

พอลิเมอร์ผสมของยางธรรมชาติกราฟต์ด้วยไกลซีดีลเมทาคริเลตและสไตรีน/พอลิเมทิลเมทาคริเลต



นางสาวพัชรียา สุริยะไชย

ศูนย์วิทยทรัพยากร

วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต  
สาขาวิชาปิโตรเคมีและวิทยาศาสตร์พอลิเมอร์ หลักสูตรปิโตรเคมีและวิทยาศาสตร์พอลิเมอร์  
คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย

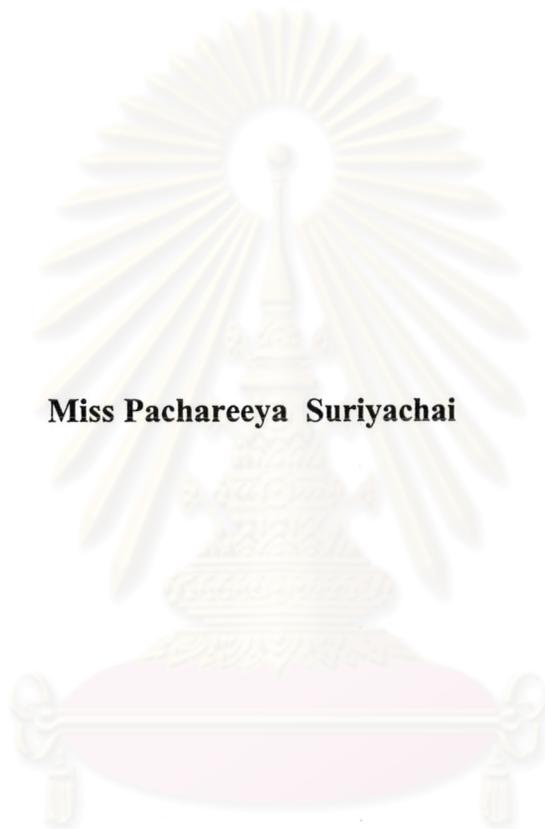
ปีการศึกษา 2544

ISBN 974-03-1047-8

ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย

**NATURAL RUBBER-G-GLYCIDYL METHACRYLATE AND STYRENE/  
POLY(METHYL METHACRYLATE) BLENDS**

**Miss Pachareeya Suriyachai**



ศูนย์วิทยทรัพยากร  
จุฬาลงกรณ์มหาวิทยาลัย

**A Thesis Submitted in Partial Fulfillment of the Requirements  
for the Degree of Master of Science in Petrochemistry and Polymer Science**

**Program of Petrochemistry and Polymer Science**

**Faculty of Science**

**Chulalongkorn University**

**Academic Year 2001**

**ISBN 974-03-1047-8**

**Thesis Title** NATURAL RUBBER-G-GLYCIDYL METHACRYLATE AND  
STYRENE/POLY(METHYL METHACRYLATE) BLENDS  
**By** Miss Pachareeya Suriyachai  
**Field of Study** Petrochemistry and Polymer Science  
**Thesis Advisor** Professor Pattarapan Prasassarakich, Ph.D.  
**Thesis Co-advisor** Professor Suda Kiatkamjornwong, Ph.D.

---

Accepted by the Faculty of Science, Chulalongkorn University in Partial  
Fulfillment of the Requirements for the Master 's Degree

..... *Pipat Karntiang* ..... Deputy Dean for Administrative Affairs  
(Associate Professor Pipat Karntiang, Ph.D.) Acting Dean, Faculty of Science

THESIS COMMITTEE

..... *Sup Tanayanon* ..... Chairman  
(Associate Professor Supawan Tantayanon, Ph.D.)

..... *Patty Pradit* ..... Thesis Advisor  
(Professor Pattarapan Prasassarakich, Ph.D.)

..... *Suda Kiatkamjornwong* ..... Thesis Co-advisor  
(Professor Suda Kiatkamjornwong, Ph.D.)

..... *W. Trakarnpruk* ..... Member  
(Associate Professor Wimonrat Trakarnpruk, Ph.D.)

..... *Nuanphun Chantarasiri* ..... Member  
(Assistant Professor Nuanphun Chantarasiri, Ph.D.)

พัชรียา สุริยะไชย : พอลิเมอร์ผสมของยางธรรมชาติกราฟต์ด้วยไกลซิดิลเมทาคริเลตและ  
สไตรีน / พอลิเมทิลเมทาคริเลต. (NATURAL RUBBER-G-GLYCIDYL  
METHACRYLATE AND STYRENE/POLY(METHYL METHACRYLATE) BLENDS) อ.  
ที่ปรึกษา : ศ.ดร.ภัทรพรหม ประศาสน์สารกิจ, อ.ที่ปรึกษาร่วม : ศ.ดร.สุดา เกียรติกำจรวงศ์;  
106 หน้า. ISBN 974-03-1047-8.

กราฟต์โคพอลิเมอร์ของไกลซิดิลเมทาคริเลตและสไตรีนบนยางธรรมชาติสังเคราะห์ด้วยวิธี  
การอิมัลชันพอลิเมอไรเซชัน โดยใช้ควิมีนไฮโดรเพอรอกไซด์และเททระเอทิลลีนเพนตะเอมีนเป็นตัว  
ริเริ่มปฏิกิริยาแบบปรีคอกซ์ งานวิจัยนี้ศึกษาผลของความเข้มข้นของตัวริเริ่มและมอนอเมอร์ อุณหภูมิของ  
ปฏิกิริยา และระยะเวลาของปฏิกิริยา ต่อประสิทธิภาพการกราฟต์และการเปลี่ยนแปลงมอนอเมอร์ ยาง  
ธรรมชาติกราฟต์ตรวจสอบด้วย FT-IR สเปกโตรสโกปีและการวิเคราะห์ NMR และศึกษาสมบัติเชิงกล  
แบบไดนามิกและสถานะยานวทยาของยางธรรมชาติกราฟต์ ภาวะที่เหมาะสมสำหรับการเตรียมยาง  
ธรรมชาติกราฟต์คือ ความเข้มข้นมอนอเมอร์ 100 ส่วน ความเข้มข้นตัวริเริ่ม 2.5 ส่วนต่อ 100 ส่วนเนื้อ  
ยาง และอุณหภูมิของปฏิกิริยา 60°C เป็นเวลา 10 ชั่วโมง

ยางธรรมชาติกราฟต์สามารถใช้ตัวประสานสำหรับพอลิเมอร์ผสมเอสทีอาร์ 5Lและพอลิเมทิล  
เมทาคริเลต ศึกษาผลของสมบัติเชิงกล เช่นความทนแรงดึง ความยืดเมื่อขาด ความทนทานการฉีกขาด  
ความแข็ง และความทนแรงกระแทก เป็นฟังก์ชันกับปริมาณยางธรรมชาติกราฟต์ พบว่าพอลิเมอร์ผสม  
เอสทีอาร์ 5Lและพอลิเมทิลเมทาคริเลตไม่วัลคาไนซ์และวัลคาไนซ์ให้สมบัติเชิงกลที่ดีที่ปริมาณยาง  
ธรรมชาติกราฟต์ 5-10 ส่วน นอกจากนี้ได้ศึกษาสมบัติเชิงกลของพอลิเมอร์ผสมยางธรรมชาติกราฟต์  
และพอลิเมทิลเมทาคริเลต พบว่าความทนแรงดึง ความทนทานการฉีกขาด และความแข็งแรงเพิ่มขึ้นเมื่อ  
เพิ่มปริมาณของพอลิเมทิลเมทาคริเลต ผิวหักโดยการดึงตรวจสอบด้วยสแกนนิ่งอิเล็กตรอนไมโคร  
สโคปแสดงว่ายางธรรมชาติกราฟต์เป็นสารเชื่อมโยงเฟสและให้การยึดที่ดีระหว่างสองเฟสของพอลิ  
เมอร์ผสม

หลักสูตร...ปีโทแรกมีและวิทยาศาสตร์พอลิเมอร์...ลายมือชื่อนิสิต.....พัชรียา สุริยะไชย  
สาขาวิชา...ปีโทแรกมีและวิทยาศาสตร์พอลิเมอร์...ลายมือชื่ออาจารย์ที่ปรึกษา.....  
ปีการศึกษา.....2544.....ลายมือชื่ออาจารย์ที่ปรึกษาร่วม.....

# # 4272349223 : MAJOR PETROCHEMISTRY AND POLYMER SCIENCE

KEY WORD: GRAFT COPOLYMERIZATION / GLYCIDYL METHACRYLATE / REDOX INITIATOR

PACHAREEYA SURIYACHAI : NATURAL RUBBER-G-GLYCIDYL METHACRYLATE AND STYRENE/POLY(METHYL METHACRYLATE) BLENDS. THESIS ADVISOR: PROF. PATTARAPAN PRASASSARAKICH, Ph.D., THESIS CO-ADVISOR PROF. SUDA KIATKAMJORNWONG, Ph.D. , 106 pp. ISBN 974-03-1047-8.

Graft copolymer of glycidyl methacrylate and styrene onto natural rubber was synthesized by emulsion polymerization using cumene hydroperoxide and tetraethylene pentamine as a redox initiator. The effects of initiator and monomer concentration, reaction temperature and reaction time on grafting efficiency and monomer conversion were investigated. The grafted natural rubber was characterized by FT-IR spectroscopy and NMR analysis. Dynamic mechanical properties and morphology of grafted natural rubber were studied. The optimum condition for the preparation of grafted natural rubber was found to be at the monomer concentration of 100 phr, the initiator concentration of 2.5 phr and temperature of 60°C for 10 hours.

The grafted natural rubber product could be used as a compatibilizer for STR5L/PMMA blends. The mechanical properties such as tensile strength, elongation at break, tear strength, hardness and impact energy were evaluated as a function of grafted natural rubber content. The good mechanical properties of unvulcanized and vulcanized of STR5L/PMMA blends were obtained at 5-10 phr of grafted natural rubber content. The mechanical properties of grafted natural rubber/PMMA blends were also studied. The tensile strength, tear strength and hardness increased with increasing of PMMA content. The tensile fracture surface examined by scanning electron microscopy showed that the grafted natural rubber acted as an interfacial agent and gave a good adhesion between the two phases of blend.

Program...Petrochemistry and Polymer Science.....Student's signature *Pachareeya Suriyachai*  
Field of study..Petrochemistry and Polymer Science...Advisor's signature *Prof. Pattarapan Prasassarakich*  
Academic year .....2001..... Co-advisor's signature *Prof. Suda Kiatkamjornwong*

## ACKNOWLEDGMENTS

The author wishes to express her deepest gratitude to her advisor, Professor Pattarapan Prasassarakich, her co-advisor, Professor Suda Kiatkamjornwong and Assistant Professor Nuanphun Chantarasiri, for their guidance, encouragement and helpful suggestion throughout this research. In addition, the author is also grateful to the members of the thesis committee for their comments and suggestions.

The author also thanks the Kouyoh Trading Co., Ltd, Tokyo, Japan for material supply as well as the Graduate School of Chulalongkorn University for the research financial supports. Many thanks are going to the Department of Chemical Technology, the Department of Chemistry, Faculty of Science, Chulalongkorn University, the Rubber Research Institute of Thailand, Department of Science Service, and the National Metal and Materials Technology Centers (MTEC) for providing research facilities throughout the research work.

Thanks go towards her friends and everyone whose names are not mentioned here for their suggestions, assistance, advice concerning the experimental techniques and the encouragement during the period of this research.

Finally, and most of all, the author would like to express her deep gratitude to her family for their love, inspiration, and encouragement.

# CONTENTS

	PAGE
ABSTRACT (IN THAI).....	iv
ABSTRACT (IN ENGLISH).....	v
ACKNOWLEDGEMENTS.....	vi
CONTENTS.....	vii
LIST OF TABLES.....	xi
LIST OF FIGURES.....	xiii
ABBREVIATION.....	xvii
CHAPTER 1 : INTRODUCTION.....	1
1.1 The Purpose of the Investigation.....	1
1.2 Objectives.....	2
1.3 Scope of the Investigation.....	2
CHAPTER 2 : THEORY AND LITERATURE REVIEW.....	4
2.1 Natural Rubber.....	4
2.1.1 Natural Rubber Latex.....	4
2.1.2 Concentration of Natural Rubber Latex.....	5
2.1.2.1 Concentration by Creaming.....	6
2.1.2.2 Concentration by Centrifugation.....	6
2.2 Standard Thai Rubber.....	7
2.3 Graft Copolymer.....	9
2.3.1 Glycidyl Methacrylate.....	10
2.4 Polymer Blends.....	11

## CONTENTS (Continued)

	PAGE
2.5 Compatibilization.....	14
2.5.1 The Method of Compatibilization.....	14
2.5.1.1 Thermodynamic Miscibility.....	14
2.5.1.2 Addition of Block and Graft Copolymers.....	15
2.5.1.3 Addition of Functional Polymers.....	16
2.5.1.4 Reactive Blending.....	16
2.6 Literature Review.....	17
CHAPTER 3 : EXPERIMENTAL.....	22
3.1 Chemicals.....	22
3.2 Glasswares.....	23
3.3 Equipments.....	23
3.4 Procedure.....	24
3.4.1 Purification of Monomer.....	24
3.4.2 Preparation Grafted Natural Rubber.....	24
3.5 Determiration of The Grafted Natural Rubber.....	25
3.5.1 Determiration the Conversion of Grafted Natural Rubber.....	25
3.5.2 Determiration the Percentage of Grafted Natural Rubber and Grafting Efficiency.....	26
3.5.3 Determiration of the Morphology of Grafted Natural Rubber.....	28
3.5.4 Characterization of Graft Copolymer Product.....	28
a. Fourier-Transform Infrared Spectroscopy.....	28
b. Nuclear Magnetic Resonance Spectroscopy.....	28
3.5.5 Determiration of Dynamic Mechanical Properties of Grafted Natural Rubber.....	29

## CONTENTS (Continued)

	PAGE
3.6 Preparation of STR5L/PMMA/Grafted Natural Rubber Blends.....	29
3.7 Preparation of Grafted Natural Rubber/PMMA Blends.....	30
3.8 Mechanical Testing.....	30
3.8.1 Tensile Properties (ASTM D638-97).....	31
3.8.2 Tear Strength (ASTM D624-98).....	31
3.8.3 Hardness (ASTM D2240-97).....	32
3.8.4 Impact Energy (ASTM D256).....	33
3.9 Oil Absorption.....	34
3.10 Scanning Electron Microscopy (SEM).....	34
CHAPTER 4 : RESULTS AND DISCUSSION.....	35
4.1 Properties of Natural Rubber Latex.....	35
4.2 Preparation of Grafted Natural Rubber.....	36
4.2.1 Effect of Initiator Concentration.....	37
4.2.2 Effect of Reaction Temperature.....	41
4.2.3 Effect of Monomer Concentration.....	42
4.2.4 Effect of Reaction Time.....	46
4.3 Mechanism of Free Radical Grafting of Styrene and Glycidyl Methacrylate onto Natural Rubber.....	48
4.4 Grafted Natural Rubber Morphology.....	53
4.5 Characterization of Grafted Natural Rubber.....	54
4.5.1 Functional Groups in the Grafted Natural Rubber.....	54
4.5.2 Functional Groups in the Samples After Solvent Extraction.....	56
4.5.3 NMR Spectroscopy.....	59
4.5.4 Dynamic Mechanical Thermal Analysis (DMTA).....	62

## CONTENTS (Continued)

	PAGE
4.6 Mechanical Properties of STR5L Blends with PMMA and Grafted Natural Rubber as a Compatibilizer.....	64
4.7 Oil Absorption.....	71
4.8 Mechanical Properties of the Grafted Natural Rubber/PMMA Blends.....	73
4.9 Scanning Electron Microscopy (SEM).....	75
CHAPTER 5 : CONCLUSIONS AND SUGGESTION.....	82
5.1 Conclusion.....	82
5.2 Suggestion for Futher Work.....	83
REFERENCES.....	85
APPENDICES.....	89
APPENDIX A.....	90
APPENDIX B.....	95
APPENDIX C.....	97
APPENDIX D.....	103
APPENDIX E.....	104
VITA.....	106

ศูนย์วิจัยทรัพยากร  
 จุฬาลงกรณ์มหาวิทยาลัย

## LISTS OF TABLES

TABLE	PAGE
2.1 Composition of Latex Sap.....	5
2.2 Processing Material of STR Grades.....	7
2.3 Standard Thai Rubber (STR) Scheme.....	8
2.4 Typical Physical Properties of GMA Monomer.....	10
2.5 Types of Polyblends.....	13
3.1 Experimental Variables for Graft Copolymerization.....	25
3.2 Compounding Formulations.....	30
4.1 The Properties of High Ammonia Natural Rubber Latex.....	36
4.2 Effect of Initiator Concentration on the Conversion, Grafting Efficiency, and Percentage Grafted Natural Rubber : (T = 50-70°C, Monomer = 100 phr, and Time = 10 h).....	40
4.3 Effect of Reaction Temperature on the Conversion, Grafting Efficiency and Percentage Grafted Natural Rubber : (Initiator Concentration = 2.5 phr, Monomer = 100 phr, and Time = 10 h).....	42
4.4 Effect of Monomer Concentration on the Conversion, Grafting Efficiency and Percentage Grafted Natural Rubber : (Initiator Concentration = 2.5 phr, T = 60°C, and Time = 10 h).....	44
4.5 Effect of Reaction Time on the Conversion, Grafting Efficiency and Percentage Grafted Natural Rubber : (Initiator Concentration = 2.5 phr, Monomer = 100 phr, and T = 60°C).....	46

## LISTS OF TABLES (Continued)

TABLE		PAGE
4.6	Glass Transition Temperature and $\tan\delta$ of the Grafted Natural Rubber by DMTA Technique.....	63
4.7	The Properties of STR5L.....	66
4.8	Typical Data of PMMA (Crystallite #MF 001).....	66
4.9	Properties of Unvulcanized STR5L/PMMA Blends.....	68
4.10	Properties of Vulcanized STR5L/PMMA Blends.....	68
4.11	The Percentage of Weight Change of Vulcanized STR5L/PMMA Blends.....	72
4.12	Properties of Grafted Natural Rubber/PMMA Blends.....	73
A-1	Effect of Initiator Concentration, Reaction Temperature, Monomer Concentration, and Reaction Time on the Conversion, Percentage of Grafted Natural Rubber, and Grafting Efficiency.....	90
A-2	The Average of the Conversion, Percentage of Grafted Natural Rubber, Percentage of Free Natural Rubber, Percentage of Free St/GMA and Grafting Efficiency.....	93
C-1	Tensile Strength of STR5L/PMMA Blends at Ratio of 70/30.....	97
C-2	Elongation at Break of STR5L/PMMA Blends at Ratio of 70/30.....	97
C-3	Stress @100% Modulus of STR5L/PMMA Blends at Ratio of 70/30.....	98
C-4	Tear Strength of STR5L/PMMA Blends at Ratio of 70/30.....	98
C-5	Hardness of STR5L/PMMA Blends at Ratio of 70/30.....	99
C-6	Tensile Strength of STR5L/PMMA Blends at Ratio of 50/50.....	99
C-7	Elongation at Break of STR5L/PMMA Blends at Ratio of 50/50.....	100
C-8	Stress @100% Modulus of STR5L/PMMA Blends at Ratio of 50/50.....	100
C-9	Tear Strength of STR5L/PMMA Blends at Ratio of 50/50.....	101
C-10	Hardness of STR5L/PMMA Blends at Ratio of 50/50.....	101

## LISTS OF TABLES (Continued)

TABLE	PAGE
C-11 Impact Energy of STR5L/PMMA Blends at Ratio of 50/50.....	102
D-1 The Percentage of Weight Change of Vulcanized STR5L/PMMA Blends....	103
E-1 Tensile Strength of Grafted Natural Rubber/PMMA Blends.....	104
E-2 Elongation at Break Grafted Natural Rubber/PMMA Blends.....	104
E-3 Stress @100% Modulus of Grafted Natural Rubber/PMMA Blends.....	105
E-4 Tear Strength of Grafted Natural Rubber/PMMA Blends.....	105
E-5 Hardness of Grafted Natural Rubber/PMMA Blends.....	105

  
 ศูนย์วิทยทรัพยากร  
 จุฬาลงกรณ์มหาวิทยาลัย

## LIST OF FIGURES

FIGURE	PAGE
2.1 Cis-1,4-Polyisoprene.....	4
2.2 Glycidyl Methacrylate Chemical Structure.....	10
2.3 Schematic Diagram Showing Location of Block and Graft Copolymers at Phase Interfaces.....	15
3.1 The Overall Schematic Experimental Process.....	27
3.2 Schematic Diagram of Tensile Test Specimen (Type IV).....	31
3.3 Schematic Diagram of Tear Test Specimen (Die C).....	32
3.4 Dimensions of Simple Beam, Charpy Type, Impact Test Specimen.....	33
4.1 Effect of Initiator Concentration on the Conversion (Monomer 100 phr, 50-70°C, and Time 10 h).....	38
4.2 Effect of Initiator Concentration on Grafting Efficiency (Monomer 100 phr, 50-70°C, and Time 10 h).....	38
4.3 Effect of Initiator Concentration on Percentage Grafted Natural Rubber, Percentage Free Natural Rubber, Percentage Free St/GMA (Monomer 100 phr, 50-70°C, and Time 10 h).....	39
4.4a Effect of Reaction Temperature on Percentage Grafted Natural Rubber, Percentage Free Natural Rubber, Percentage Free St/GMA : T = 50-70°C, Monomer = 100 phr and Time = 10 h.....	43
4.4b Effect of Reaction Temperature on the Conversion and Grafting Efficiency : T = 50-70°C, Monomer = 100 phr and Time 10 h.....	43
4.5a Effect of Monomer Concentration on Percentage Grafted Natural Rubber, Percentage Free Natural Rubber , Percentage free St/GMA : T = 60°C, Monomer = 50-150 phr, and Time = 10 h.....	45

## LIST OF FIGURES (Continued)

FIGURE	PAGE
4.5b Effect of Monomer Concentration on the Conversion and Grafting Efficiency : T = 60°C, Monomer = 50-150 phr, and Time = 10 h.....	45
4.6a Effect of Reaction Time on Percentage Grafted Natural Rubber, Percentage Free Natural Rubber, Percentage Free St/GMA : T = 60°C, Monomer = 100 phr, and Time = 4-10 h.....	47
4.6b Effect of Reaction Time on the Conversion and Grafting Efficiency : T = 60°C, Monomer = 100 phr, and Time = 4-10 h.....	47
4.7 Pathway to Prepare Core/Shell Particle.....	54
4.8 Transmission Electron Micrographs of the Particle Morphologies of Grafted Natural Rubber.....	55
4.9 The FT-IR Spectrum of Natural Rubber.....	57
4.10 The FT-IR Spectrum of Grafted Natural Rubber.....	57
4.11 The Schematic Experimental Process of Samples After Solvent Extraction....	58
4.12 The FT-IR Spectrum of Petroleum Ether Extract.....	58
4.13 The FT-IR Spectrum of Acetone Extract.....	59
4.14 The <sup>1</sup> H-NMR Spectrum of Grafted Natural Rubber.....	60
4.15 The <sup>13</sup> C-NMR Spectrum of Grafted Natural Rubber.....	61
4.16 Loss Factor (tanδ) of Grafted Natural Rubber at Different Monomer Concentration.....	63
4.17 Storage Modulus of Grafted Natural Rubber at Different Monomer Concentration.....	64
4.18 Effect of Grafted Natural Rubber Content on the Mechanical Properties of Unvulcanized STR5L/PMMA Blends.....	70
4.19 Effect of Grafted Natural Rubber Content on the Mechanical Properties of Vulcanized STR5L/PMMA Blends.....	71

## LIST OF FIGURES (Continued)

FIGURE	PAGE
4.20 The Percentage of Weight Change of Vulcanized STR5L/PMMA Blends.....	72
4.21 Effect of PMMA Content on the Mechanical Properties of Grafted Natural Rubber/PMMA Blends.....	75
4.22 SEM Photographs of Unvulcanized of STR5L/PMMA Blends at Ratio of 70/30 with Grafted Natural Rubber.....	77
4.23 SEM Photographs of Vulcanized of STR5L/PMMA Blends at Ratio of 70/30 with Grafted Natural Rubber.....	78
4.24 SEM Photographs of Unvulcanized of STR5L/PMMA Blends at Ratio of 50/50 with Grafted Natural Rubber.....	79
4.25 SEM Photographs of Vulcanized of STR5L/PMMA Blends at Ratio of 50/50 with Grafted Natural Rubber.....	80
4.26 SEM Photographs of NR-g-St/GMA/PMMA Blends.....	81
B-1 Dynamic Mechanical Properties of Natural Rubber.....	95
B-2 Dynamic Mechanical Properties of Grafted Natural Rubber (Monomer 50 phr).....	95
B-3 Dynamic Mechanical Properties of Grafted Natural Rubber (Monomer 100 phr).....	96
B-4 Dynamic Mechanical Properties of Grafted Natural Rubber (Monomer 150 phr).....	96

## ABBREVIATIONS

HANR	:	High ammonia natural rubber
PMMA	:	Poly (methyl methacrylate)
St	:	Styrene
GMA	:	Glycidyl methacrylate
STR	:	Standard Thai Rubber
DRC	:	Dry rubber content
$\bar{M}_w$	:	Weight average molecular weight
$\bar{M}_n$	:	Number average molecular weight
$T_g$	:	Glass transition temperature
$T_m$	:	Melting temperature
EPR	:	Ethylene - propylene rubber
TEPA	:	Tetraethylene pentamine
CHPO	:	Cumene hydroperoxide
TEM	:	Transmission electron microscopy
SEM	:	Scanning electron microscopy
NMR	:	Nuclear magnetic resonance spectroscopy
FT-IR	:	Fourier–transform infrared spectroscopy
DMTA	:	Dynamic mechanical thermal analysis
ASTM	:	The American Society for Testing and Material
phr	:	Part per hundred
h	:	Hour
Avg.	:	Average
GE	:	Grafting efficiency
S.D.	:	Standard deviation