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ศูนย์วิทยทรัพยากร
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APPENDICES

ศูนย์วิทยทรัพยากร
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APPENDIX A

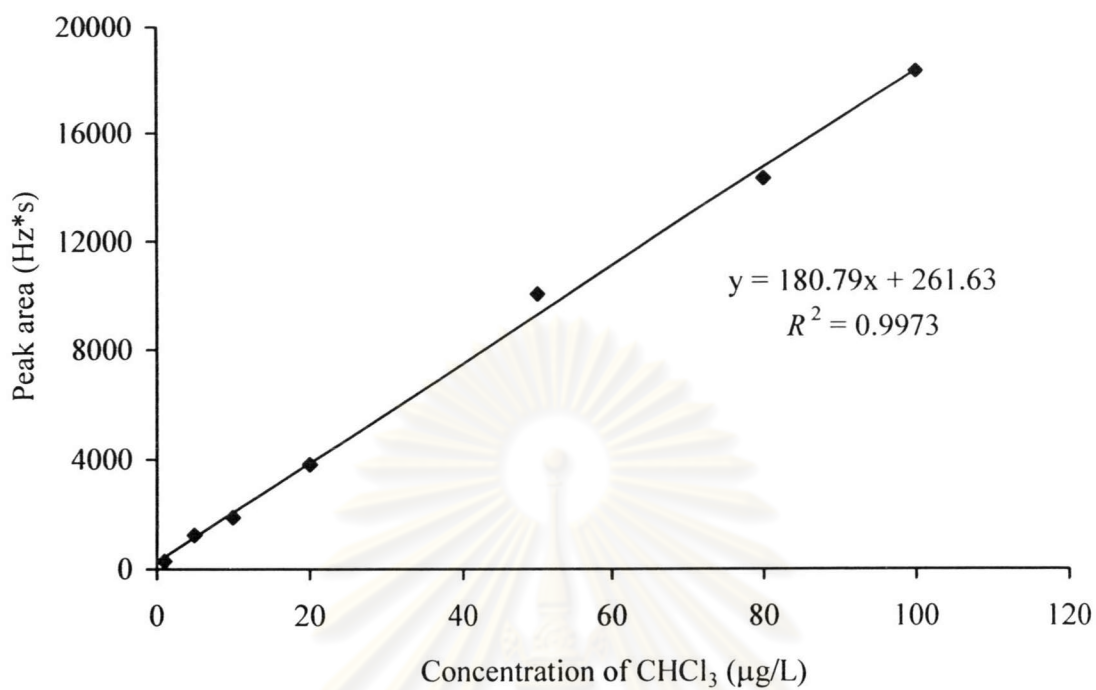


Figure A-1 The calibration curve of chloroform by condition in Table 3.4.

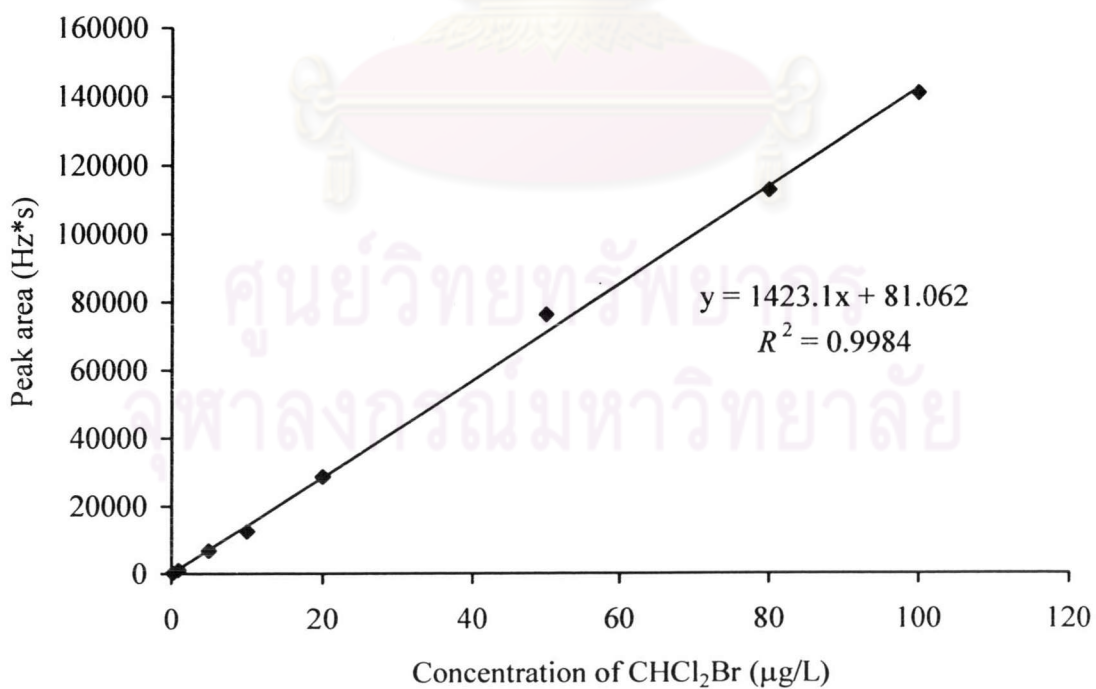


Figure A-2 The calibration curve of bromodichloromethane by condition in Table 3.4.

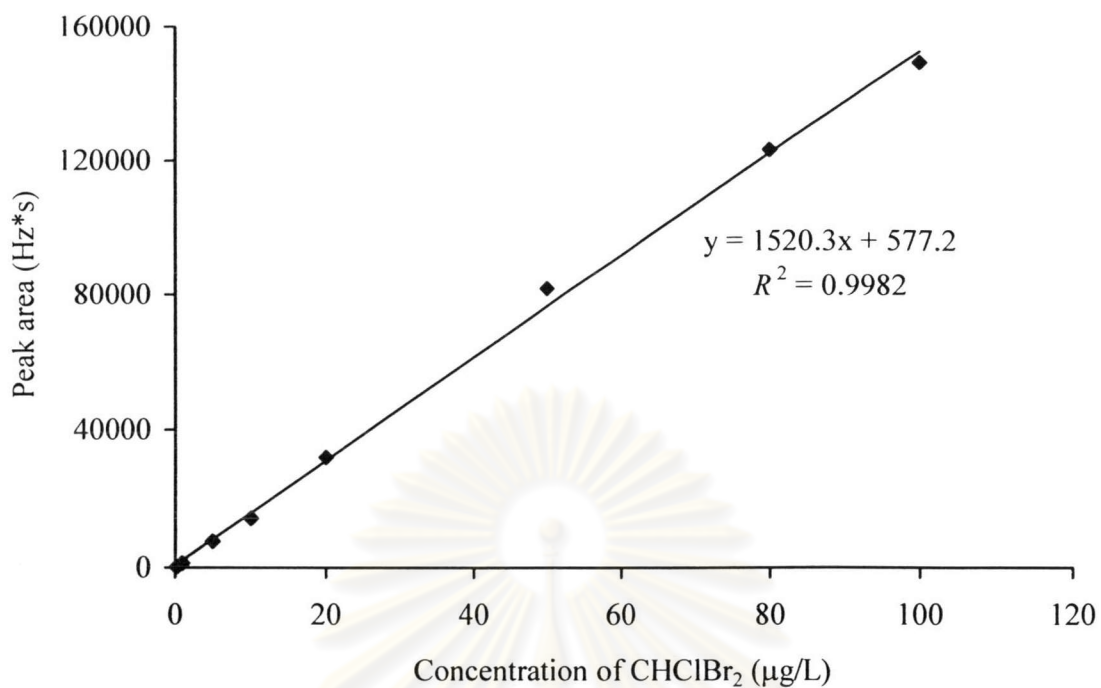


Figure A-3 The calibration curve of chlorodibromomethane by condition in Table 3.4.

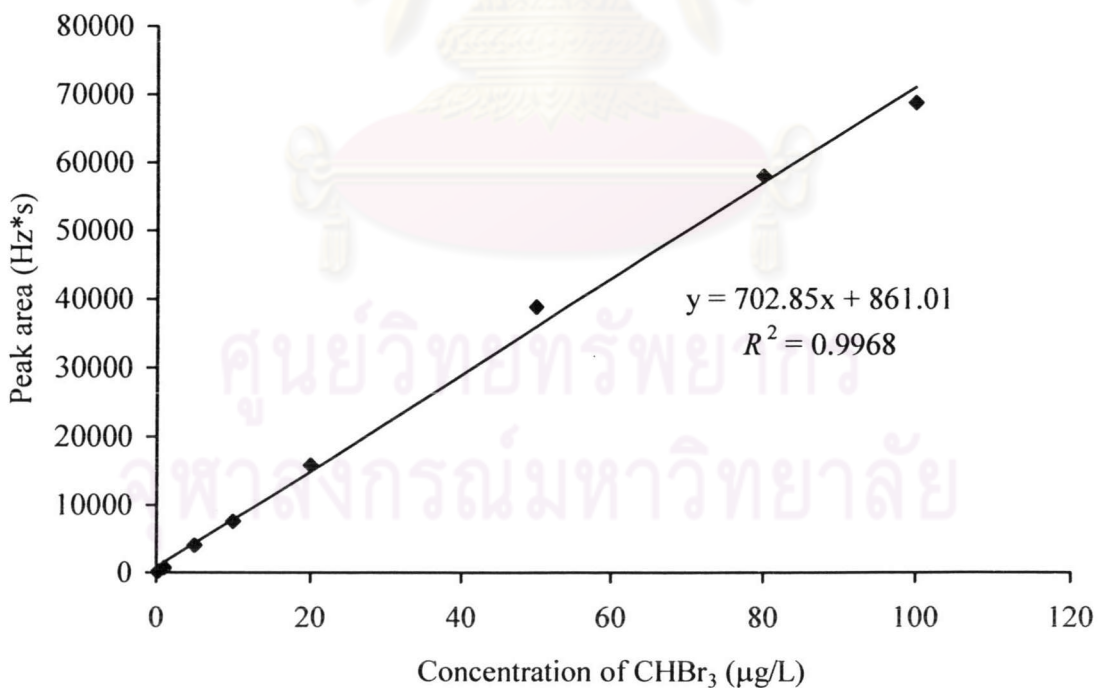


Figure A-4 The calibration curve of bromoform by condition in Table 3.4.

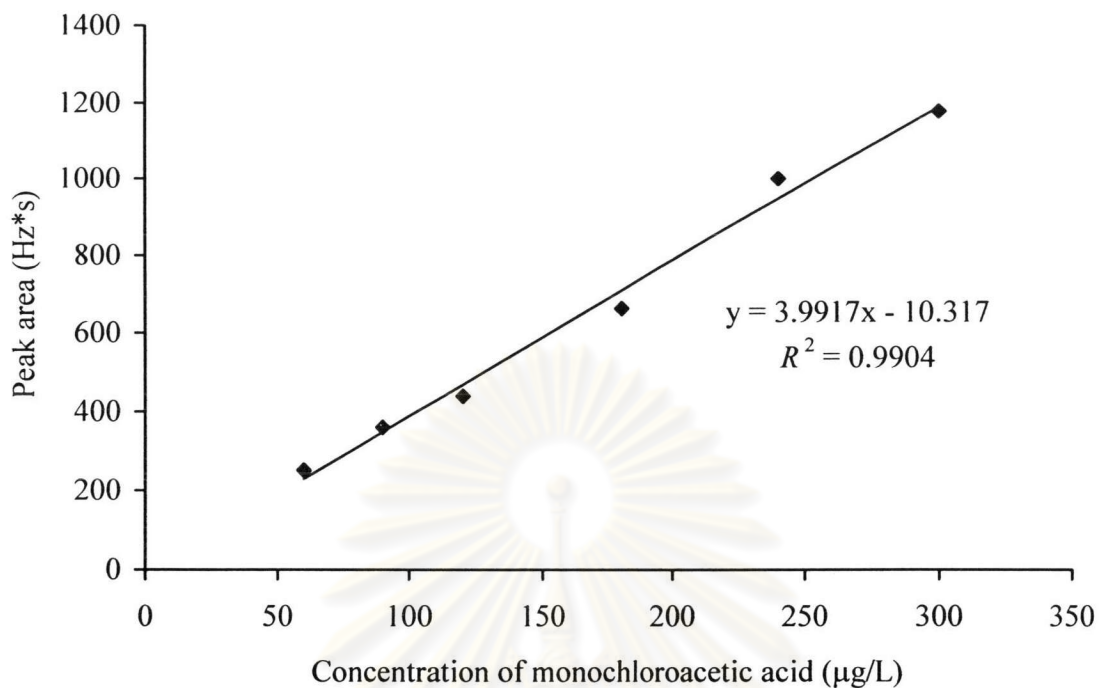


Figure A-5 The calibration curve of monochloroacetic acid by condition in Table 3.5.

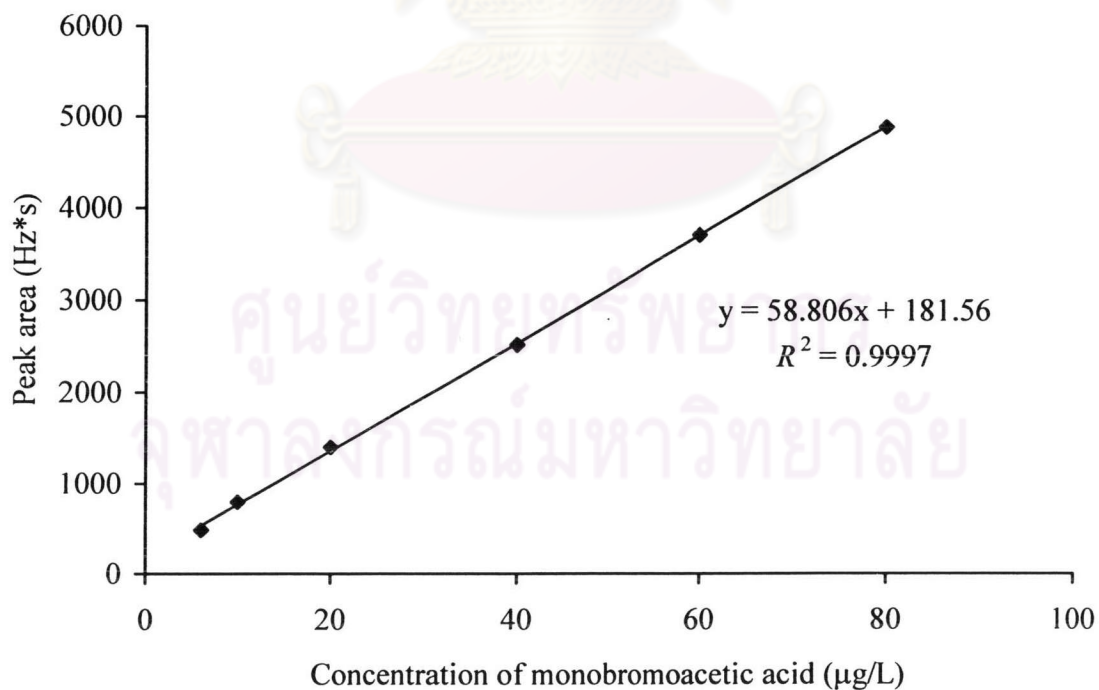


Figure A-6 The calibration curve of monobromoacetic acid by condition in Table 3.5.

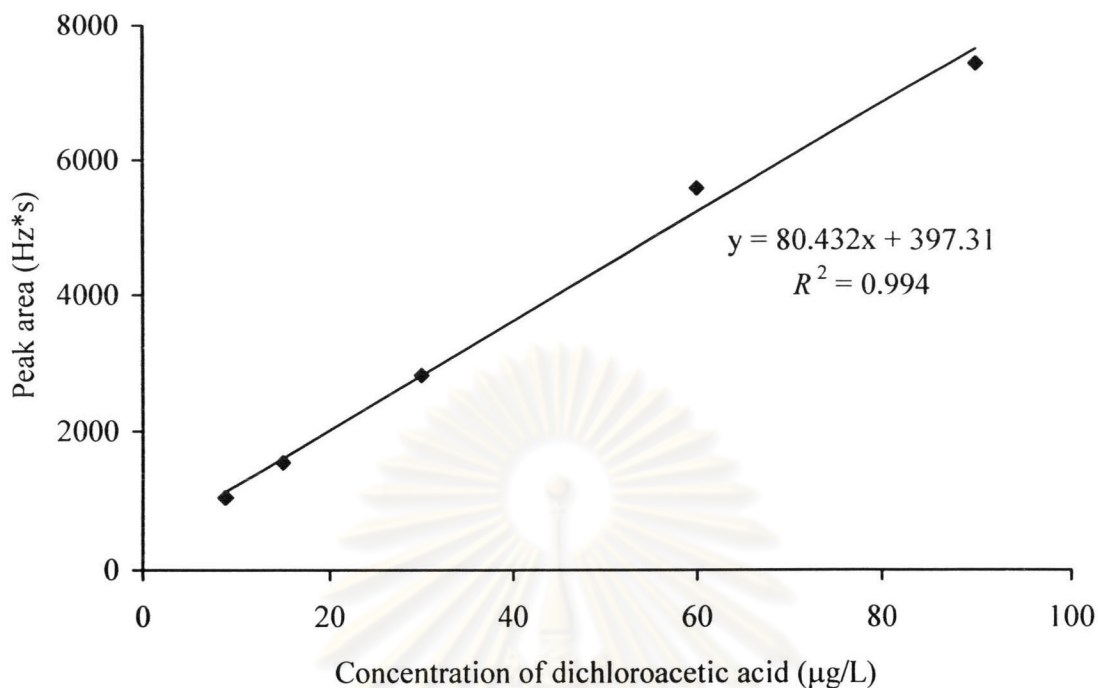


Figure A-7 The calibration curve of dichloroacetic acid by condition in Table 3.5.

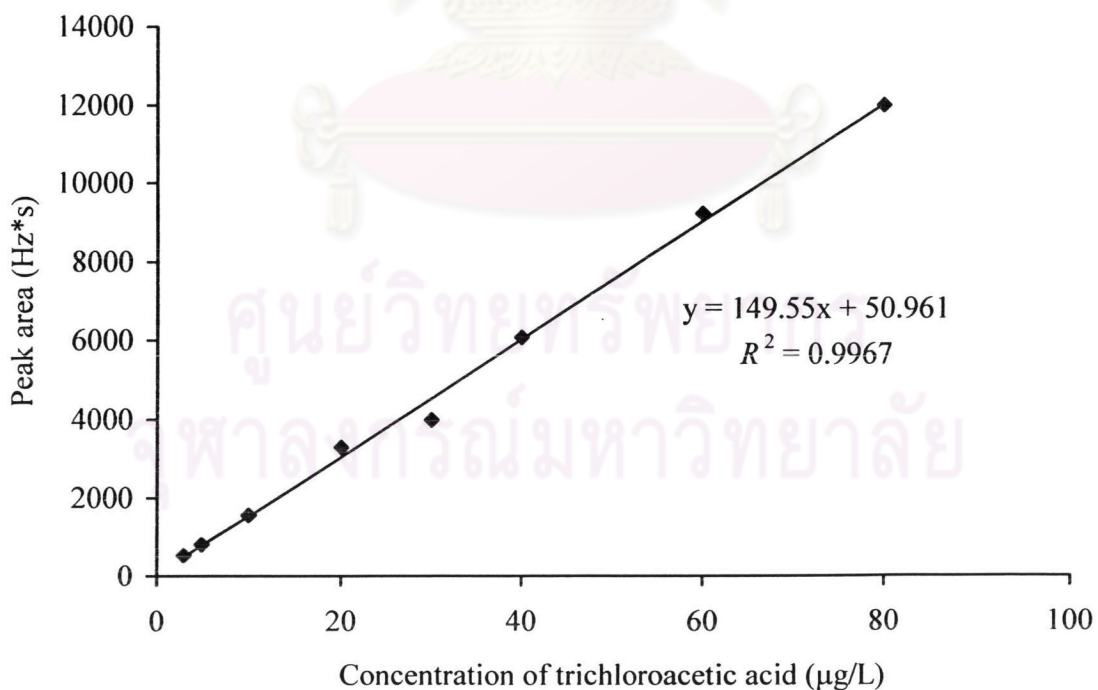


Figure A-8 The calibration curve of trichloroacetic acid by condition in Table 3.5.

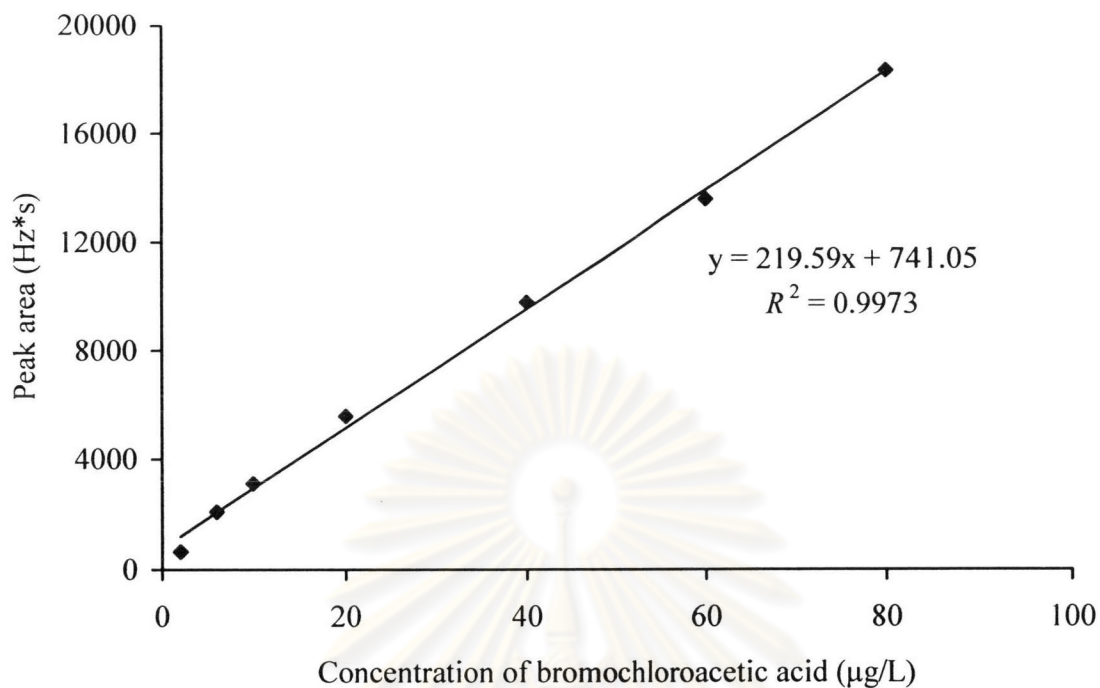


Figure A-9 The calibration curve of bromochloroacetic acid by condition in Table 3.5.

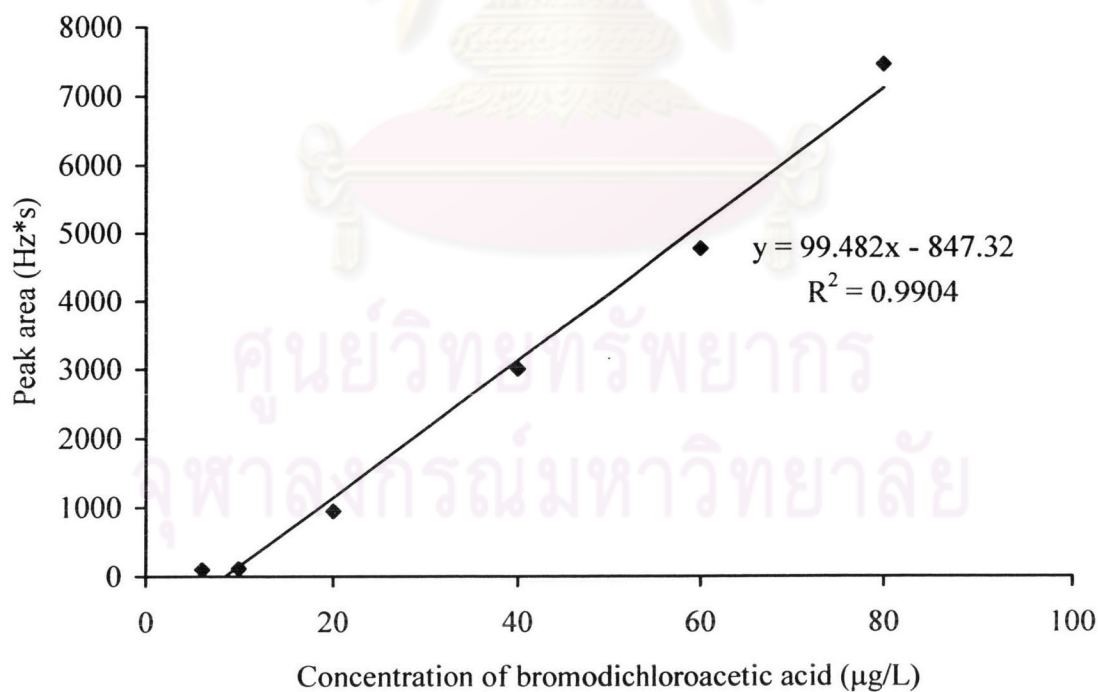


Figure A-10 The calibration curve of bromodichloroacetic acid by condition in Table 3.5.

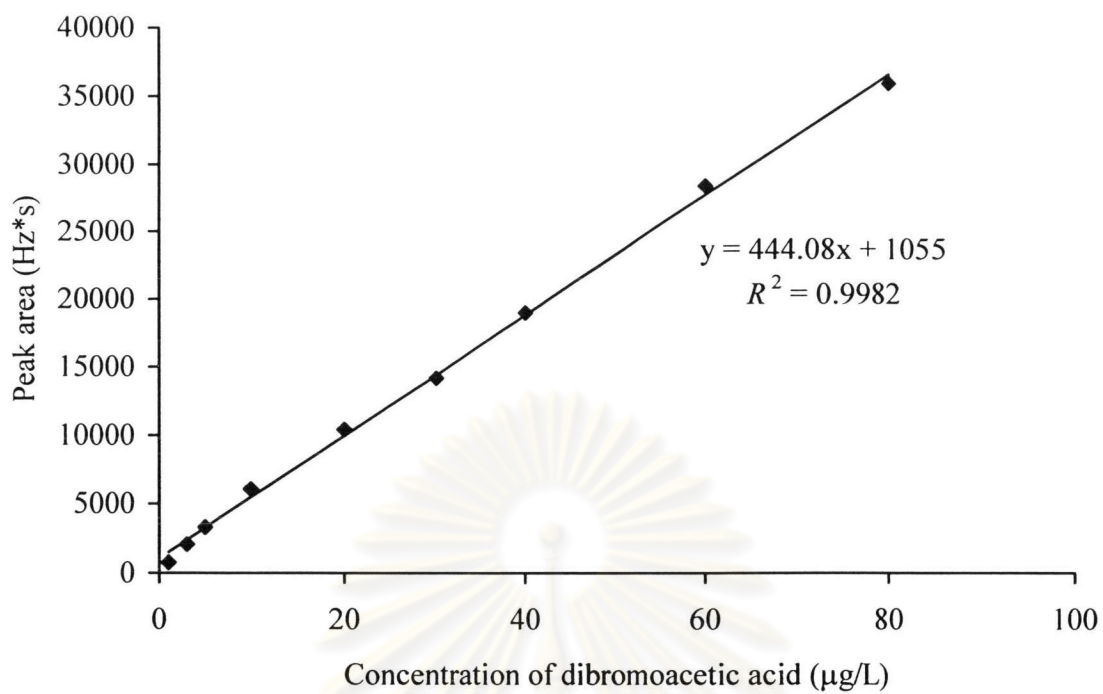


Figure A-11 The calibration curve of dibromoacetic acid by condition in Table 3.5.

ศูนย์วิทยทรัพยากร
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APPENDIX B

Table B-1 The peak area of stirring and non stirring solution after extracted with LPME.

Analyte	Batch	Peak area (Hz*s)	
		Stir	Non stir
MCAA	1	370	422
	2	362	339
	3	354	395
BCAA	1	4744	4771
	2	4392	3981
	3	4551	5017
DCAA	1	10056	9057
	2	8206	7836
	3	10753	9401
TCAA	1	6491	5226
	2	4752	4399
	3	6410	5249
BCAA	1	18773	16229
	2	14451	15573
	3	19945	16956
BDCAA	1	6679	5063
	2	4051	3183
	3	6415	5079
DBAA	1	21519	17455
	2	14871	18234
	3	21669	19163

Table B-2 The comparison of peak area of MCAA between stirring and non stirring solution using *F*-test and *t*-test at 95% confidence level.

F-Test Two-Sample for Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	362.4567	385.8667
Variance	59.14263	1812.295
Observations	3	3
df	2	2
F	0.032634	
P(F<=f) one-tail	0.031603	
F Critical one-tail	0.052632	

t-Test: Two-Sample Assuming Equal Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	362.4567	385.8667
Variance	59.14263	1812.295
Observations	3	3
Pooled Variance	935.719	
Hypothesized Mean Difference	0	
df	4	
t Stat	-0.93729	
P(T<=t) one-tail	0.200837	
t Critical one-tail	2.131846	
P(T<=t) two-tail	0.401674	
t Critical two-tail	2.776451	

Table B-3 The comparison of peak area of MBAA between stirring and non stirring solution using F -test and t -test at 95% confidence level.

F -Test Two-Sample for Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	4562.663	4590.177
Variance	30937.94	293168.9
Observations	3	3
df	2	2
F	0.105529	
P(F<=f) one-tail	0.095456	
F Critical one-tail	0.052632	

t -Test: Two-Sample Assuming Unequal Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	4562.663	4590.177
Variance	30937.94	293168.9
Observations	3	3
Hypothesized Mean Difference	0	
df	2	
t Stat	-0.08371	
P(T<=t) one-tail	0.470457	
t Critical one-tail	2.919987	
P(T<=t) two-tail	0.940914	
t Critical two-tail	4.302656	

Table B-4 The comparison of peak area of DCAA between stirring and non stirring solution using F -test and t -test at 95% confidence level.

F-Test Two-Sample for Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	9671.87	8765.217
Variance	1732387	676121.8
Observations	3	3
df	2	2
F	2.562241	
P(F<=f) one-tail	0.280722	
F Critical one-tail	19.00003	

t-Test: Two-Sample Assuming Equal Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	9671.87	8765.217
Variance	1732387	676121.8
Observations	3	3
Pooled Variance	1204254	
Hypothesized Mean Difference	0	
df	4	
t Stat	1.011877	
P(T<=t) one-tail	0.184416	
t Critical one-tail	2.131846	
P(T<=t) two-tail	0.368832	
t Critical two-tail	2.776451	

Table B-5 The comparison of peak area of TCAA between stirring and non stirring solution using *F*-test and *t*-test at 95% confidence level.

F-Test Two-Sample for Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	5884.693	4958.383
Variance	963587.8	234603.3
Observations	3	3
df	2	2
F	4.107307	
P(F<=f) one-tail	0.195798	
F Critical one-tail	19.00003	

t-Test: Two-Sample Assuming Equal Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	5884.693	4958.383
Variance	963587.8	234603.3
Observations	3	3
Pooled Variance	599095.5	
Hypothesized Mean Difference	0	
df	4	
t Stat	1.46573	
P(T<=t) one-tail	0.108299	
t Critical one-tail	2.131846	
P(T<=t) two-tail	0.216597	
t Critical two-tail	2.776451	

Table B-6 The comparison of peak area of BCAA between stirring and non stirring solution using F -test and t -test at 95% confidence level.

F -Test Two-Sample for Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	17723.77	16253.17
Variance	8373211	478745
Observations	3	3
df	2	2
F	17.48992	
P(F<=f) one-tail	0.054084	
F Critical one-tail	19.00003	

t -Test: Two-Sample Assuming Equal Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	17723.77	16253.17
Variance	8373211	478745
Observations	3	3
Pooled Variance	4425978	
Hypothesized Mean Difference	0	
df	4	
t Stat	0.856122	
P(T<=t) one-tail	0.220093	
t Critical one-tail	2.131846	
P(T<=t) two-tail	0.440186	
t Critical two-tail	2.776451	

Table B-7 The comparison of peak area of BDCAA between stirring and non stirring solution using *F*-test and *t*-test at 95% confidence level.

F-Test Two-Sample for Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	5715.48	4442.267
Variance	2093620	1188751
Observations	3	3
df	2	2
F	1.761193	
P(F<=f) one-tail	0.362162	
F Critical one-tail	19.00003	

t-Test: Two-Sample Assuming Equal Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	5715.48	4442.267
Variance	2093620	1188751
Observations	3	3
Pooled Variance	1641186	
Hypothesized Mean Difference	0	
df	4	
t Stat	1.217217	
P(T<=t) one-tail	0.145211	
t Critical one-tail	2.131846	
P(T<=t) two-tail	0.290422	
t Critical two-tail	2.776451	

Table B-8 The comparison of peak area of DBAA between stirring and non stirring solution using *F*-test and *t*-test at 95% confidence level.

F-Test Two-Sample for Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	19353.5	18284.6
Variance	15073877	731143.3
Observations	3	3
df	2	2
F	20.61686	
P(F<=f) one-tail	0.04626	
F Critical one-tail	19.00003	

t-Test: Two-Sample Assuming Unequal Variances

	<i>Variable 1</i>	<i>Variable 2</i>
Mean	19353.5	18284.6
Variance	15073877	731143.3
Observations	3	3
Hypothesized Mean Difference	0	
df	2	
t Stat	0.465693	
P(T<=t) one-tail	0.343613	
t Critical one-tail	2.919987	
P(T<=t) two-tail	0.687226	
t Critical two-tail	4.302656	

APPENDIX C

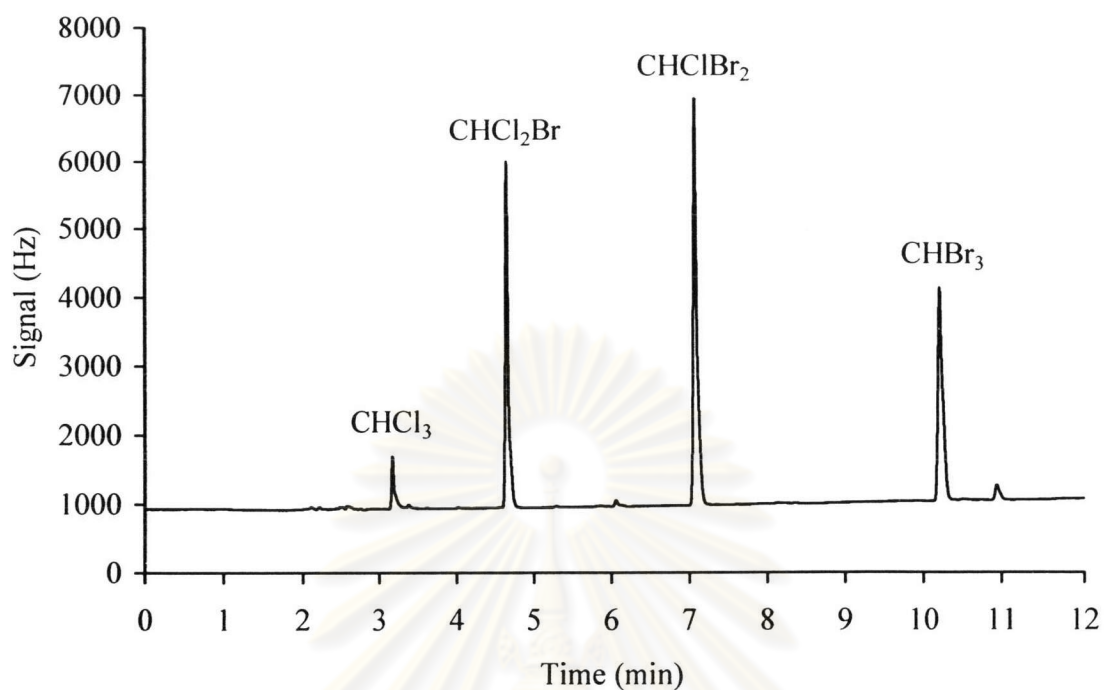


Figure C-1 Chromatogram of THMs in dodecane after extracted using liquid-phase microextraction.

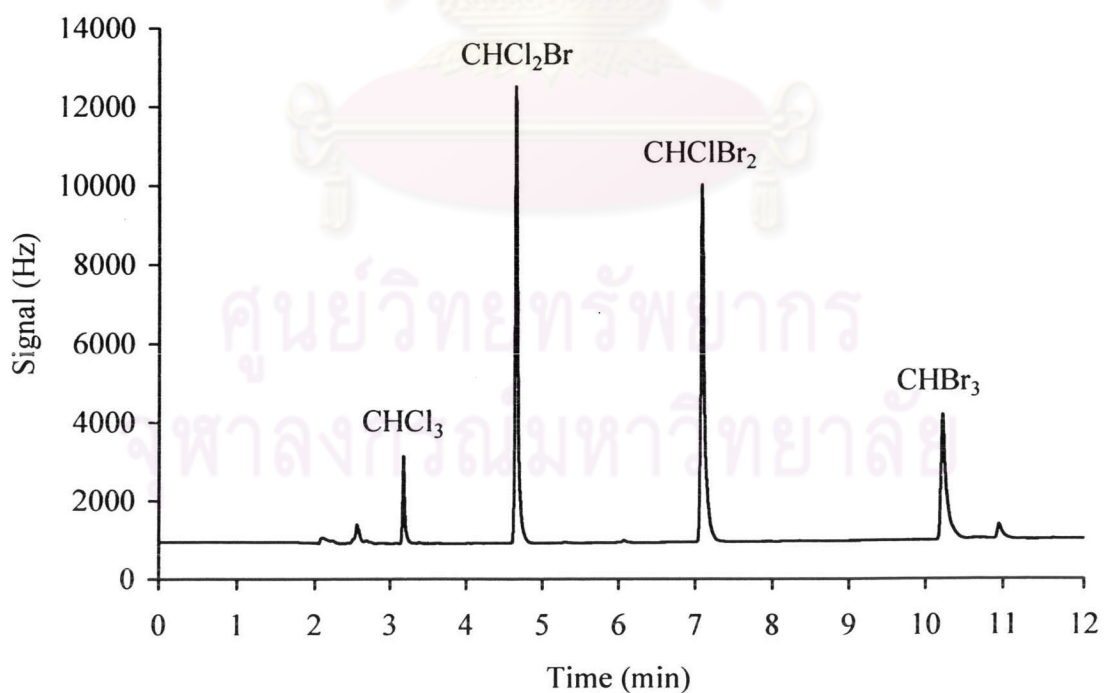


Figure C-2 Chromatogram of THMs in 1-octanol after extracted using liquid-phase microextraction.

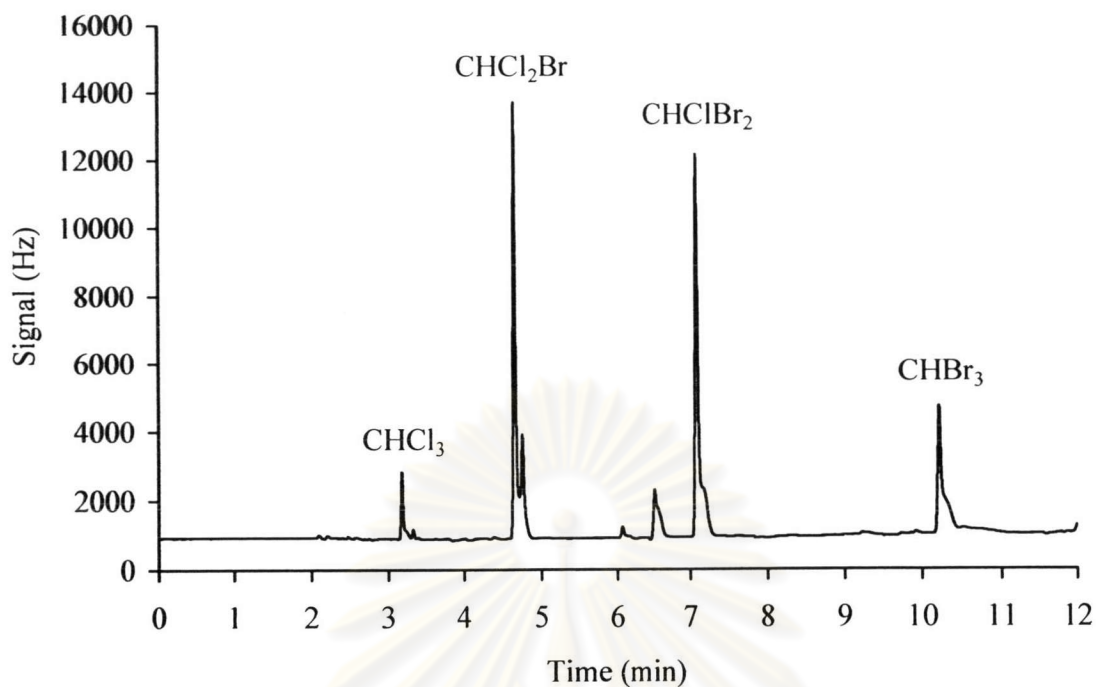


Figure C-3 Chromatogram of THMs in dihexyl ether after extracted using liquid-phase microextraction.

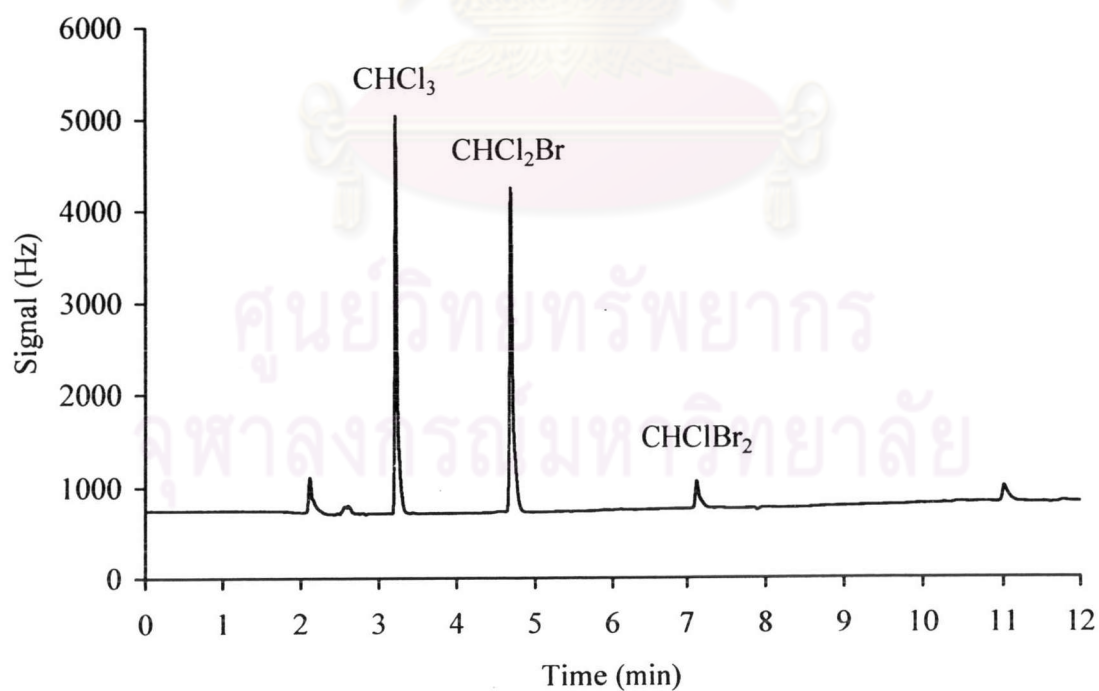


Figure C-4 Chromatogram of THMs in drinking water 1 after extracted using liquid-phase microextraction.

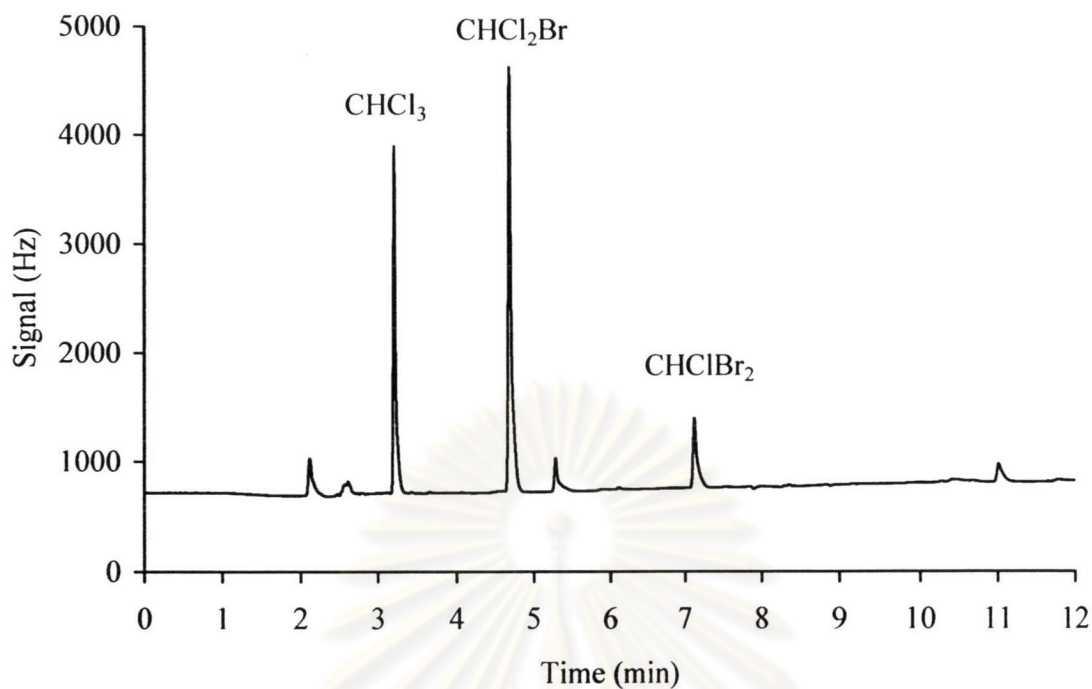


Figure C-5 Chromatogram of THMs in drinking water 2 after extracted using liquid-phase microextraction.

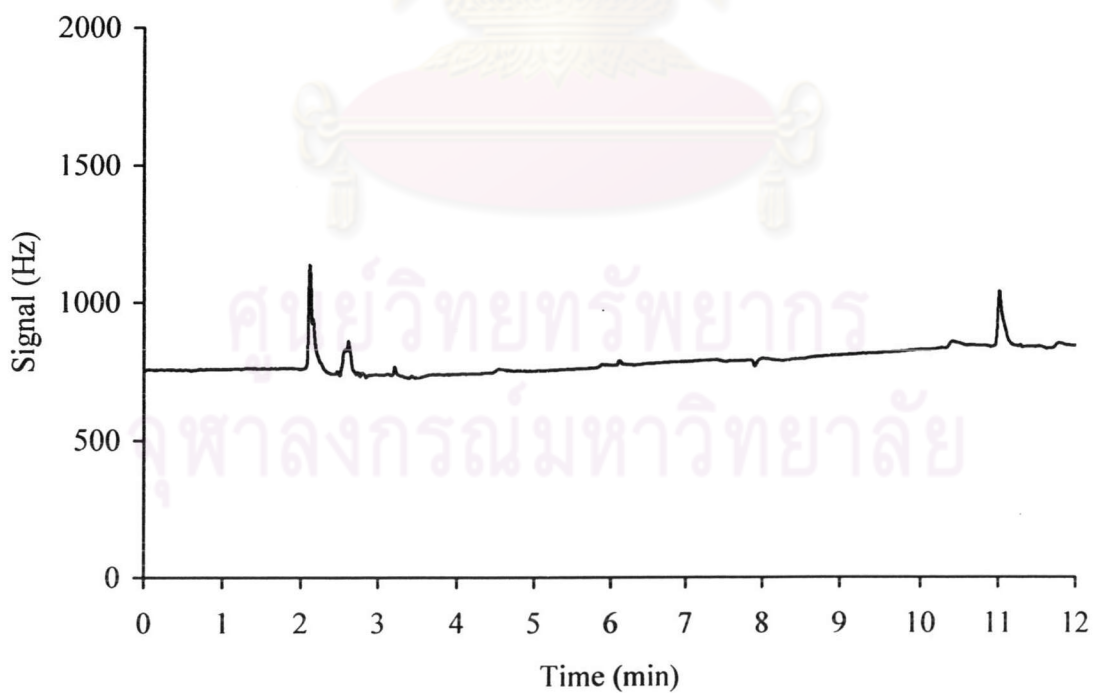


Figure C-6 Chromatogram of THMs in drinking water 3 after extracted using liquid-phase microextraction.

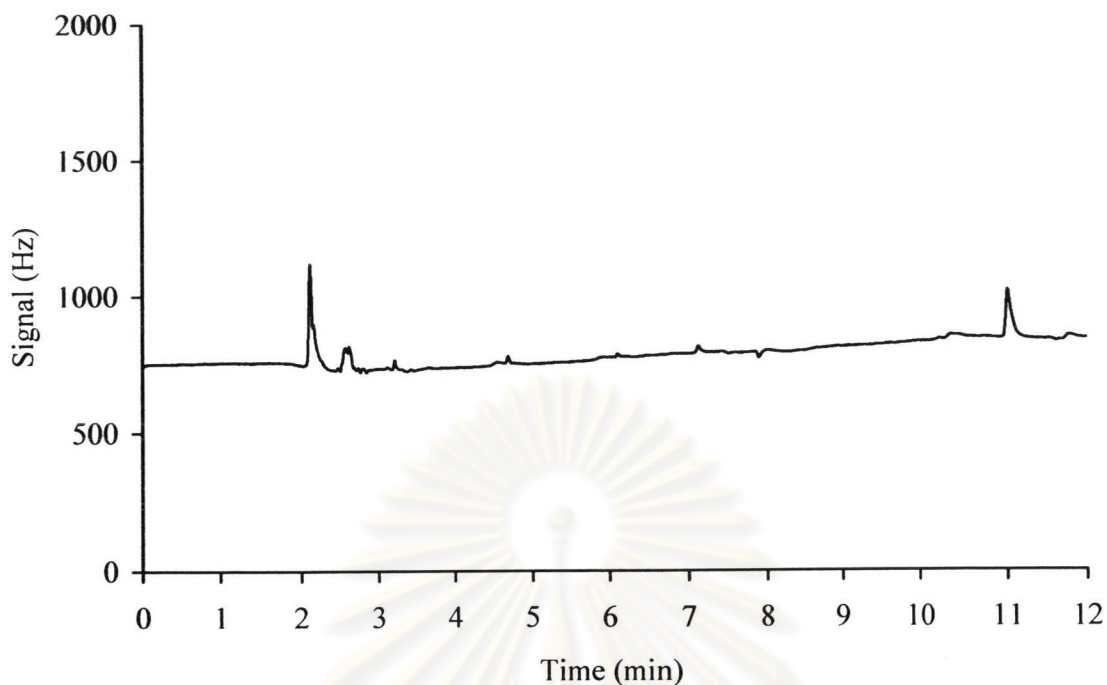


Figure C-7 Chromatogram of THMs in drinking water 4 after extracted using liquid-phase microextraction.

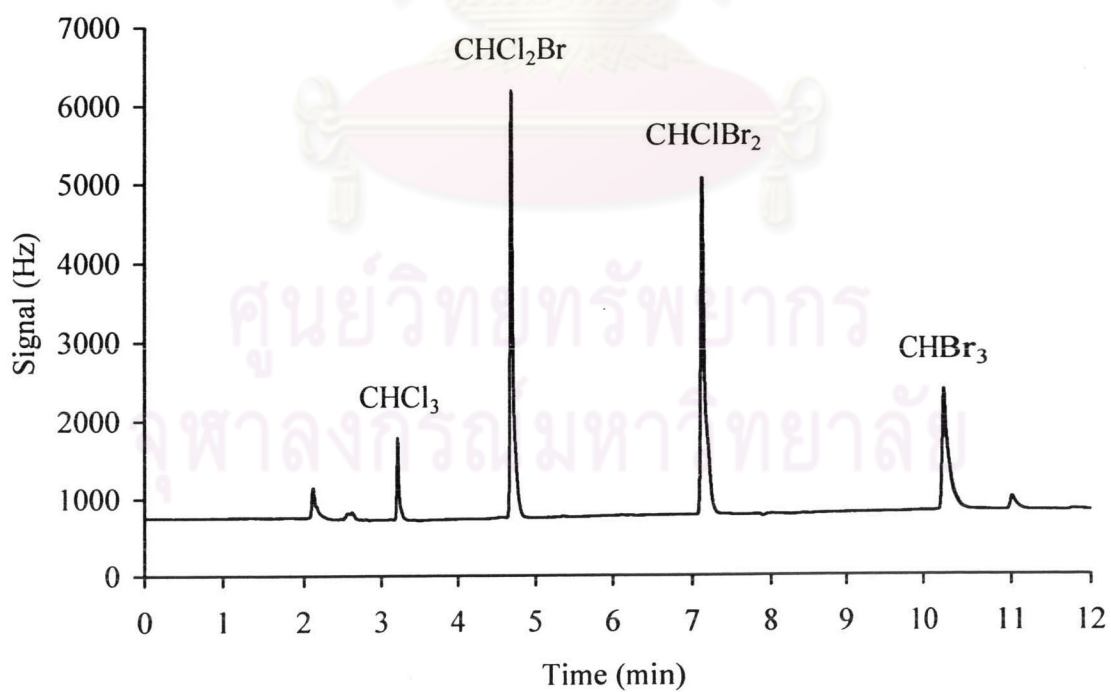


Figure C-8 Chromatogram of THMs in drinking water 4 + spiked THMs 10 $\mu\text{g/L}$ after extracted using liquid-phase microextraction.

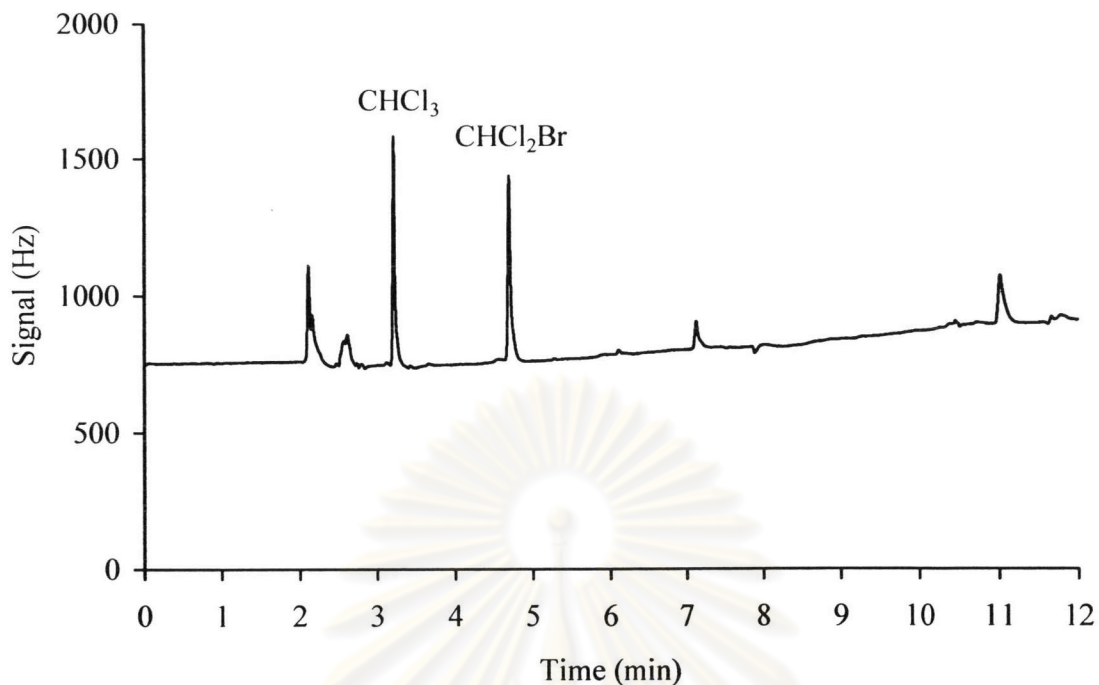


Figure C-9 Chromatogram of THMs in drinking water 5 after extracted using liquid-phase microextraction.

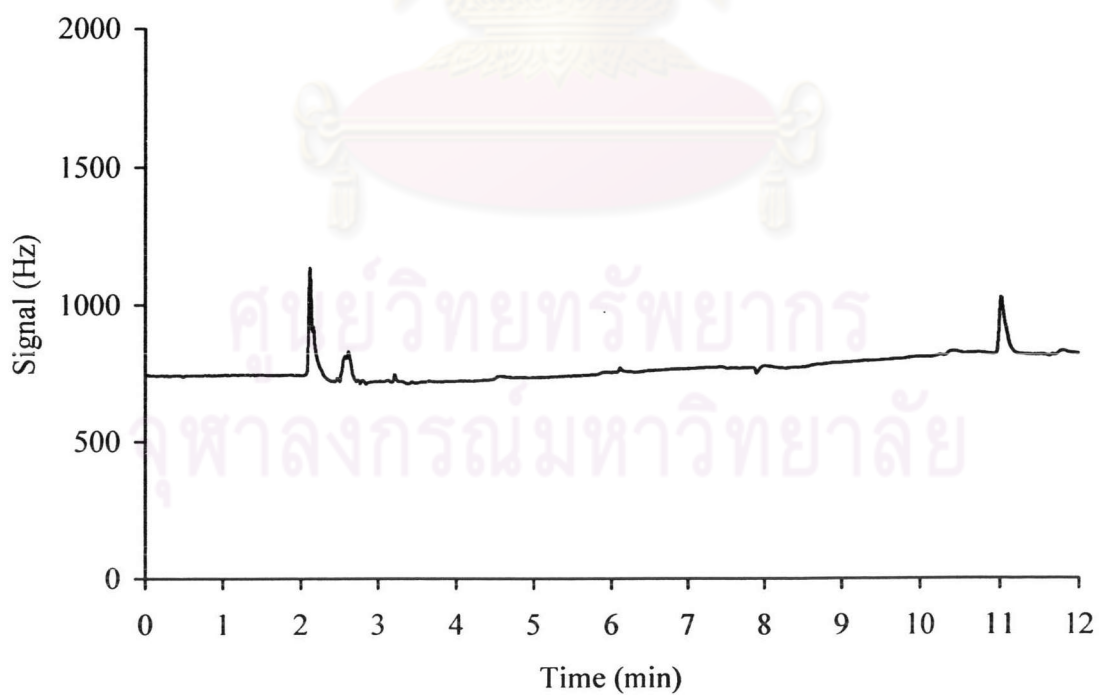


Figure C-10 Chromatogram of THMs in mineral water 1 after extracted using liquid-phase microextraction.

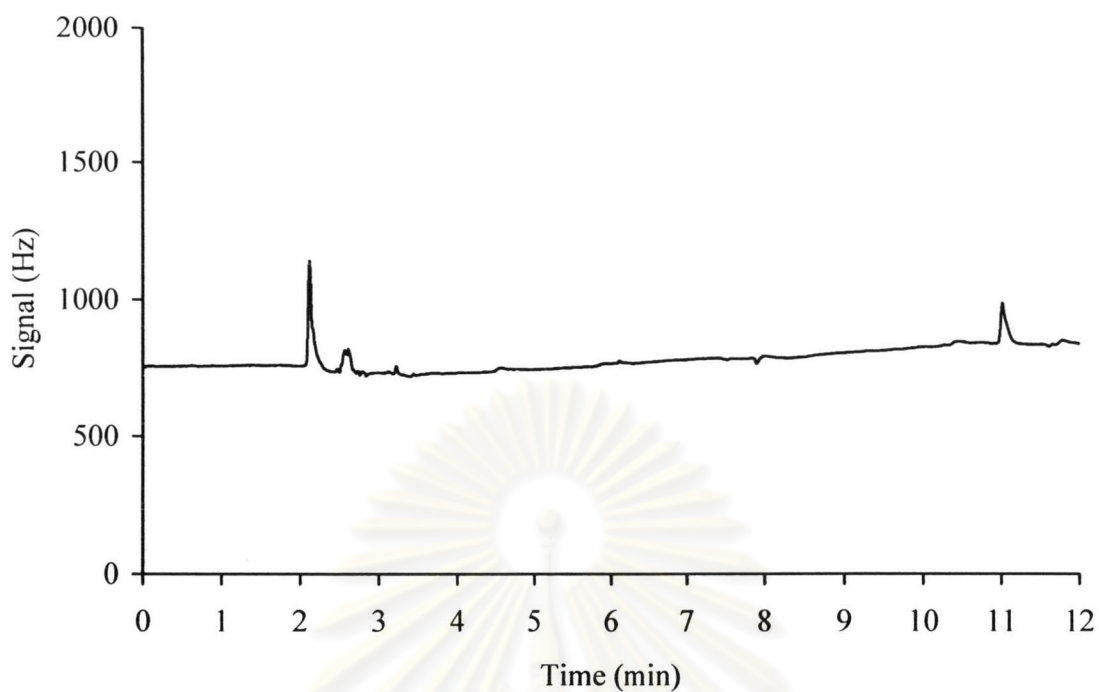


Figure C-11 Chromatogram of THMs in mineral water 2 after extracted using liquid-phase microextraction.

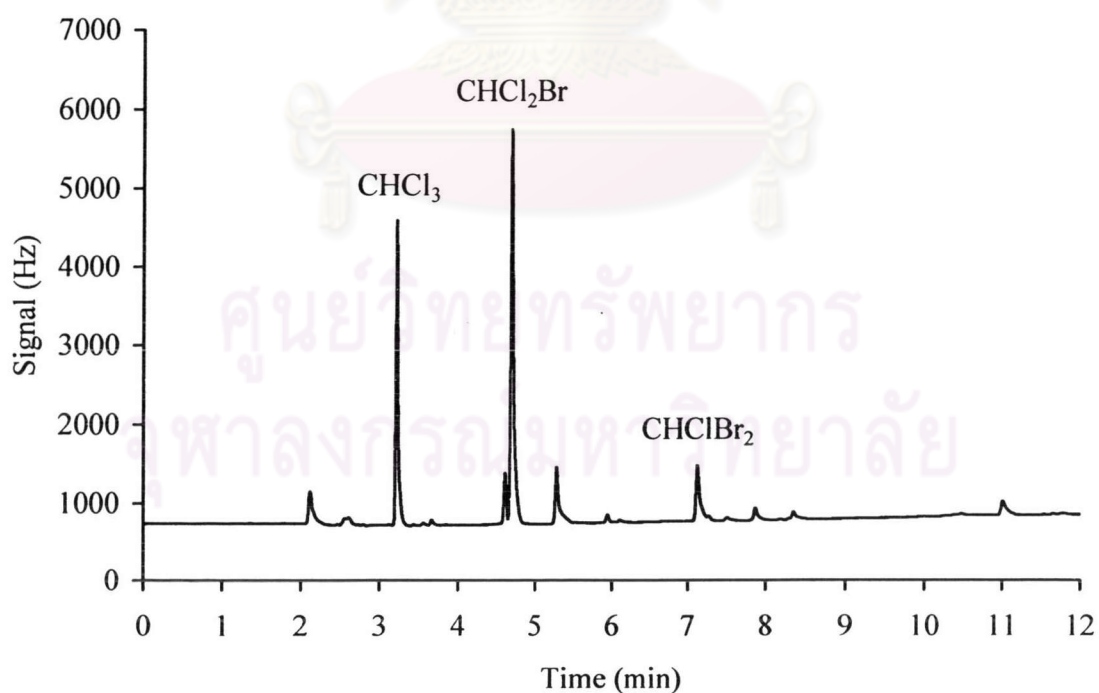


Figure C-12 Chromatogram of THMs in tap water 1 after extracted using liquid-phase microextraction.

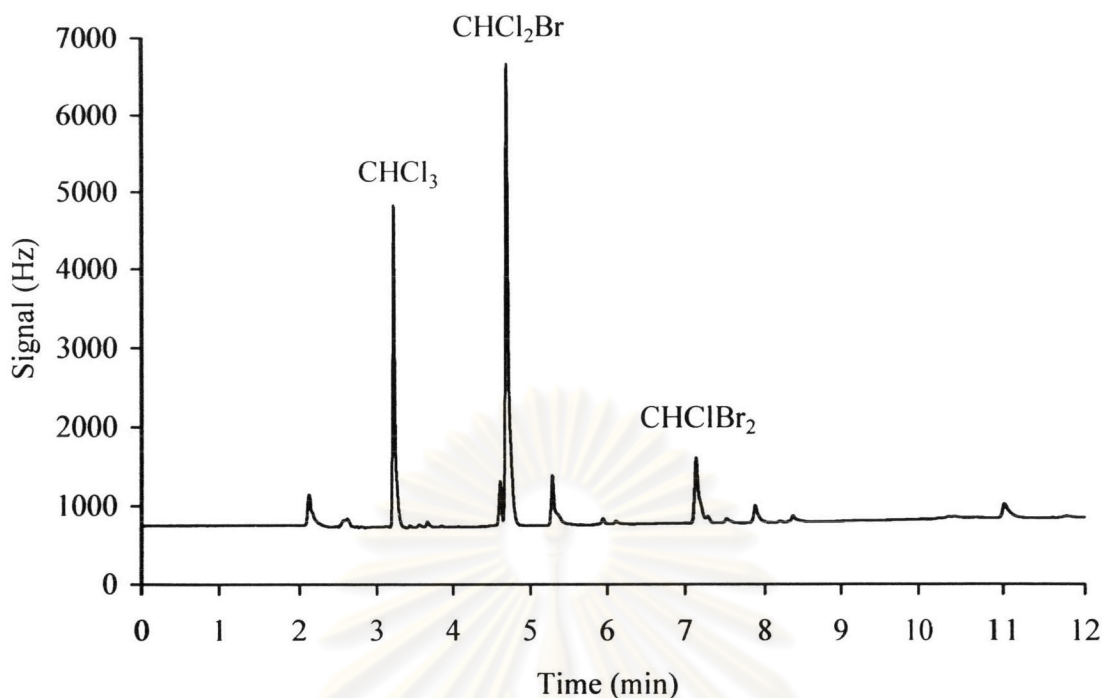


Figure C-13 Chromatogram of THMs in tap water 2 after extracted using liquid-phase microextraction.

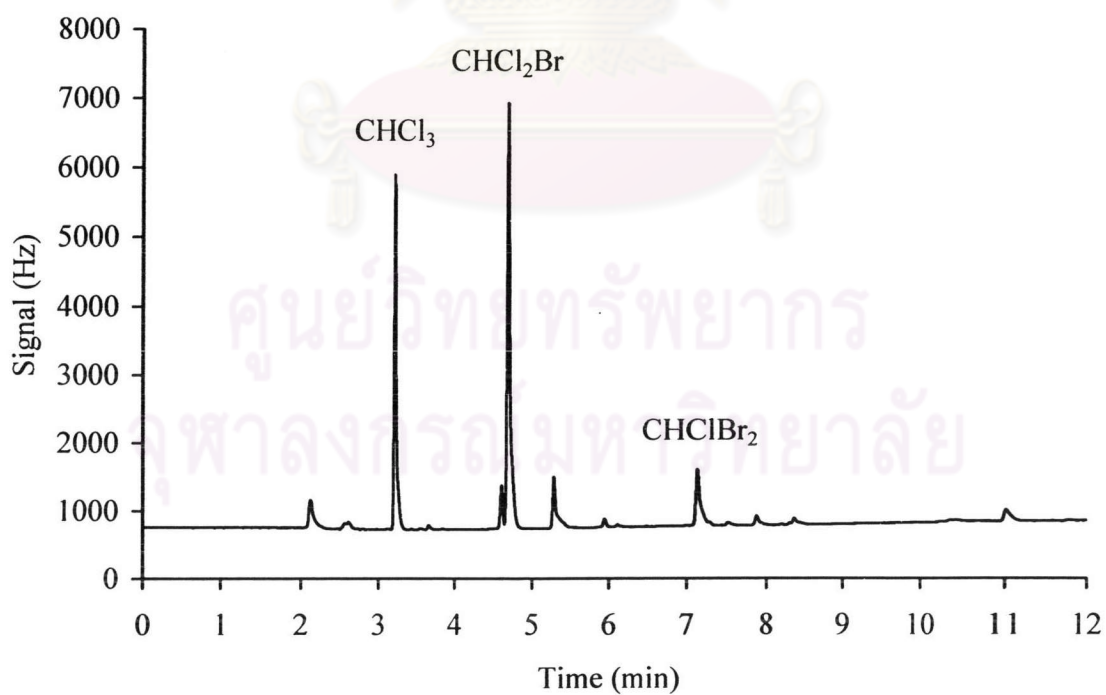


Figure C-14 Chromatogram of THMs in tap water 3 after extracted using liquid-phase microextraction.

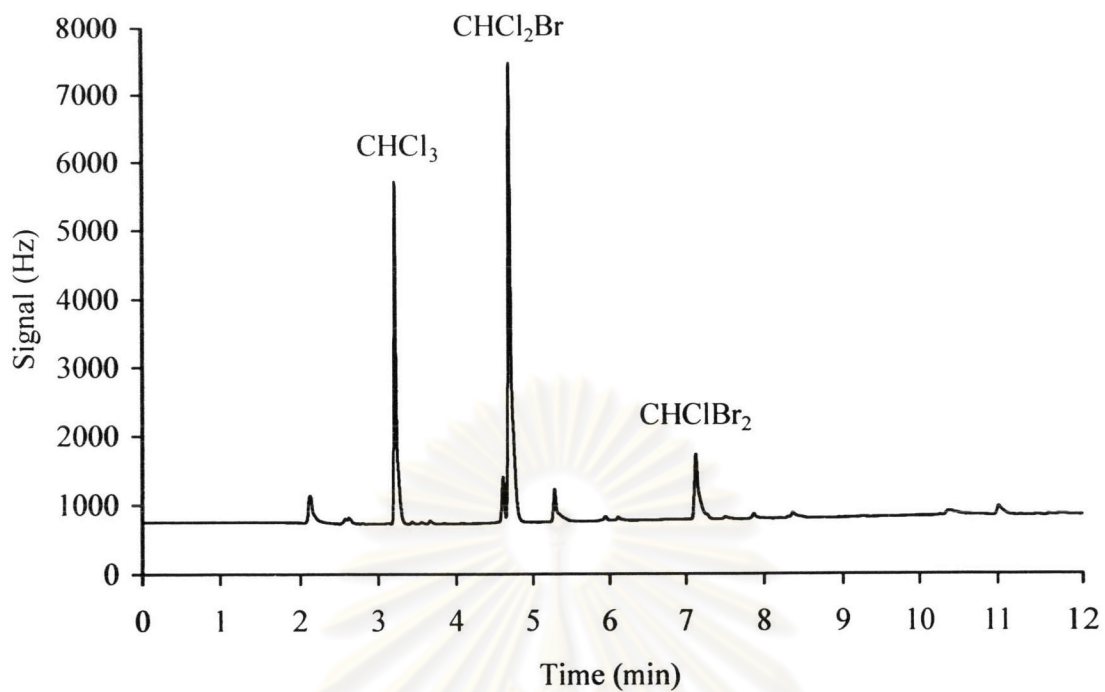


Figure C-15 Chromatogram of THMs in tap water 4 after extracted using liquid-phase microextraction.

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

APPENDIX D

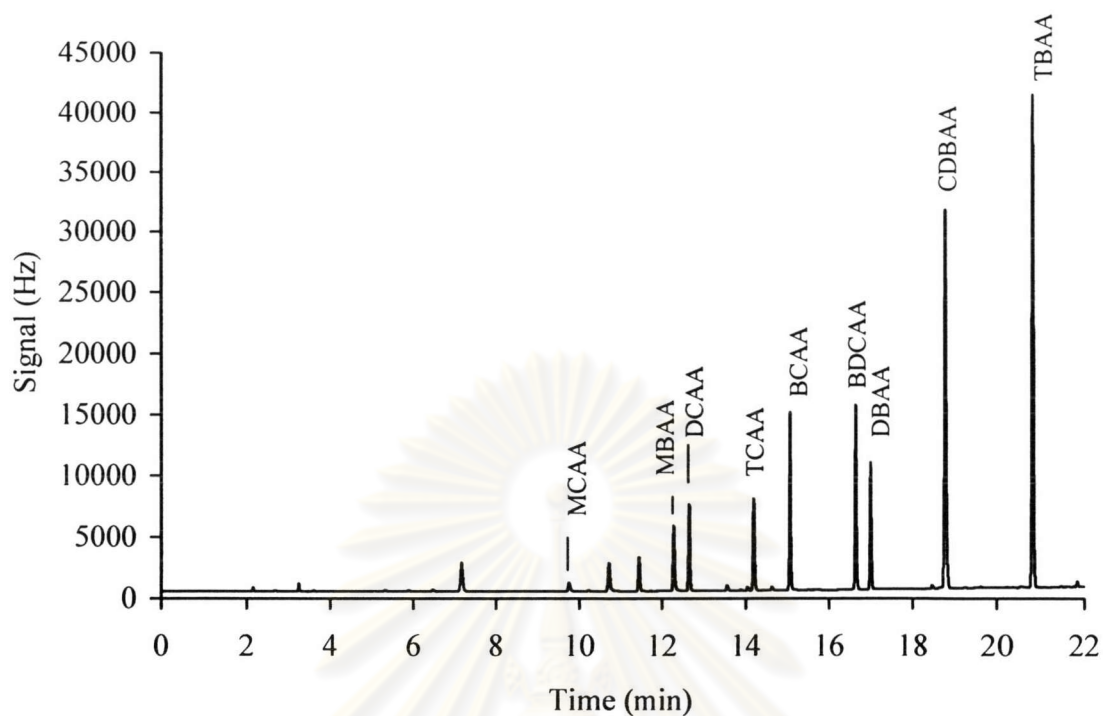


Figure D-1 Chromatograms of standard haloacetic methyl ester in dodecane.

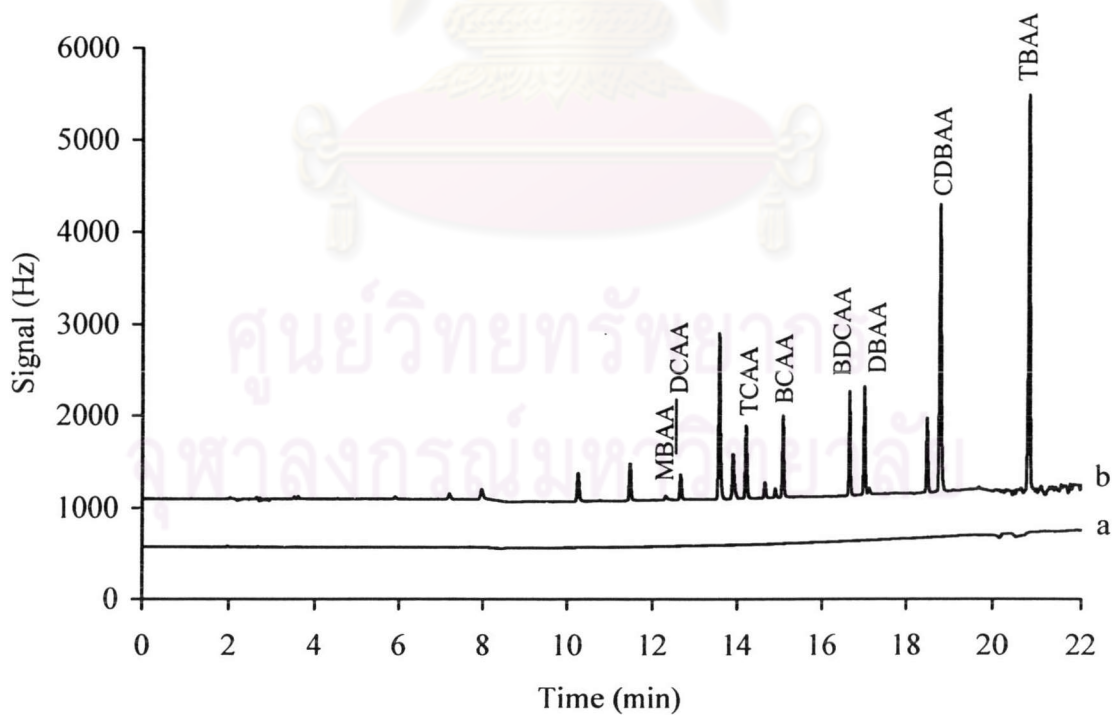


Figure D-2 Chromatograms of HAAs in dodecane after extracted using liquid-phase microextraction (a) blank; (b) spike water sample with HAAs.

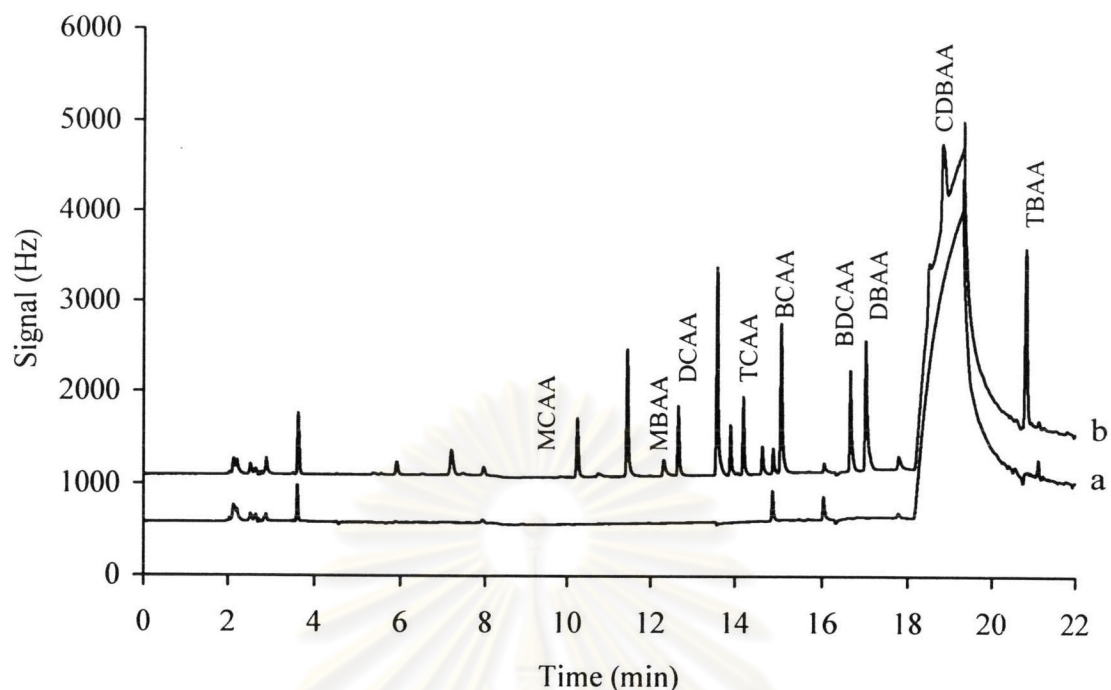


Figure D-3 Chromatograms of HAAs in 1-octanol after extracted using liquid-phase microextraction (a) blank; (b) spike water sample with HAAs.

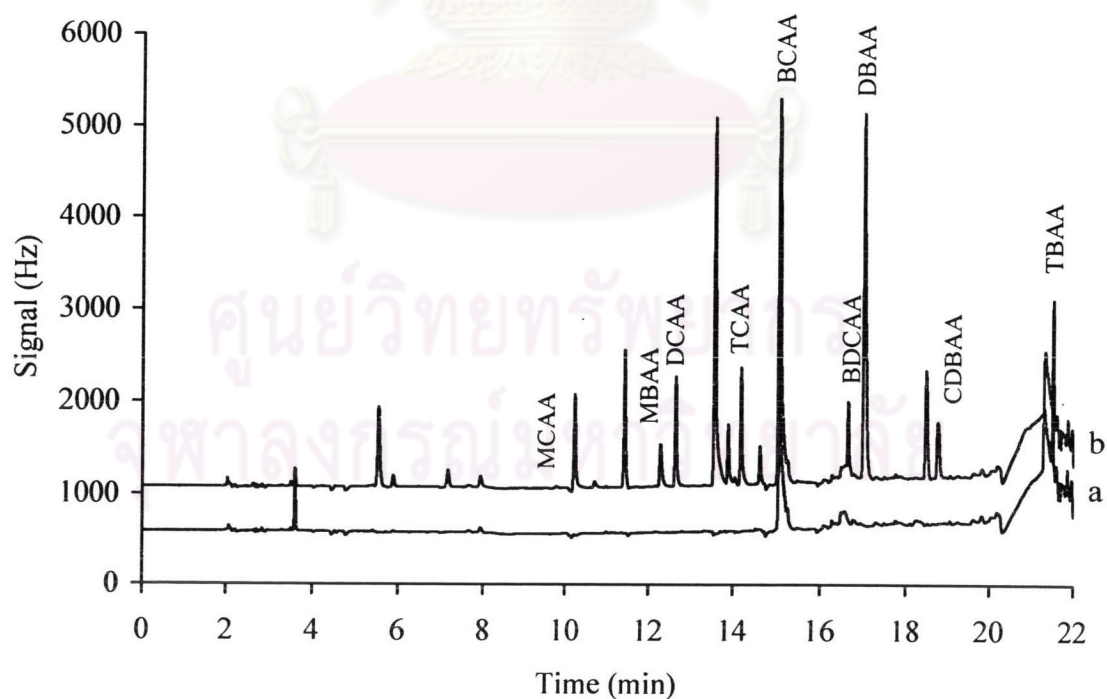


Figure D-4 Chromatograms of HAAs in dihexyl ether after extracted using liquid-phase microextraction (a) blank; (b) spike water sample with HAAs.

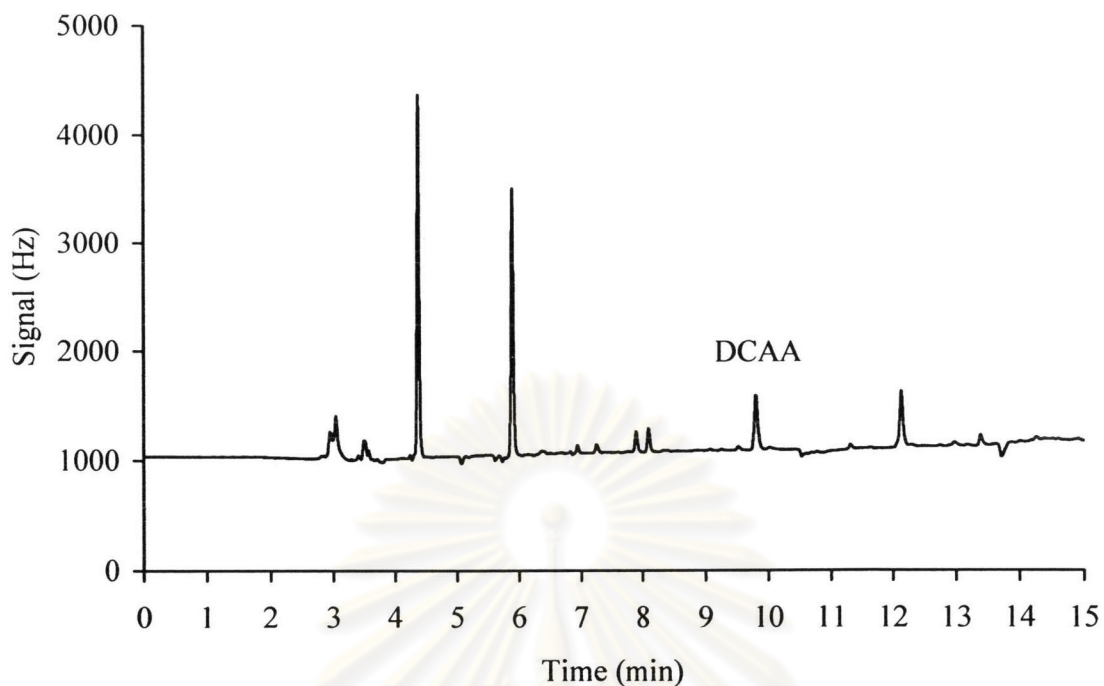


Figure D-5 Chromatogram of HAAs in drinking water 1 after extracted using liquid-phase microextraction.

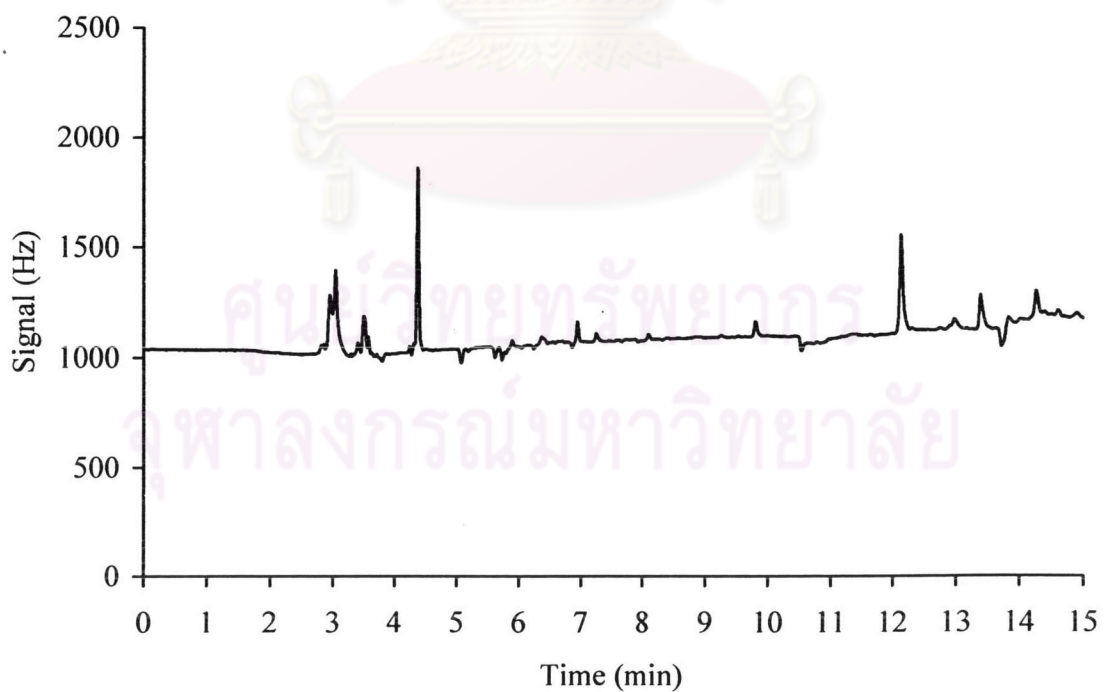


Figure D-6 Chromatogram of HAAs in drinking water 4 after extracted using liquid-phase microextraction.

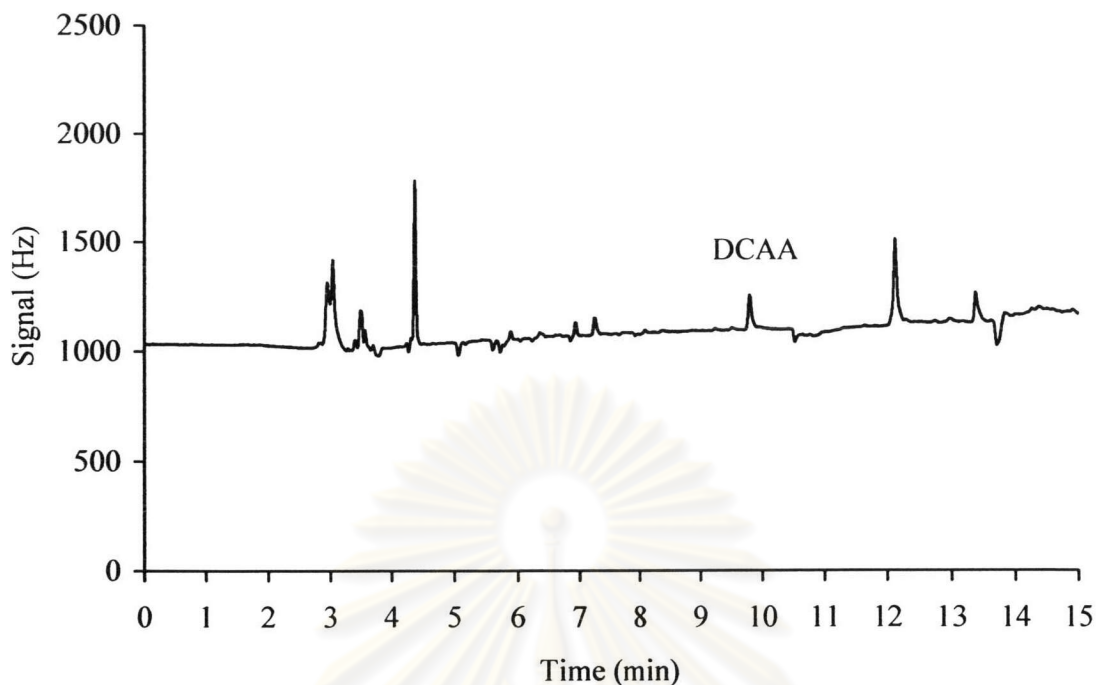


Figure D-7 Chromatogram of HAAs in drinking water 5 after extracted using liquid-phase microextraction.

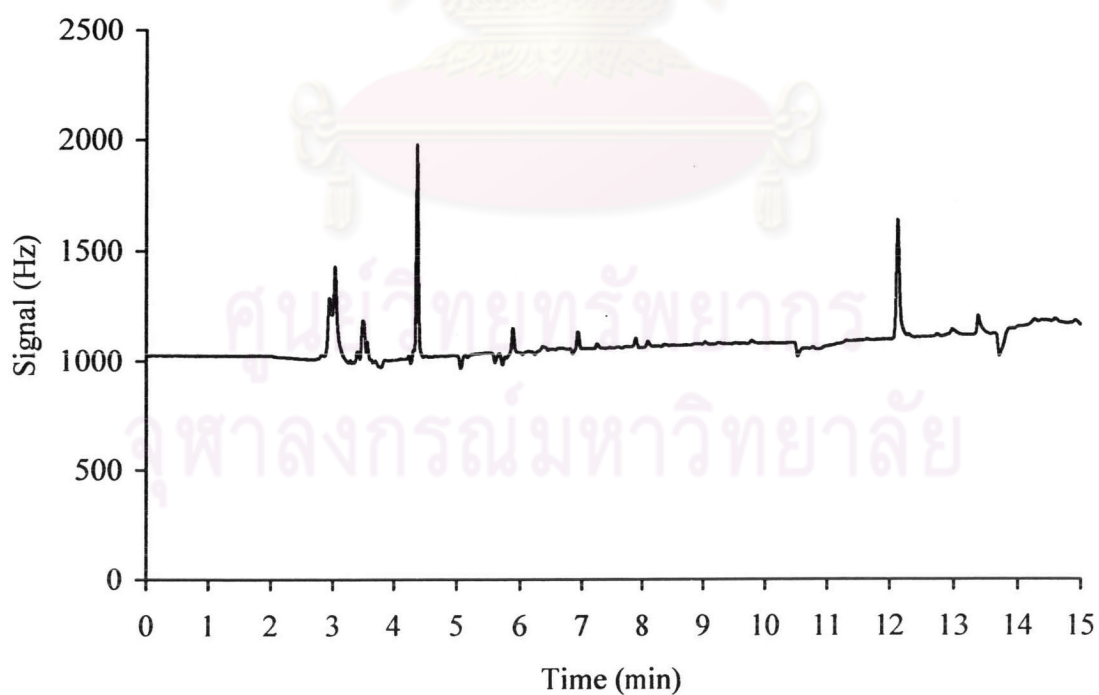


Figure D-8 Chromatogram of HAAs in drinking water 6 after extracted using liquid-phase microextraction.

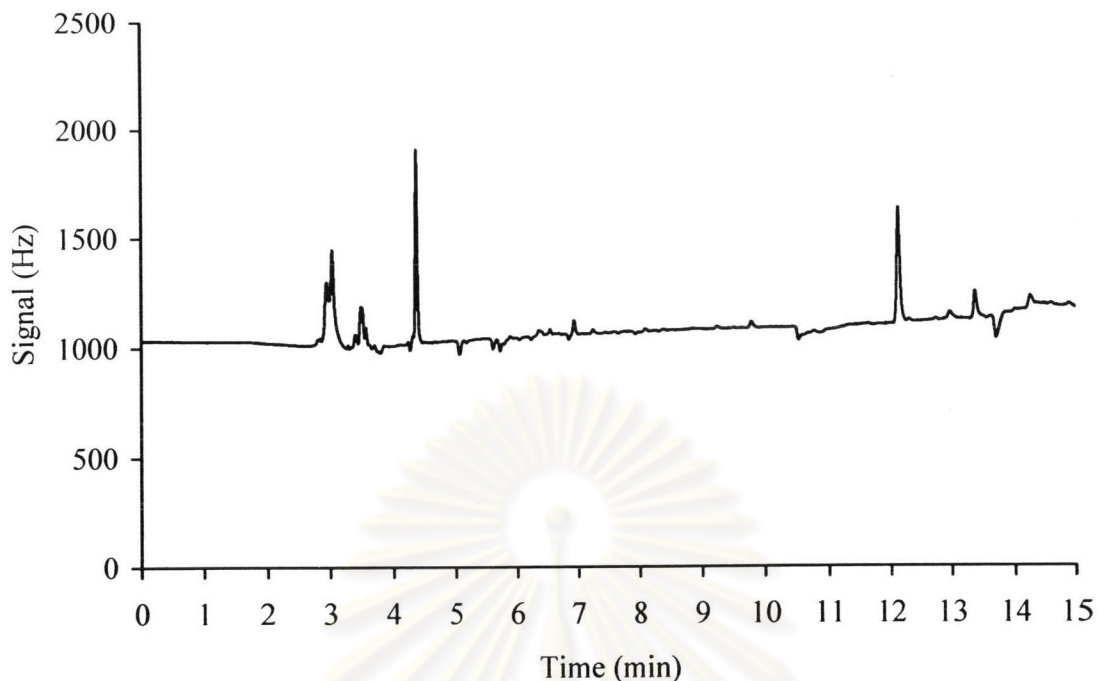


Figure D-9 Chromatogram of HAAs in mineral water 1 after extracted using liquid-phase microextraction.

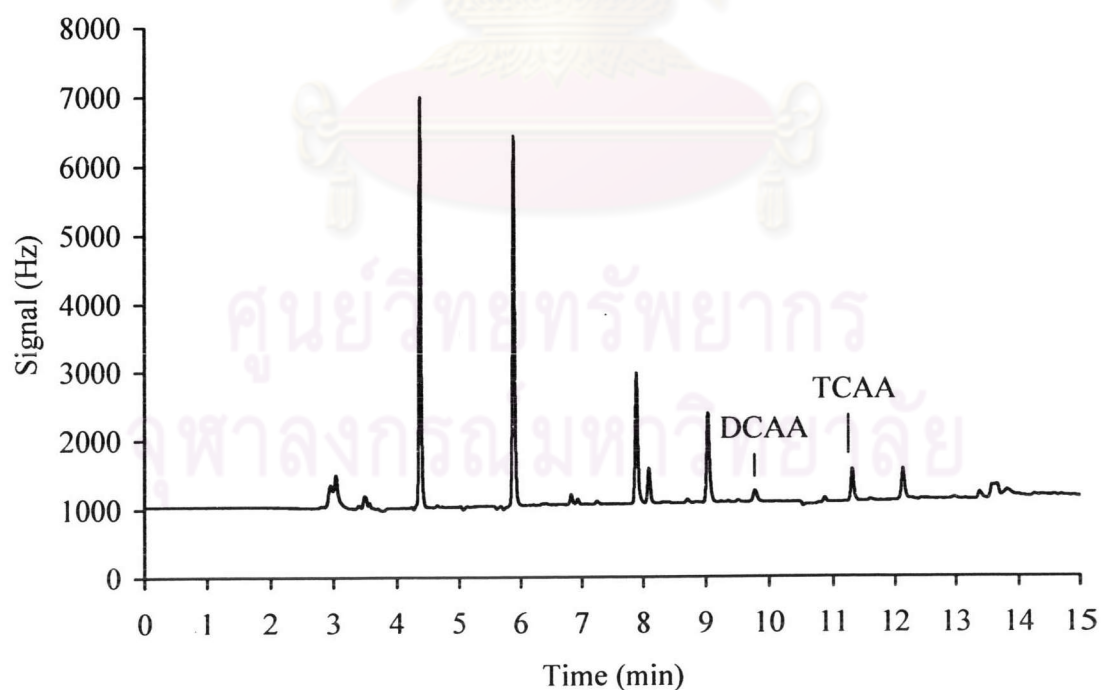


Figure D-10 Chromatogram of HAAs in tap water 1 after extracted using liquid-phase microextraction.

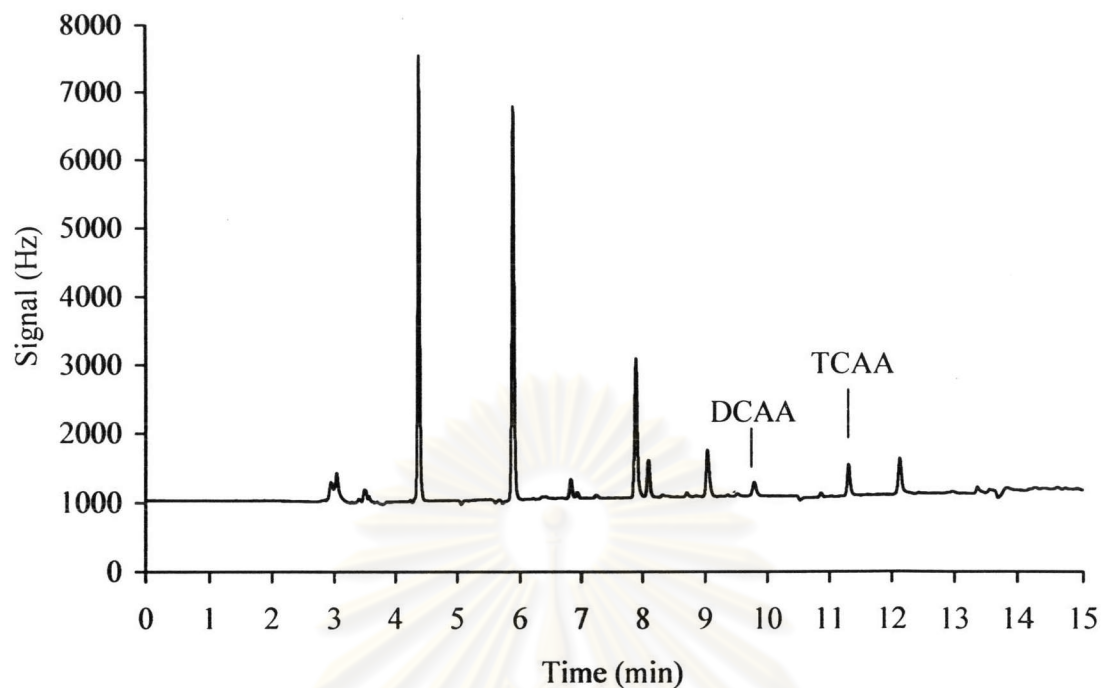


Figure D-11 Chromatogram of HAAs in tap water 4 after extracted using liquid-phase microextraction.

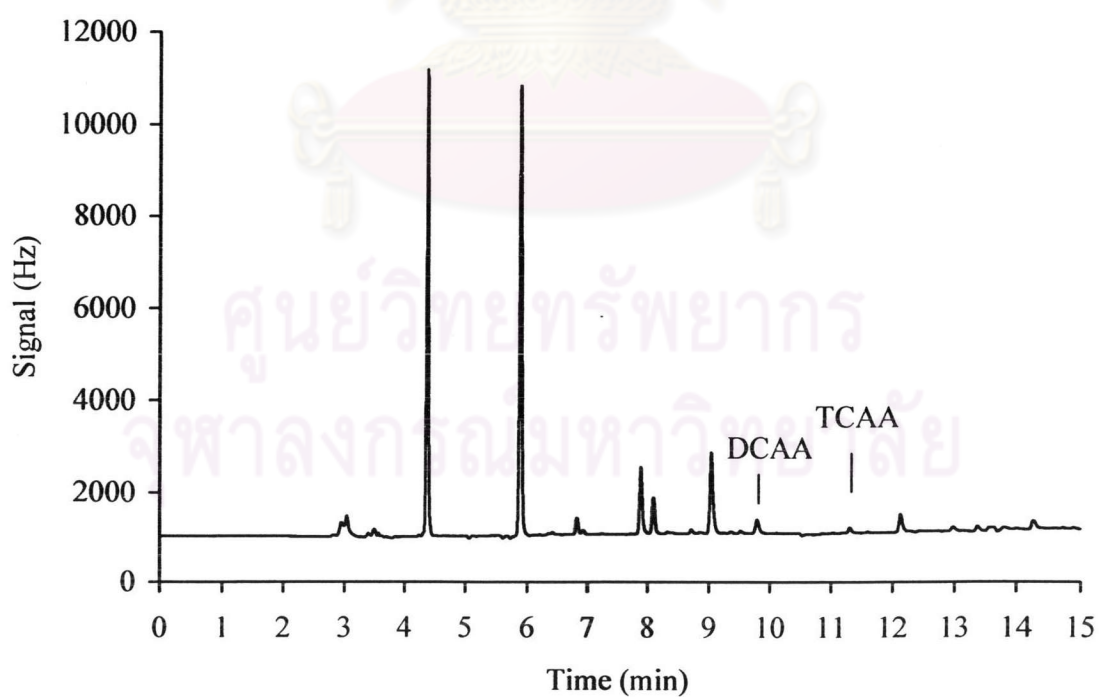


Figure D-12 Chromatogram of HAAs in tap water 5 after extracted using liquid-phase microextraction.

VITA

Name : Mr. Narongchai Vora-adisak

Date of birth : 28th January 1981

Place of birth : Bangkok, THAILAND

Education : 2003-2005 Chulalongkorn University
Master's Degree of Science (Analytical Chemistry)

1999-2002 Chulalongkorn University
Bachelor's Degree of Science (Chemistry)

1996-1998 Wat Suthivararam School
Senior High School

1993-1995 Wat Suthivararam School
Junior High School

Presentation :

1. N. Vora-adisak and P. Varanusupakul, "Liquid phase microextraction using hollow fiber membrane for gas chromatographic determination of trihalomethanes in water samples", Presented at *The First Graduate Congress of Mathematics and Physical Science 2005*, 6-7 December 2005, Bangkok, Thailand. (International)
2. N. Vora-adisak and P. Varanusupakul, "Liquid phase microextraction with supported liquid hollow fiber membrane for sample preparation of trihalomethanes in water samples", Presented at *31st Congress on Science and Technology of Thailand*, 18-20 October 2005, Nakhon Ratchasima, Thailand. (National)

Publication :

1. N. Vora-adisak and P. Varanusupakul, "A simple supported liquid hollow fiber membrane microextraction for sample preparation of trihalomethanes in water samples", *J. Chromatogr. A*, In press.