

REFERENCES

- Salaneck, W.R., I, Lundstrom , Ranby, B (1993) *Conjugated Polymers and Related Materials*, Oxford University Press Inc, New York .
- Mark, J.E (1996) *Handbook of the physical properties of Polymers*. AIP Press, Woodbury, CT.
- Xia, Y., Macdiarmid, A.G., Epstein A.J. (1994) Camphorsulfonic Acid Fully Doped Polyaniline Emeraldine Salt : In Situ Observation of Electronic and Conformational Changes Induced by Organic Vapors by an Ultraviolet/visible/Near-Infraed Spectroscopy Method.
Macromolecules 27, 7212.
- Kang, E.T., Neoh K.G., Tan K.L. (1998) Polyaniline:A polymer with many interesting intrinsic redox states Progress in Polymer Science 23 , 277.
- Show-An Chen.,Liang-Chang Lin. (1995) Polyaniline Doped by the New Class of Dopant, Ionic Salt : Structure and Properties.
Macromolecules 28 , 1239.
- Levon, K., Ho.H.K., Zheng, W.Y., Laakso, J., Karna, T., Taka, T., Osterholm, J.E. (1995) Thermal doping of polyaniline with dodecylbenzene sulfonic acid without auxillary solvents. Polymer 36, 2733.
- Joo, J., Song, H.G., Chung, Y.C., Baeck, J.S., Jeong, S.K., Sun, J.S., Oh,E.T. (1997) The Effects of Dopant and Solvent on Charge Transport of Doped Polyanilines. Journal of Korean Physical Society 30 , 230.
- Geng, Y., Li, J., Jing. X., Wang, F. (1997) Polyaniline doped with macromolecule. Synthetic Metals 84 , 81.
- Shimizu, S., Saitoh, T., Uzawa, M., Yuasa, M., Yano, K., Maruyama, T., Watanabe, K. (1997) Synthesis and Applications of Sulfonated Polyaniline. Synthetic Metals 85 , 1337.
- Tsutsumi, H., Yamashita, S., Oishi, T. (1997) Preparation of polyaniline-poly (*p*-styrenesulfonic acid) composite by post-polymerizationand

- application as positive active material for a rechargeble lithium battery. Journal of Applied Electrochemistry 27 , 477.
- Sertova, N., Geffroy, B., Nunzi, J.M., Petkov, I. (1998) PVC as photodonor of HCl for protonation of polyaniline. Journal of Photochemistry and photobiology A: Chemistry 113 , 99.
- Liu, W., Kumar, J., Tripathy, S., Senecal, K.J., Samuelson, L. (1999) Enzymatically synthesized Conducting Polyaniline. Journal of Chemical Society 121 , 71.
- Hue, M.Y., Hwang, G.W., Chung, Y.H., Chen, S.A., Tsai, R.Y. (2000) Soluble n-Doping Polyaniline : Synthesis and Characterization. Macromolecules 33 , 6235.
- Luzny, W., Banka, E. (2000) Relation between the structure and Electric Conductivity of Polyaniline Protonated with Camphorsulfonic Acid. Macromolecules 33 , 425.
- Lee, Y.M., Kim, J.H., Kang, J.S., Ha, S.Y. (2000) Annealing Effects of Dilute Polyaniline / NMP Solution. Macromolecules 33 , 7431.
- Davey, J.M., Too,C.O., Ralph, S.F., Kane, L.A., Wallace, G.G. (2000) Conducting Polyaniline / Calixarene Salt : Synthesis and Properties. Macromolecules 33 , 7044.
- Olinga, T.E., Freysse, J., Travers, J.P., Dufresne, A., Pron, A. (2000) Highly Conducting and Solution-Processable Polyaniline Obtained via Protonation with a New Sulfonic Acid Containing Plasticizing Functional Groups. Macromolecules 33 , 2107.
- Hidaka, M., Aizawa, M. (1995) Electrochemically Synthesized Polyaniline/Enzyme Membrane for a Choline Biosensor, Biosensors 63 , 1113
- Kukla, A.L., Shirshov, Y.M., Piletsky, S.A. (1996) Ammonia sensors based on sensitive polyaniline films. Sensor and Actuators Part B 37, 135.

- Sangodkar, H., Sukeerthi, S., Srinivasa, R.S., Lai, R., Contractor, A.Q. (1996) A biosensor Array Based on Polyaniline. Analytical Chemistry 68 , 779.
- Mu, S., Xue, H. (1996) Bioelectrochemical characteristics of glucose oxidase immobilized in a polyaniline film. Sensors and Actuators Part B 31 , 155.
- Laranjeira,J.M.G., Azevedo, W.M., Araujo, M.C.U. (1997) A conductimetric system based on polyaniline for determination of ammonia in fertilizers. Analytical Letters 30 , 2189.
- Dhawan, S.K., Kurnur, D., Ram, M.K., Chandra, S., Trivedi, D.C. (1997) Application of conducting polyaniline as sensor material for ammonia Sensors and Actuators Part B 40 , 99.
- Yoon, C.O., Kim, J.H., Sung, H.K., Lee, H. (1997) Electrical conductivity and thermopower of phosphoric acid doped polyaniline. Synthetic Metals 84 , 789.
- Luckchova, L.V., Karyakin, A.A., Karyakina, E.E., Gorton, L. (1997) The improvement of polyaniline glucose biosensor stability using enzyme immobilization from water-organic mixtures with a high content of organic solvent. Sensors and Actuators B 44 , 356.
- Cho, W.J., Huang, H.S. (1998) An amperometric Urea Biosensor Based on a Polyaniline-Perfluorosulfonated Ionomer Composite Electrode. Analytical Chemistry 70 , 3946.
- Min, H. K., Yang, H.S., Cho, S.M. (2000) Extremely sensitive optical sensing of ethanol using porous silicon. Sensors and Actuators B 67 , 199.
- Stella, R., Barisci, .N., Serra, G., Wallace, G.G., Rossi, D.D. (2000) Characterisation of olive oil by an electronic nose based on conducting polymer sensors. Sensor and Actuator B 63 , 1.
- Pouget, J.P., Jozefowicz, M.E., Epstein, A.J., Tang, X., Macdiarmid, A.G. (1991) X-ray Structure of Polyaniline. Macromolecule 24 , 779.

- Xing-Rong Zeng and Tze-Man Ko. (1998) Structures and properties of chemically reduced polyanilines. Polymer 39 (5),1187.
- Wan, M. (1992) Absorption spectra of Thin Film of Polyaniline. Journal of Polymer Science: Part A: Polymer Chemistry 30, 543.
- Li, W and Wan, M. (1999) Stability of Polyaniline Synthesized by a Doping-Dedoping-Redoping Method. Journal of Applied Polymer Science 71, 615.
- Blackwood, D and Josowicz, M.(1991) Work Function and Spectroscopic Studies of Interactions Between Conducting Polymers and Organic Vapors. Journal of Physical Chemistry 95, 493.

APPENDICES

APPENDIX A Determination of ohmic linear regime.

Linear regime or ohmic regime is the regime that applied voltage depends directly on apply current according to ohmic law in equation (A-1)

In this work, linear regime was determined by plotting applied voltage (V_a) versus current(I). The range that gives the straight line is acceptable for using in conductivity measurement. Figure A-1 and A-2 are the plots of V_a and I that using silicon wafer as the standard material and polyaniline, respectively. This experiment was carried out under 1 atm , 26 °C and 40% humidity.

$$V = IR \quad (A-1)$$

where :

V = applied voltage (mV)

I = current (mA)

R = resistance (Ω)

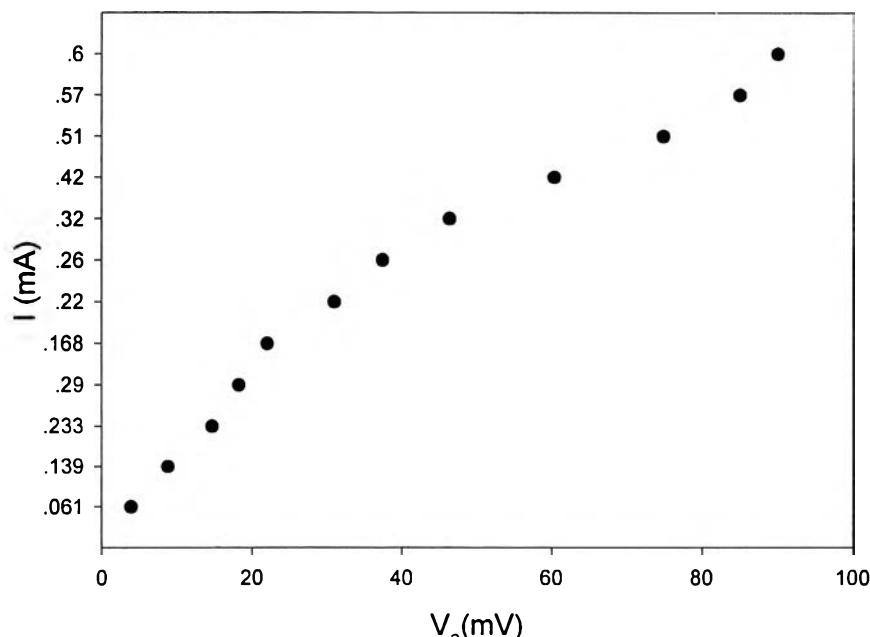


Figure A-1 Linear Regime of V_a and I used Silicon wafer as standard material.

According to Figure A-1, straight line is indicated the range of applied voltage and current corresponding to the ohmic law. The accepted range of those for using in conductivity measurement are 0 to 20 mV and 0 to 0.25 mA, respectively.

Table A-1 Raw data of determination of linear regime (Silicon Wafer).

Applied voltage (mV)	Current (mA)	Volt drop (mV)
3.95	0.06	16.5
8.84	0.14	18.3
14.70	0.23	15.2
18.30	0.29	14.6
22.10	0.17	15.1
31.10	0.22	14.7
37.50	0.26	15.6
46.40	0.32	15.9
60.40	0.42	16.1
74.90	0.51	18.2
85.00	0.57	16.3
90.00	0.60	17.2

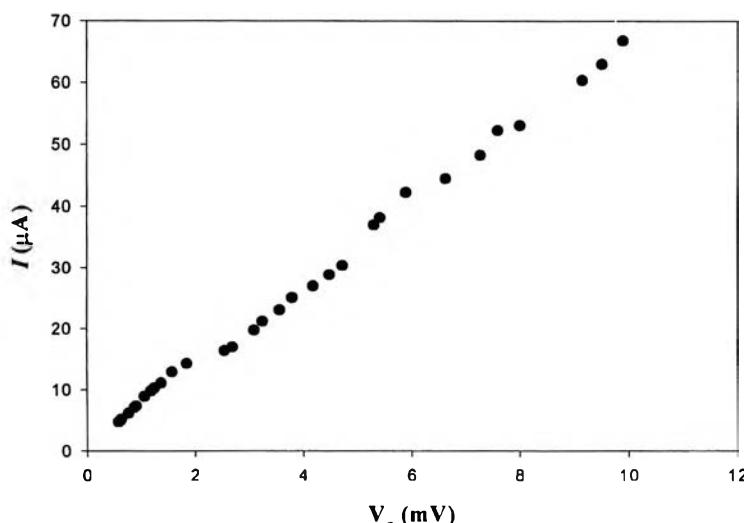


Figure A-2 Linear Regime of V_a and I used polyaniline.

In the case of polyaniline, a straight line can be seen in the range of 0 to 2 mV of applied voltage and 0 to 15 μ A of current , respectively. The resulting value is lower than in the case of silicon wafer due to a lower resistance value.

Table A-2 Raw data of determination of linear regime (polyaniline)

Applied voltage (mV)	Current(μ A)	Volt drop (mV)
0.592	4.780	224.2
0.620	5.000	223.6
0.640	5.250	225.4
0.771	6.190	226.2
0.879	7.090	226.4
0.915	7.350	227.1
1.070	8.900	226.3
1.188	9.800	225.8
1.249	10.30	228.1
1.371	11.09	226.4
1.581	12.90	226.8
1.851	14.30	225.9
2.550	16.40	224.8
2.700	17.00	221.9
3.100	19.80	222.6
3.250	21.20	226.7
3.570	23.10	228.6
3.800	25.10	227.3
4.190	27.00	225.5
4.490	28.85	225.6
4.730	30.40	224.9
5.310	37.00	223.8
5.420	38.20	226.7

Applied voltage (mV)	Current(µA)	Volt drop (mV)
5.900	42.30	225.5
6.440	44.50	226.1
7.280	48.30	227.4
7.600	52.00	225.2
8.010	53.10	224.9

APPENDIX B Determination of Geometric Correction Factor (K).

Geometric correction factor (K) is a correction factor that takes into account of geometric effects. It depends on the configuration and probe tip spacing. K factor can be determined by using the following equation (B-1). The resistivity of standard materials were calibrated from a using four point probe at King Mongkut's Institute Technology of Lad Krabang.

In this experiment, silicon wafer will be used as a standard material. Resistivities of materials were measured by using the four-point probe. The geometric correction factor was calculated and tabulated in the Table B-1 below:

$$K = \frac{\rho_{ref}}{R \times t} = \frac{w}{l} \quad (B-1)$$

where K = geometric correction Factor

ρ_{ref} = known resistivity from the polymer handbook ($\Omega \cdot \text{cm}$)

R = resistivity (Ω)

t = sheet thickness (cm)

w = width (cm)

l = length (cm)

Table B-1 Data of K correction factor determination

Material	t(cm)	$\rho_{ref}(\Omega \cdot \text{cm})$	Probe no.	K
SiO_2/TaA	0.0715	9.3×10^3	1	0.09151
			2	0.10550
Si 10-28A	0.0522	3.5×10^1	1	0.11471
			2	0.09457

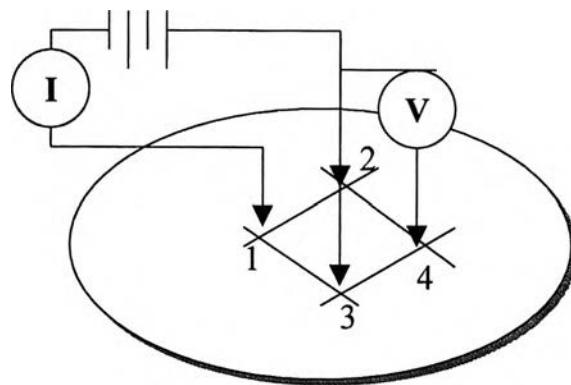


Figure B-1 Schematic draw of four-point probe.

Table B-2 Raw data of K of probe 1 using silicon wafers as the standard materials (26°C , 40% relative humidity and Applied voltage = 0.30V).

Si 10-28A			SiO ₂ /TaA		
I (mA)	V (mV)	K	I (mA)	V (mV)	K
0.000355	2.5	0.09521	0.0198	0.031	0.08420
0.000415	2.4	0.11590	0.0442	0.070	0.08272
0.000572	2.9	0.13225	0.0522	0.083	0.08172
0.000600	3.2	0.12570	0.0735	0.116	0.08206
0.000944	4.5	0.14065	0.0970	0.151	0.08383
0.001887	9.8	0.12900	0.1160	0.166	0.09023
0.001721	9.7	0.11896	0.1260	0.176	0.09338
0.002351	13.5	0.11670	0.1350	0.192	0.09135
0.002378	14.1	0.11308	0.1430	0.200	0.09323
0.002584	16.3	0.10629	0.1480	0.201	0.09609
0.002744	17.8	0.10336	0.1540	0.207	0.09700
0.002915	19.9	0.09822	0.1600	0.209	0.09926
0.003014	21.1	0.09577	0.1650	0.213	0.10130
0.003043	21.9	0.09317	0.1660	0.206	0.10471
	Average	0.11471		Average	0.09151

Table B-3 Raw data of K of probe 2 using silicon wafers as the standard material (26°C , 40% relative humidity and Applied voltage = 0.30 V).

Si 10-28A			SiO ₂ /TaA		
I (mA)	V (mV)	K	I (mA)	V (mV)	K
0.000665	5.20	0.08574	0.045	0.473	0.1237
0.000650	5.10	0.08546	0.050	0.0530	0.1227
0.000650	4.90	0.08894	0.065	0.0601	0.1406
0.000700	5.10	0.09203	0.075	0.0753	0.1295
0.000700	5.20	0.09026	0.075	0.0810	0.1204
0.000630	5.00	0.08448	0.085	0.0993	0.1113
0.000600	4.90	0.08210	0.100	0.1235	0.1053
0.000700	4.70	0.09415	0.105	0.1471	0.0928
0.000800	5.40	0.09986	0.115	0.1537	0.0973
0.000800	5.60	0.09578	0.125	0.1694	0.0960
0.000800	5.10	0.10518	0.130	0.1975	0.0856
0.000580	4.20	0.09259	0.135	0.2120	0.0828
0.000670	4.50	0.09983	0.140	0.2180	0.0835
0.000780	4.10	0.12756	0.145	0.2210	0.0853
	Average	0.09457		Average	0.1055

APPENDIX C Determination of % bipolaron and polaron by UV-Visible spectrometer.

According to the Beer's law (Chambell and White,1989),

$$A_i = a_i b_i c_i \quad (C-1)$$

where A_i = area of each peak

a_i = absorptivity (cm^2/g)

b_i = path length (cm)

c_i = concentration of emeraldine base in solution (g/cm^3)

The calibration curves in which the areas are plotted as a function of the concentration of emeraldine base in the solution can give some important peaks. These are ~ 325 nm representing the benzenoid part, ~ 440 nm showing the bipolaron part, ~ 625 nm representing the quinoid part, and ~ 700 - 900 nm giving the polaron part. The slopes of the calibration curves, thus, provide the product of absorptivity of particular species, a_i and b_i . To obtain the amount of the polaron and bipolaron in an unknown sample, the concentration can be calculated by the following equation;

$$C_i = A_i/a_i b_i \quad (C-2)$$

Hence, the % bipolaron and % polaron could be calculated by equations if the areas of the benzenoid, quinoid, bipolaron and polaron part are known

$$A_{BZ} + A_{BP} + A_Q + A_P = A_{total} \quad (C-3)$$

$$\% \text{ bipolaron} = (A_{BP}/A_{total}) * 100 \quad (C-4)$$

$$\% \text{ polaron} = (A_P/A_{total}) * 100 \quad (C-5)$$

The area of each peaks in a UV-Visible spectra could be calculated by using the Gaussian's Equation as shown in equation C-6.

$$\text{Gaussian equation} = \left(1/(SD * ((2 * (22/7))^{0.5}))\right) * \exp(0.5(((x-\text{avg})/SD)^2)) * \text{area}$$

(C-6)

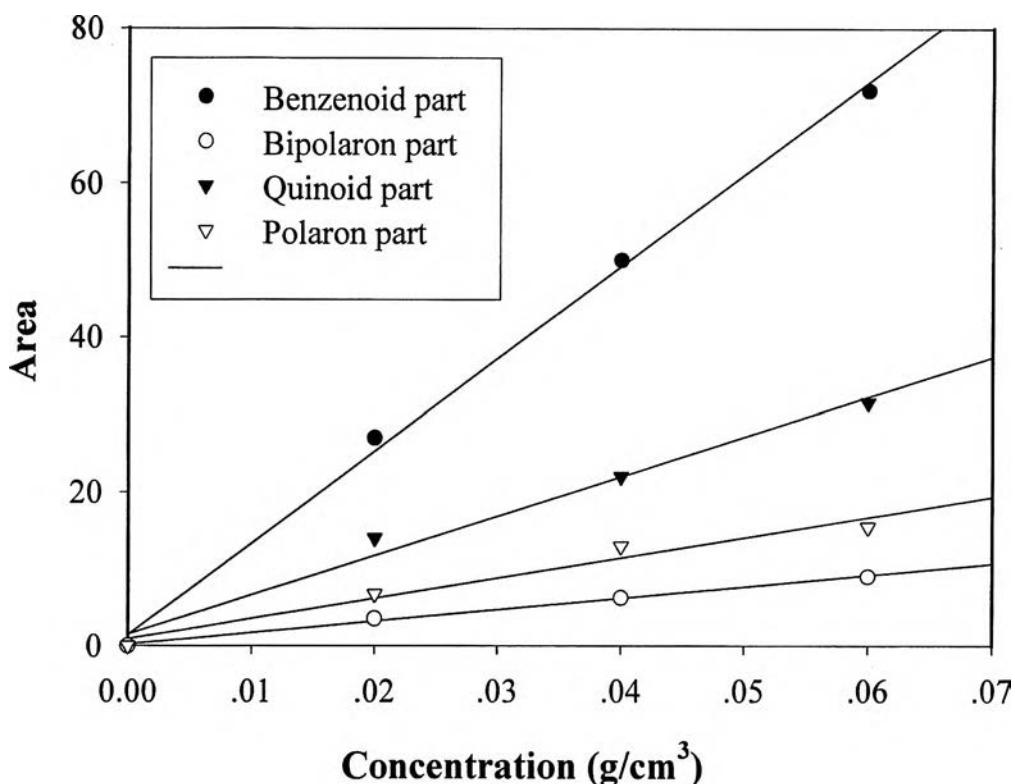


Figure C-1 The calibration curve of PANI-H₂SO₄/CSA.

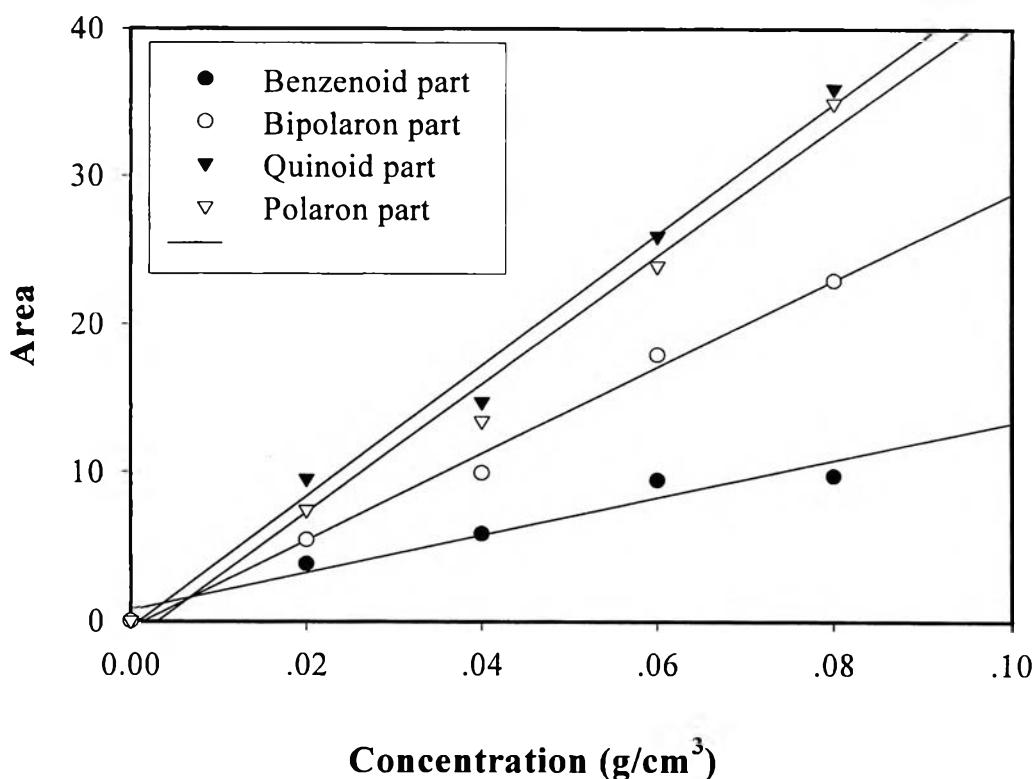


Figure C-2 The calibration curve of $\text{PANI-H}_2\text{SO}_4/\text{HNO}_3$.

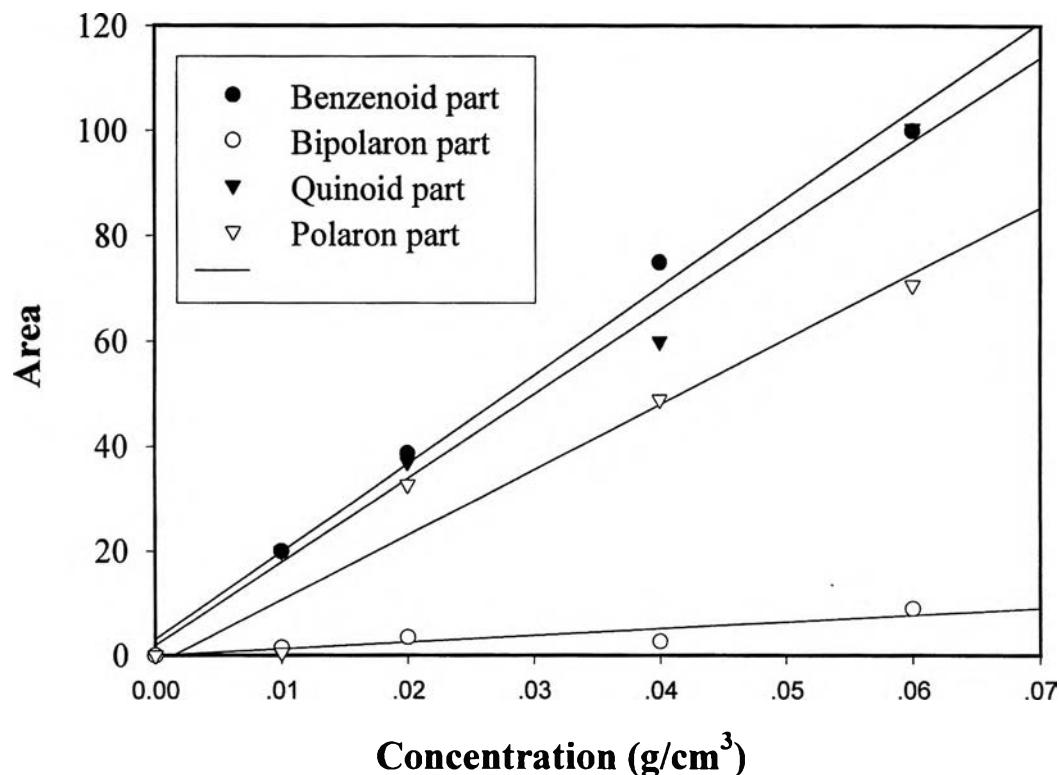


Figure C-3 The calibration curve of $\text{PANI-H}_2\text{SO}_4/\text{HCOOH}$.

From Equation C-1, the slopes of the calibration curves, thus , provide the product of particular species, a_i and b_i . Table C-1 shows the molar absorption coefficient of particular species of doped polyaniline.

Table C-1 The product of absorptivity (a_i) of particular species and b_i of doped polyaniline.

Doped PANI	$a_{\text{benzenoid}} * b_i$	$a_{\text{bipolaron}} * b_i$	$a_{\text{quinoid}} * b_i$	$a_{\text{ polaron}} * b_i$
PANI/CSA	1.20e+3	1.49e+2	5.13e+2	2.64e+2
PANI/HNO ₃	1.26e+2	2.93e+2	4.42e+2	4.33e+2
PANI/HCOOH	1.68e+3	1.29e+2	1.60e+3	1.24e+3

Table C-2 Raw data of % bipolaron and % polaron determination.

Acid	Doping ratio	Area				Concentration				% Area				%BP+%P
		BZ	BP	Q	P	BZ	BP	Q	P	BZ	BP	Q	P	
HCOOH	1	220	6.8	470	15	0.13	0.05	0.29	0.01	30.9	0.96	66.0	2.11	3.07
	2	205	7.15	448	17.5	0.12	0.06	0.3	0.01	30.3	1.06	66.1	2.58	3.64
	4	188	7.47	421	18	0.11	0.06	0.26	0.01	29.6	1.18	66.4	2.84	4.01
	10	175	7.55	398	25	0.10	0.06	0.27	0.02	28.9	1.25	65.7	4.13	5.38
	20	150	7.6	289	32.5	0.10	0.06	0.20	0.03	31.3	1.59	60.3	6.78	8.37
	40	170	7.5	321	35.7	0.10	0.06	0.20	0.03	31.8	1.40	60.1	6.68	8.09
	50	160	7	276	25	0.10	0.05	0.70	0.02	34.2	1.50	59.0	5.34	6.84
	180	45	3.8	230	7.45	0.03	0.03	0.14	0.01	15.7	1.33	80.3	2.6	3.93
	200	39.4	1.8	2.28	0.72	0.02	0	0	0	71.7	4.07	4.15	1.63	5.70
HNO ₃	1	235	78	725	451	1.87	0.27	1.64	1.04	15.8	5.24	48.7	30.3	35.5
	2	235	85	711	469	1.87	0.29	1.61	1.08	15.7	5.67	47.4	31.3	36.9
	4	212	100	684	486	1.68	0.30	1.55	1.12	14.3	6.75	46.2	32.8	39.5
	10	215	120	600	500	1.71	0.41	1.36	1.15	15.0	8.36	31.8	34.8	43.2
	20	220	135	500	539	1.75	0.46	1.13	1.24	15.8	9.68	35.9	38.7	48.4
	30	220	145	200	625	1.75	0.49	0.45	1.44	18.5	12.2	16.8	52.5	64.7
	40	234	366	180	648	1.86	1.25	0.41	1.50	16.4	23.5	12.6	45.4	68.9
	CSA	1	170	28.7	7	170	0.14	0.19	0.01	0.64	45.2	7.64	1.86	45.3
	2	173	33	68	359	0.14	0.22	0.13	1.36	27.3	5.21	10.7	56.7	61.9
	4	173	31.1	7.5	220	0.14	0.21	0.01	0.83	40.1	7.21	1.74	51.0	58.2
	10	179	35	32.9	415	0.15	0.24	0.06	1.57	27.0	5.29	4.97	62.7	68.0
	20	183	33	30.5	425	0.15	0.22	0.06	1.61	27.3	4.91	4.54	63.3	68.2
	30	34	9.42	6.8	120	0.03	0.06	0.01	0.45	20.0	5.53	3.99	70.5	76.0
	40	177	38	38.4	398	0.15	0.26	0.07	1.51	25.1	8.50	5.44	62.4	70.89
	80	178	33	72	315	0.15	0.22	0.14	1.19	29.8	5.52	12.0	52.7	58.2
	160	185	32.5	29.6	300	0.15	0.22	0.06	1.14	33.8	5.94	5.41	54.8	60.8
	250	165	35	65	395	0.14	0.24	0.13	1.50	25.0	5.30	9.85	59.9	65.2

Molecular weight of synthesized PANI was equal to 22,000 g/mol (Matt, 1991)

Number of charge carriers (#) can be calculated from the following equation:

$$\text{MW} = N \times \text{MW of repeating unit}$$

$$\# \text{ of charge carriers}_{\text{Total}} = \# \text{ of charge carriers}_{\text{BP}} + \# \text{ of charge carriers}_{\text{P}} \quad (\text{C-7})$$

$$\# \text{ of charge carriers}_{\text{BP}} = 2 \times N \times (\% \text{bipolaron}) \quad (\text{C-8})$$

$$\# \text{ of charge carriers}_{\text{P}} = N \times (\% \text{ polaron}) \quad (\text{C-9})$$

where N is degree of polymerization.

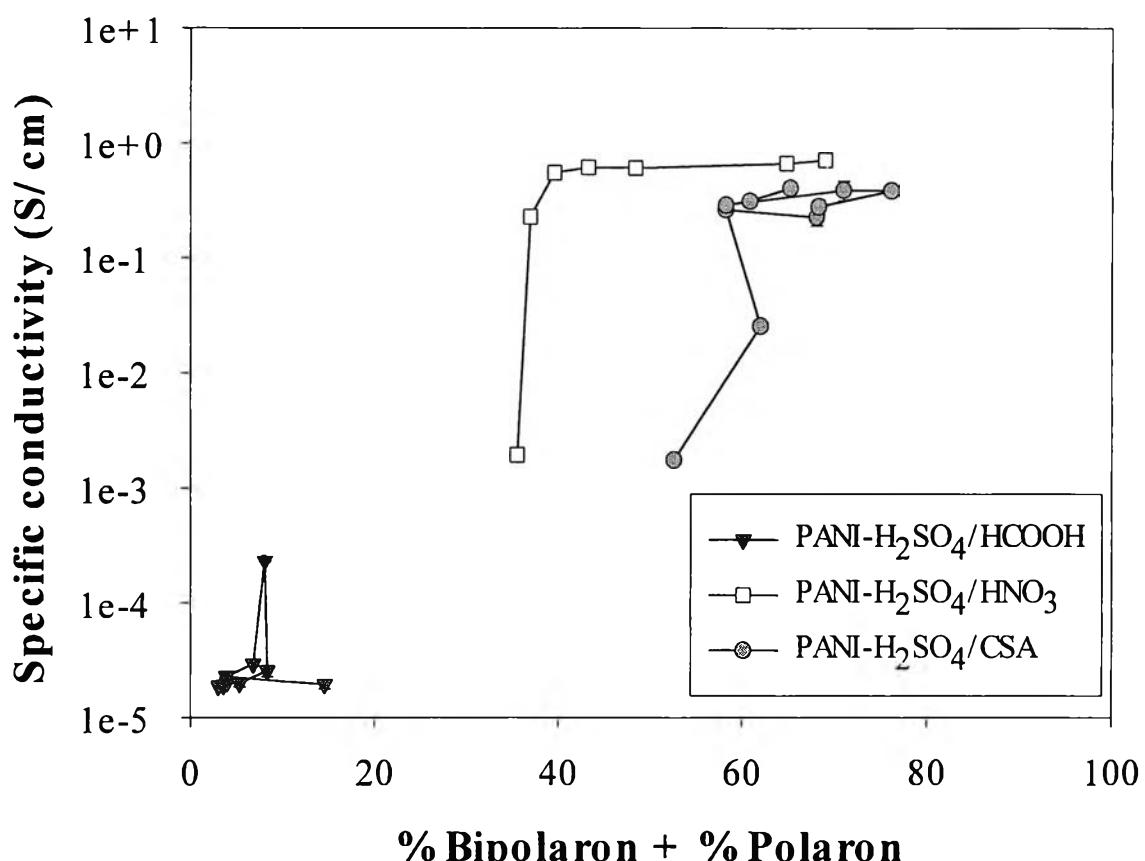
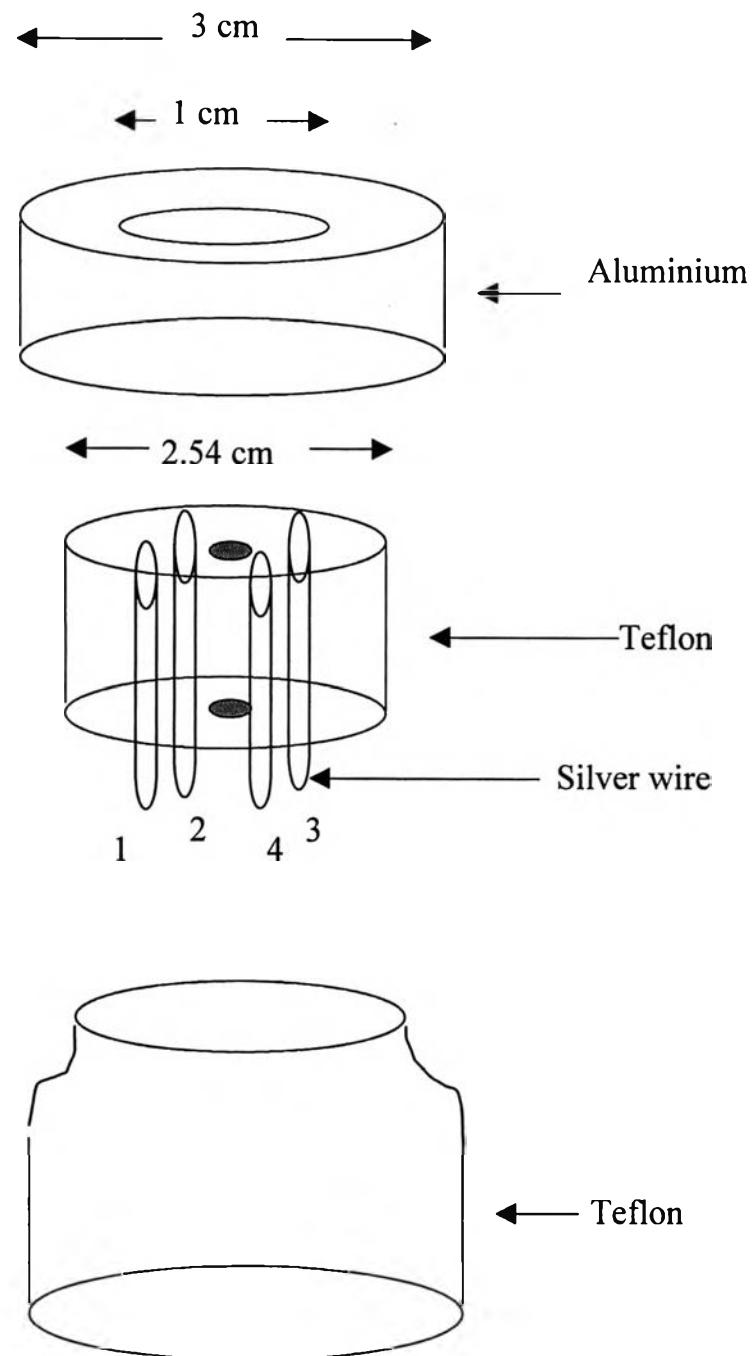


Figure C-4 The effect of %bipolaron + %polaron on the specific conductivity.

Table C-3 Raw data of number of carrier molecules.

Acid	Doping ratio	%BP	%P	n_{BP}	n_P	$n_{BP} + n_P$	σ (S/cm)
HCOOH	1	0.96	2.11	1.16684928	1.28231874	2.44916802	1.89E-05
	2	1.1	2.58	1.28839608	1.56795372	2.8563498	1.93E-05
	4	1.18	2.84	1.43425224	1.72596456	3.1602168	1.95E-05
	10	1.25	4.13	1.519335	2.50994142	4.02927642	2.51E-05
	20	1.59	6.78	1.93259412	4.12043652	6.05303064	2.98E-05
	40	1.4	6.68	1.7016552	4.05966312	5.76131832	2.32E-04
	50	1.5	5.34	1.823202	3.24529956	5.06850156	2.22E-05
	180	1.33	2.6	1.61657244	1.5801084	3.19668084	2.00E-05
	200	4.07	1.63	4.94695476	0.99060642	5.93756118	2.00E-05
HNO ₃	1	5.24	30.3	6.36905232	18.4143402	24.7833925	1.93E-03
	2	5.67	31.3	6.89170356	19.0220742	25.9137778	2.39E-01
	4	6.75	32.8	8.204409	19.9336752	28.1380842	5.85E-01
	10	8.36	34.8	10.1613125	21.1491432	31.3104557	6.16E-01
	20	9.68	38.7	11.7657302	23.5193058	35.285036	6.38E-01
	30	12.2	52.5	14.8287096	31.906035	46.7347446	6.92E-01
	40	23.5	45.4	28.563498	27.5911236	56.1546216	7.17E-01
	CSA	7.64	45.3	9.28617552	27.5303502	36.8165257	1.79E-03
CSA	2	5.21	56.7	6.33258828	34.4585178	40.7911061	2.56E-02
	4	7.21	51	8.76352428	30.994434	39.7579583	2.65E-01
	10	5.29	62.7	6.42982572	38.1049218	44.5347475	2.89E-01
	20	4.91	63.3	5.96794788	38.4695622	44.4375101	3.27E-01
	30	5.53	70.5	6.72153804	42.845247	49.566785	3.95E-01
	40	8.5	62.4	10.331478	37.9226016	48.2540796	4.20E-01
	80	5.52	52.7	6.70938336	32.0275818	38.7369652	3.95E-01
	160	5.94	54.8	7.21987992	33.3038232	40.5237031	2.58E-01
	250	5.3	59.9	6.4419804	36.4032666	42.845247	2.49E-01

APPENDIX D Probe Configuration.

APPENDIX E Conductivity Measurement of Doped PANI under the condition of 1 atm, 60% relative humidity and 28° C.

Table E-1 Raw data of conductivity measurement in air.

Doping ratio	Doping Acid	Thickness (cm)	Applied voltage (V)	Current (mA)	Avg.	SD	Voltage drop (mV)	SD	σ (S/cm)	SD
0	Undoped (1)	0.01065	0.55	1.86E-03	1.17E-04	97382.5	1324.8	1.74E-05	1.09E-06	
	Undoped (2)	0.01018	0.55	1.97E-03	1.34E-04	97082.5	953.1	1.93E-05	1.48E-06	
	Undoped (3)	0.01127	0.55	2.12E-03	1.48E-04	96832.5	1115.6	1.88E-05	1.30E-06	
1	HCOOH (1)	0.01121	0.55	2.14E-03	1.43E-04	96932.5	1077.1	1.91E-05	1.30E-06	
	HCOOH (2)	0.00991	0.35	1.83E-03	1.89E-04	97682.5	1989.3	1.83E-05	2.02E-06	
	HCOOH (3)	0.01114	0.35	2.16E-03	2.22E-04	97382.5	1819.5	1.93E-05	1.97E-06	
2	HCOOH (1)	0.01093	0.35	2.15E-03	1.51E-04	97732.5	1209.3	1.95E-05	1.42E-06	
	HCOOH (2)	0.01112	0.35	2.10E-03	2.31E-04	97682.5	1767.5	1.87E-05	1.99E-06	
	HCOOH (3)	0.01012	0.35	1.99E-03	2.18E-04	97382.5	1693	1.96E-05	2.10E-06	
4	HCOOH (1)	0.01129	0.55	2.35E-03	1.08E-04	97882.5	1353.8	2.06E-05	1.08E-06	
	HCOOH (2)	0.01029	0.55	2.06E-03	1.58E-04	98382.5	1090.2	1.97E-05	1.47E-06	
	HCOOH (3)	0.01225	0.55	2.29E-03	1.37E-04	99192.50	1953	1.83E-05	1.02E-06	
10	HCOOH (1)	0.00981	0.55	2.29E-03	8.76E-05	99842.5	2270.2	2.27E-05	5.39E-07	
	HCOOH (2)	0.01065	0.55	2.61E-03	1.30E-04	97682.5	2228	2.43E-05	1.13E-06	
	HCOOH (3)	0.01225	0.55	3.49E-03	9.94E-05	97592.5	2175.9	2.83E-05	7.14E-07	
20	HCOOH (1)	0.01029	0.55	3.22E-03	1.62E-04	97272.5	1786.4	3.12E-05	1.57E-06	
	HCOOH (2)	0.00925	0.55	2.64E-03	1.77E-04	98382.5	1090.2	2.81E-05	1.21E-06	
	HCOOH (3)	0.01129	0.55	3.42E-03	1.32E-04	97472.5	2291.1	3.02E-05	1.25E-06	
40	HCOOH (1)	0.01181	0.35	2.86E-02	3.98E-04	95732.5	3036.1	2.45E-04	3.34E-06	
	HCOOH (2)	0.01216	0.35	2.73E-02	3.02E-04	95172.5	1320.8	2.29E-04	2.56E-06	
	HCOOH (3)	0.01177	0.25	2.59E-02	8.76E-05	96532.5	1697.7	2.21E-04	8.02E-06	
50	HCOOH (1)	0.01452	0.38	3.45E-03	2.01E-04	96332.5	1081.2	2.39E-05	1.49E-06	
	HCOOH (2)	0.01543	0.38	3.28E-03	2.90E-04	96732.5	1068.8	2.13E-05	1.90E-06	
	HCOOH (3)	0.01216	0.38	2.61E-03	1.10E-04	97212.5	1818.7	2.14E-05	1.03E-06	
180	HCOOH (1)	0.01454	0.38	2.67E-03	2.31E-04	97272.5	1203.7	1.83E-05	1.62E-06	
	HCOOH (2)	0.01520	0.27	3.26E-03	2.22E-04	96132.5	1276.3	2.16E-05	1.43E-06	
	HCOOH (3)	0.01225	0.27	2.47E-03	2.36E-04	97172.5	1104.6	2.01E-05	1.92E-06	
200	HCOOH (1)	0.01205	0.27	2.13E-03	2.41E-04	97252.5	1037.4	1.76E-05	2.01E-06	
	HCOOH (2)	0.01264	0.27	2.68E-03	2.10E-04	97222.5	921.1	2.11E-05	1.56E-06	
	HCOOH (3)	0.01197	0.30	2.56E-03	2.59E-04	97372.5	894.5	2.13E-05	2.13E-06	

Doping		Thickness	Applied voltage (V)	Current (mA)		Voltage (mV)	drop	σ	
ratio	Acid	(cm)	voltage	(mA)	SD	Avg.	SD	Avg.	SD
1	CSA (1)	0.01108	0.30	1.65E-01	1.84E-02	80612.5	1966.6	1.79E-03	1.67E-04
	CSA (2)	0.01205	0.33	1.75E-01	1.08E-02	80462.5	2257.4	1.75E-03	1.39E-04
	CSA (3)	0.01095	0.33	1.63E-01	1.16E-02	79902.5	2937.1	1.81E-03	1.08E-04
2	CSA (1)	0.01135	0.23	4.98E-01	1.93E-02	15932.5	343.3	2.67E-02	1.17E-02
	CSA (2)	0.01149	0.23	4.78E-01	2.52E-02	16342.5	529.6	2.47E-02	1.71E-01
	CSA (3)	0.01098	0.31	4.66E-01	2.88E-01	16072.5	412.8	2.56E-02	1.68E-02
4	CSA (1)	0.01356	0.31	7.44	1.90E-01	16162.5	457.5	3.30E-01	1.56E-02
	CSA (2)	0.01068	0.28	5.06	2.37E-01	19282.5	1650.5	2.40E-01	2.83E-02
	CSA (3)	0.01387	0.28	5.84	4.38E-01	18562.5	1230.4	2.21E-01	1.96E-02
10	CSA (1)	0.01659	0.34	9.13	3.37E-01	19822.5	1478.6	2.71E-01	2.61E-02
	CSA (2)	0.01353	0.34	8.99	3.28E-01	22002.5	763.0	2.93E-01	8.75E-02
	CSA (3)	0.01038	0.37	8.11	3.31E-01	24772.5	2867.5	3.10E-01	3.30E-02
20	CSA (1)	0.00996	0.37	8.34	3.63E-01	24012.5	2705.8	3.42E-01	3.76E-02
	CSA (2)	0.00978	0.42	7.09	2.56E-01	24852.5	3263.0	2.87E-01	3.78E-02
	CSA (3)	0.00861	0.42	7.47	5.23E-01	24712.50	3411.2	3.47E-01	5.76E-01
30	CSA (1)	0.01097	0.42	12.41	2.69E-01	25092.5	3006.5	4.44E-01	6.19E-02
	CSA (2)	0.00970	0.31	6.87	2.56E-01	18665.5	2126.3	3.72E-01	4.40E-02
	CSA (3)	0.00970	0.31	7.00	2.62E-01	18592.5	2082.4	3.81E-01	4.35E-02
40	CSA (1)	0.01044	0.31	9.52	4.98E-01	18892.5	806.2	4.69E-01	3.32E-02
	CSA (2)	0.01227	0.31	10.95	4.77E-01	18082.5	742.1	4.79E-01	1.90E-02
	CSA (3)	0.01249	0.36	7.90	2.82E-01	19622.5	1151.6	3.14E-01	2.00E-02
80	CSA (1)	0.01253	0.36	9.65	8.48E-01	18262.5	895.9	4.10E-01	4.38E-02
	CSA (2)	0.01252	0.37	9.95	3.40E-01	18212.5	919.5	4.24E-01	1.98E-02
	CSA (3)	0.01290	0.37	8.89	2.81E-01	19082.5	1039.0	3.51E-01	3.69E-02
160	CSA (1)	0.01344	0.55	1.31	8.19E-03	4602.5	115.5	2.05E-01	5.03E-02
	CSA (2)	0.01514	0.55	2.41	4.24E-02	4772.5	92.6	3.23E-01	5.49E-02
	CSA (3)	0.01044	0.55	1.30	1.20E-02	4602.5	115.5	2.63E-01	6.05E-02
250	CSA (1)	0.01264	0.27	6.70	7.86E-01	19672.5	1249.9	2.62E-01	3.63E-02
	CSA (2)	0.01093	0.27	4.66	7.04E-01	21602.5	1465.2	1.91E-01	2.34E-02
	CSA (3)	0.01304	0.30	7.55	3.72E-01	18512.5	1461.7	3.05E-01	2.35E-02
1	HNO ₃ (1)	0.01007	0.4	1.94E-02	1.70E-02	91102.5	1523.9	2.05E-03	2.00E-04
	HNO ₃ (2)	0.00910	0.4	1.60E-01	8.17E-03	92172.5	1699.7	1.85E-03	8.66E-05
	HNO ₃ (3)	0.01011	0.35	1.80E-01	1.56E-02	91822.5	3424.0	1.88E-03	1.92E-04

Table E-2 Raw data of conductivity measurement of PANI-H₂SO₄/CSA when exposed to 100%ethanol.

	Doping	Thickness	Applied voltage	Current (mA)		Voltage drop (mV)		σ (S/cm)	
ratio	Acid	(cm)	(V)	Avg.	SD	Avg.	SD	Avg.	SD
1	CSA (1)	0.01108	0.30	2.33E-01	1.40E-02	70424.3	1407.3	2.90E-03	1.87E-04
	CSA (2)	0.01205	0.33	2.31E-01	7.40E-03	6662.5	1885.9	2.79E-03	5.73E-05
2	CSA (1)	0.01135	0.23	1.21E+00	6.43E-02	18202.5	920.1	5.66E-02	8.58E-04
	CSA (2)	0.01149	0.23	1.14E+00	8.53E-02	18102.5	784.6	5.30E-02	2.16E-03
4	CSA (1)	0.01356	0.31	8.31E+00	2.39E-01	17762.5	356.5	3.35E-01	1.20E-02
	CSA (2)	0.01068	0.28	7.97E+00	3.30E-01	21302.5	828.7	3.40E-01	1.09E-02
10	CSA (1)	0.01659	0.34	1.05E+01	2.72E-01	16667.5	464.3	3.70E-01	5.63E-03
	CSA (2)	0.01353	0.34	9.69E+00	2.60E-01	18487.5	452.2	3.76E-01	3.10E-03
20	CSA (1)	0.00996	0.37	1.02E+01	2.50E-01	23822.5	1521.5	4.20E-01	2.65E-02
	CSA (2)	0.00978	0.42	1.03E+01	3.03E-01	23362.5	982.2	4.37E-01	1.46E-02
30	CSA (1)	0.01097	0.42	1.29E+01	5.32E-01	23612.5	1143.5	4.81E-01	6.68E-03
	CSA (2)	0.0097	0.31	1.06E+01	3.13E-01	22522.5	661.3	4.69E-01	5.45E-03
40	CSA (1)	0.01044	0.31	1.31E+01	1.94E-01	16342.5	469.5	6.23E-01	1.39E-02
	CSA (2)	0.01227	0.31	1.31E+01	1.55E-01	16642.5	327.3	6.24E-01	1.11E-02
80	CSA (1)	0.01253	0.36	1.22E+01	2.72E-01	18022.5	887.9	5.23E-01	2.35E-02
	CSA (2)	0.01252	0.37	1.24E+01	2.64E-01	18712.5	652.3	5.12E-01	1.98E-02
160	CSA (1)	0.01344	0.55	1.29E+00	3.00E-02	3602.5	115.5	2.59E-01	1.12E-02
	CSA (2)	0.01514	0.55	1.49E+00	9.18E-02	3587.5	184.2	2.67E-01	2.55E-02

Table E-3 Raw data of sensitivity of PANI-H₂SO₄/CSA at N_A/N_{EB} equal to 1:40 when exposed to ethanol.

Ethanol	Thickness	Applied	Current	(mA)	Volt drop	before (mV)	Volt drop	after (mV)	σ (S/cm)	before	σ (S/cm)	after	$\Delta\sigma$ (S/cm)
(%)	(cm)	Voltage (V)	Avg.	SD.	Avg.	SD.	Avg.	SD.	Avg.	SD.	Avg.	SD.	
0	0.00997	0.35	1.0142	0.0214	4559.0	26.9	2924.0	65.3	0.2164	0.00153	0.3374	0.0023	0.1210
	0.01104	0.35	1.1057	0.0196	4888.2	46.1	3042.9	42.3	0.1987	0.00215	0.3192	0.0048	0.1205
1	0.00993	0.55	1.1629	0.0265	6082.5	44.7	3879.2	40.4	0.1800	0.0012	0.2943	0.0025	0.1142
	0.01223	0.55	0.7056	0.0192	3327.8	29.3	1960.6	16.1	0.1681	0.0012	0.2845	0.0029	0.1164
2.3	0.01014	0.55	1.1002	0.0281	5582.5	83.7	4167.5	35.4	0.1835	0.0014	0.2628	0.0048	0.0826
	0.01429	0.55	1.2157	0.0201	5021.8	71.3	3276.7	25.7	0.1643	0.0012	0.2518	0.0053	0.0875
4.6	0.01344	0.55	1.4106	0.0201	4702.5	27.9	4202.5	253.9	0.2108	0.0067	0.2436	0.0165	0.0662
	0.01426	0.55	1.4283	0.0119	5355.5	37.2	3844.4	21.3	0.1810	0.0005	0.2527	0.0143	0.0713
9.2	0.01249	0.55	1.4112	0.0183	4515.0	138.4	3865.0	197.4	0.2380	0.0091	0.2878	0.0146	0.0629
	0.01149	0.52	1.5675	0.0720	6111.2	100.1	4797.3	107.3	0.2165	0.0074	0.2758	0.0153	0.0593

Ethanol	Thickness	Applied	Current	(mA)	Volt drop	before (mV)	Volt drop	after (mV)	σ (S/cm)	before	σ (S/cm)	after	$\Delta\sigma$ (S/cm)
(%)	(cm)	Voltage (V)	Avg.	SD.	Avg.	SD.	Avg.	SD.	Avg.	SD.	Avg.	SD.	
15	0.01302	0.55	1.2819	0.0116	5982.5	44.7	5424.1	89.4	0.1570	0.0012	0.1758	0.0017	0.0188
	0.01416	0.46	1.3231	0.0284	5223.3	40.6	4724.9	74.2	0.1735	0.0016	0.1918	0.0011	0.0183
20	0.01314	0.46	1.3707	0.0156	5835.8	81.6	5513.7	85.4	0.1724	0.0035	0.1835	0.0025	0.0110
	0.01171	0.46	1.2144	0.0149	5549.1	46.8	5218.1	39.1	0.1812	0.0015	0.1928	0.0017	0.0115
30	0.01038	0.29	1.1238	0.0169	5534.7	56.1	5255.9	42.5	0.1897	0.0017	0.1998	0.0013	0.00976
	0.01159	0.29	1.1179	0.0117	4595.5	72.3	4384.2	39.7	0.2036	0.0021	0.2134	0.0027	0.00981

TableE-4 Raw data of sensitivity of PANI-H₂SO₄/CSA at N_A/N_{EB} equal to 1:160 when exposed to ethanol.

Ethanol (%)	Thickness (cm)	Applied Voltage (V)	Current Avg.	(mA)	Volt drop before (mV)	Volt drop after (mV)	σ (S/cm)	before SD.	σ (S/cm)	after SD.	$\Delta\sigma$ (S/cm)		
			SD.	Avg.	SD.	Avg.	SD.	Avg.	SD.	Avg.			
0	0.01271	0.35	1.0173	0.0121	4136.3	36.2	2514.8	30.1	0.1877	0.0027	0.3087	0.0014	0.1210
	0.01151	0.35	1.0694	0.0097	4520.3	59.2	2791.1	67.3	0.1993	0.0063	0.3228	0.0011	0.1235
1	0.01162	0.64	0.9409	0.0908	4054.4	17.8	2484.5	46.3	0.1937	0.0071	0.3161	0.0018	0.1224
	0.01042	0.64	1.0176	0.0867	4419.6	26.9	2820.5	21.1	0.2143	0.0050	0.3358	0.0021	0.1215
2.3	0.01182	0.39	1.1703	0.0238	4207.7	39.3	3154.4	47.2	0.2282	0.0035	0.3044	0.0026	0.0762
	0.01239	0.48	1.1777	0.0375	3782.2	45.1	3039.1	49.7	0.2437	0.0032	0.3033	0.0028	0.0596
4.6	0.01149	0.48	1.2439	0.0069	4522.6	102.3	3800.6	99.7	0.2322	0.0032	0.2763	0.0019	0.0441
	0.01064	0.25	0.7015	0.0116	3068.6	66.4	2491.0	69.9	0.2084	0.0050	0.2517	0.0164	0.0433

Ethanol	Thickness	Applied	Current	(mA)	Volt drop	before (mV)	Volt drop	after (mV)	σ (S/cm)	before	σ (S/cm)	after	$\Delta\sigma$ (S/cm)
(%)	(cm)	Voltage (V)	Avg.	SD.	Avg.	SD.	Avg.	SD.	Avg.	SD.	Avg.	SD.	
9.2	0.01084	0.48	1.2123	0.0443	5102.5	60.8	4552.5	70.7	0.2120	0.0014	0.2547	0.0051	0.0427
	0.01134	0.48	1.1182	0.0307	4335.3	79.8	3651.6	63.1	0.2206	0.0057	0.2619	0.0052	0.0413
15	0.00780	0.53	0.9246	0.0222	7505.3	52.3	6813.2	49.7	0.1531	0.0042	0.1687	0.0036	0.0156
	0.00830	0.53	0.9942	0.0051	5406.6	67.7	5002.3	41.4	0.2149	0.0115	0.2319	0.0048	0.0170
20	0.01064	0.53	1.0696	0.0407	5415.0	72.1	5200.6	63.1	0.1800	0.0007	0.1875	0.0032	0.00742
	0.00964	0.47	1.0754	0.0398	4819.2	43.2	4617.3	56.2	0.2245	0.0062	0.2343	0.0071	0.00982
30	0.01017	0.30	1.0534	0.0097	4987.1	41.3	4764.1	49.1	0.2014	0.0011	0.2109	0.0025	0.00943
	0.01062	0.30	1.0050	0.0128	4651.3	39.2	4473.3	49.9	0.1973	0.0022	0.2052	0.0031	0.00785

CURRICULUM VITAE

Name : Jutharat Amornlertratanatada

Date of Birth: April 4, 1978

Nationality : Thai

University Education:

1995-1998 : Bachelor Degree of Science in Material
Science, Faculty of Science, Chulalongkorn University,
Bangkok, Thailand.