

## REFERENCES

- Azemar, N., Carrera, I., and Solans, C. (1993) Studies on textile detergency at low temperature. Journal of Dispersion Science and Technology, 4(6), 645-660.
- Azemar, N. (1997) The role of microemulsions in detergency process. In Solans, C., and Kunieda, H. (Eds.). Industrial Applications of Microemulsions. New York: Marcel Dekker.
- Bourrel, M. and Schechter, R.S. (1988) Formulation, solvency and physical properties. Microemulsions and Related System. New York: Marcel Dekker.
- Broze, G. and Lange, K.R. (1994) A Handbook for Formulators. Munich: Hanser.
- Carroll, B. (1996) The Direct Study of Oily Soil removal from Solid Substrates in Detergency. Colloids and Surfaces, 114, 161-164.
- Chi, Y.S. and Obendorf, S.K. (1999) Detergency of used motor oil from cotton and polyester fabrics. Journal of Surfactants and Detergents, 2(1), 1-11.
- Chi, Y.S. and Obendorf, S.K. (2001) Effect of fiber substrates on appearance and removal of aged oily soils. Journal of Surfactants and Detergents, 4(1), 35-41.
- Christ, T., Morgenthaler, W.W., and Pacholec, F. (1994) Laundry products. In Lange, K.R. (Eds.), Detergents and Cleaners: A Handbook for Formulators. New York: Hanser.
- Dörfler, H.D., Grosse, A., and Krüssmann, H. (1996) The use of microemulsions as cleaning media. Tenside Surfactants and Detergents, 33(6), 432-440.
- Germain, T. (2002) Understanding Terg-O-Tomerter testing variables, The paper presented on American Oil Chemist's Society Annual Meeting, Montreal, Canada.
- Gordon, B.E., Roddewig, J., and Shebs, W.T. (1967) A double label radiotracer approach to detergency studies. Journal of American Oil Chemist's Society, 44, 289-294.
- Goel, S.K. (1998) Measuring detergency of oily soil in the vicinity of phase inversion temperatures of commercial nonionic surfactants using an oily-soluble dye. Journal of Surfactants and Detergents, 1(2), 221-225.

- Goel, S.K. (1998) Selecting the optimal linear alcohol ethoxylate for enhanced oily soil removal. Journal of Surfactants and Detergents, 1(2), 213-219.
- Kissa, E. (1987) Evaluation of detergency. In Culter, W.G., and Kissa, E. (Eds.). Detergency Theory and Technology. New York: Marcel Dekker.
- Kissa, E. (1971) Kinetics of oily soils release. Textile Research Journal, 47, 760.
- Linfield, W.M., Jungermann, E., and Sherrill, J.C. (1962) Establishment of a standardized detergency evaluation method. Journal of American Oil Chemist's Society, 34, 47-55.
- Morris, M.A. and Prato, H.H. (1982) The effect of wash temperature on removal of particulate and oily soil from fabrics of varying fiber content. Textile Research Journal, 52, 280-286.
- Obendorf, S.K. and Klemash, N.A. (1982) Electron microscopical analysis of oily soil penetration into cotton and polyester/cotton fabrics. Textile Research Journal, 434-442.
- Obendorf, S.K. and Borsa, J. (2001) Lipid soil removal from cotton fabric after mercerization and carboxymethylation finishing. Journal of Surfactants and Detergents, 4(3), 247-256.
- Raney, K.H. and Benson, H.L. (1990) The effect of polar soil components on the phase inversion temperature and optimum detergency conditions. Journal of American Oil Chemist's Society, 67(11), 722-729.
- Raney, K.H., Benton, W.J., and Miller, C.A. (1987) Optimum detergency conditions with nonionic surfactants. Journal of Colloid and Interface Science, 117(1), 282-290.
- Rosen, M.J. (1988) Surfactants and Interfacial Phenomena. 2<sup>nd</sup> ed. New York: John Wiley.
- Scott, B.A. (1963). Mechanism of fatty soil removal. Journal of Applied Chemistry, 13, 133.
- Solans, C., Domínguez, J.G., and Friberg, S.E. (1985) Evaluation of textile detergent efficiency of microemulsions in systems of water nonionic surfactant and hydrocarbon at low temperature. Journal of Dispersion Science and Technology, 6(5), 523-537.

- Solans, C., Azemar, N., and Parra, J.L. (1988) Proceeding of CESIO 2<sup>nd</sup> world surfactants congress. Journal of Calbet, 2(6), 421-429.
- Solans, C. and Azemar, N. (1992) Detergency and the HLB temperature. In Friberg, S.E., and Linman, B. (Eds.). Organized Solutions. New York: Marcel Dekker.
- Thompson, L. (1994) The role of oil detachment mechanisms in determining optimum detergency conditions. Journal of Colloid Interface Science, 163, 61-73.
- Tungsubutra, T. and Miller, C.A. (1992) Equilibrium and dynamic behavior of a system containing C12E6, water, triolein and oleyl alcohol. In Friberg, S.E., and Linman, B. (Eds.). Organized Solutions. New York: Marcel Dekker.
- Verma, S. and Kumar, V.V. (1998) Relationship between oil-water interfacial tension and oily soil removal in mixed surfactant systems. Journal of Colloid and Interface Science, 207, 1-10.
- Webb, J.J. and Obendorf, S.K. (1988) Detergency study of the synergism between oily and particulate soil on polyester/cotton fabric. Journal of American Oil Chemist's Society, 67(11), 722-729.
- Whang, H.S., Kim, Y.J., and Ko, S.W. (2001) Effect of hydrophile-lipophile balance values of surfactant mixtures on the detergency of oily-soiled single fiber. Textile Research Journal, 71(7), 650-654.
- Winsor, P. (1954) Solvent Properties of Amphiphilic Compounds. London: Butterworth.
- Wu, B., Cheng, H., Childs, J.D., and Sabatani, D.A. (2000) Surfactant-enhanced removal of hydrophobic oil contamination. In Smith, J.A., and Burns, S.E. (Eds.). Physicochemical Ground Water Remediation. New York: Plenum.

## APPENDICES

### Appendix A Experimental Data of Microemulsion Formation

#### 1. Solubilization parameter (SP)

The solubilization parameter of oil (SP<sub>o</sub>) and water (SP<sub>w</sub>) are designed as :

$$SP_o = \frac{V_o}{M_s} \quad \text{and} \quad SP_w = \frac{V_w}{M_s} \quad (\text{A.1})$$

Where  $V_o$  = volume of oil solubilized  
 $M_s$  = weight of surfactants  
 $V_w$  = volume of water solubilized

#### 2. Interfacial tension (IFT)

The interfacial tension of each phase of microemulsion is calculated by following formulation :

$$IFT = e (Vd)^3 n^2 \Delta\rho \quad (\text{A.2})$$

where  $\sigma$  = interfacial tension (mN/m)  
 $n$  = number of revolution (rpm)  
 $e = 3.427 \times 10^{-4}$  (mN cm<sup>3</sup> min<sup>2</sup>/m g mm<sup>3</sup>)  
 $V = 0.31$  (mm/sdv)  
 $D$  = measured drop diameter (sdv)  
 $\Delta\rho$  = density difference of two liquids (g/cm<sup>3</sup>)

### 3. Experiment data of solubilization parameter

**Table A-1** Volume fractions of water, middle, and oil phase and solubilization parameters of oil and water phases in microemulsion formation with 0.5wt% ADPODS, 5wt% AOT and 5wt% Span80 at different NaCl concentrations by using an oil to water ratio of 1:1

Sample number	%NaCl	Phase height (cm)			Relative volume			Ms	Solubilization		Rel. vol. Water+Mid	Rel. vol. Water
		Water	Middle	Oil	Water	Middle	Oil		Spw	Spo		
1	0	3.78	0	3.00	0.5577	0	0.4429	0.5025	9.9502	1.1486	0.5577	0
2	0.05	4.02	0	2.81	0.5889	0	0.4112	0.5025	9.9502	1.7679	0.5889	0
3	0.07	4.14	0	2.60	0.6144	0	0.3856	0.5025	9.9502	2.2758	0.6144	0
4	0.15	2.52	1.75	2.49	0.3730	0.2587	0.3684	0.5025	2.5280	2.6192	0.6316	0.3730
5	0.2	2.56	1.64	2.60	0.3768	0.2414	0.3818	0.5025	2.4521	2.3525	0.6182	0.3768
6	0.3	2.84	1.46	2.57	0.4134	0.2122	0.3745	0.5025	1.7239	2.4981	0.6256	0.4134
7	0.31	2.89	1.58	2.36	0.4236	0.2316	0.3448	0.5025	1.5205	3.0877	0.6552	0.4236
8	0.32	2.97	1.21	2.63	0.4364	0.1778	0.3858	0.5025	1.2659	2.2733	0.6142	0.4364
9	0.35	3.05	1.20	2.57	0.4474	0.1756	0.3770	0.5025	1.0466	2.4478	0.6230	0.4474
10	0.43	3.00	1.50	2.29	0.4417	0.2206	0.3377	0.5025	1.1606	3.2300	0.6623	0.4417
11	0.45	2.94	1.77	2.06	0.4345	0.2615	0.3039	0.5025	1.3027	3.9023	0.6961	0.4345
12	0.44	2.99	1.53	2.29	0.4365	0.2245	0.3360	0.5025	1.2041	3.2641	0.6640	0.4395
13	0.47	2.760	2.11	1.99	0.4022	0.3074	0.2904	0.5025	1.9465	4.1716	0.7096	0.4022
14	0.48	2.73	2.21	1.84	0.4028	0.3259	0.2713	0.5025	1.9348	4.5517	0.7287	0.4028
15	0.49	2.82	2.29	1.74	0.4112	0.3345	0.2543	0.5025	1.7670	4.8887	0.7457	0.4112
16	0.5	2.76	2.23	1.87	0.4026	0.3251	0.2723	0.5025	1.9384	4.5307	0.7277	0.4026
17	0.6	2.73	2.54	1.53	0.4018	0.3734	0.2248	0.5025	1.9550	5.4760	0.7752	0.4018
18	0.75	2.80	2.55	1.46	0.4114	0.3746	0.2140	0.5025	1.7628	5.6915	0.7860	0.4114
19	0.8	2.80	2.70	1.29	0.4129	0.3971	0.1900	0.5025	1.7333	6.1700	0.8100	0.4129
20	1.0	2.87	2.60	1.31	0.4233	0.3833	0.1935	0.5025	1.5269	6.1000	0.8065	0.4233

**Table A-2** Volume fractions of water, middle, and oil phase and solubilization parameters of oil and water phases in microemulsion formation with 0.75wt% ADPODS, 5wt% AOT and 5wt% Span80 at different NaCl concentrations by using initial oil to water ratio of 1:1

Sample number	%NaCl	Phase height (cm)			Relative volume			Ms	Solubilization		Rel. vol. Water+Mid	Rel. vol. Water
		Water	Middle	Oil	Water	Middle	Oil		Spw	Spo		
1	0	3.75	0	3.20	0.5391	0	0.4609	0.5375	9.3023	0.7283	0.5412	0
2	0.1	3.80	0	3.00	0.5588	0	0.4412	0.5375	9.3023	1.0941	0.5436	0
3	0.2	3.93	0	2.88	0.5775	0	0.4225	0.5375	9.3023	1.4421	0.5467	0
4	0.3	3.86	0	2.93	0.5685	0	0.4315	0.5375	9.3023	1.2742	0.5483	0
5	0.4	2.66	1.79	2.33	0.3930	0.2637	0.3434	0.5375	1.9912	2.9140	0.6566	0.3930
6	0.5	2.82	1.50	2.41	0.4185	0.2230	0.3585	0.5375	1.5168	2.6329	0.6415	0.4185
7	0.6	2.60	2.36	1.85	0.3825	0.3460	0.2715	0.5375	2.1869	4.2570	0.72848	0.3825
8	0.7	2.58	2.59	1.62	0.3801	0.3817	0.2381	0.5375	2.2304	4.8717	0.7619	0.3801
9	0.8	2.66	2.63	1.51	0.3910	0.3866	0.2224	0.5375	2.0275	5.1652	0.7776	0.3910
10	0.9	2.75	2.65	1.43	0.4025	0.3877	0.2099	0.5375	1.8147	5.3979	0.79014	0.4025
11	1.0	3.00	2.34	1.51	0.4384	0.3416	0.2200	0.5375	1.1460	5.2092	0.7800	0.4384

**Table A-3** Volume fractions of water, middle, and oil phase and solubilization parameters of oil and water phases in microemulsion formation with 1wt% ADPODS, 5wt% AOT and 5wt% Span80 at different NaCl concentrations by using initial oil to water ratio of 1:1

Sample number	%NaCl	Phase height (cm)			Relative volume			Ms	Solubilization		Rel. vol. Water+Mid	Rel. vol. Water
		Water	Middle	Oil	Water	Middle	Oil		Spw	Spo		
1	0	3.61	0	3.32	0.5216	0	0.4784	0.55	9.0909	0.3934	0.5216	0
2	0.2	3.62	0	3.30	0.5233	0	0.4767	0.55	9.0909	0.4245	0.5233	0
3	0.4	3.60	0	3.24	0.5262	0	0.4738	0.55	9.0909	0.4767	0.5262	0
4	0.6	2.76	1.18	2.90	0.4037	0.1722	0.4240	0.55	1.7502	1.3810	0.5760	0.4037
5	0.7	2.34	2.53	1.94	0.3434	0.3718	0.2848	0.55	2.8470	3.9121	0.7152	0.3434
6	0.72	2.30	2.73	1.81	0.3358	0.3989	0.2653	0.55	2.9851	4.2669	0.7347	0.3358
7	0.75	2.42	2.55	1.86	0.3549	0.3731	0.2720	0.55	2.6381	4.1448	0.7280	0.3549
8	0.77	2.45	2.63	1.75	0.3584	0.3858	0.2558	0.55	2.5746	4.4402	0.7442	0.3584
9	0.8	2.44	2.60	1.77	0.3582	0.3815	0.2604	0.55	2.5791	4.3572	0.7396	0.3582
10	0.83	2.45	2.52	1.86	0.3587	0.3917	0.2725	0.55	2.5687	4.1368	0.7504	0.3587
11	0.85	2.47	2.49	1.92	0.3591	0.3620	0.2788	0.55	2.5611	4.0216	0.7212	0.3591
12	0.87	2.43	2.52	1.81	0.3598	0.3731	0.2671	0.55	2.5492	4.2344	0.7329	0.3598
13	0.89	2.48	2.58	1.72	0.3662	0.3802	0.2536	0.55	2.4327	4.4804	0.7464	0.3662
14	1.0	2.52	2.49	1.74	0.3729	0.3694	0.2577	0.55	2.3118	4.4053	0.7423	0.3729
15	1.2	2.55	2.58	1.65	0.3768	0.3800	0.2432	0.55	2.2405	4.6689	0.7568	0.3768
16	1.4	2.63	2.52	1.66	0.3864	0.3702	0.2434	0.55	2.0647	4.6655	0.7566	0.3864
17	1.5	2.61	2.51	1.66	0.3848	0.3700	0.2451	0.55	2.0942	4.6341	0.7549	0.3848
18	1.6	2.65	2.50	1.69	0.3875	0.3657	0.2469	0.55	2.0457	4.6027	0.7531	0.3875
19	1.8	2.72	2.48	1.60	0.3998	0.3652	0.2351	0.55	1.8225	4.8171	0.7649	0.3998
20	2.0	2.70	2.44	1.92	0.3995	0.3608	0.2397	0.55	1.8268	4.7330	0.7603	0.3995

**Table A-4** Volume fractions of water, middle, and oil phase and solubilization parameters of oil and water phases in microemulsion formation with 0.5wt% ADPODS, 3wt% AOT and 5wt% Span80 at different NaCl concentrations by using initial oil to water ratio of 1:1

Sample number	%NaCl	Phase height (cm)			Relative volume			Ms	Solubilization		Rel. vol. Water	Rel. vol. Oil
		Water	Middle	Oil	Water	Middle	Oil		Spw	Spo		
1	0	3.60	0	3.25	0.5260	0	0.4740	0.4025	12.4224	0.6456	0.5260	0.4740
2	0.1	3.38	0	3.49	0.4924	0	0.5076	0.4025	12.4224	0	0.4924	0.5076
3	0.2	3.44	0	3.43	0.5004	0	0.4996	0.4025	12.4224	0.0090	0.5004	0.4996
4	0.3	3.14	0	3.72	0.4573	0	0.5427	0.4025	1.0608	2.0125	0.4573	0.5427
5	0.4	3.20	0	3.67	0.4656	0	0.5344	0.4025	0.8552	2.0125	0.4656	0.5344
6	0.5	3.04	0	3.86	0.4409	0	0.5591	0.4025	1.4683	2.0125	0.4409	0.5591
7	0.6	3.07	0	3.78	0.4484	0	0.5516	0.4025	1.2811	2.0125	0.4484	0.5516
8	0.7	3.05	0	3.84	0.4430	0	0.5570	0.4025	1.4155	2.0125	0.4430	0.5570
9	0.8	3.09	0	3.81	0.4478	0	0.5522	0.4025	1.2961	2.0125	0.4478	0.5522
10	0.9	3.07	0	3.79	0.4478	0	0.5522	0.4025	1.2973	2.0125	0.4478	0.5522
11	1.0	3.08	0	3.81	0.4469	0	0.5531	0.4025	1.3197	2.0125	0.4469	0.5531
12	1.2	3.04	0	3.82	0.4430	0	0.5570	0.4025	1.4169	2.0125	0.4430	0.5570
13	1.3	2.99	0	3.88	0.4357	0	0.5643	0.4025	1.5985	2.0125	0.4357	0.5643
14	1.4	2.99	0	3.71	0.4337	0	0.5663	0.4025	1.6472	2.0125	0.4337	0.5663
15	1.5	3.20	0	3.66	0.4667	0	0.5333	0.4025	0.8284	2.0125	0.4667	0.5333
16	1.6	3.21	0	3.67	0.4661	0	0.5339	0.4025	0.8412	2.0125	0.4661	0.5339
17	1.7	3.24	0	3.65	0.4704	0	0.5296	0.4025	0.7345	2.0125	0.4704	0.5296
18	1.8	3.24	0	3.60	0.4733	0	0.5267	0.4025	0.6634	2.0125	0.4733	0.5267
19	1.9	3.19	0	3.66	0.4653	0	0.5347	0.4025	0.8610	2.0125	0.4653	0.5347
20	2.0	3.21	0	3.68	0.4652	0	0.5348	0.4025	0.8649	2.0125	0.4652	0.5348



**Table A-5** Volume fractions of water, middle, and oil phase and solubilization parameters of oil and water phases in microemulsion formation with 0.5wt% ADPODS, 4wt% AOT and 5wt% Span80 at different NaCl concentrations by using initial oil to water ratio of 1:1

Sample number	%NaCl	Phase height (cm)			Relative volume			Ms	Solubilization		Rel. vol. Water	Rel. vol. Oil
		Water	Middle	Oil	Water	Middle	Oil		Spw	Spo		
1	0	3.08	0	3.77	0.4498	0	0.5502	0.4525	2.2625	0	0.4498	0.5502
2	0.1	4.15	0	2.68	0.6073	0	0.3927	0.4525	2.2625	2.3705	0.6073	0.3927
3	0.3	3.00	0	3.83	0.4395	0	0.5605	0.4525	1.374	11.0497	0.4395	0.5605
4	0.4	3.17	0	3.68	0.4627	0	0.5373	0.4525	0.8252	11.0497	0.4627	0.5373
5	0.5	2.89	0	3.90	0.4260	0	0.5740	0.4525	1.6357	11.0497	0.4260	0.5740
6	0.6	2.95	0	2.88	0.4322	0	0.5678	0.4525	1.4985	11.0497	0.4322	0.5678
7	0.7	2.91	0	3.90	0.4272	0	0.5728	0.4525	1.6082	11.0497	0.4272	0.5728
8	0.75	2.83	0	3.93	0.4189	0	0.5811	0.4525	1.7918	11.0497	0.4189	0.5811
9	0.8	2.91	0	3.90	0.4272	0	0.5728	0.4525	1.6087	11.0497	0.4272	0.5728
10	0.9	2.92	0	3.89	0.4284	0	0.5716	0.4525	1.5822	11.0497	0.4284	0.5716
11	1.0	2.86	0	4.01	0.4161	0	0.5839	0.4525	1.8548	11.0497	0.4161	0.5839
12	1.5	3.12	0	3.59	0.4649	0	0.5351	0.4525	0.7750	11.0497	0.4649	0.5351
13	2.0	3.24	0	3.53	0.4789	0	0.5211	0.4525	0.4669	11.0497	0.4789	0.5211

**Table A-6** Volume fractions of water, middle, and oil phase and solubilization parameters of oil and water phases in microemulsion formation with 0.5wt% ADPODS, 5wt% AOT and 1wt% Span80 at different NaCl concentrations by using initial oil to water ratio by 1:1

Sample number	%NaCl	Phase height (cm)			Relative volume			Ms 0.3025	Solubilization		Rel. vol. Water+Mid	Rel. vol. Water
		Water	Middle	Oil	Water	Middle	Oil		Spw	Spo		
1	0	3.71	0	3.14	0.5414	0	0.4586	0.3025	16.5289	1.3670	0.5414	0
2	0.1	3.62	0	3.24	0.5280	0	0.4720	0.3025	16.5289	0.9247	0.5280	0
3	0.2	3.66	0	3.15	0.5369	0	0.4631	0.3025	16.5289	1.2188	0.5369	0
4	0.3	3.72	0	3.12	0.5434	0	0.4566	0.3025	16.5289	1.4362	0.5434	0
5	0.4	3.65	0	3.17	0.5357	0	0.4643	0.3025	16.5289	1.1786	0.5357	0
6	0.5	3.61	0	3.29	0.5231	0	0.4769	0.3025	16.5289	0.7627	0.5231	0
7	0.6	3.62	0	3.25	0.5269	0	0.4731	0.3025	16.5289	0.8877	0.5269	0
8	0.7	3.69	0	3.19	0.5369	0	0.4631	0.3025	16.5289	1.2186	0.5369	0
9	0.8	3.60	0	3.22	0.5278	0	0.4722	0.3025	16.5289	0.9202	0.5278	0
10	0.9	3.62	0	3.22	0.5292	0	0.4708	0.3025	16.5289	0.9646	0.5292	0
11	1.0	3.63	0	3.27	0.5258	0	0.4742	0.3025	16.5289	0.8535	0.5258	0
12	1.1	3.43	0.17	3.25	0.5006	0.0253	0.4841	0.3025	0.2361	0.8549	0.5259	0.5006
13	1.2	3.45	0.27	3.16	0.5018	0.0392	0.4589	0.3025	0.2352	1.3576	0.5411	0.5018
14	1.3	3.37	0.28	3.18	0.4928	0.0413	0.4659	0.3025	0.2371	1.1274	0.5341	0.4928
15	1.4	3.38	0.17	3.31	0.4931	0.0243	0.4826	0.3025	0.2289	0.5758	0.5174	0.4931
16	1.5	3.41	0.17	3.28	0.4974	0.0248	0.4778	0.3025	0.0868	0.7327	0.5222	0.4974
17	1.6	3.47	0.12	3.27	0.5058	0.0174	0.4759	0.3025	0.0985	0.7644	0.5231	0.5058
18	1.7	3.40	0.14	3.31	0.4966	0.0207	0.4826	0.3025	0.1110	0.5743	0.5174	0.4966
19	1.8	3.47	0.14	3.23	0.5075	0.0203	0.4722	0.3025	0.1259	0.9183	0.5278	0.5075
20	2.0	3.38	0.22	3.26	0.4934	0.0318	0.4748	0.3025	0.2170	0.8341	0.5252	0.4934

**Table A-7** Volume fractions of water, middle, and oil phase and solubilization parameters of oil and water phases in microemulsion formation with 0.5wt% ADPODS, 5wt% AOT and 2wt% Span80 at different NaCl concentrations by using initial oil to water ratio of 1:1

Sample number	%NaCl	Phase height (cm)			Relative volume			Ms	Solubilization		Rel. vol. Water+Mid	Rel. vol. Water
		Water	Middle	Oil	Water	Middle	Oil		Spw	Spo		
1	0	3.66	0	3.25	0.5299	0	0.4701	0.3525	14.1844	0.8472	0.5299	0
2	0.1	3.65	0	3.13	0.5383	0	0.4617	0.3525	14.1844	1.0789	0.5383	0
3	0.2	3.63	0	3.16	0.5351	0	0.4649	0.3525	14.1844	0.9944	0.5351	0
4	0.3	3.62	0	3.16	0.5337	0	0.4663	0.3525	14.1844	0.9559	0.5337	0
5	0.4	3.69	0	3.11	0.5426	0	0.4574	0.3525	14.1844	1.2091	0.5426	0
6	0.5	3.66	0	3.13	0.5391	0	0.4609	0.3525	14.1844	1.1078	0.5391	0
7	0.6	3.70	0	3.09	0.5451	0	0.4549	0.3525	14.1844	1.2789	0.5451	0
8	0.7	3.66	0	3.10	0.5417	0	0.4583	0.3525	14.1844	1.1841	0.5417	0
9	0.8	2.79	1.18	2.85	0.4087	0.1732	0.4181	0.3525	2.5890	2.3231	0.5819	0.4087
10	0.9	2.69	1.16	2.71	0.4104	0.1765	0.4131	0.3525	2.5420	2.4642	0.5869	0.4104
11	1.0	2.72	1.52	2.56	0.4004	0.2231	0.3765	0.3525	2.8269	3.5033	0.6235	0.4004
12	1.5	2.88	1.44	2.52	0.4209	0.2104	0.3687	0.3525	2.2440	3.7262	0.6313	0.4209
13	2.0	2.89	1.69	2.22	0.4251	0.2490	0.3259	0.3525	2.1247	4.9382	0.6741	0.4251

**Table A-8** Volume fractions of water, middle, and oil phase and solubilization parameters of oil and water phases in microemulsion formation with 0.5wt% ADPODS, 5wt% AOT and 3wt% Span80 at different NaCl concentrations by using initial oil to water ratio of 1:1

Sample number	%NaCl	Phase height (cm)			Relative volume			Ms	Solubilization		Rel. vol. Water+Mid	Rel. vol. Water
		Water	Middle	Oil	Water	Middle	Oil		Spw	Spo		
1	0	3.64	0	3.17	0.5345	0	0.4655	0.4025	12.4224	0.8563	0.5345	0
2	0.1	3.66	0	3.12	0.5403	0	0.4597	0.4025	12.4224	1.0001	0.5403	0
3	0.2	3.70	0	3.07	0.5470	0	0.4530	0.4025	12.4224	1.1673	0.5470	0
4	0.3	2.08	1.87	2.99	0.2996	0.2691	0.4313	0.4025	4.9779	1.7071	0.5687	0.2996
5	0.4	2.36	1.73	2.66	0.3492	0.2566	0.392	0.4025	3.7460	2.6279	0.6058	0.3492
6	0.5	2.57	1.70	2.49	0.3803	0.2512	0.3686	0.4025	2.9749	3.2657	0.6315	0.3803
7	0.6	2.64	2.29	1.82	0.3909	0.3392	0.2699	0.4025	2.7111	5.7164	0.7301	0.3909
8	0.7	2.67	2.11	1.99	0.3950	0.3117	0.2932	0.4025	2.6078	5.1367	0.7067	0.3950
9	0.8	2.68	2.20	1.92	0.3935	0.3240	0.2825	0.4025	2.6452	5.4037	0.7175	0.3935
10	0.9	2.64	2.14	1.95	0.3928	0.3173	0.2899	0.4025	2.6639	5.2189	0.7101	0.3928
11	1.0	2.72	2.13	1.94	0.4008	0.3134	0.2858	0.4025	2.4640	5.3213	0.7142	0.4008
12	1.1	2.89	2.02	1.80	0.4308	0.3005	0.2686	0.4025	1.7185	5.7482	0.7313	0.4308
13	1.2	2.81	2.15	1.83	0.4137	0.3162	0.2701	0.4025	2.1446	5.7121	0.7299	0.4137
14	1.3	2.84	2.12	1.93	0.4153	0.3086	0.2761	0.4025	2.1033	5.5635	0.7239	0.4153
15	1.4	2.88	2.08	1.86	0.4224	0.3049	0.2726	0.4025	1.9268	5.6495	0.7273	0.4224
16	1.5	2.87	2.01	1.78	0.4181	0.2930	0.2888	0.4025	2.0336	5.2468	0.7111	0.4181
17	1.6	2.97	1.84	1.98	0.4374	0.2707	0.2919	0.4025	1.5558	5.1694	0.7081	0.4374
18	1.7	2.92	1.70	2.24	0.4264	0.2474	0.3262	0.4025	1.8295	4.3173	0.6738	0.4264
19	1.8	2.95	1.83	2.03	0.4331	0.2682	0.2988	0.4025	1.6630	5.0000	0.7013	0.4331
20	2.0	3.06	1.64	2.10	0.4503	0.2404	0.3093	0.4025	1.2349	4.7388	0.6907	0.4503

## Appendix B Experimental Data of Detergency Experiment

### 1. %Detergency (%D)

The detergency performance can be determined by %Detergency, it is calculated from the following equation :

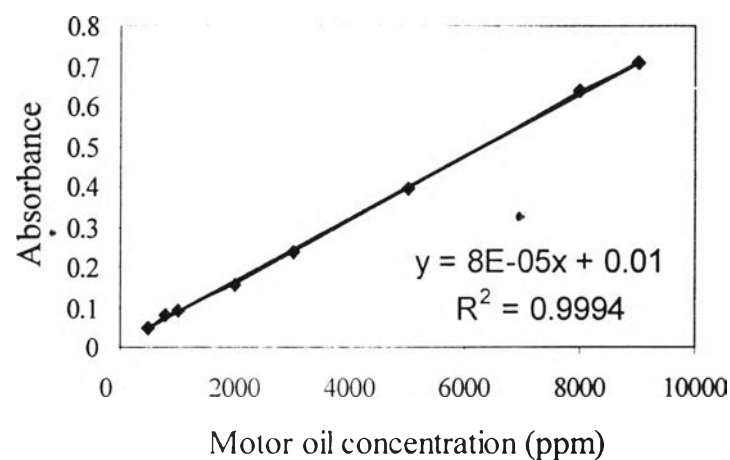
$$\% \text{ Detergency} = |(A-B)/(Co-Bo)| \times 100$$

where

- A = average reflectance of the soiled swatches after washing
- B = average reflectance of the soiled swatches before washing
- Co = average reflectance of the unsoiled swatches before washing

### 2. %Oil Removal

The oil removal is calculated from the calibration curve for colored motor oil.



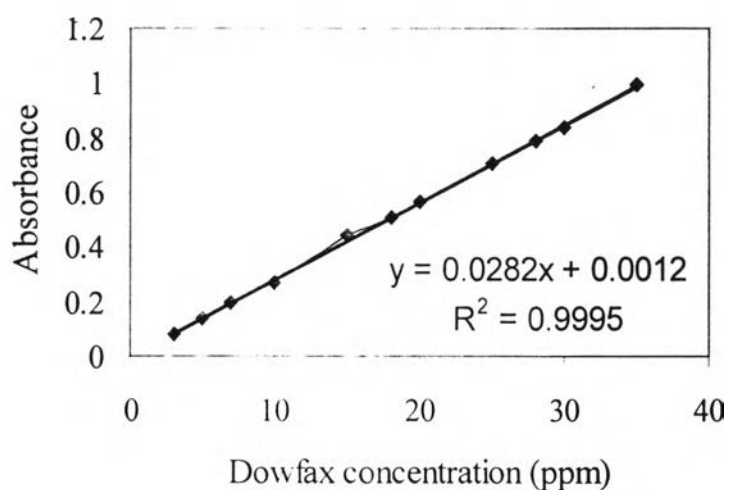
**Figure B-1** Relationship between colored motor oil concentration and the absorbance measured at 520 nm.

**Table B-1** Relationship between colored motor oil concentration and the absorbance measured at 520 nm

Motor oil concentration (ppm)	Absorbance
500	0.051
800	0.081
1000	0.090
2000	0.157
3000	0.236
5000	0.392
8000	0.637
9000	0.708

### 3. % ADPODS Concentration

The ADPODS concentration is calculated from the calibration curve for Dowfax8390.



**Figure B-2** Relationship between ADPODS concentration and the absorbance measured at 235 nm.

**Table B-2** Relationship between ADPODS concentration and the absorbance measured at 235 nm

Motor oil concentration (ppm)	Absorbance
3	0.086
5	0.142
7	0.195
10	0.274
15	0.441
18	0.513
20	0.567
25	0.710
28	0.790
30	0.839
35	0.992

## CURRICULUM VITAE

**Name:** Ms. Pantipa Ratchatawetchakul

**Date of birth:** January 6, 1981

**Nationality:** Thai

**University Education:**

1999-2003 Bachelor Degree of Engineering in Chemical Engineering,  
Faculty of Engineering, Thammasat University, Bangkok, Thailand.