REFERENCES

- Rudoy, V. M.; Dement'eva, O. V.; Yaminskii, I. V.; Sukhov, V. M.; Kartseva, M. E.; Ogarev, V. A. Metal Nanoparticles on Polymer Surfaces: 1. A New Method of Determining Glass Transition Temperature of the Surface Layer. Colloid J. 64 (2002): 746-754.
- 2. Chan, E. W. L.; Yu, L. Chemoselective Immobilization of Gold Nanoparticles onto Self-Assembled Monolayers. <u>Langmuir</u> 18 (2002): 311-313.
- 3. Yong Yang, Mamiko Hori, Tomokatsu Hayakawa and Masayuki Nogami. Self-assembled 3-dimensional arrays of Au@SiO₂ core-shell nanoparticles for enhanced optical nonlinearities. <u>Surface Science</u> 579 (2005): 215–224.
- Tilley, R. D.; Saito, S. Preparation of Large Scale Monolayers of Gold
 Nanoparticles on Modified Silicon Substrates Using a Controlled Pulling
 Methodology. <u>Langmuir</u> 19 (2003): 5115-5120.
- 5. Bryce Sadtler and Alexander Wei. Spherical ensembles of gold nanoparticles on silica: electrostatic and size effects. <u>Chem. Commun.</u> (2002): 1604-1605.
- 6. Zanella, R.; Giorgio, S.; Henry, C. R.; Louis, C. Alternative Methods for the Preparation of Gold Nanoparticles Supported on TiO₂. <u>J. Phys. Chem. B</u> 106 (2002): 7634-7642.
- 7. Schaaff, T. G.; Blom, D. A. Deposition of Au-Nanocrystals on TiO₂ Crystallites.

 Nano Lett. 2 (2002): 507-511.
- 8. Hulteen, J. C.; Patrissi, C. J.; Miner, D. L.; Crosthwait, E. R.; Oberhauser, E. B.; Martin, C. R. Changes in the Shape and Optical Properties of Gold Nanoparticles Contained Within Alumina Membranes Due to Low-Temperature Annealing. J. Phys. Chem. B 101 (1997): 7727-7731.

- Yonezawa, T.; Onoue, S-y.; Kunitake, T. Growth of Closely Packed Layers of Gold Nanoparticles on an Aligned Ammonium Surface. <u>Adv. Mater.</u> 10 (1998): 414-416.
- Shon, Y.-S.; Choo, H. [60]Fullerene-Linked Gold Nanoparticles: Synthesis and Layer-by-Layer Growth on a Solid Surface. <u>Chem. Commun.</u> (2002): 2560-2561.
- 11. Liu, L.; Wang, T.; Li, J.; Guo, Z.-X.; Dai, L.; Zhang, D.; Zhu, D. Self-Assembly of Gold Nanoparticles to Carbon Nanotubes Using a Thiol-Terminated Pyrene as Interlinker. Chem. Phys. Lett. 22 (2003): 747-752.
- 12. Box ..., H.-G.; Herzog, Th.; Kaqrle, G.; Weigl, F.; Zieman., Ya-Ray

 Photoelectron Spectroscopy Study on Gold Nanoparticles Sup, Fred on
 Diamond. Phys. Rev. B 65 (2002): 0754121-.0754125.
- 13. J. Dutta, H. Hofmann, Self-organization of colloidal nanoparticles in: H.S. Nalwa (Ed.), Encyclopedia of Nanoscience and Nanotechnology, <u>American Scientific</u>
 Publisher, California, CA, (2004): pp. 617–640.
- 14. M.S. Chen, D.W. Goodman. The structure of catalytically active gold on titania, Science 306 (2004): 252–255.
- 15. Pulker, H.K. Coating on Glass. Elsevier 1987
- Alexandre Dokoutchaev, J. Thomas James, Shannon C. Koene, Srikant Pathak, G.
 K. Surya Prakash, and Mark E. Thompson. Colloidal Metal Deposition onto Functionalized Polystyrene Microspheres. <u>Chem. Mater.</u> 11 (1999):2389-2399
- Akamatsu, K.; Deki, S. Characterization and Optical Properties of Gold
 Nanoparticles Dispersed in Nylon 11 Thin Films. <u>J. Mater. Chem.</u> 7 (1997)
 1773-1777.

- Akamatsu, K.; Deki, S. TEM Investigation and Electron Diffraction Study on Dispersion of Gold Nanoparticles into a Nylon 11 Thin Film During Heat Treatment. <u>J. Colloid Interface Sci.</u> 214 (1999): 353-361.
- 19. Inoue, Y.; Fujii, M.; Inata, M.; Hayashi, S.; Yamamoto, K.; Akamatsu, K.; Deki, S. Single-Electron Tunneling Effects in Thin Nylon 11 Films Containing Gold Nanoparticles. <u>Thin Solid Films</u> 372 (2000): 169-172.
- 20. O.D. Velev, P.M. Tessier, A.M. Lenhoff, E.W. Kaler, Nature 401 (1999): 548.
- 21. G. Kumaraswamy, A.M. Dibaj, F. Caruso. Photonic Materials from Self-Assembly of "Tolerant" Core-Shell Coated Colloids. <u>Langmuir</u> 18 (2002): 4150.
- 22. J.E.G.J. Wijnhoven, S.J.M. Zevenhuizen, M.A. Hendriks, D. Vanmaekelbergh, J.J. Kelly, W.L. Vos, <u>Adv. Mater.</u> 12 (2000): 888.
- 23. J J Diao, F S Qiu, G D Chen and M E Reeves. Surface vertical deposition for gold nanoparticle film. J. Phys. D: Appl. Phys. 36 (2003): L25–L27.
- 24. P. Jiang, J. Cizeron, J.F. Bertone, V.L. Colvin. Preparation of Macroporous Metal Films from Colloidal Crystals. <u>J. Am. Chem. Soc.</u> 121 (1999): 7957.
- 25. Haiyan Wang, Yugai Huang, Zhian Tan, Xiaoya Hu. Fabrication and characterization of copper nanoparticle thin-films. <u>Analytica Chimica Acta</u> 526 (2004): 13–17.
- 26. H.H. Read, Rutley's Elements of Mineralogy, Thomas Murby & Co, London, 26th Edition, (1970).
- 27. C.L. Cleveland, U. Landman, T.G. Schaaf and M.N. Shafigullin, <u>Phys. Rev. Letters</u>, 79 (1997): 1873

- 28. M. Cortie and E. van der Lingen, <u>Gold 2003</u>, 28th September-1st October 2003, Vancouver.
- 29. A.L. Mackay, Acta Cryst., 15 (1962): 916.
- 30. J. Turkevich, Gold Bull, 18(3) (1985): 86.
- 31. Mie, G. Beitra"ge zur Optik Tru" ber Medien, Speziell Kolloidaler Metallo"sungen. Ann. Phys. 25 (1908): 377-445.
- 32. Logunov, S. L.; Ahmadi, T. S.; El-Sayed, M. A.; Khoury, J. T.; Whetten, R. L. Electron Dynamics of Passivated Gold Nanocrystals Probed by Subpicosecond Transient Absorption Spectroscopy. <u>J. Phys. Chem. B</u> 101 (1997): 3713-3719.
- 33. Schaaf, T. G.; Shafigullen, M. N.; Khoury, J. T.; Vezmar, I.; Whetten, R. L.; Cullen, W. G.; First, P. N.; Guttierrez-Wing, C.; Ascencio, J.; Jose-Yacamun, M. J. Isolation of Smaller Nanocrystal Au Molecules: Robust Quantum Effects in Optical Spectra. <u>J. Phys. Chem. B</u> 101 (1997): 7885-7891.
- 34. S. Link et al. Size and Temperature Dependence of the Plasmon Absorption of Colloidal Gold Nanoparticles, <u>J. Phys. Chem. B</u> 103 (1999): 4212-4217.
- 35. C. Cretu and E van der Lingen, Gold Bull. 32(4) (1999): 115.
- 36. M. Quinten, J. of Cluster Science, 10(2) (1999): 319.
- 37. H.B. Liu, J.A. Ascencio, M. Perez-Alvarez and M.J. Yacaman, <u>Surf. Sci.</u> 491 (2001): 88.
- 38. K. Dick, T. Dhanasekaran, Z. Xhang and D. Meisel, <u>J. Am. Chem. Soc</u>, 124(10) (2002): 2312.

- 39. S. Ismail, K. Deppert, T. Junno, C. Kortegaard, H. Larne, M.H. Magnusson, C. Thelander and I. Samuelson, *J. of Nanoparticle Research*, 4 (2002): 351.
- 40. R.C. Ashoori, Nature, 1st February 379 (1996): 413.
- 41. R.P. Andres, T. Bein, M. Dorogi, S. Feng, J.I. Henderson, C.P. Kubiak, W. Mahoney, R.G. Osifchin and R. Reifenberger, <u>Science</u>, 272 (1996): 1323.
- 42. S.A. Stockman et al, MRS Fall Meeting. 2-6th December 2002, Boston, USA, L8.1.
- 43. D.N. Madsen, K. Malhave, R. Mateiu, A. Rasmussen, M. Brorson, C.J.H. Jacobsen and P. Baggild, Nano Lett, 3(1), (2003): 47.
- 44. Nuzzo, R.G.; Allara, D.L. Adsorption of bifunctional organic disulfides on gold surfaces. J. Am. Chem. Soc. 105 (1983): 4481.
- 45. Bishop, A.R.; Nuzzo, R.G. Self-assembled monolayers: recent developments and applications. <u>Current Opinion in Colloid and Interface Science</u> 1 (1996): 127.
- 46. Bain, C.D.; Troughton, E.B.; Tao, Y.-T.; Evall, J.; Whitesides, G.M.; Nuzzo, R.G. Formation of monolayer films by the spontaneous assembly of organic thiols from solution onto gold. J. Am. Chem. Soc. 111 (1989): 321.
- 47. Xia, Y.; Whitesides, G.M. AR. Soft Lithiography. Mater. Res. 28 (1998): 153.
- 48. Mandenius, C. F.; Welin, S.; Danielsson, B.; Lundstrom, I.; Mosbach, K. The interaction of protein and cells with affinity ligands covalently coupled to silicon surfaces as monitored by ellipsometry. <u>Anal. Biochem.</u> 137 (1984): 106.
- 49. Sigal, G.B.; Mrksich, M.; Whitesides, G.M., Using surface plasmon resonance spectroscopy to measure the association of detergents with self-assembled monolayers of hexadecanethiolate on gold, <u>Langmuir</u> 13 (1997): 2749.

- 50. Feldman, L.C.; Mayer, J.W. Fundamentals of Surface and Thin Film Analysis, <u>PTR Prentice Hall</u>, Englewood Cliffs, 1986.
- 51. Weck, M.; Jackiw, J.J.; Rossi, R.R.; Weiss, P.S.; Grubbs, R.H. Ring-opening metathesis polymerization from surfaces. <u>J. Am. Chem. Soc.</u> 121 (1999): 4088.
- 52. Li, H.; Luk, Y.-Y.; Mrksich, M. Catalytic asymmetric dihydroxylation by gold colloids functionalized with self-assembled monolayers. <u>Langmuir</u> 15 (1999): 4957.
- 53. Bartz, M.; Küther, J.; Seshadri, R.; Tremel, W. Angew.. Colloid-bound catalysts for ring-opening metathesis polymerization: a combination of homogeneous and heterogeneous properties. <u>Chem. Int. Ed.</u> 37 (1998): 2466.
- 54. Ostuni, E.; Whitesides, G.M. The interaction of proteins and cells with self-assembled monolayers of alkanethiolates on gold and silver. <u>Colloids and Surfaces B: Biointerfaces</u> 15 (1999): 3.
- 55. Storhoff, J.J.; Elghanian, R.; Mucic, R.C.; Mirkin, C.A.; Letsinger, R.L. One-pot colorimetric differentiation of polynucleotides with single base imperfections using gold nanoparticle probes. J. Am. Chem. Soc. 120 (1998): 1959.
- 56. Spinke, J.; Liley, M.; Guder, H.-J.; Angermaier, L.; Knoll, W. Molecular recognition at self-assembled monolayers: the construction of multicomponent multilayers. <u>Langmuir</u> 9 (1993): 1821.
- 57. Chen, C.S.; Mrksich, M.; Huang, S.; Whitesides, G.M.; Ingber, D.E. Micropatterned surfaces for control of cell shape, position and function. Biotechnol. Prog. 14 (1998): 356.
- 58 Jame J. Licari, Plastic coatings for electronics, Robert E. Krieger <u>Publishing</u>

 <u>Company, Inc.</u> Malabar, Florida, 1981.

- 59 Kirk-Othmer, Encyclopedia of Chemical Technology, Volume 19, 4th edition 832.
- 60. Turkevitch, J.; Stevenson, P. C.; Hillier, J. Nucleation and Growth Process in the Synthesis of Colloidal Gold. <u>Discuss. Faraday Soc.</u> 11 (1951): 55-75.
- 61. Frens, G. Controlled Nucleation for the Regulation of the Particle Size in Monodisperse Gold Suspensions. <u>Nature: Phys. Sci.</u> 241 (1973): 20-22.
- 62. Yonezawa, T.; Kunitake, T. Practical Preparation of Anionic Mercapto Ligand-Stabilized Gold Nanoparticles and Their Immobilization. <u>Colloids Surf. A:</u>

 <u>Physicochem. Eng. Asp.</u> 149 (1999): 193-199.
- 63. J. Turkevich, Colloidal Gold. Part 1: Historical and preparative aspects, morphology and structure, Gold Bull., 18 (1985): 3.
- 64. J. Turkevich, J. Hillier and P.C. Stevenson, <u>Discussion of the Faraday Society</u> 11 (1951): 55-74.
- 65. F.H. Fry, G.A. Hamilton and J. Torkevich, <u>J. Inorg. Chem.</u>, 5 (1966): 1943.
- 66. M. Hu et al., Structure and morphology of self-assembled 3-mercaptopropyltri Methoxysilane layers on silicon oxide, <u>Appl. Surf. Sc.</u> 181 (2001): 307-316.
- 67. K.C. Vrancken, K. Possemier, P. Van Der Voort, E.F. Vansant, Colloid Serf. A 98 (1995): 235.
- 68. M.E. McGovern, K.M.R. Kallury, M. Thompson, Langmuir 10 (1994): 3607.
- 69. Zheng-Chun Liu et al., Self-assembly monolayer of mercaptopropyltrimethoxy silane for electroless deposition of Ag, <u>Materials Chemistry and Physics</u> 82 (2003): 301–305.

- 70. Z.D. Xiao, J.H. Gu, D. Huang, Z.-H. Lu, Y. Wei, Appl. Surf. Sci. 125 (1998): 85.
- 71. Y.L. Chen, L. Gan, Z.-H. Lu, Mol. <u>Cryst. Liq. Cryst.</u> 337 (1999): 477.
- 72. I. Piwanski et al., Investigate of 3-mercaptopropyltrimethoxysilane self-assembled monolayers on Au(111) surface, <u>Appl. Surf. Sci.</u> 242 (2005): 147-153.
- 73. G.E. Poirier, Characterization of Organosulfur Molecular Monolayers on Au(111) using Scanning Tunneling Microscopy, Chem. Rev. 97 (1997): 1117–1127.
- 74. Yong Yang, Mamiko Hori, Tomokatsu Hayakawa and Masayuki Nogami. Self-assembled 3-dimensional arrays of Au@SiO₂ core-shell nanoparticles for enhanced optical nonlinearities. <u>Surf. Sci.</u> 579 (2005): 215–224.
- 75. J. Wang et al., Gold nanoparticulate film bound to silicon surface with self-assembled manolayers, Thin Solid Film 327-329 (1998): 591-594.
- 76. I. Haller, J. Am. Chem. Soc. 100 (1978) 8050.
- 77. C.S. Dulcey, J.H. Georger, V. Krauthamar Jr., D.A. Stenger, T.L. Fare, J.M. Calvert, <u>Science</u> 252 (1991): 551.
- 78. D.J. Dunaway, R.L. McCarley, <u>Langmuir</u> 10 (1994): 3598.
- 79. K.D. Hermanson, S.O. Lumsdon, J.P. Williams, E.W. Kaler and O.D. Velev, Science, 294 (2001): 1082.
- 80. Haruta, M., 'Gold as a Novel Catalyst in the 21st Century: Preparation, Working Mechanisms and Applications' Gold Bull. 37 (2004): 1.
- 81. A.S.K. Hashmi, T.M. Frost and J.W. Bats, Catal. Today, 72 (2002): 19.

- 82. Christopher W Corti, Richard J Holliday and David T Thompson, Developing New Industrial Applications for Gold: Gold Nanotechnology, <u>Gold Bull</u>, 35 (2002): 4.
- 83. Dr Christopher W. Corti and Dr Richard J. Holliday, Commercial Aspects of Gold Applications: From Materials Science to Chemical Science, <u>Gold Bull</u>, 37, (2004): 1-2.
- 84. D. Cameron, D.T. Thompson and R.J. Holliday, <u>Journal of Power Sources</u>, 118 (2003): 298-303.
- 85. Z. Hao, D. Cheng, Y. Guo and Y. Liang, Appl. Catal. B: Env. 33(3) (2001): 217.
- 86. O. Kajikawa, X-S. Wang, T. Tabata, and O. Okada, Organohalogen Compounds, 40 (1999): 581.
- 87. Chandler, J., Gurmin, T., Robinson, N. 'The Place of Gold in Rapid Tests' IVD Technology, 37, March-April 2000.
- 88. Wilson, M., Kannangara, K., Smith, G., Simmons, M., Raguse, B.,

 'Nanotechnology Basic Science and Emerging Technologies', <u>Chapman</u>
 Hall, 2002.
- 89. T. Kobayashi, Nippon Paint Co. Ltd., Private communication, 2002.
- 90. Iwakoshi, A., Nanke, T., Kobayashi, T., 'Coating Materials Containing Gold.'
 Nanoparticles' <u>Proceedings of the International Conference</u> 'Gold2003: New Industrial Applications for Gold', Vancouver, Canada, Sept 28 Oct 1, 2003.

- 91. Turkevitch, J., Stevenson, P.C. and Hillier, J., <u>Discussions of the Faraday</u>
 Society.11, (1951): 55-75.
- 92. Zhuravlev, L.T., Langmuir 3 (1987): 316-318.
- 93. L.T. Zhuravlev, <u>Langmuir</u> 3 (1987): 316.
- 94. J.M. Madeley, C.R. Richmond, Z. Anorg. Allg. Chem. 389 (1972): 82.
- 95. David I Gittins, Andrei S. Susha, Bjoern Schoeler amd Frank Caruso, Dense Nanoparticulates Thin Films via Gold Nanoparticles Self-Assembly, <u>Adv. Mat4r</u>, 14 (2002): 7.
- 96. Zhuravlev, L.T., <u>Langmuir</u> 3 (1987): 316-318.

· ·

- 97. E.T. Vandenberg, L. Bertilsson, B. Liedberg, K. Uvdal, R. Erlandsson, H. Elwing, Lundstrom, J.Colloid Interf. Sci. 17 (1994): 492.
- 98. Hu, M.H., Noda, S., Okubo, T., Yamaguchi, Y. and Komiyama, H., <u>Appl. Surf.</u>
 <u>Sci.</u>181 (2001): 307-316.



บันทึกข้อความ

ส่วนราชการ บัณฑิตศึกษา ภาควิชาวิศวกรรมเคมี โทร. 86880 ที่ วคบ. วันที่

เรื่อง โปรคลงนาม

เรียน คณบดีคณะวิศวกรรมศาสตร์ (ผ่านรองคณบดีฝ่ายวิชาการ)

ตามที่ภาควิชาวิศวกรรมเคมี ได้จัดสอบวิทยานิพนธ์นิสิตบัณฑิตศึกษา ของนิสิตชื่อ นาย ขวัญพงศ์ เหมือนโพธิ์ เลขประจำตัว 4770229721 ระดับปริญญาโท ในหัวข้อวิทยานิพนธ์ เรื่อง "การสร้างฟิล์มบางของอนุภาคนาโนของทองบนพอลิอิไมด์" ซึ่งผลการสอบที่ได้ ดีมาก

ภาควิชาวิศวกรรมเคมี ขอส่งวิทยานิพนธ์ฉบับสมบูรณ์ จำนวน 1 ชุด เพื่อโปรคให้ท่าน คณบคืลงนามตามเอกสารที่แนบมา

จึงเรียนมาเพื่อโปรคลงนามในวิทยานิพนธ์ฉบับสมบูรณ์คังกล่าวด้วย จักขอบคุณยิ่ง

(ศาสตราจารย์ คร.ปียะสาร ประเสริฐธรรม) หัวหน้าภาควิชาวิศวกรรมเคมี

APPENDIX A

Calculation of Percent Coverage of Gold Nanoparticles on Substrate

We can calculate the percent coverage of AuNPs on modified surfaces from the following equation:

% Coverage of AuNPs
$$= \left(\frac{\left(\frac{1}{4}\pi d^2 N\right)}{A}\right) x 100$$

Where d is the average diameter of 50 gold nanoparticles (nm)

N is the amount of AuNPs on surface, being 1000x1000 nm² in area

A is the total area in surface

Assumption 1. Gold nanoparticle is regular sphere.

2. The differention in AuNPs diameter isn't significant.

Because the standard deviation of AuNPs diameter is small, the average diameter of gold nanoparticle was calculated from the average diameter of 50 gold nanoparticles. The value was then multiplied by the amount of AuNPs on total area to calculate the area covered by gold nanoparticles.

Figure A1 shows the examples of size distribution of gold nanoparticles on the surface area. The size distribution of AuNPs on APTMS-modified glass slide and polyimide are shown in Figure A1(a-b), respectively. The average diameter and standard deviation of gold nanoparticles in APTMS-modified glass slide are 18.84 nm and 1.94, respectively. The average diameter and standard deviation of gold nanoparticles in APTMS-modified polyimide are 28.56 nm and 3.38, respectively. Those standard deviation are small. Therefore, the area of surface covered by AuNPs can be calculated from the average value of 50 gold nanoparticles.

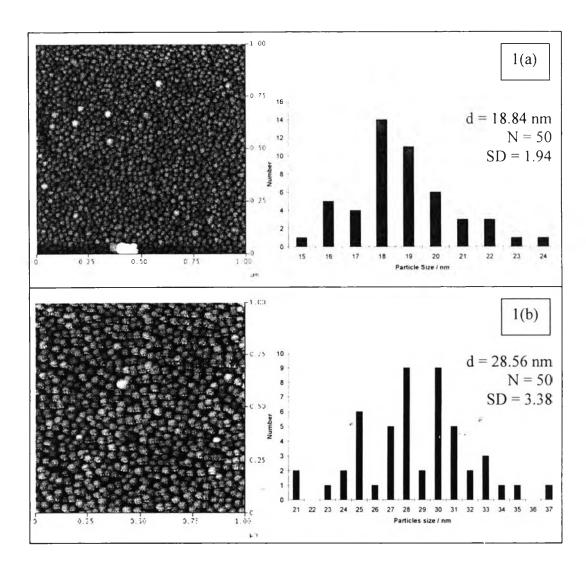


Figure A1 size distribution of gold nanoparticles on various : (a) APTMS-modified glass slide and (b) APTMS-modified polyimide

Calculation exemple for the percent AuNPs coverage on APTMS-modified glass

Diameter (nm)	15	16	17	18	19	20	21	22	23	24
Number	1	5	4	14	11	6	3	3	I	1

Calculating the average diameter

$$d = \frac{(15x1) + (16x5) + (17x4) + (18x14) + (19x11) + (20x6) + (21x3) + (22x3) + (23x1) + (24x1)}{50}$$

= 18.84

The average diameter is 18.84 nm. (d = 18.84)

Counting the number of AuNPs on surface

The total number of AuNPs on $1000x1000 \text{ nm}^2$ of APTMS-modified glass, is 1373 nanoparticles. (N = 1373)

Determination of the total area

Because scan size of AFM is 1
$$\mu m$$
, the total area is $1000 \times 1000 = 10^6 \text{ nm}^2$ (A = 10^6)

These values are substitute in

% Coverage of AuNPs
$$= \frac{\left(\frac{1}{4}\pi d^{2}N\right)}{A}x100$$

$$= \frac{\left(\frac{1}{4}\pi(18.84)^{2}(1373)}{10^{6}}\right)x100$$

$$= 38.26$$

The percent coverage of AuNPs on APTMS-modified glass slide is 38.26

APPENDIX B

Standard Test Method for Measuring Adhesion by Tape Test (Destination: D 3359 - 02)

1. Scope

- 1.1 These test methods cover procedures for assessing the adhesion of coating films to metallic substrates by applying and removing pressure-sensitive tape over cuts made in the film.
- 1.2 Test Method A is primarily intended for use at job sites while Test Method B is more suitable for use in the laboratory. Also, Test Method B is not considered suitable for films thicker than 5 mils (125µm).
- 1.3 These test methods are used to establish whether the adhesion of a coating to a substrate is at a generally adequate level. They do not distinguish between higher levels of adhesion for which more sophisticated methods of measurement are required.
- 1.4 In multicoated systems adhesion failure may occur between coats so that the adhesion of the coating system to the substrate is not determined.
- 1.5 The values stated in SI units are to be regarded as the standard. The values given in parentheses are for information only.

2. Summary of Test Method

- 2.1 **Test Method** A -An X-cut is made through the film to the substrate, pressure-sensitive tape is applied over the cut and then removed, and adhesion is assessed qualitatively on the 0 to 5 scale.
- 2.2 **Test Method B** A lattice pattern with either six or eleven cuts in each direction is made in the film to the substrate, pressure-sensitive tape is applied over the lattice and then removed and adhesion is evaluated by comparison with descriptions and illustrations.

Test Method B – Cross-Cut Tape Test

Procedures

- 1. Select an area free of blemishes and minor surface imperfections, place on a firm base, and under the illuminated magnifier, make parallel cuts as follows:
- 1.1 For coating having a dry film thickness between 2.0 mils (50 \(\mu m\)) space the cuts 2 mm apart and make six cuts. For films thicker than 5 mils use Test Method A.
- 1.2 Make all cuts about 20 mm (3/4 in.) long. Cut through the film to the substrate in one steady motion using just sufficient pressure on the cutting tool to have the cutting edge reach the substrate. When making successive single cuts with the aid of a guide, place the guide on the uncut area.
- 2. After making the required cuts brush the film lightly with a soft brush or tissue to remove any detached flakes or ribbons of coatings.
- 3. Examine the cutting edge and, if necessary, remove any flat spots or wire-edge by abrading lightly an a fine oil stone. Make the addition number of cuts at 90° to and centered on the original cuts.
- 4. Brush the area as before and inspect the incisions for reflection of light from the substrate. If the metal has not been reached make another grid in a different location.
- 5. Remove two complete laps of tape and discard. Remove an additional length at a steady (that is, not jerked) rate and cut a piece about 75 mm (3 in.) long.
- 6. Place the center of the tape over the grid and in the area of the grid smooth into place by a finger. To ensure good contact with the film rub the tape firmly with the eraser on the end of a pencil. The color under the tape is a useful indication of when good contact has been made.
- 7. Within 90 ± 30 s of application, remove the tape by seizing the free end and rapidly (not jerked) back upon itself at as close to an angle of 180° as possible.
- 8. Inspect the grid area for removal of coating from the substrate of from a previous coating using the illuminated magnifier. Rate the adhesion in accordance with the following scale illustrated in Table B1:

Table B1 Classification of Adhesion Test Results

CLASSIFICATION OF ADHESION TEST RESULTS							
CLASSIFICATION	PERCENT AREA REMOVED	SURFACE OF CROSS-CUT AREA FROM WHICH FLAKING HAS OCCURRED FOR SIX PARALLEL CUTS AND ADHESION RANGE BY PRECENT					
5B	0 % None						
4B	Less than 5 %						
3B	5 - 15 %						
2B	15 – 35 %						
1B	35 – 65 %						
0B	Greater than 65 %						

- 5B The edges of cuts are completely smooth; none of the squares of the lattice is detached.
- 4B Small flakes of the coating are detached at intersections; less than 5 % of the area is affected.
- 3B Small flakes of the coating are detached along edge and at intersection of cuts. The area affected is 5 to 15 % of the lattice.
- 2B The coating has flaked along the edges and on parts of the squares. The area affected is 15 to 35 % of the lattice.
- 1B The coating has flaked along the edge of cuts in large ribbons and whole squares have detached. The area affected is 35 to 65 % of the lattice.
- 0B Flaking and detachment worse than Grade 1.

APPENDIX C

LIST OF PUBLICATION

1. Kwarnpong Muanpho, Piyasan Praserthdam and Varong Pavarajarn. "Surface Modification for Fabrication of Gold Nanoparticles Thin Film on Glass Substrate", 4th Thailand Materials Science and Technology Conference, Bangkok, Thailand, 31 March – 1 April 2005.



VITA

Mr.Kwarnpong Muanpho was born on 21st June, 1982, in Bangkok, Thailand. He received her Bachelor degree of Engineering with a major in Chemical Engineering from Chulalongkorn University, Thailand in April 2004. He continued her Master study in the same major at Chulalongkorn University, Bangkok, Thailand in June 2004.