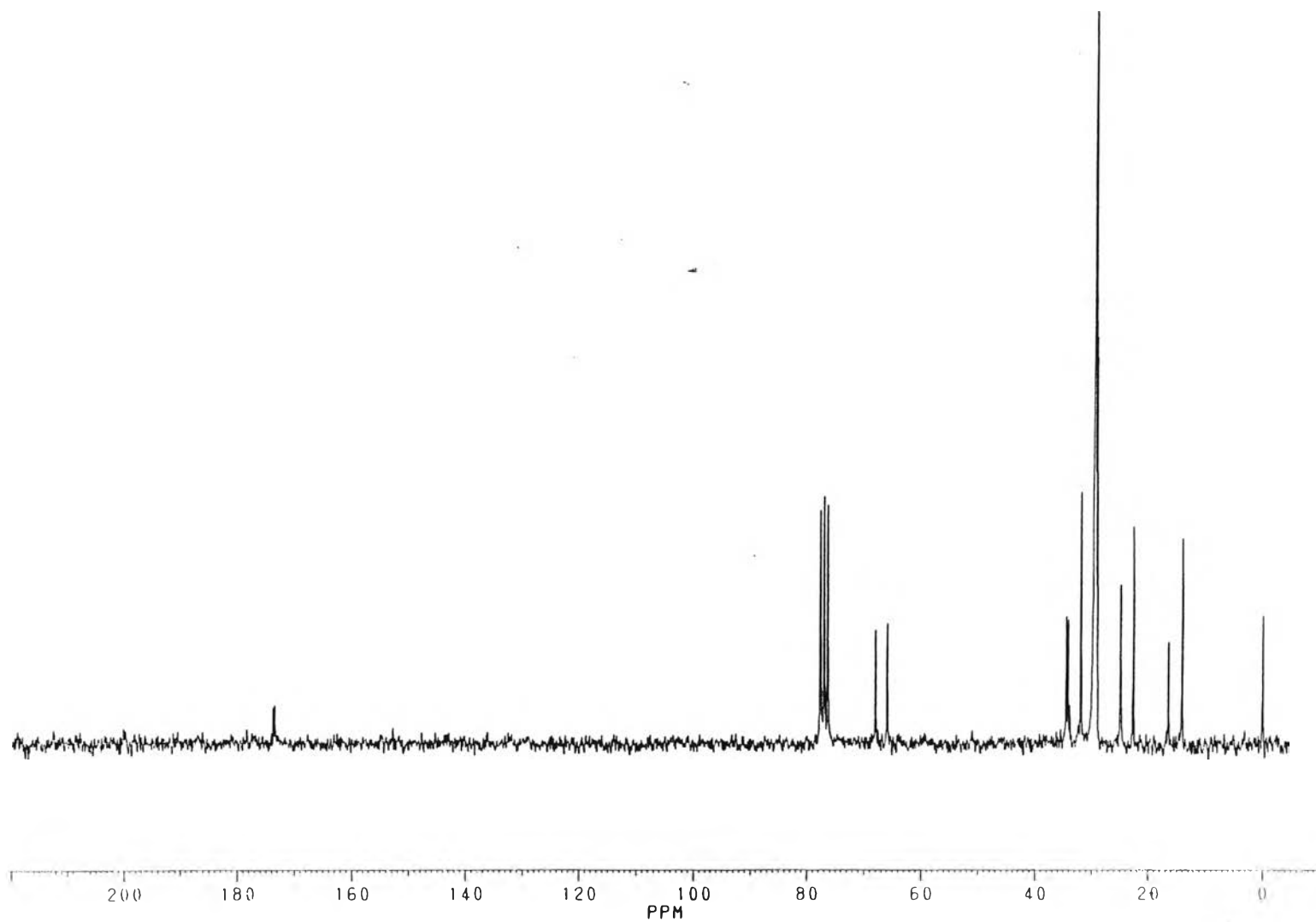


**Figure A12**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-propanedimyristate



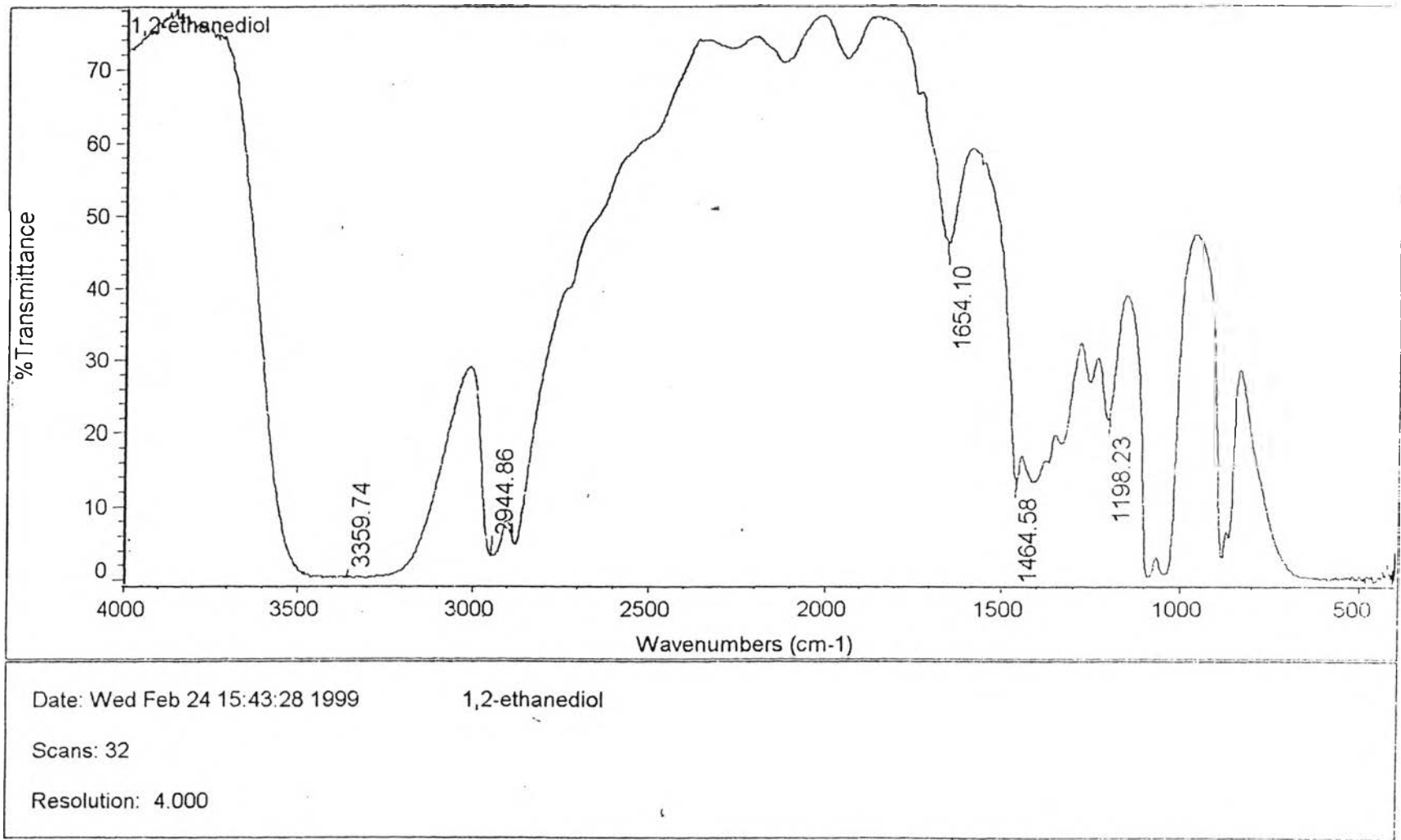


**Figure A13**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-propanedipalmitate



**Figure A14**  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 1,2-propanedistearate

## **APPENDIX B**



**Figure B1** IR spectrum of 1,2-ethanediol

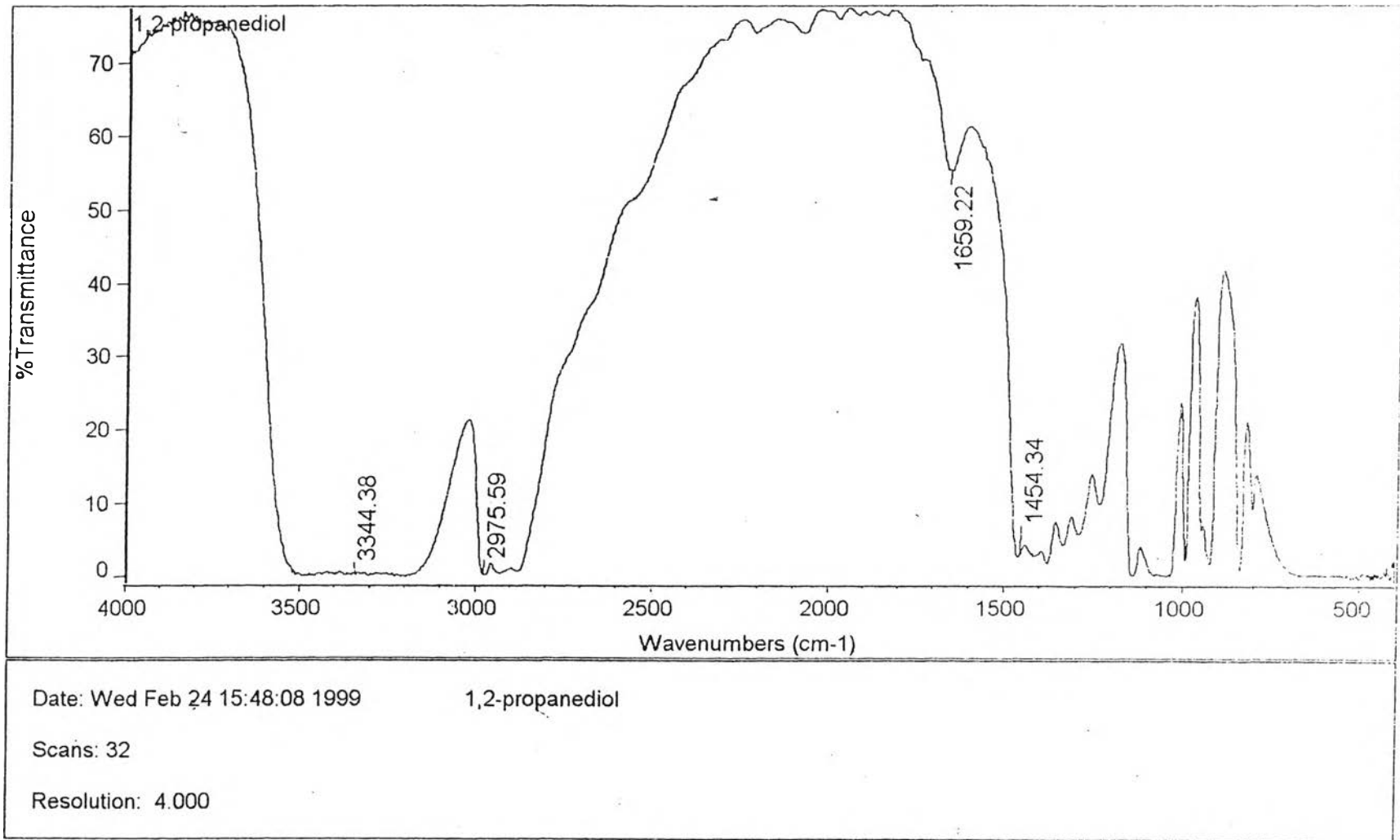
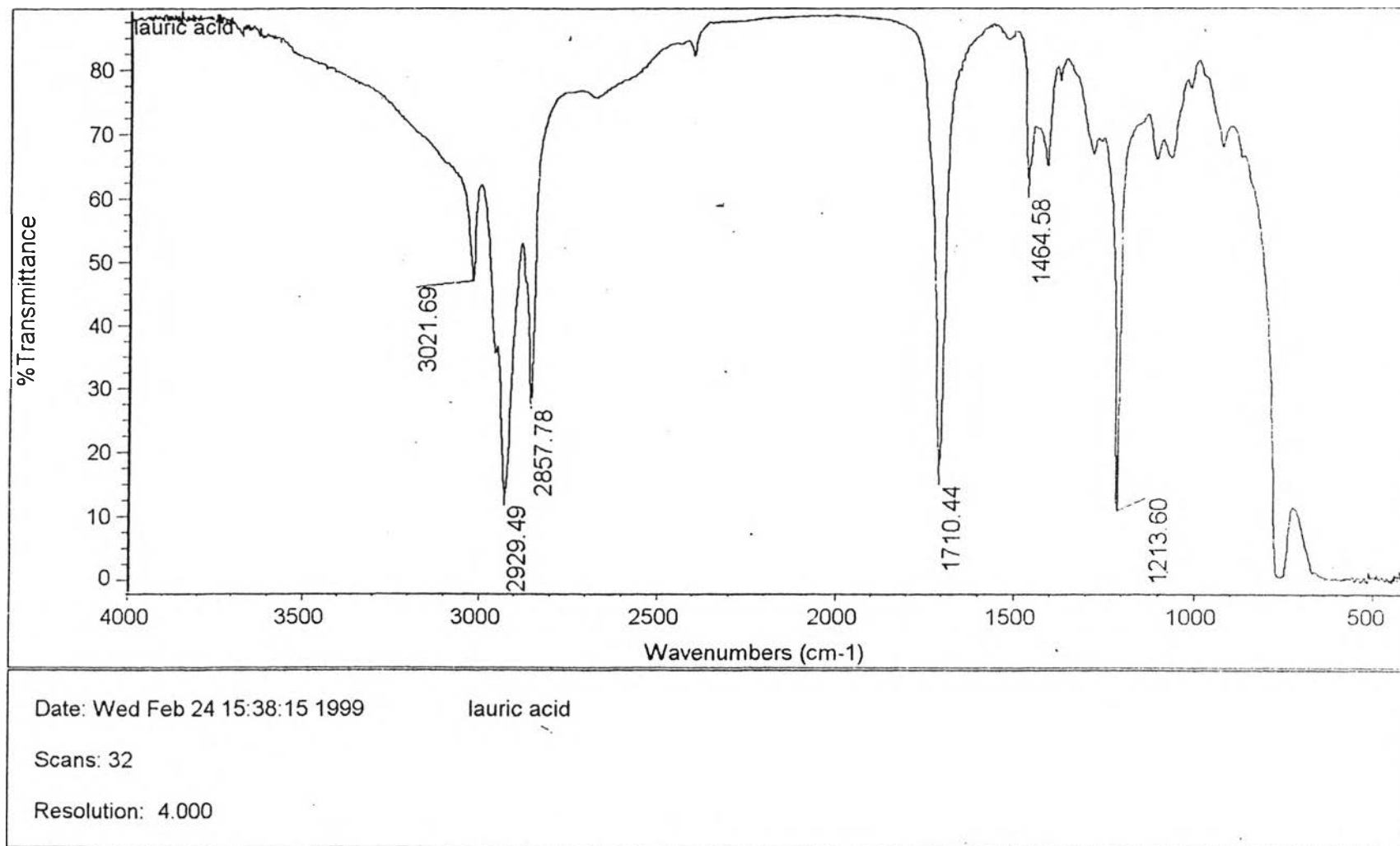
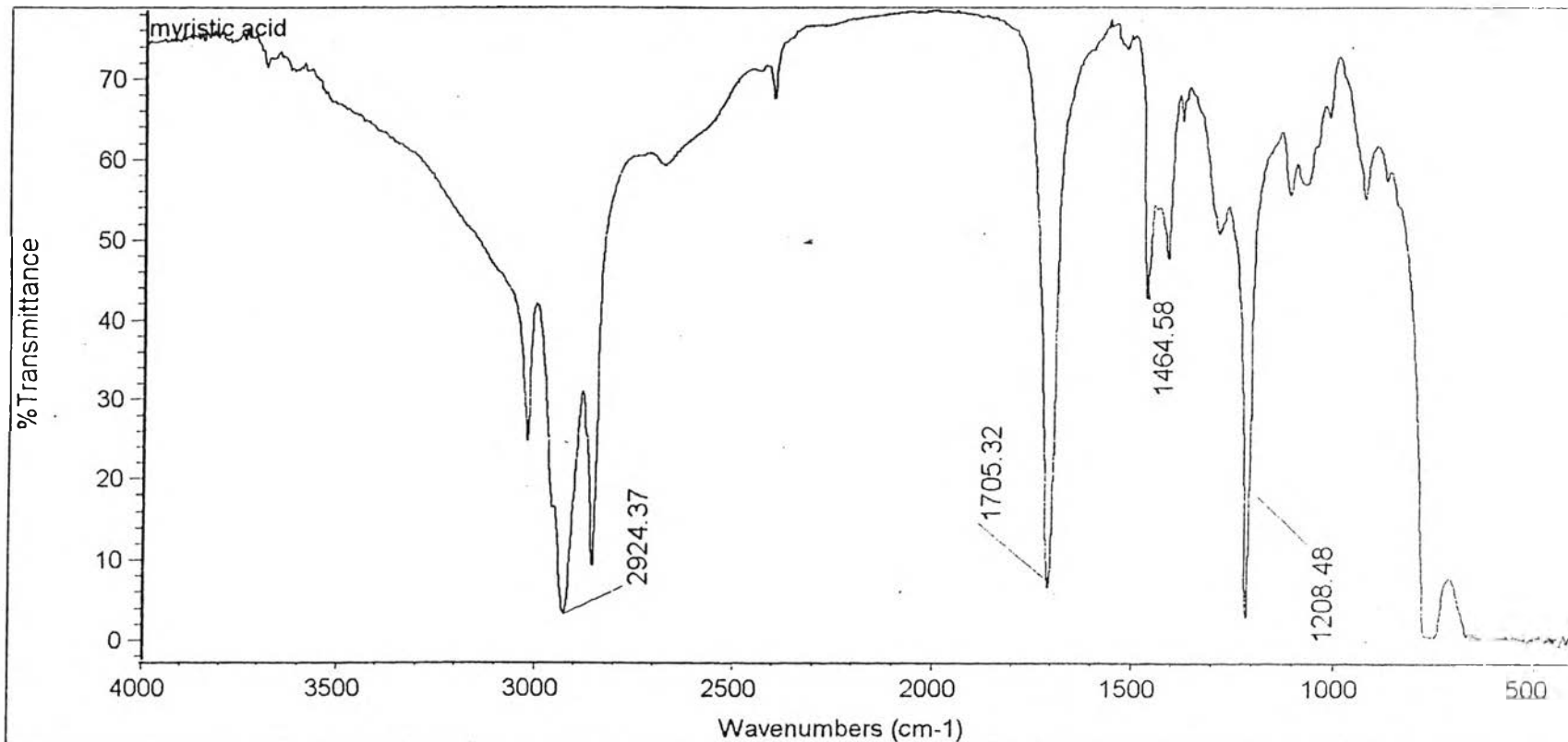


Figure B2 IR spectrum of 1,2-propanediol



**Figure B3** IR spectrum of lauric acid



Date: Wed Feb 24 14:11:12 1999

myristic acid

Scans: 32

Resolution: 4.000

**Figure B4** IR spectrum of myristic acid

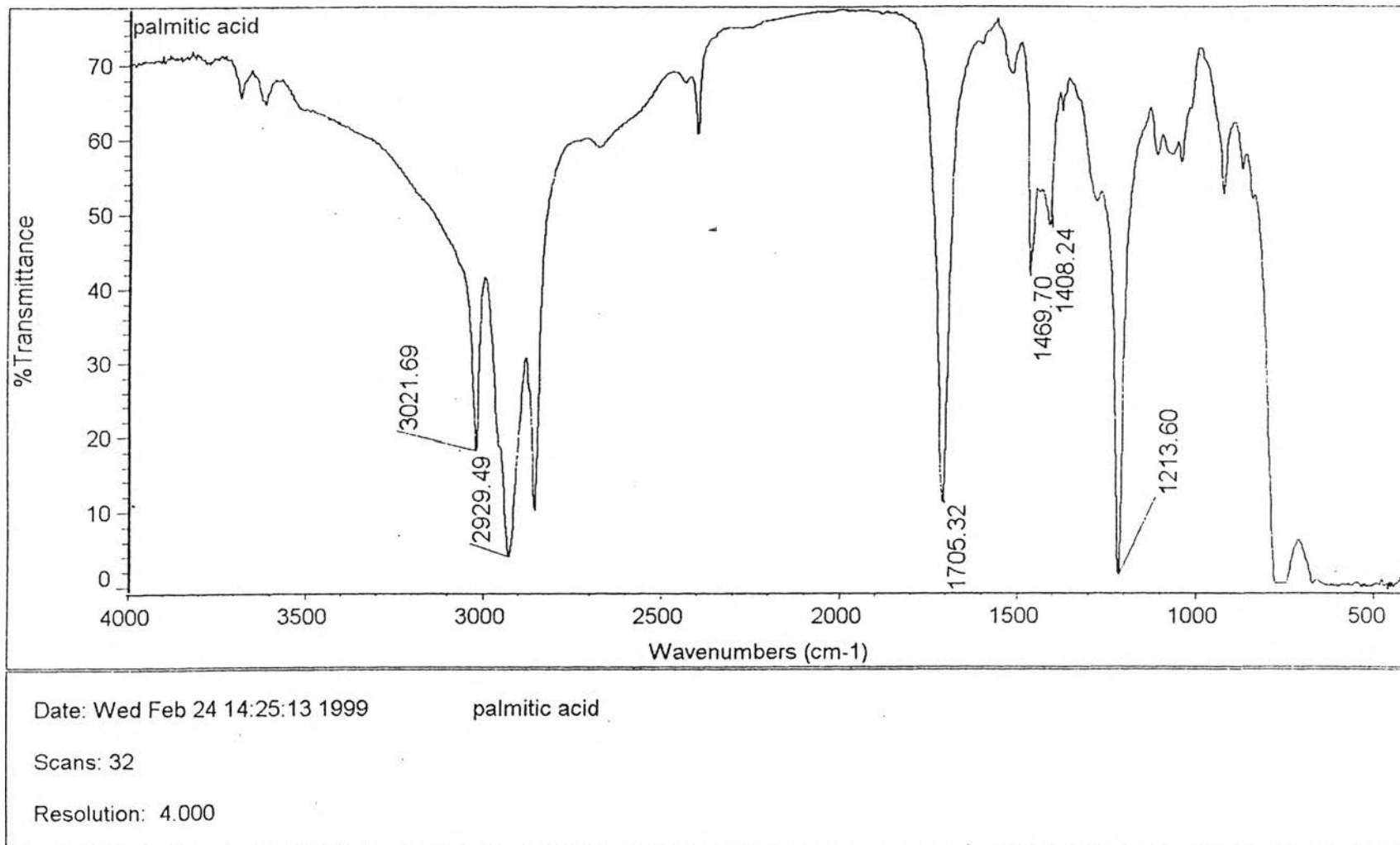


Figure B5 IR spectrum of palmitic acid



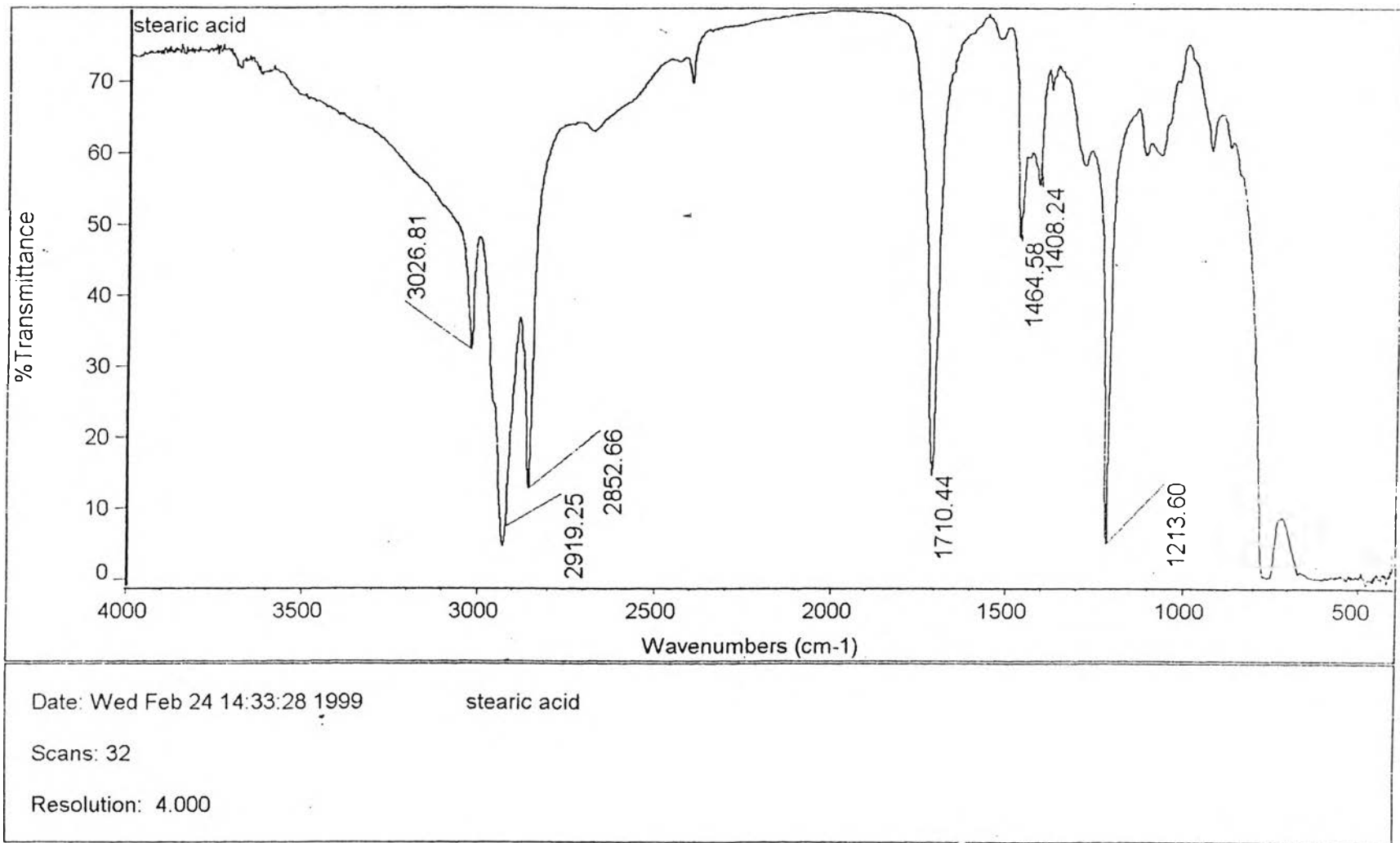
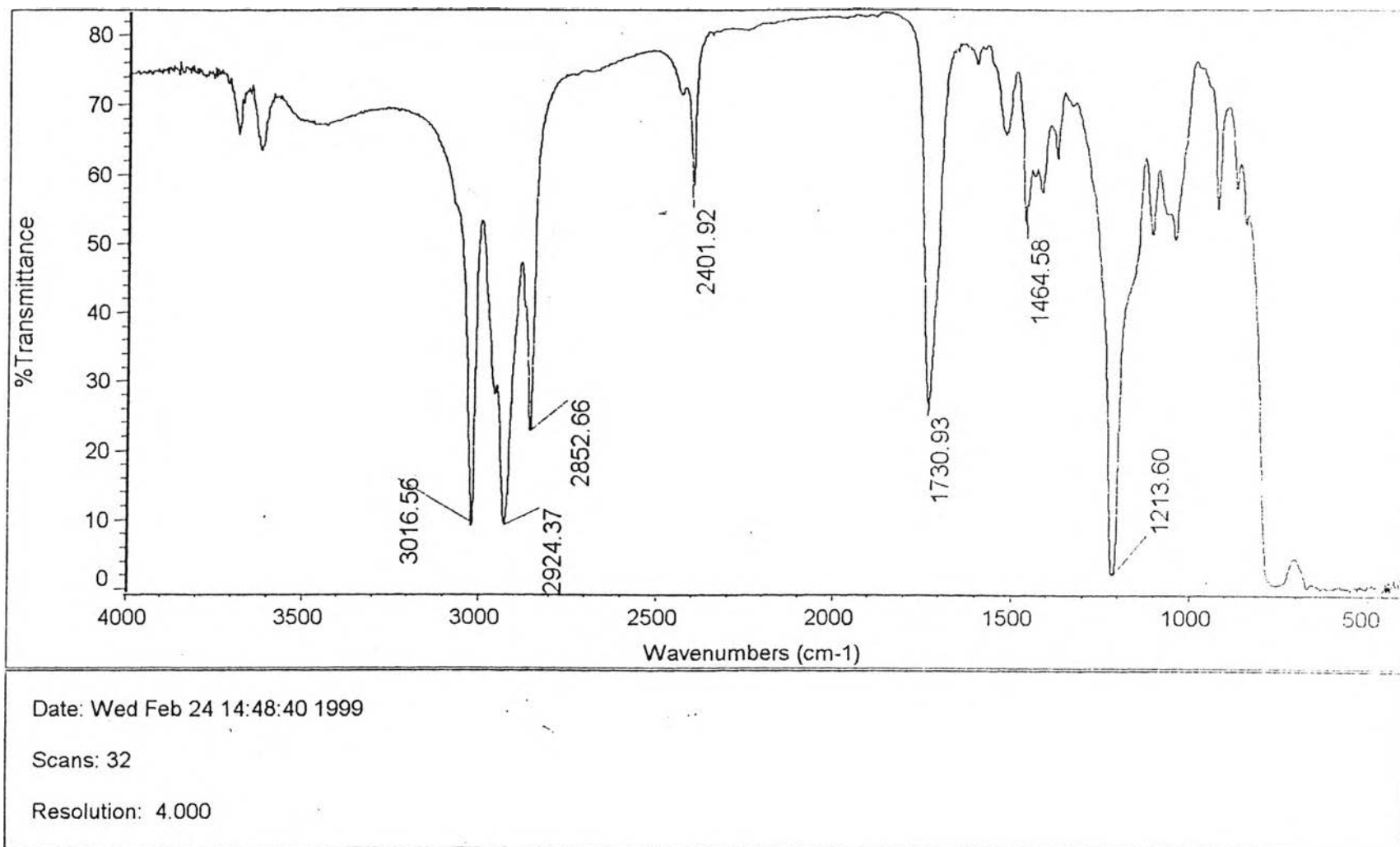
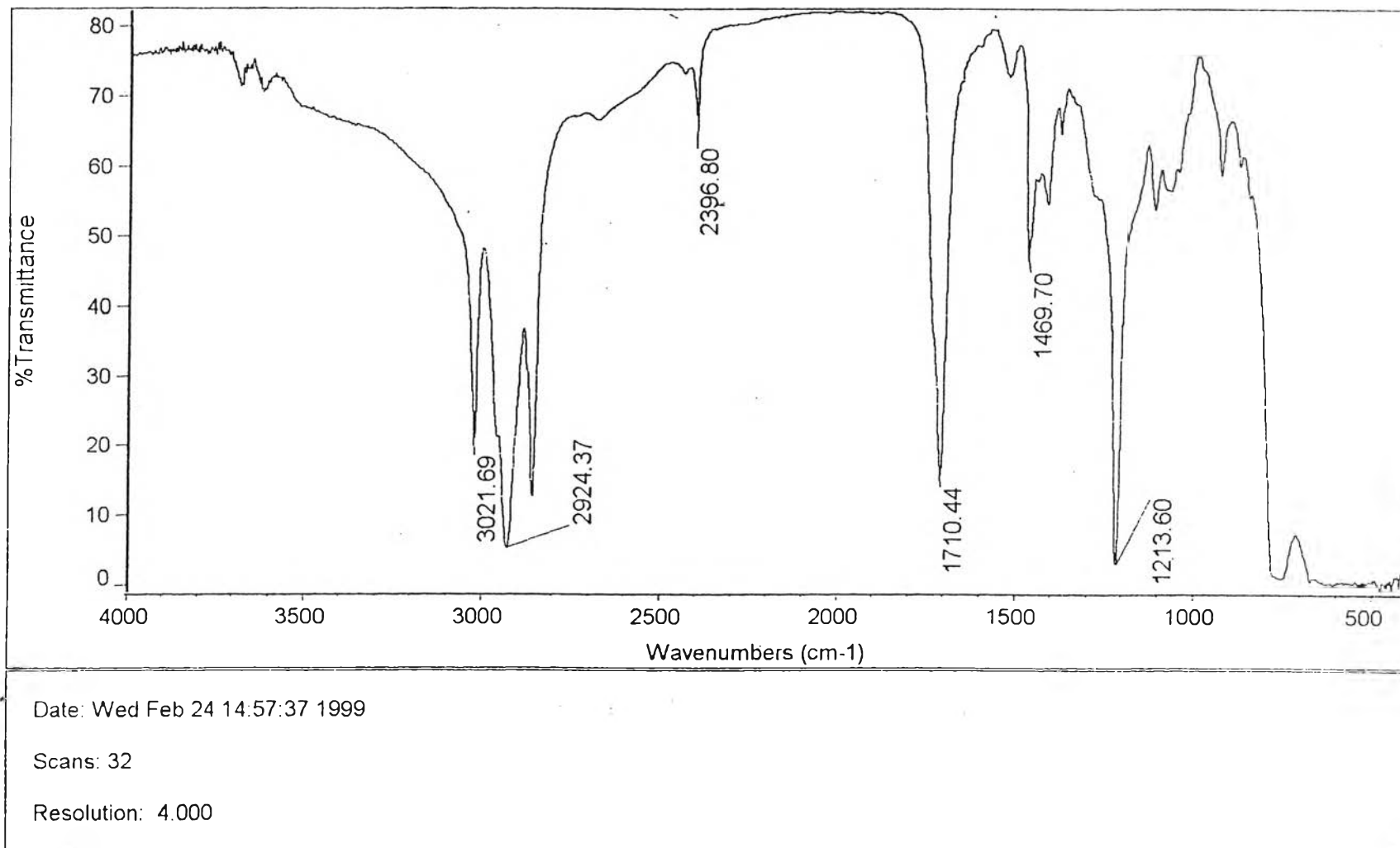


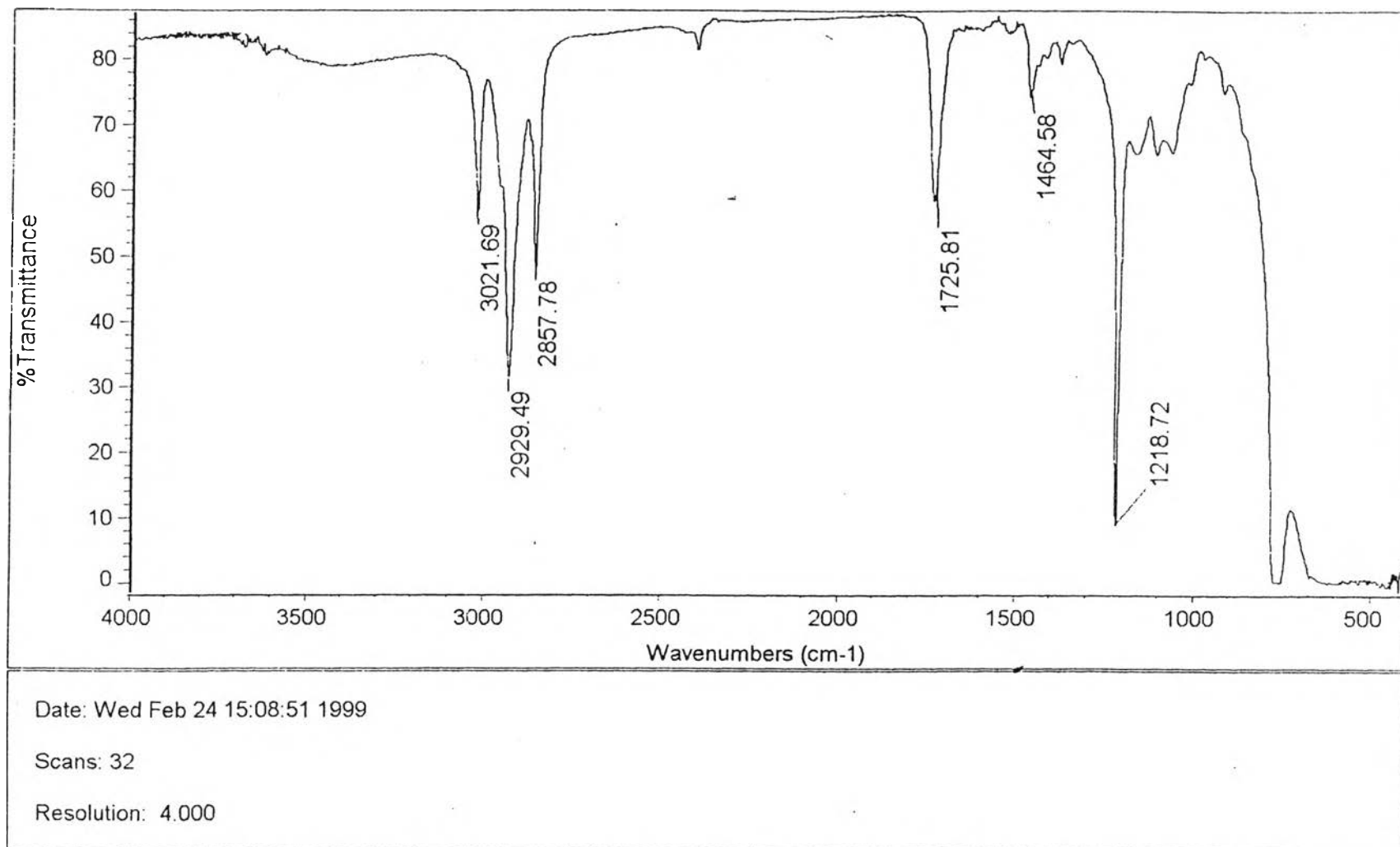
Figure B6 IR spectrum of stearic acid



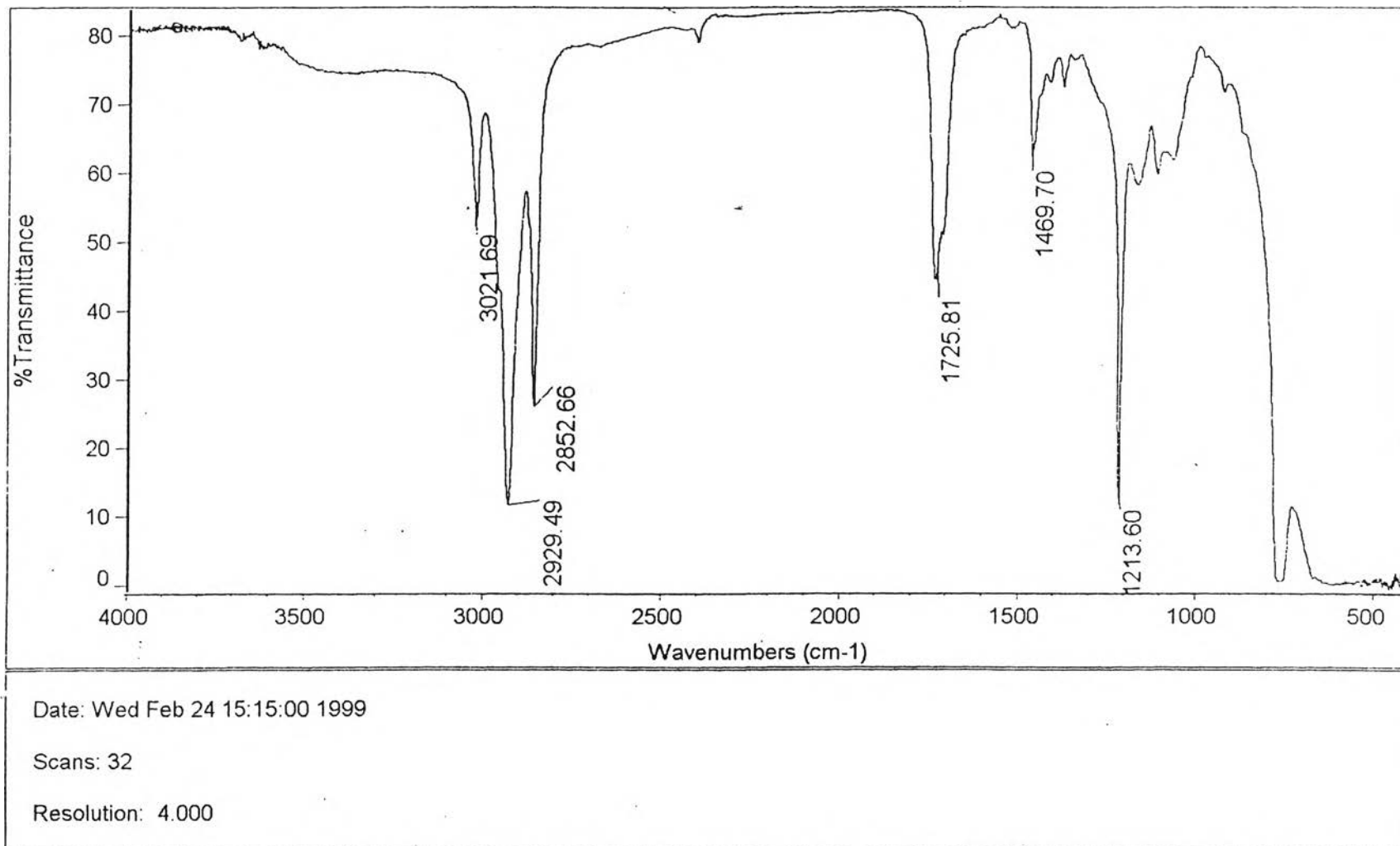
**Figure B7** IR spectrum of 1,2-ethanedilaurate



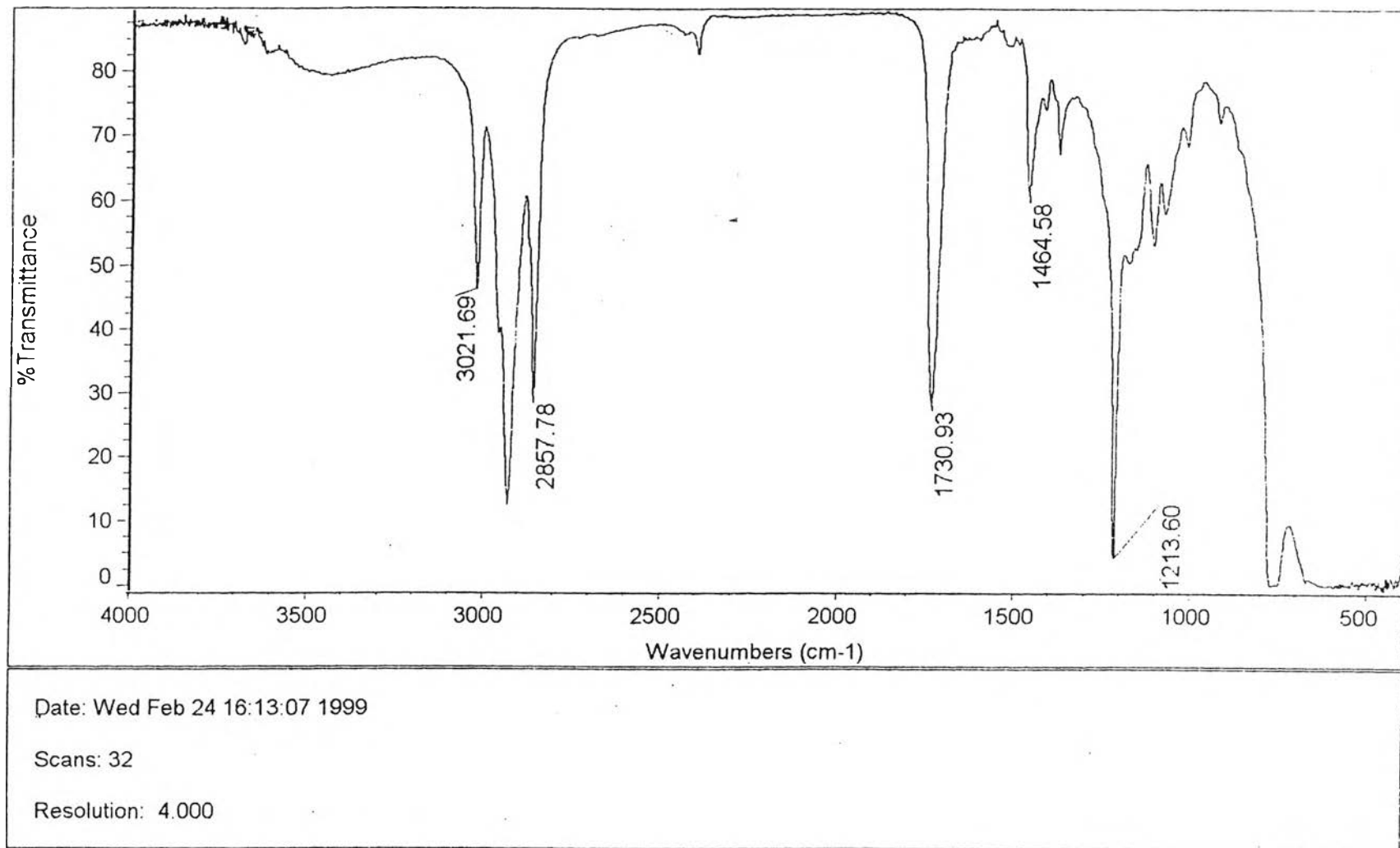
**Figure B8** IR spectrum of 1,2-ethanedimyristate



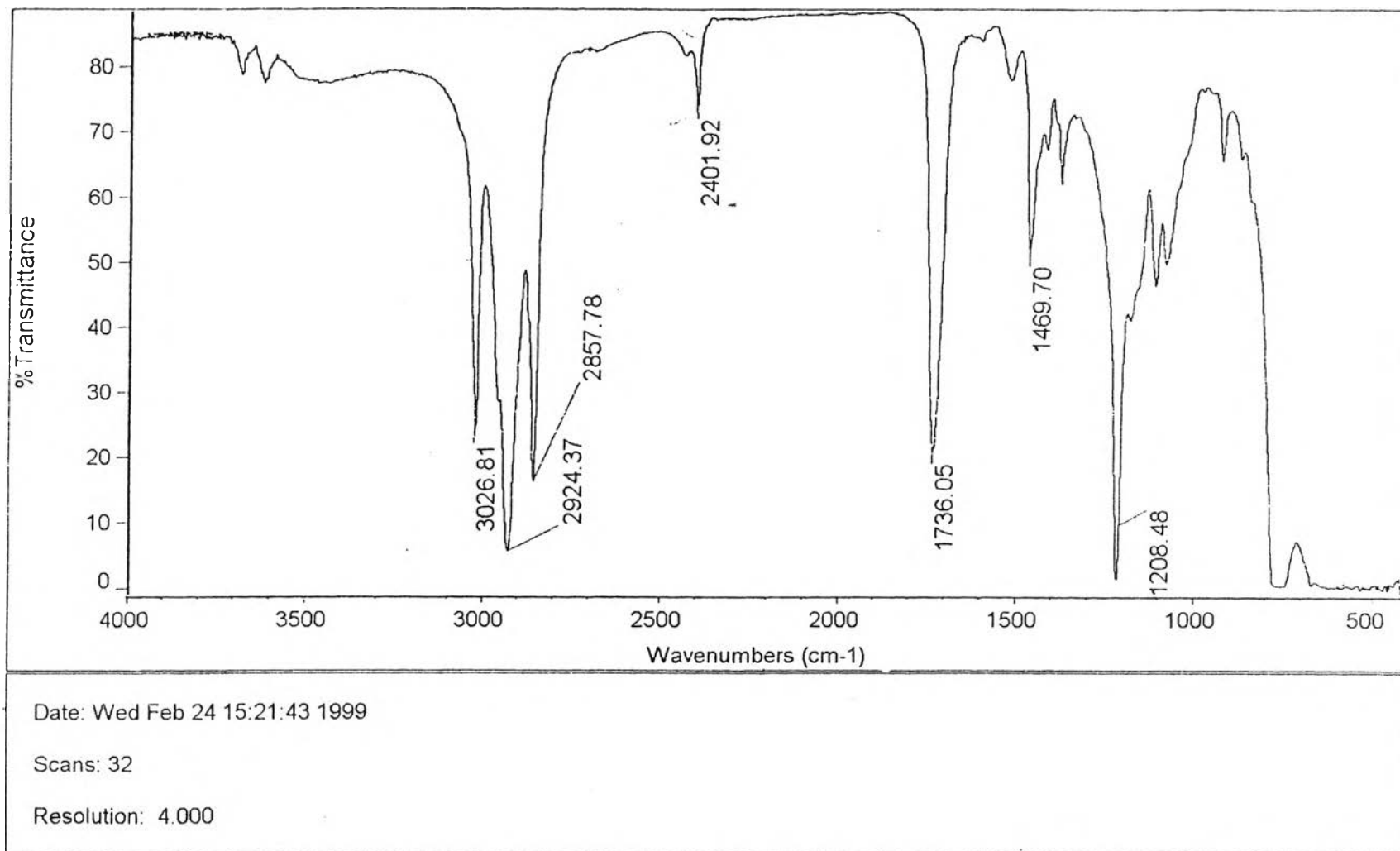
**Figure B9** IR spectrum of 1,2-ethanedipalmitate



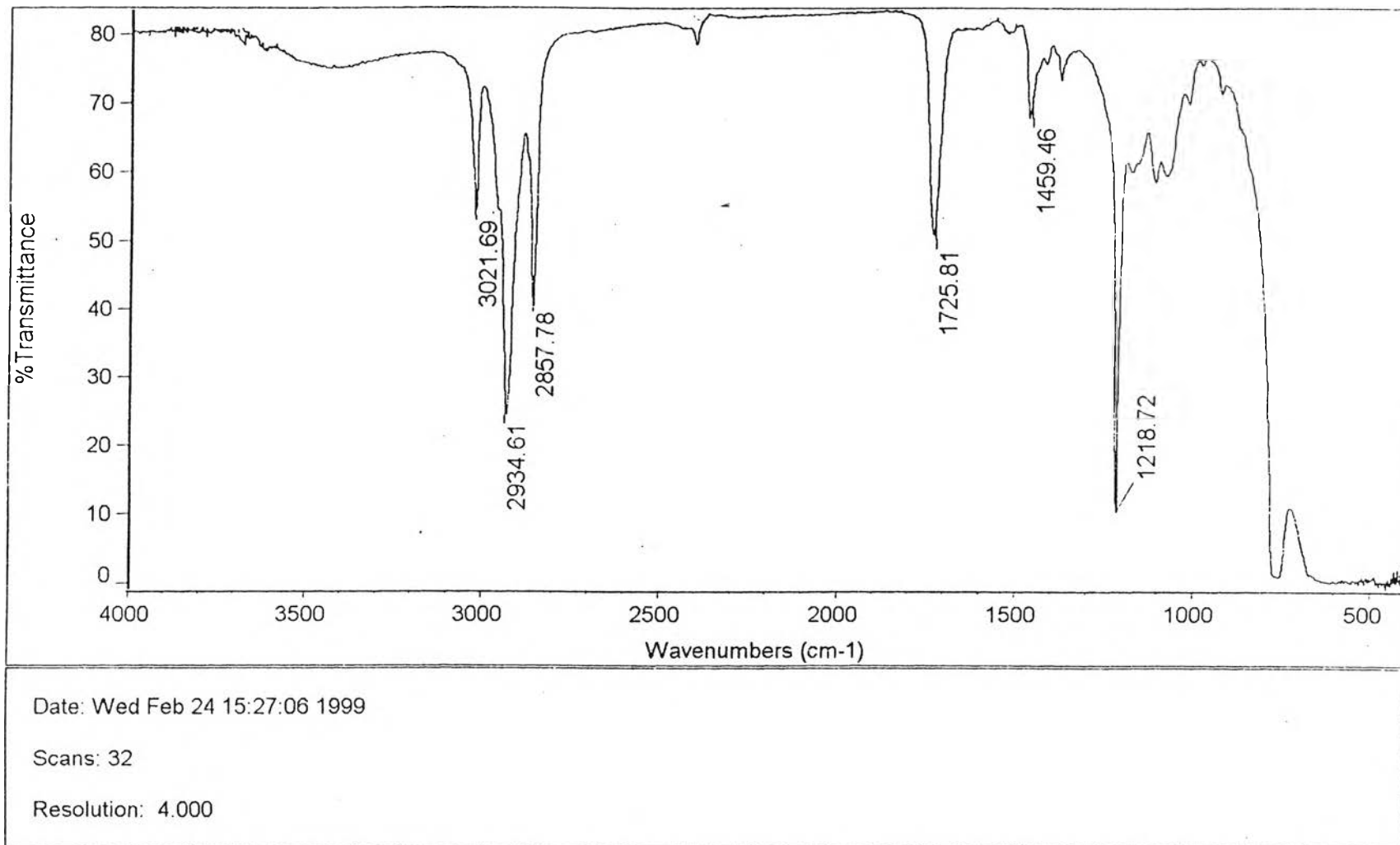
**Figure B10** IR spectrum of 1,2-ethanedistearate



**Figure B11** IR spectrum of 1,2-propanedilaurate

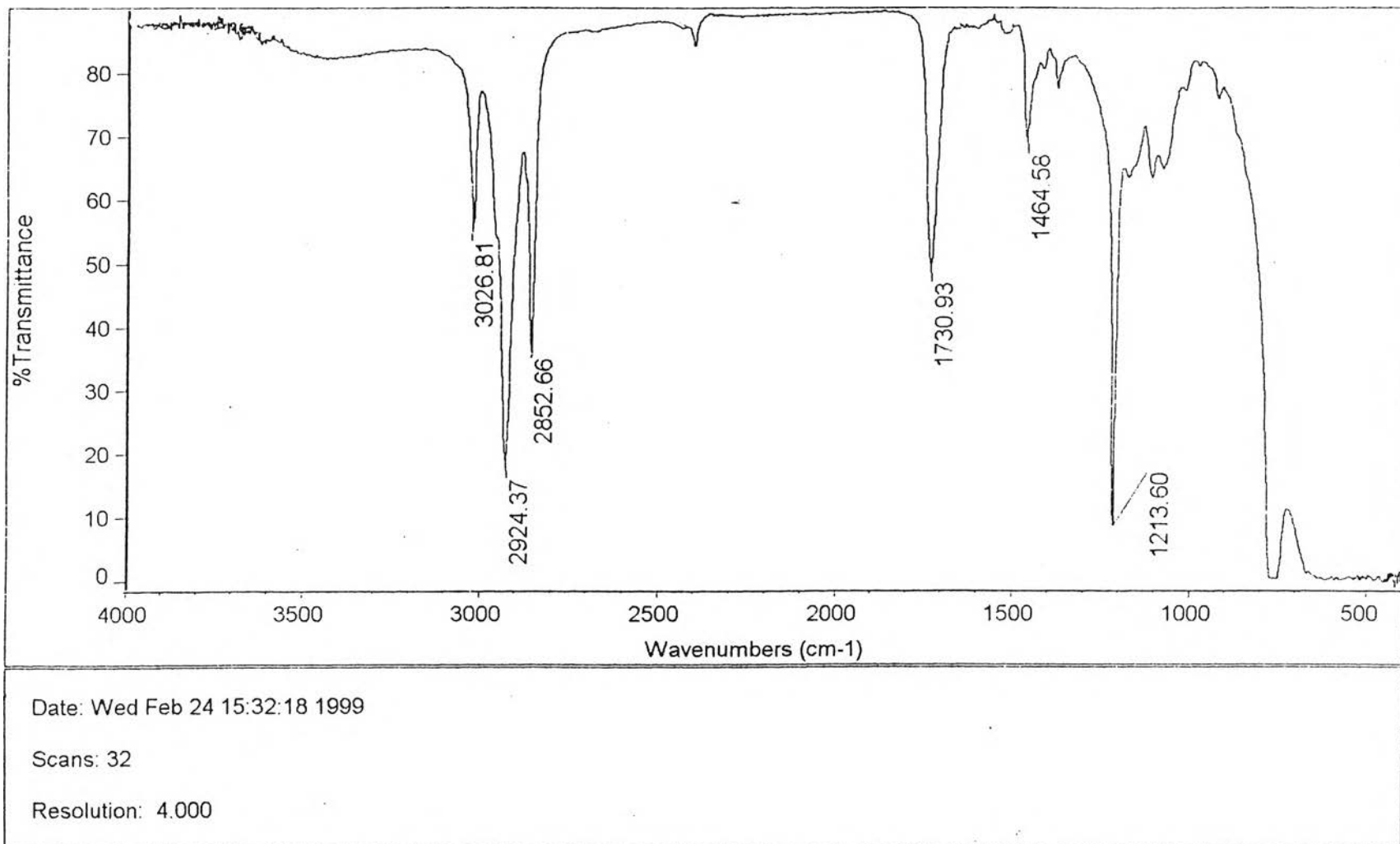


**Figure B12** IR spectrum of 1,2-propanedimyristate



**Figure B13** IR spectrum of 1,2-propanedipalmitate



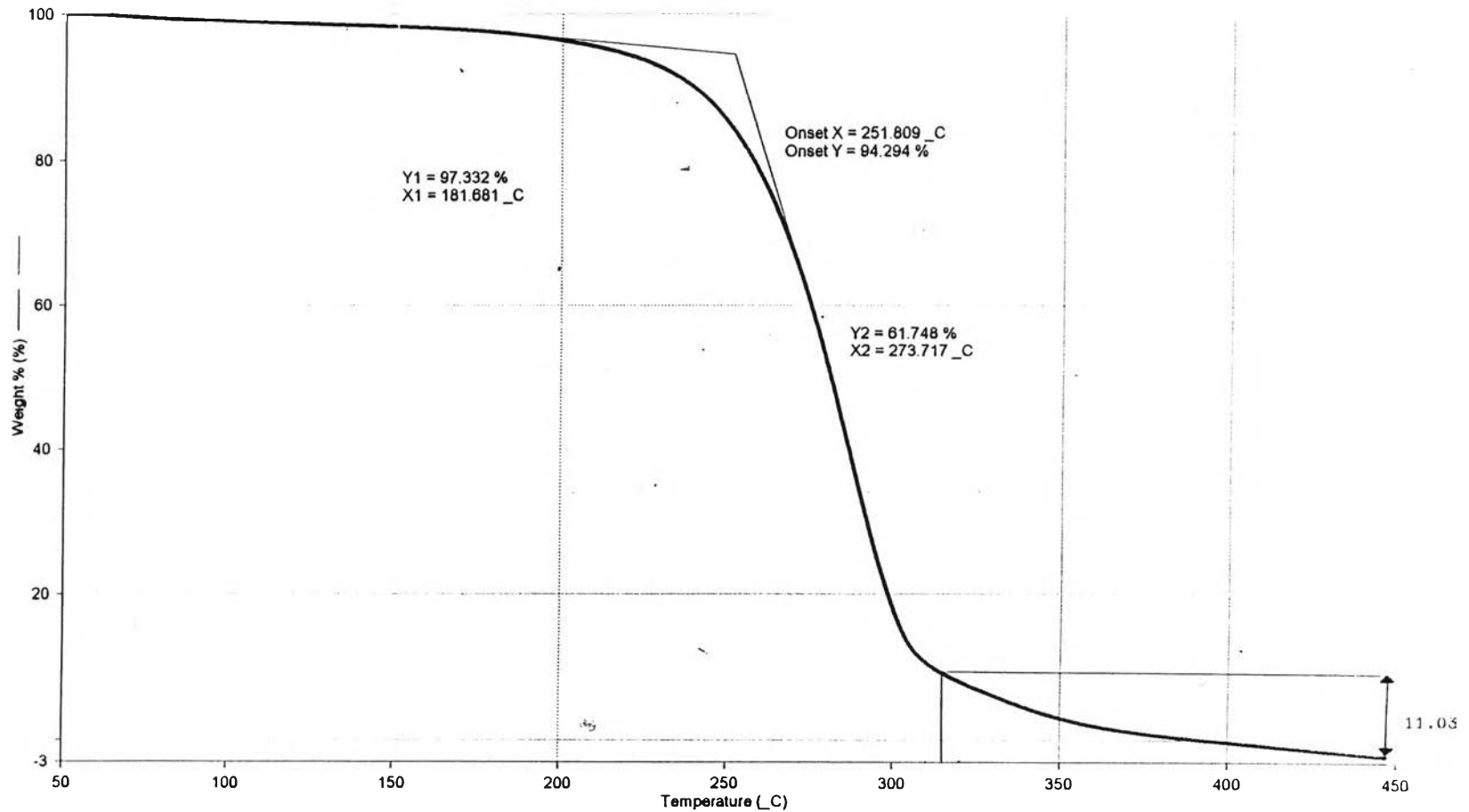


**Figure B14** IR spectrum of 1,2-propanedistearate

## APPENDIX C

Data Collected: 19/2/42 10:58:28  
Operator ID: khanit  
Sample ID: /e  
Sample Weight: 2.573 mg  
Total Points in Run: 1201  
Comment: Scanning rate 20C/min 40/10 by air

/e: le.tgd  
Unsubtracted Weight % (%) : Step: 1



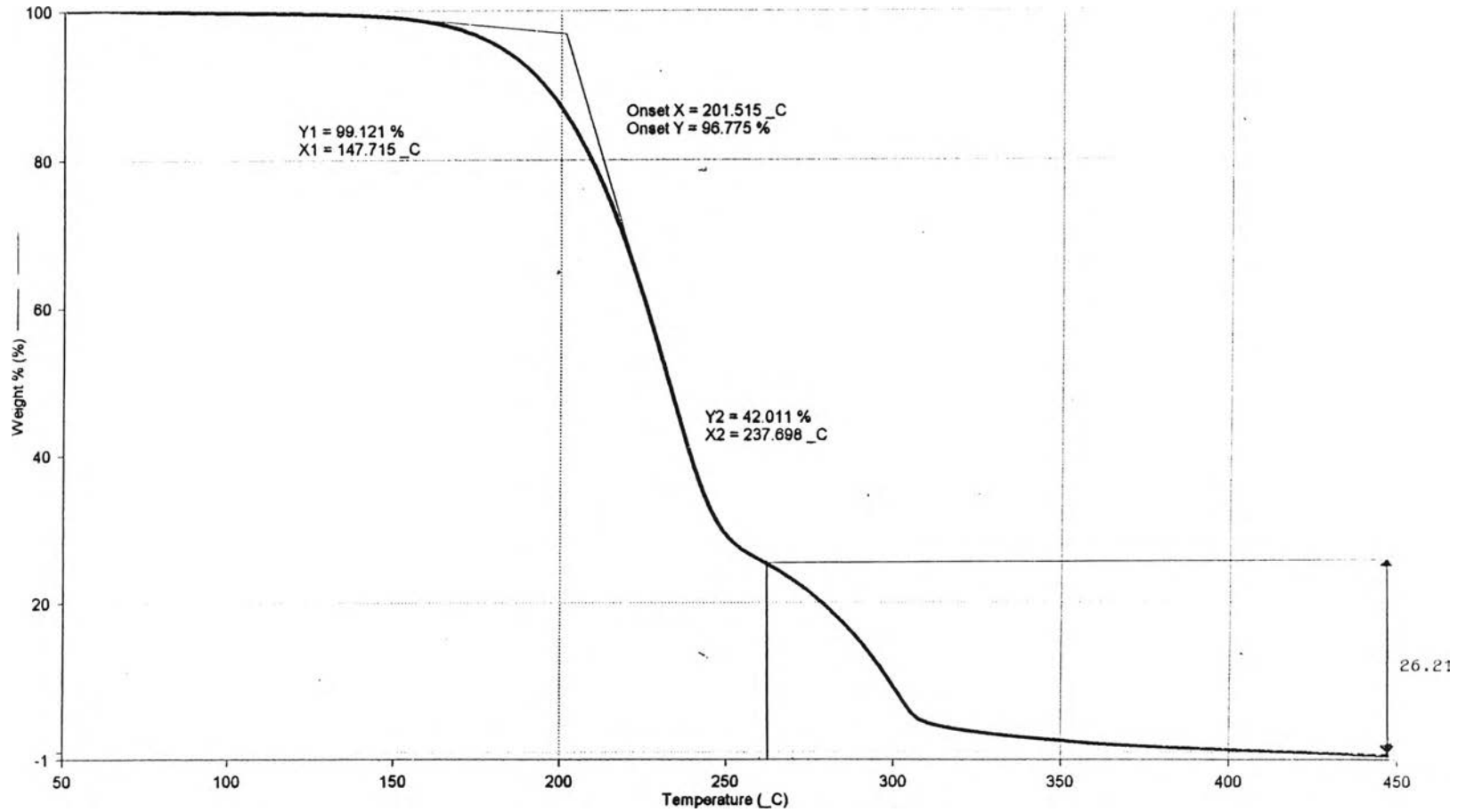
1) Heat from 50.00\_C to 450.00\_C at 20.00\_C/min

22/2/42 11:33:32

Figure C1 Thermogram of 1,2-ethanedilaurate

Data Collected: 19/2/42 11:37:31  
Operator ID: khanit  
Sample ID: m/e  
Sample Weight: 1.933 mg  
Total Points in Run: 1201  
Comment: Scanning rate 20C/min 40/10 by air

m/e: me.tgd  
Unsubtracted Weight % (%) : Step 1



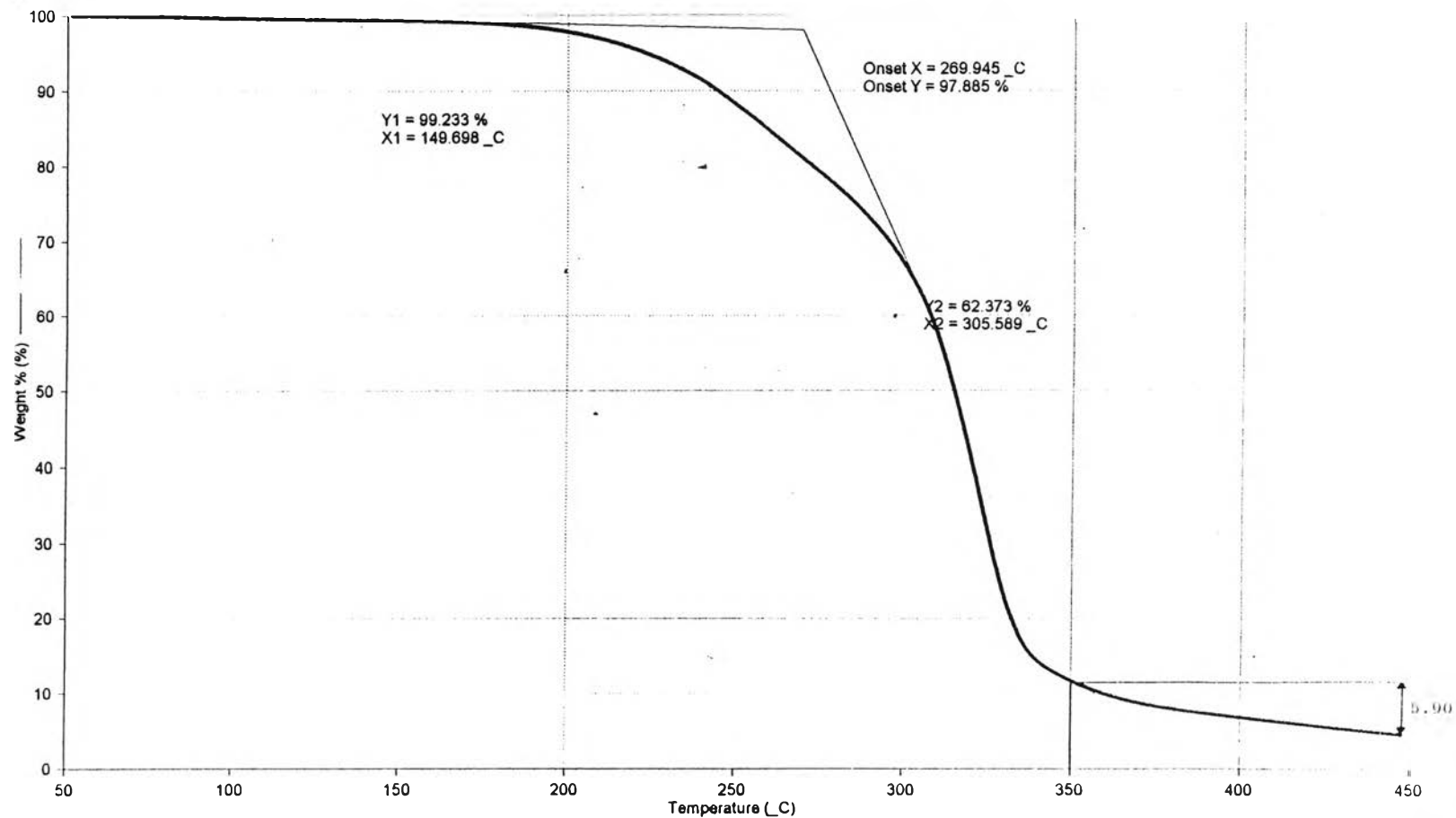
1) Heat from 50.00\_C to 450.00\_C at 20.00\_C/min

22/2/42 12:03:37

Figure C2 Thermogram of 1,2-ethanedimyristate

Data Collected: 19/2/42 12:25:34  
Operator ID: khanit  
Sample ID: p/e  
Sample Weight: 3.469 mg  
Total Points in Run: 1201  
Comment: Scanning rate 20C/min 40/10 by air

p/e: pe.tgd  
Unsubtracted Weight % (%) : Step: 1



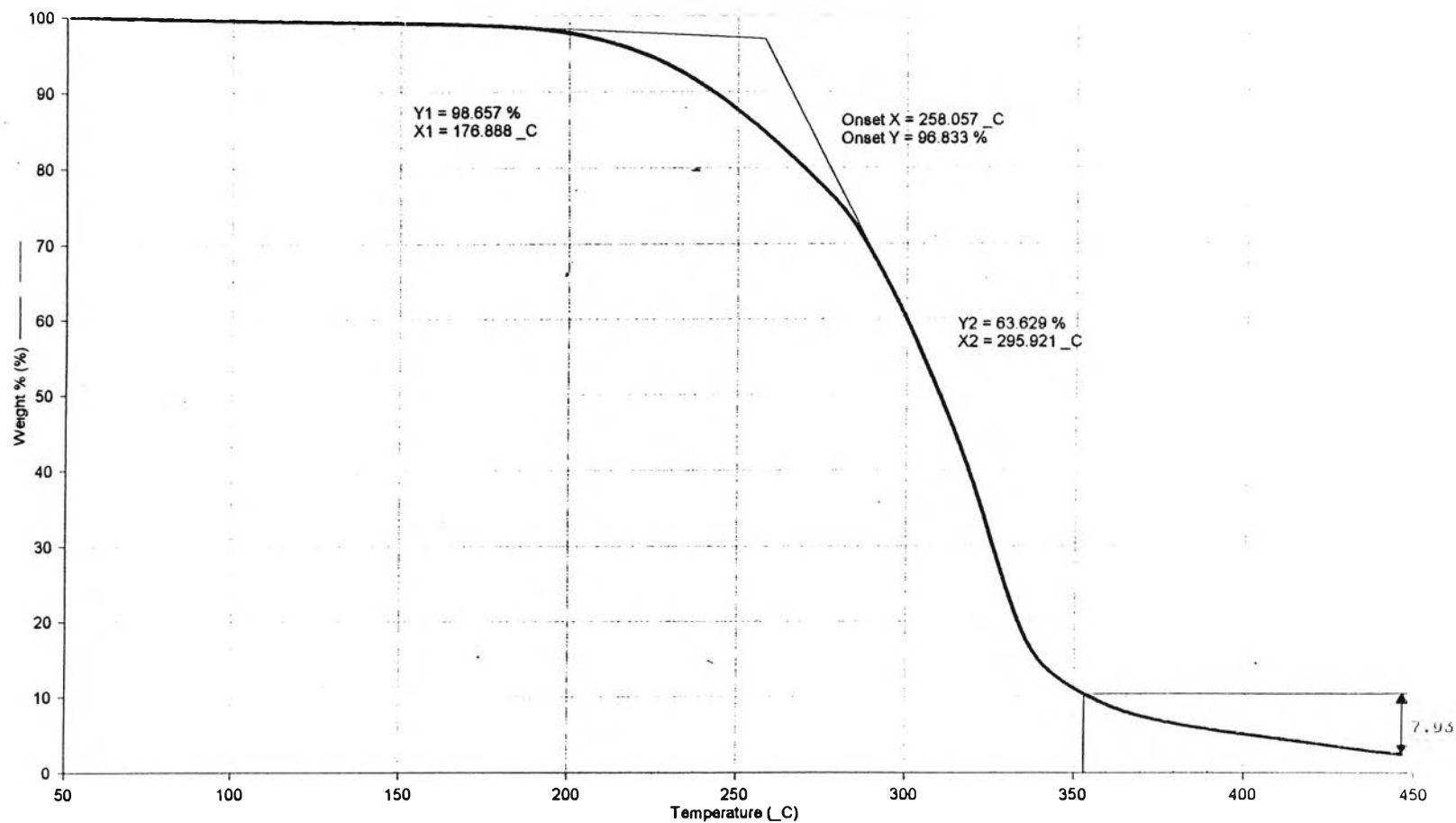
1) Heat from 50.00\_C to 450.00\_C at 20.00\_C/min

22/2/42 11 11 50

Figure C3 Thermogram of 1,2-ethanedipalmitate

Data Collected: 19/2/42 13:09:20  
Operator ID: kharit  
Sample ID: a/e  
Sample Weight: 2.386 mg  
Total Points in Run: 1201  
Comment: Scanning rate 20C/min 40/10 by air

a/e: se tgd  
Unsubtracted Weight % (%): Step: 1



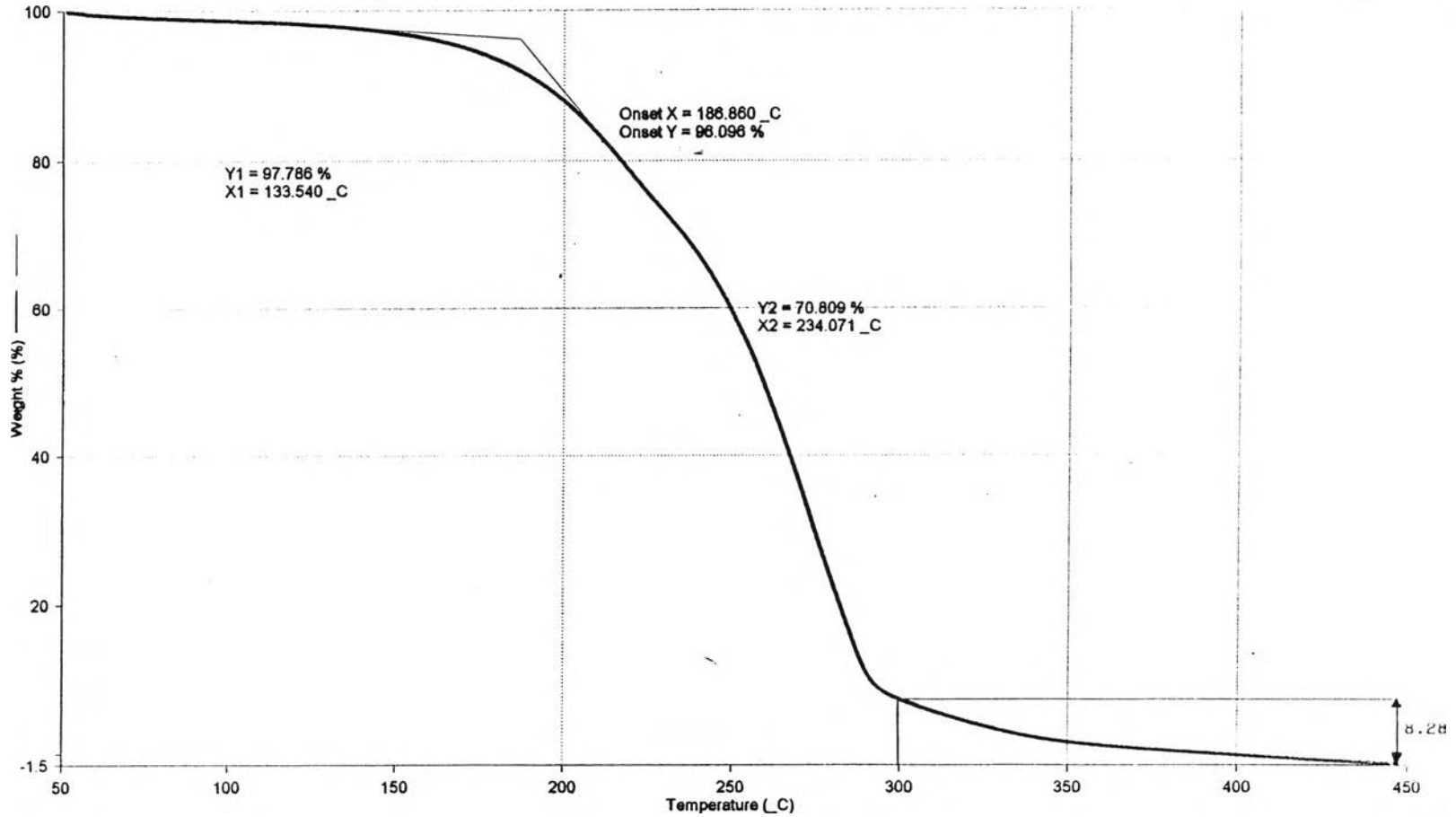
1) Heat from 50.00\_C to 450.00\_C at 20.00\_C/min

22/2/42 11:13:57

Figure C4 Thermogram of 1,2-ethanedistearate

Data Collected: 19/2/42 13:49:20  
Operator ID: khanit  
Sample ID: lp  
Sample Weight: 2.318 mg  
Total Points in Run: 1201  
Comment: Scanning rate 20C/min 40/10 by air

lp: lp.tgd  
Unsubtracted Weight % (%) : Step: 1



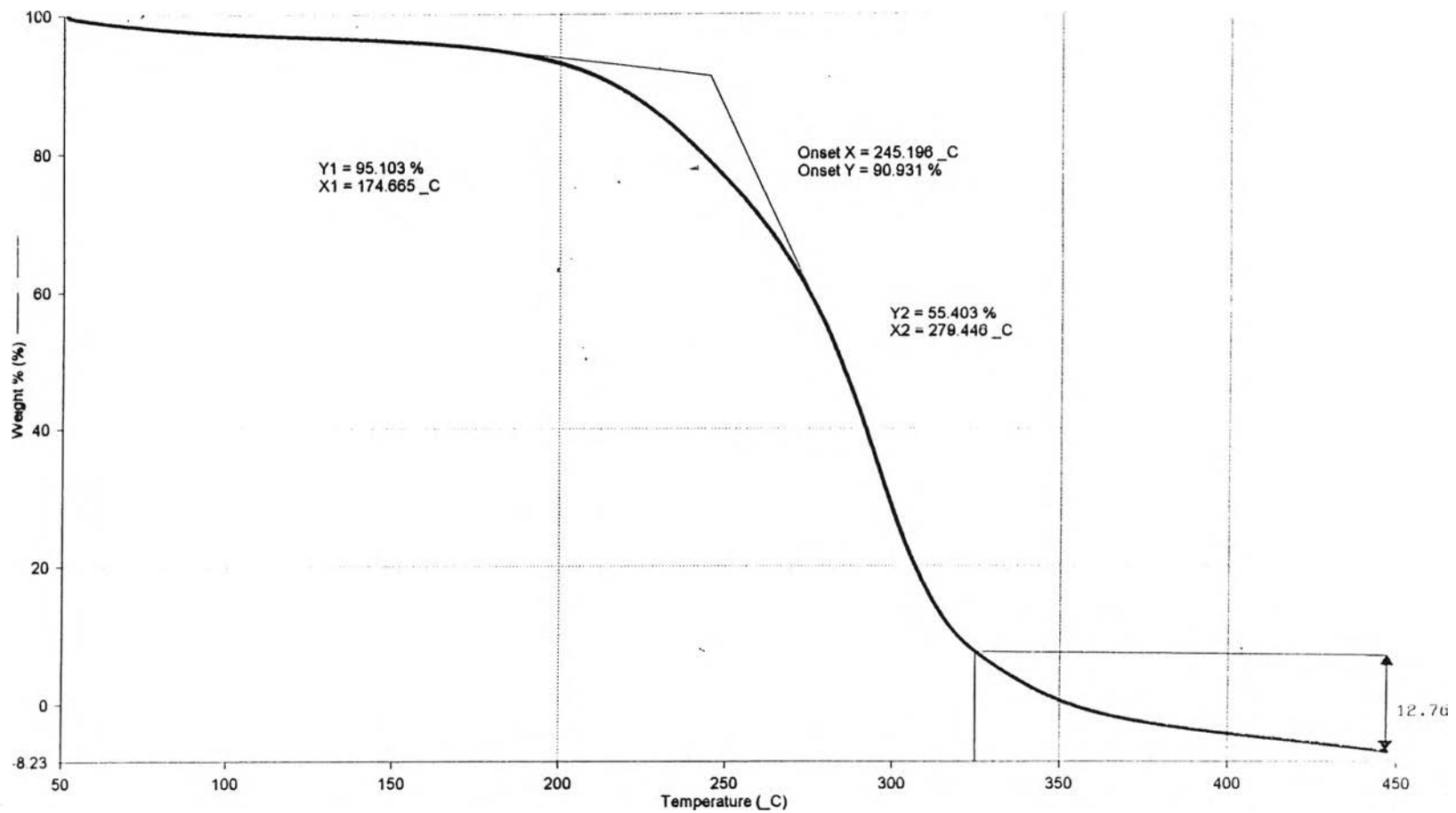
1) Heat from 50.00\_C to 450.00\_C at 20.00\_C/min

22/2/42 12:48:46

Figure C5 Thermogram of 1,2-propanedilaurate

Data Collected: 19/2/42 15:59:27  
Operator ID: khanit  
Sample ID: m/p  
Sample Weight: 2 588 mg  
Total Points in Run: 1201  
Comment: Scanning rate 20C/min 40/10 by air

m/p: mp.tgd  
Unsubtracted Weight % (%) : Step: 1



1) Heat from 50.00\_C to 450.00\_C at 20.00\_C/min

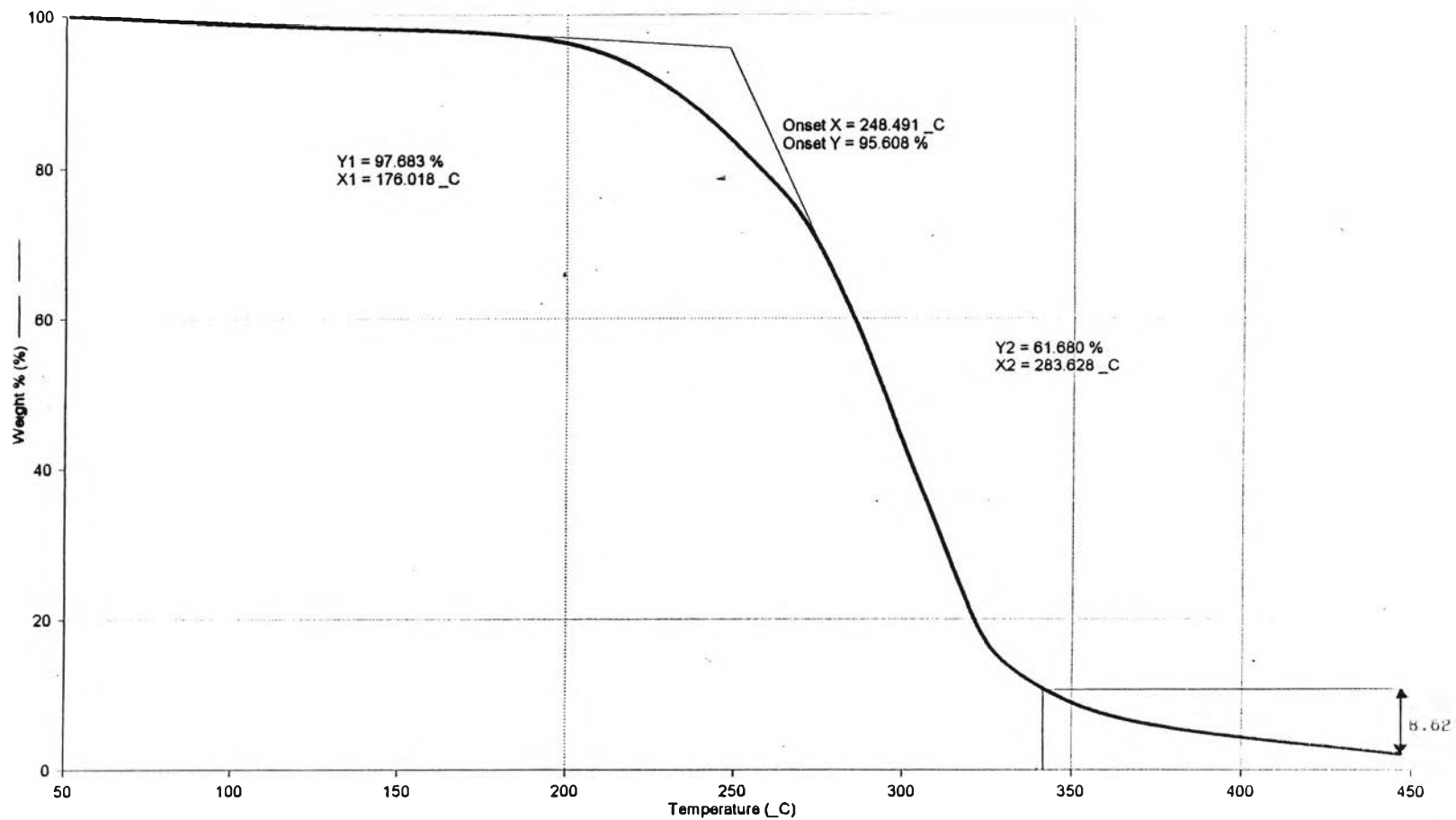
22/2/42 11 25 07

Figure C6 Thermogram of 1,2-propanedimyrystate



Data Collected: 19/2/42 18:41:29  
Operator ID: khanit  
Sample ID: p/p  
Sample Weight: 2.286 mg  
Total Points in Run: 1201  
Comment: Scanning rate 20C/min 40/10 by air

p/p: pp.tgd  
Unsubtracted Weight % (Step: 1)



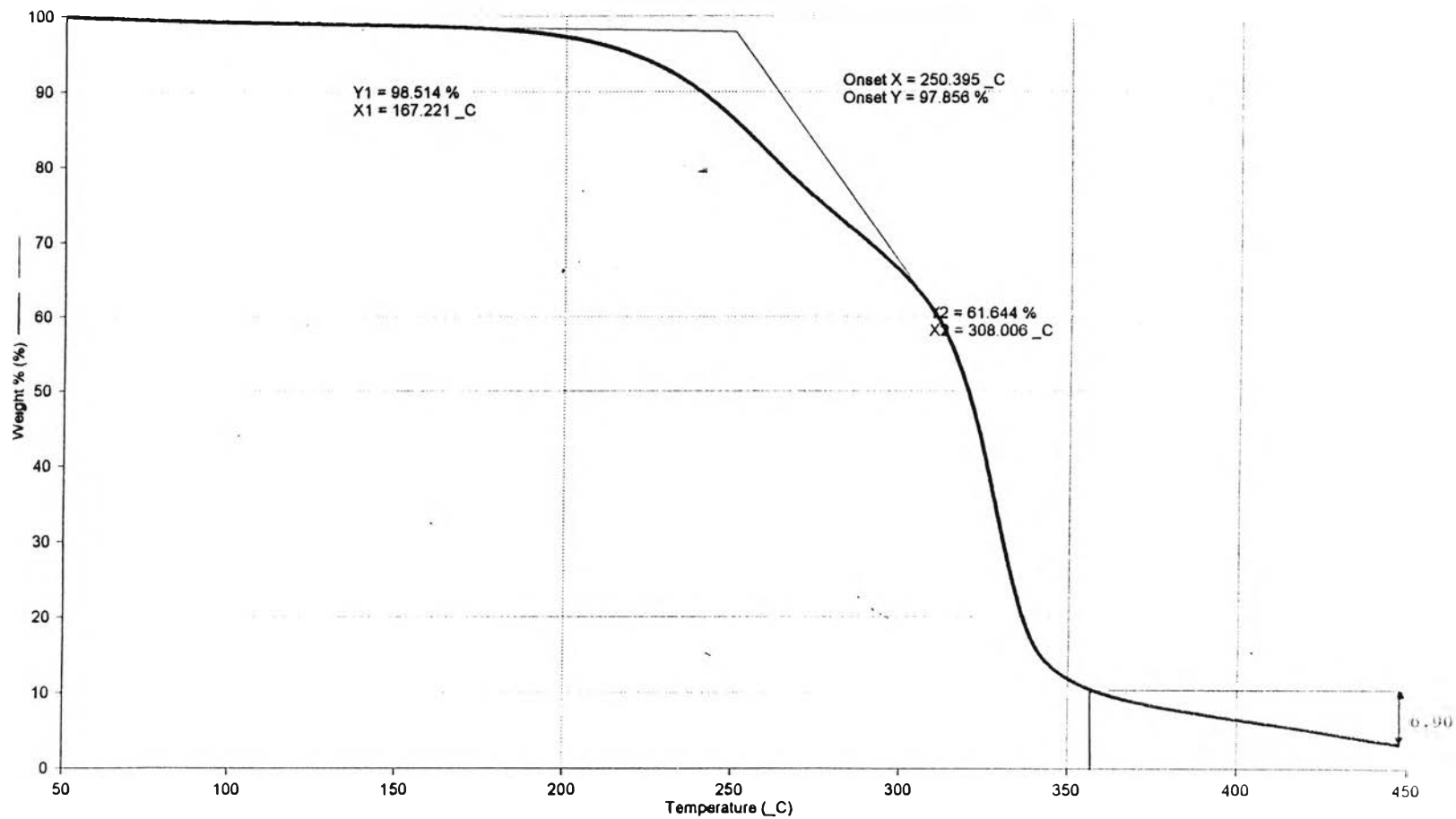
1) Heat from 50.00\_C to 450.00\_C at 20.00\_C/min

22/2/42 11:29:18

Figure C7 Thermogram of 1,2-propanedipalmitate

Data Collected: 22/2/42 9:48:54  
Operator ID:  
Sample ID: s/p  
Sample Weight: 3.378 mg  
Total Points in Run: 1201  
Comment: Air zero 40/10psi/psi

s/p: sp.tgd  
Unsubtracted Weight % (%) : Step: 1



1) Heat from 50.00\_C to 450.00\_C at 20.00\_C/min

22/2/42 11 06 55

Figure C8 Thermogram of 1,2-propanedistearate

## APPENDIX D

### *Antifoaming agent* [8, 23]

Foaming, the formation of a stable dispersion of a gas in a liquid, is common occurrence in chemical industries. Which undesirable foam can be a costly problem that reduces plant capacity, causes environmental problems and safety hazards, and raw materials. Chemical additives to reduce foam problem are called antifoam agents, foam inhibitors, foam control agents, foam suppressants and air release agent. Antifoaming agents have many properties in common. Most antifoaming agents are surface - active materials, have hydrophobic - hydrophilic characteristics within the same molecule and have very limited solubility in the bulk liquid to be defoamed. Table D1 demonstrate the foam destabilizers and field of applications.

Table D1 Antifoaming agents [24]

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
Alcohols	2-Ethyl-hexanol (octyl alcohol)	Carbide and Carbon Co.	Beet sugar, paper textile printing, glue spreading	0.005-1%
	Dodecanol		Fermentation	0.8 cm <sup>3</sup> /ft <sup>2</sup>
	Dodecanol and tetradecanol or hexadecanol (1:1)		Fermentation	
	Di-isobutyl carbinol (nonyl alcohol)	Carbide and Carbon Co.	Neutralization of waste H <sub>2</sub> SO <sub>4</sub> with CaCO <sub>3</sub> , paper, printing ink, glue wire drawing (controls the foaming of a soap solution used as a lubricant)	0.5%

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
Fatty acids and fatty acid ester	Polyalkylene glycols and derivatives (Ucon. 50-HB-660, 50-HB-3520, 50-HB-5100, LB-525, LB-1145)	Carbide and Carbon Co.	Antifoaming in hot aqueous solution, suspended solids, petroleum oils, and in other applications	
	Nopco Vegifat Y composed of a mixture of long- and short-chain fatty acids	Nopco	Fermentation, yeast manufacturing, production of penicillin	
	Span 20 (sorbitan monoluarate)	Atlas Powder Co.	Phosphoric acid production	20-40 cm <sup>3</sup> in 1300 lb of acid
	Span 20 (sorbitan monoluarate)	Atlas Powder Co.	Evaporation of aqueous milk sugar solution Speed up the drying of egg white	0.05% 0.1%

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
			Evaporation step in the concentration of molasses	0.002%
	Span 85 (sorbitan trioleate)	Atlas Powder Co.	Yeast manufacture	1 pint of Span 85 in 1000 gal of yeast slurry
			Cooking of soybean casein	5 ppm
			Cooking, evaporation of glue	0.0025%
	Tween 20 (polyoxyethylene-sorbitan monolaurate)	Atlas Powder Co.	Food manufacture	
	Nopco KF <sup>27</sup> (a mixture of wax and partial glycerides)	Nopco	Paper, textile	0.1% (on dry weight of the wood pulp)

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
Amines	Nopco 1907	Nopco	Animal glue	0.05-2%
	Isoamyl stearate and diglycol laurate	Glyco Product Co.	Gas cutting during the drilling of deep wells Reduce the stability of bubbles	
	Fatty acid esters of glycols and polymerized glycols	National	Glue solution paper, Fermentation, starch pastes	0.05-0.1%
	Alkyl ester of sulfonated ricinoleic acid	Mid-Continent Petroleum Co.	Lubricating oils	
	Diamylamine	Dearborn	Aqueous solution of wetting agents such as Gardinol W.A. (a mixture of alkyl sulfates of aliphatic alcohols C <sub>12</sub> - C <sub>18</sub> )	0.02%

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
Amides	Di-and trialkyl amines (C <sub>8</sub> in each alkyl group)	Dearborn	Dye bath, dye pastes, discharge pastes, textiles	1-2%
	Palmitylamino-octadecyl-caproamide and palmitylaminohexadecyl-P-aminophenylacetamide	Imperial	Boiler frothing	<1 g/gal
	Distearoylethylene-diamide: C <sub>17</sub> H <sub>35</sub> CONH — CH <sub>2</sub> — NHCOC <sub>17</sub> H <sub>35</sub>	Dearborn	Boiler water	
	Anti-foam Nos. 666 and 659 (polyamides)	Dearborn	Power plant boiler application	
	Formula No. 662 (low melting point polyamide)	Dearborn	Textile, paper detergency applications, automotive cooling	



Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
	Cationic surface-active agents: Oleic acid with diethylene triamine Oleic acid with ethylene diamine Rice oil with diethylene triamine Soybean oil with diethylene triamine	Nation	Antifoaming agents	
Ethers	Foamicide L, Foamicide A (ditertiary amyl-phenoxyethanol) in two preparations	Wyandotte	Washing or cleaning machines, textile plants, mercerization baths, sizing, kier boiling and dyeing operations, pulp and paper plants, glue and casein plants, chemical and dyestuff	0.025% starting amount

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
			plants, paper coating mills, power plant boilers, cannery wastes, tanneries, sewage disposal plants	
	Polymerized triethanolamine	Petrolite Co	Aqueous hypochlorite bleaching solution	25 cm <sup>3</sup> /8000 gal
			Contaminated diesel fuel	10 gal/ 1000 barrel tank
	Alkaterge O (oxazoline, from oleic acid and 2-amino-2-methyl-1,3-pentanediol)	Commercial Solvents Co.	Fermentation	
	Alkaterge C (substituted oxazoline)		Fermentation	Mixture of 5 gal each of Alkaterge C and a white mineral oil, inject this mixture into

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
Phosphate esters	Tributyl phosphate	Commercial Solvents Co.	Manufacture and use of adhesives, paper, casein solutions, ink, rubber latex, textiles	a 10,000-gal tank 100-1000 ppm
	Trioctyl phosphate (Flexol plasticizer ToF)	Carbide and Carbon Co.	Alkaline cleaning baths of bottle washers	
	Sodium octyl phosphate	Victor	Industrial cleaners	
	Dimethylaniline	Gulf Oil Co.	Mineral oils	0.01-0.04% in a highly viscous, refined Pennsylvania oil
	Potassium trioctylethylene diphosphate	Gulf Oil Co.	Lubricating oil	
	Sterox	Monsanto Chemical Co.	Automatic washing machine	

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
Metallic soaps of fatty acids	Renex (polyoxyethylene esters of mixed fatty acid and resin)	Atlas Powder Co.	Automatic washing machine	18% of the alkyl aryl sulfonate
	Stearates and palmitates of Al Ca, Mg, Zn	National Oil Co.	Paper making	5%
	A mixture of 10-20 parts of polyethylene glycol mono-ester, 100 parts of H <sub>2</sub> O, and 40 parts of metallic soap	Nopco	Paper coating Paper making, glue solution	1%
	Nopco 1333, Nopco 1497, Nopco 1497B, (aluminum stearate)	Nopco	Animal glue, textile warp sizing, preparation of defoamed glue stock	0.5%
	Alkali-metal soap	West Virginia Pulp and Paper Co.	Paper stock	

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
Silicones and other silicon materials	Calcium salt of wool olein	Anglo-Saxon Petroleum Co.	Lubrication oil	0.01-0.10%
	Copper palmitate, lead palmitate		Lubricating oil in aircraft engines	
	Potassium oleate	Gulf Oil Co.	Lubricating oil	0.5%
	Soaps of alkali metals	Industrial Patents Co.	Deep frying	0.0001-0.5%
	Polysilixanes	Gulf Research & Development Co.	Lubricating oils	0.5 ppm
	Polyalkoxysiloxanes	Shell Development Co.	Lubricating oils	
	Diethyltrisiloxane		Aqueous system oil-in-water emulsions	
	Dimethylpolysiloxane		Hydrocarbon oil, water-in-oil emulsions	
Dc antifoam A	Dow-Corning Co.	Adhesives, distillation of alcohols, beet sugar, chewing gum, skim milk, soft drink tallow,	1 part in 6 millions parts of asphalt	

Table D1 (Continued)

Compound Category	Chemical Compound	Manufacturer	Field of Application	Concentration
			yeast, wine, perfume solubilizers, resins, varnishes, lacquers, soaps, wetting agent, pharmaceuticals	
	Anti Foam 81066 G.E.		Solution of soaps and wetting agent	25-100 ppm
			Phenolic resins	100-500 ppm
			Paints	25-100 ppm
			Distillation, fermenta- tion, paper, rubber latices	

## VITA

Miss Khanit Panchoowong was born on February 20, 1974 in Rayong, Thailand. She received her Bachelor of Science degree in Chemistry, Faculty of Science, Thammasat University in 1995. She began her Master study at Petrochemistry and Polymer Science, Graduate School, Chulalongkorn University, in 1996 and completed the program in 1999.

