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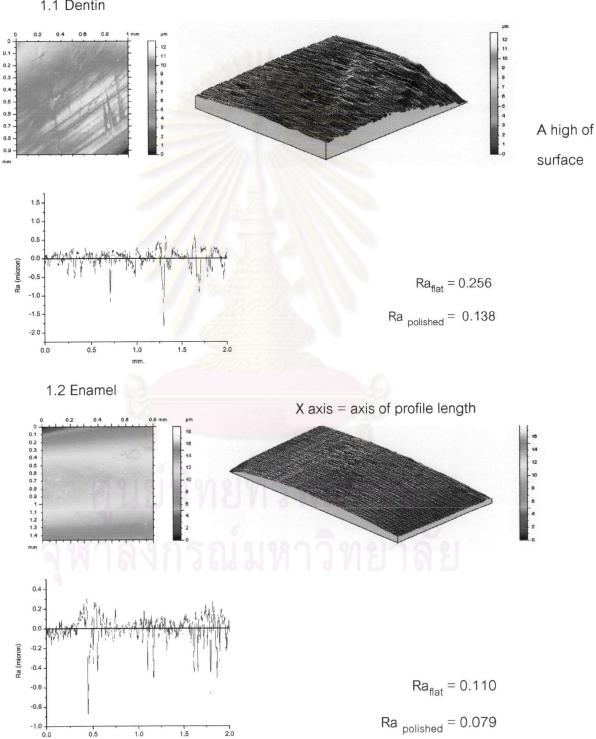


Appendix A

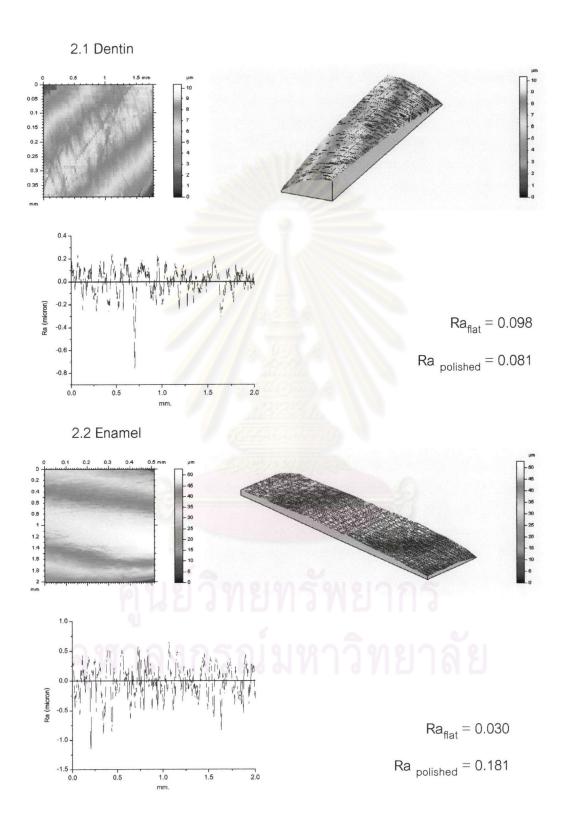
The result of surface roughness induced was recorded as average roughness (Ra in µm) with a profilometer (Taylor Hobson Precision, Talyscan 150).

1. Dentin and enamel surfaces polished with Nupro prophylaxis paste.

1.1 Dentin

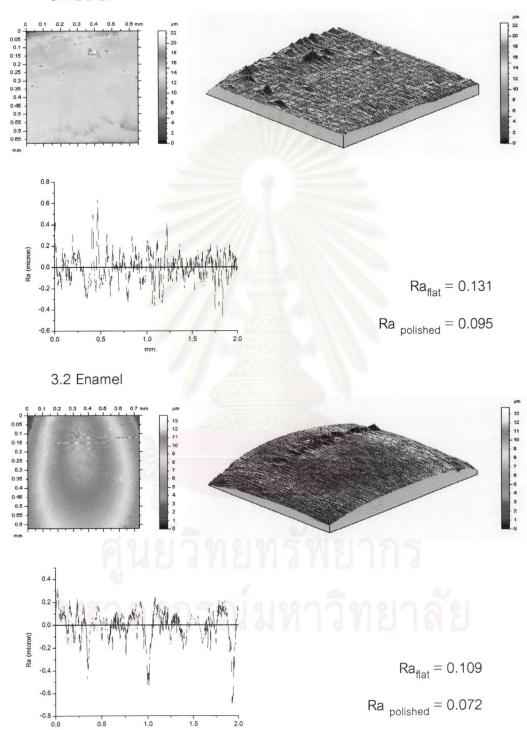


2. Dentin and enamel surfaces polished with silica -16 μm

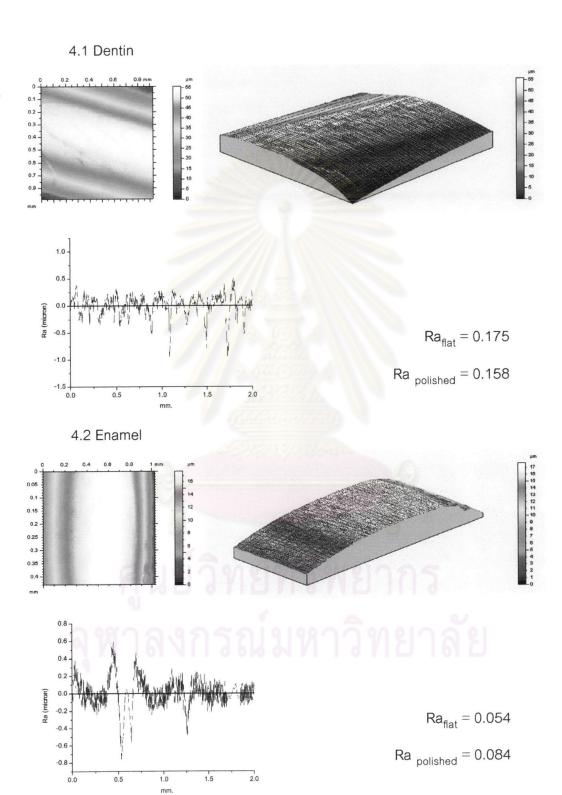


3. Dentin and enamel surfaces polished with perlite -16 μm





4. Dentin and enamel surfaces polished with pumice -16 μm

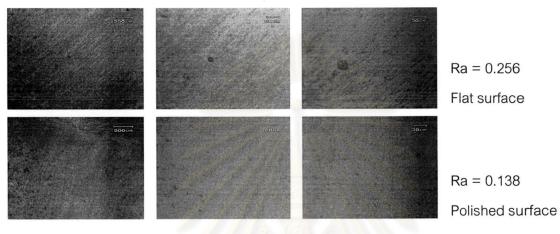


Appendix B

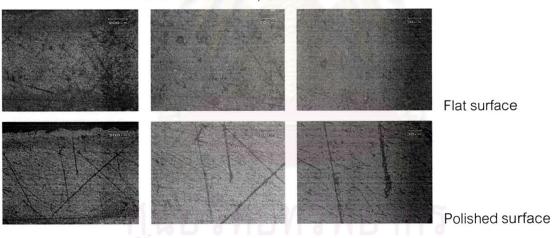
The OM of the some dentin and enamel in 5X, 10X, 20X

- 1. Dentin and enamel surface of flat surface and polished surface with Nupro.
 - 1.1 Dentin

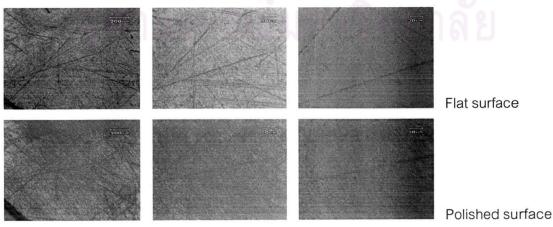
1.1.1 Dentin surface was polished 30 second.



1.1.2 Dentin surface was polished 60 second.



1.1.3 Dentin surface was polished 90 second.



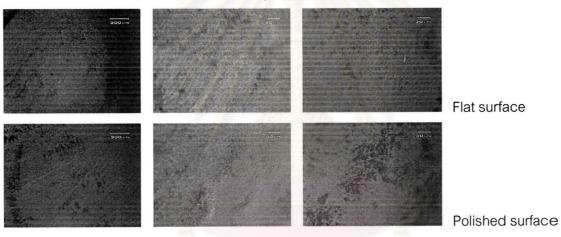
20X

5X 10X

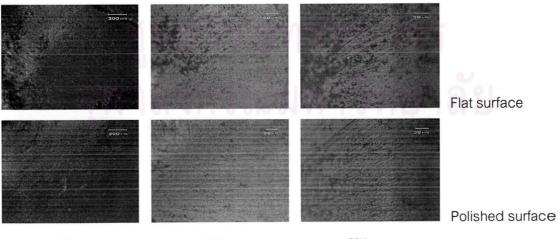
1.2.1 Enamel surface was polished 30 second.



1.2.2 Enamel surface was polished 60 second.



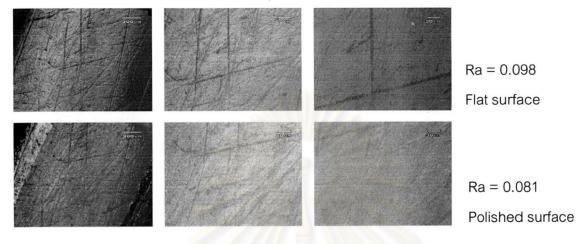
1.2.3 Enamel surface was polished 90 second.



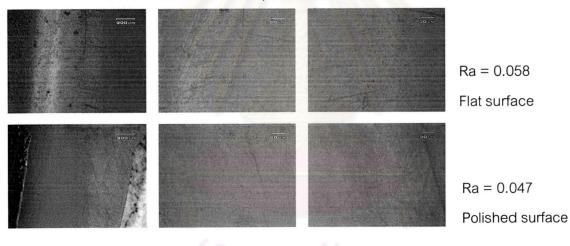
5X 10X 20X

2. Dentin and enamel surface of flat surface and polished surface with silica -16 μm 2.1 Dentin

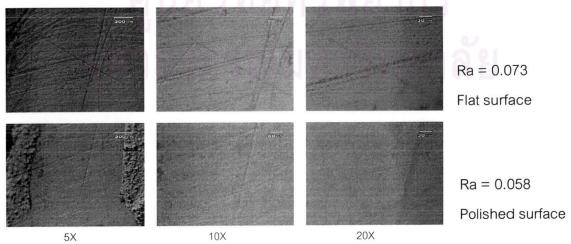
2.1.1 Dentin surface was polished 30 second.



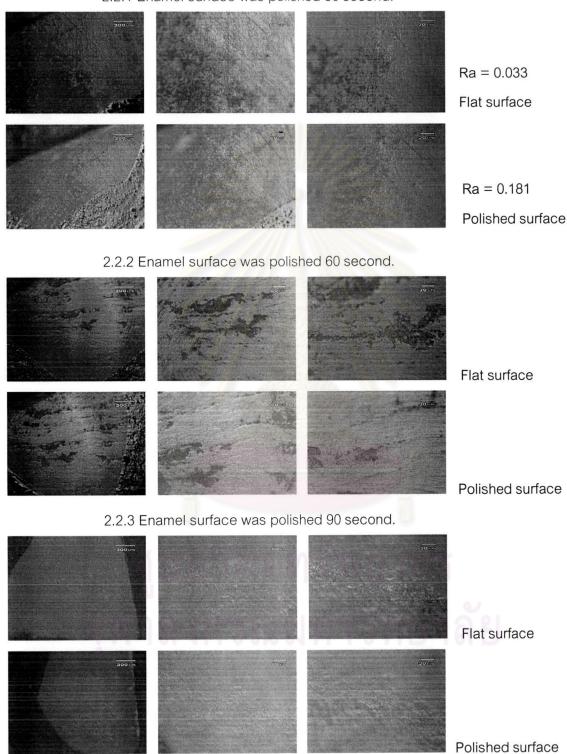
2.1.2 Dentin surface was polished 60 second.



2.1.3 Dentin surface was polished 90 second.



2.2.1 Enamel surface was polished 30 second.



10X

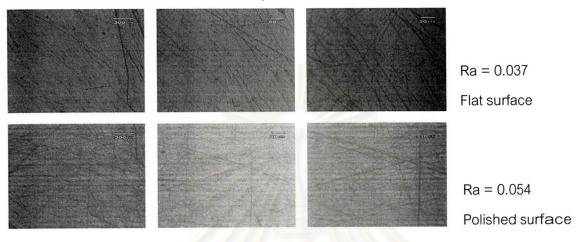
5X

20X

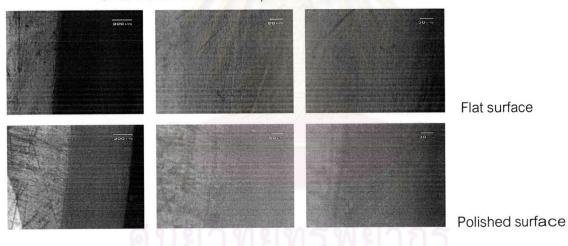
3. Dentin and enamel surface of flat surface and polished surface with calcined diatomite waste 45-63 μm

3.1 Dentin

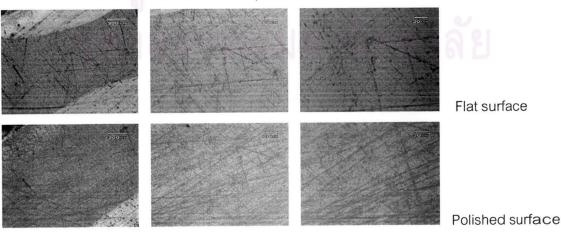
3.1.1 Dentin surface was polished 30 second.



3.1.2 Dentin surface was polished 60 second.

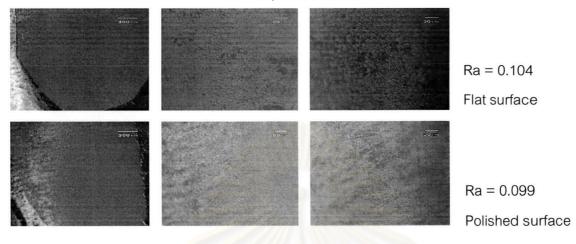


3.1.3 Dentin surface was polished 90 second.

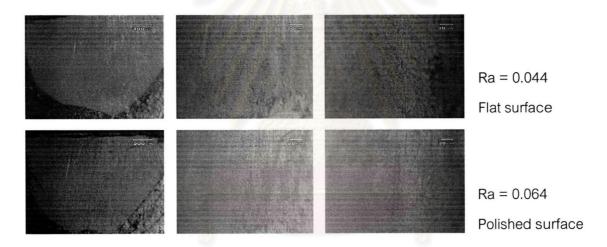


5X 10X 20X

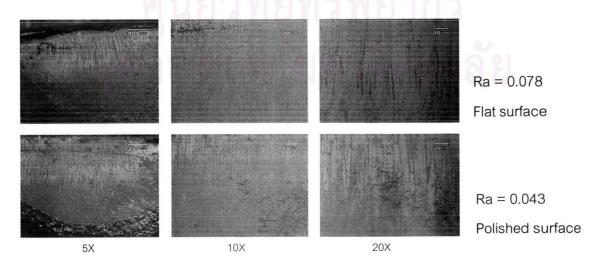
3.2.1 Enamel surface was polished 30 second.



3.2.2 Enamel surface was polished 60 second.

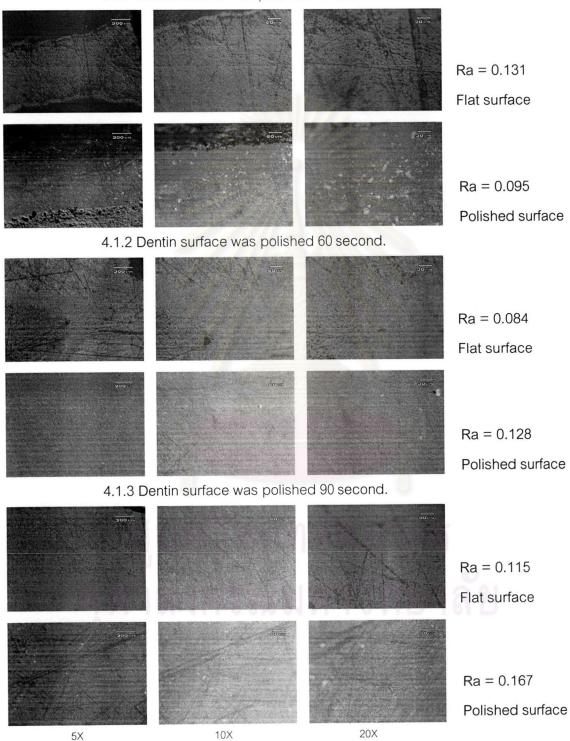


3.2.3 Enamel surface was polished 90 second.

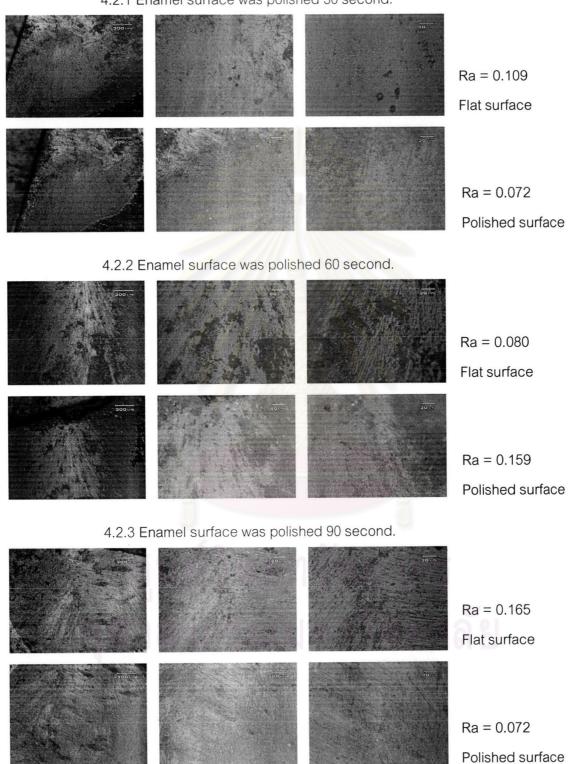


4. Dentin and enamel surface of flat surface and polished surface with perlite -16 μm 4.1 Dentin

4.1.1 Dentin surface was polished 30 second.



4.2.1 Enamel surface was polished 30 second.



10X

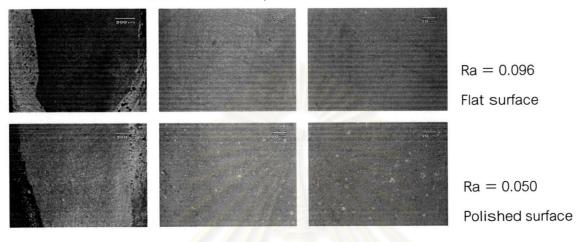
5X

20X

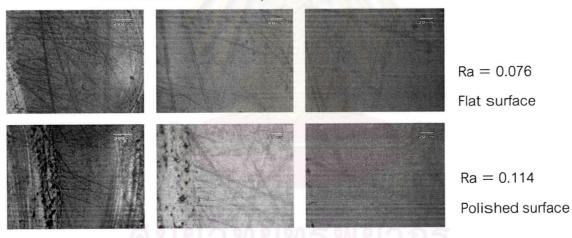
5. Dentin and enamel surface of flat surface and polished surface with perlite + diatomite 505 (1:1) (Both $-16 \mu m$)

5.1 Dentin

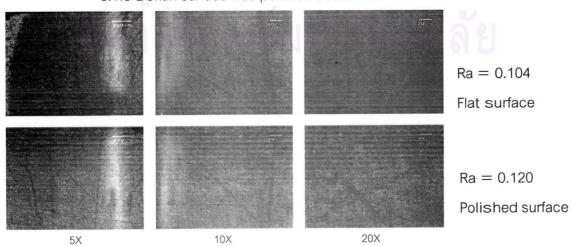
5.1.1 Dentin surface was polished 30 second.



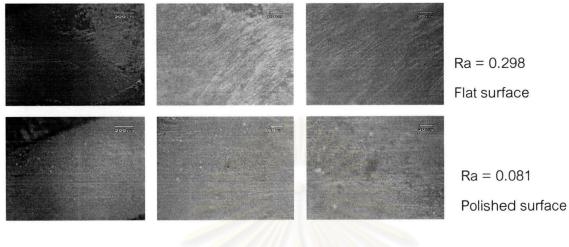
5.1.2 Dentin surface was polished 60 second.



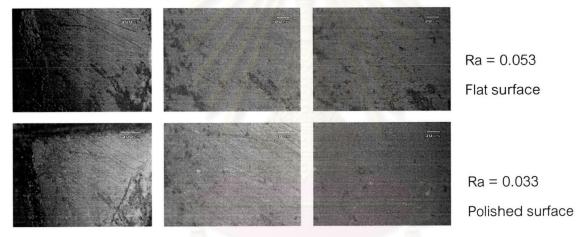
5.1.3 Dentin surface was polished 90 second.



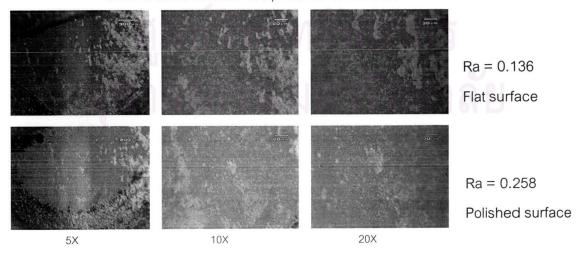
5.2.1 Enamel surface was polished 30 second.



5.2.2 Enamel surface was polished 60 second.

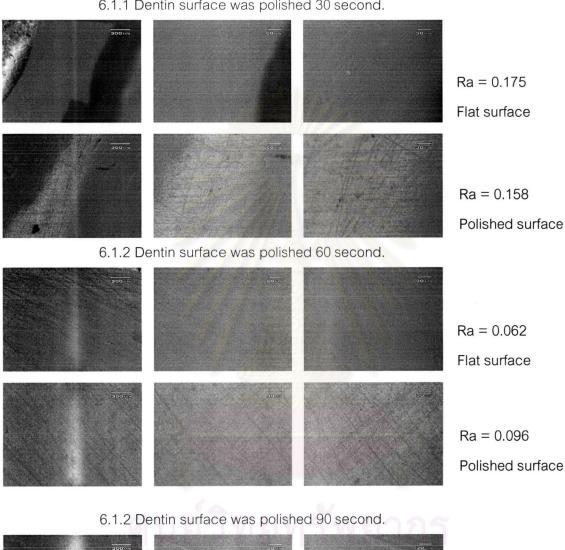


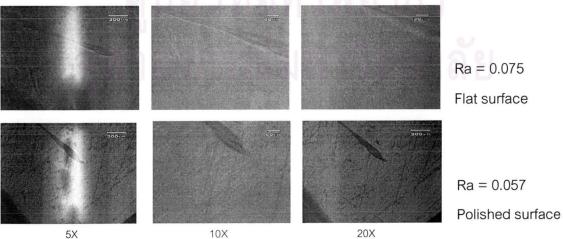
5.2.3 Enamel surface was polished 90 second.



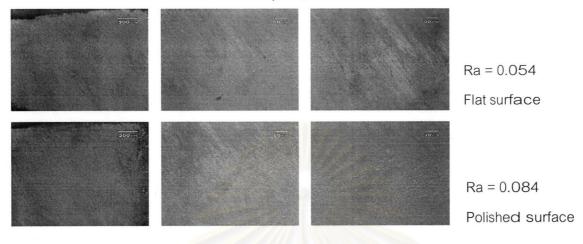
6. Dentin and enamel surface of flat surface and polished surface with pumice -16 μm 6.1 Dentin

6.1.1 Dentin surface was polished 30 second.

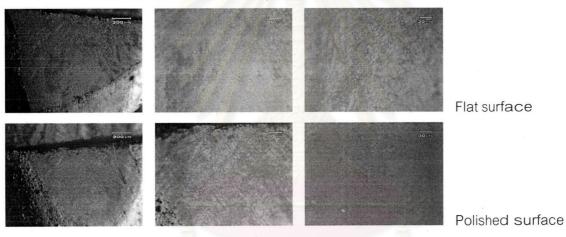




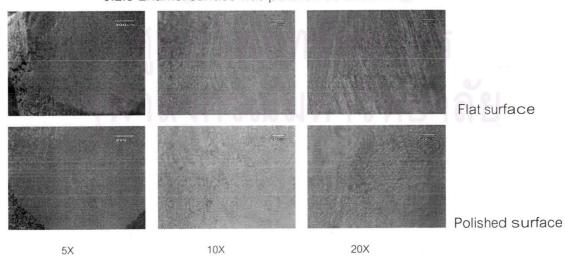
6.2.1 Enamel surface was polished 30 second.



6.2.2 Enamel surface was polished 60 second.



6.2.3 Enamel surface was polished 90 second.



Appendix C

The effect of time on Ra

Dentin surface

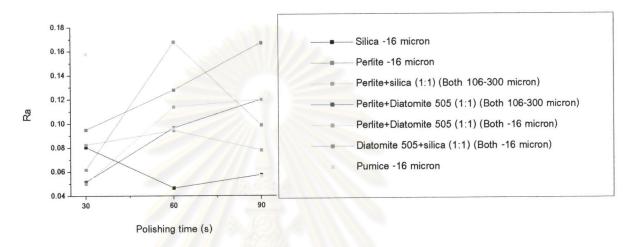


Fig. 1 Average roughness after polishing dentin surface with prepared prophylaxis paste at 30, 60, 90 s.

Enamel surface

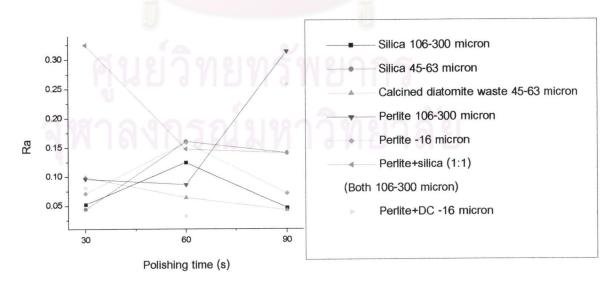
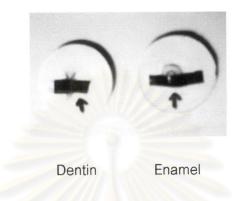


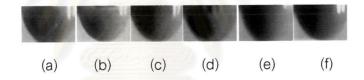
Fig. 2 Average roughness after polishing enamel surface with prepared prophylaxis paste at 30, 60, 90 s.

Appendix D

The prepared of dentin and enamel



The prepared prophylaxis paste

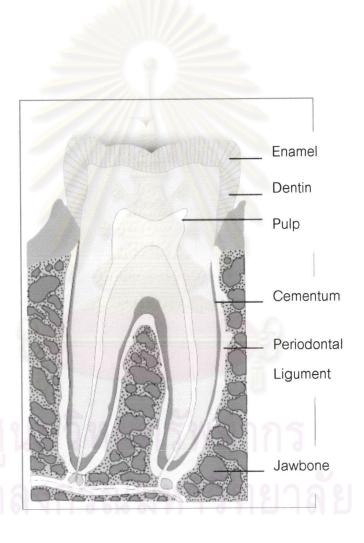


- (a) Silica
- (b) Diatomite hyflow
- (c) Nupro
- (d) Diatomite waste
- (e) Perlite AP-120
- (f) Pumice

(Most - 16 micron)

Appendix E

The tooth structure



Appendix F

Abrasion

Abrasion is the process of wear on the surface of one material by another material by scratching, gouging, chiseling, tumbling, or other mechanical means. The material that causes the wear is called an abrasive; the material being abraded is called the substrate.

Polishing

Polishing is the process of making a rough surface smooth to the touch and glossy (mostly specular reflection of incident light) in Fig 1 (a). Polishing is usually performed with very small particle size (sub-micron) abrasives.

Grinding

Grinding is the gross reduction of the surface of a substrate by abrasion. It is usually performed with large particle size abrasives. The surface texture of the substrate after a grinding procedure is usually rough to the touch and diffuse reflection to incident light in Fig 1 (b).

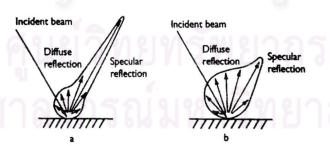


Fig. 1 Schematic diagram of diffuse and spectular reflection:

(a) high gloss, (b) Low gross

Factors affecting of abrasion

A large difference in hardness between the abrasive and substrate. Mohs
values indicate one material's resistance to scratching by another shown in
Table 1.

Table 1 Hardness of dental abrasives and substrates

Material	Hardness scale
	(Mohs)
Abrasives	
Pumice	6
Sand	7
Zirconium silicate	7-7.5
Aluminum oxide	9
Silicon carbide	9-10
Substrates	
Dentin	1-4
Enamel	5-6
Porcelain	6-7

- 2. The particle size of an abrasive may be expressed in micrometers. By convention, particles are classified as fine (0-10 μm), medium (10-100 μm), and coarse (100-500 μm), according to average particle size. Large abrasive particles will abrade a surface more rapidly than smaller particles. However, they tend to leave coarser scratches in the abrade surface than fine particles. Equivalent-size scratches can be produce by difference sizes of particles by varying the applied pressure in Fig 2 (a).
- 3. The particle shape also has an effect on rate of abrasion. Sharp, irregular shaped particles will abrade surface more rapidly than rounded particles having duller cutting angles. However, the former will produce deeper scratches than the latter in Fig 2 (b). The abrasion rate of an abrasive decrease during use; this

- is due partly to rounding of the particles and partly to contamination of the abrasive with some of the substrate material (debris).
- 4. The greater the speed at with the abrasive travels across the surface of the substrate, the greater abrasion. The greater friction at higher speed. However, tend to create higher temperatures at the surface of the substrate.
- 5. The greater the pressure applied. The greater the pressure produces deeper and wider scratches in Fig 2 (c) and creates higher temperatures.
- 6. Lubricants (Silicone grease, water spray, and glycerol) are used during abrasion for two purposes: to reduce heat buildup, and to wash away debris to prevent clogging or blinding of the abrasive instrument. Too much lubrication can reduce the abrasion rate because it may prevent some of the abrasive from coming in contact with the substrate.

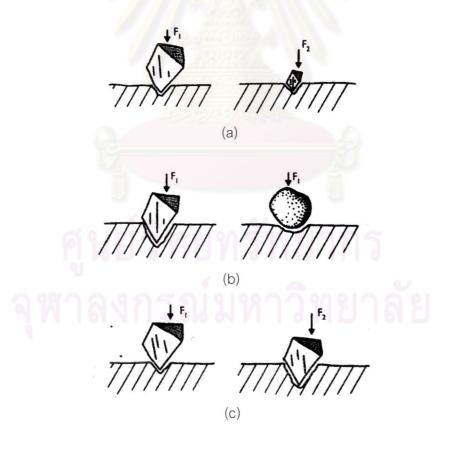


Fig 2 Factors affecting abrasion. (a) Large particle produces similar scratches with lower applied force (F_1) than smaller particle. (b) Sharp particle produces deeper

abrasion than rounder particle under applied force. (c) Deeper and wider scratches are produced by increasing the applied force from F_1 to F_2

Applications of abrasives

Prophylaxis paste

Prophylaxis paste should be chosen carefully and applied to remove exogenous stains without damaging the under lying tooth structure or adjacent restorative materials.

Dentifrices

Dentifrices are used for removing debris and minor stains from teeth and for polish tooth surface. The most commonly used abrasives are dibasic calcium phosphate, tricalcium phosphate, calcium pyrophosphate, and hydrated alumina.

Many dentifrices contain therapeutic agents, such as sodium fluoride, stannous fluoride, or sodium monofluorophosphate, to decrease the acid solubility of tooth enamel, decrease hypersensitivity, and interrupt the mechanisms of plaque attachment and calculus formation on tooth structure.

Denture cleaners

Food debris, plaque, calculus, and stains may accumulate on denture base materials in the same way as natural teeth and are best removed on a regular basis by the patient. Daily soaking in the denture cleanser solution or brushing with or without a paste or powder is usually effective.

Chemical cleansers may contain sodium perborate, which releases peroxide. As peroxide decomposes, oxygen is released, resulting in effervescence. Effervescence, along with the oxidizing ability of peroxide, is assumed to be responsible for the cleaning action.

Biography

Miss Parawee Chaono was born on November 20th, 1980 in Bangkok Province.

After graduation with a bachelor degree in Industrial Chemistry from Faculty of Science,

Chaingmai University in 2001. She continued a further study for master's degree in the
field of Ceramic Technology at Chulalongkorn University in 2002.

