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## APPENDIX

**Table A1** Molar ratio and attraction constant for solubility calculation

Group	Molar Volume	Molar attraction
	Constant $V^*$ ( $\text{cm}^3 \text{ mole}^{-1}$ )	Constant $F^*$ ( $(\text{cal.cm}^3)^{0.5} \text{ mole}^{-1}$ )
-CH <sub>3</sub>	31.8	218
-CH <sub>2</sub> -	16.5	132
>CH-	1.9	23
>C<	-14.8	-97
C <sub>6</sub> H <sub>3</sub>	41.4	562
C <sub>6</sub> H <sub>4</sub>	58.5	652
C <sub>6</sub> H <sub>5</sub>	75.5	735
CH <sub>2</sub> =	29.7	203
-CH=	13.7	113
>C=	-2.4	18
-OCO-	19.6	298
-CO-	10.7	262
-O-	5.1	95
-Cl	23.9	264
-CN	23.6	426
-NH <sub>2</sub>	18.6	275
>NH	8.5	143
>N-	-5.0	-3

Solubility parameter = Molar attraction constant / Molar volume constant

**Table A2** Solubility parameter

	Solubility parameter
H <sub>2</sub> O	18.63
CH <sub>3</sub> CH <sub>2</sub> OH	8.33
Poly (BA-hda)	11.11
Poly (BA-eda)	12.65

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2. Tungsattabud, J.; Pakkethati, K.; Chaisuwan, T.; and Wongkasemjit, S., (2010, April 22) Development of Polybenzoxazine (PBZ) Membranes for Ethanol/Water Separation. Paper presented at the 16<sup>th</sup> PPC Symposium on Petroleum, Petrochemicals, and Polymers, Bangkok, Thailand.