

REFERENCES

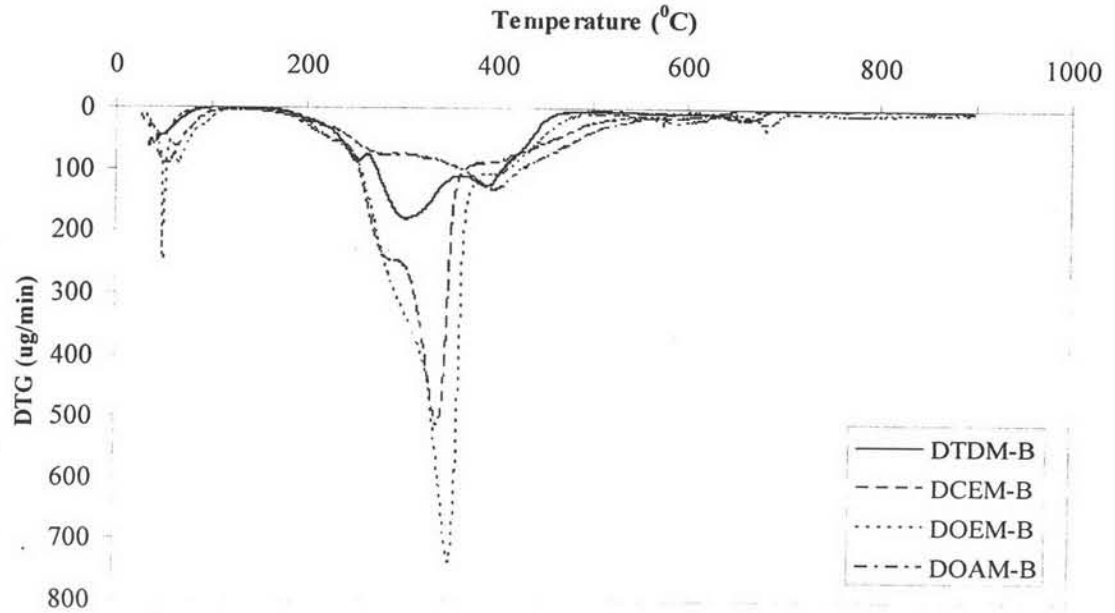
- Sinha Ray, S., and Okamoto, M. (2003) Polymer/layered silicate nanocomposites: a review from preparation to processing. Prog. Polym. Sci., 28, 1539-1641.
- Minisini, B. and Tsobnang, F. (2005) Molecular mechanics studies of specific interactions in organomodified clay nanocomposite. Composites: Part A, 36, 531-537.
- Fornes, T.D., Hunter, D.L. and Paula, D.R. (2004) Nylon-6 nanocomposites from alkylammonium-modified clay: The role of alkyl tails on exfoliation. Macromolecules, 37, 1793-1798.
- García-López, D., Picazo, O., Merino, J. C. and Pastor, J. M. (2003) Polypropylene-clay nanocomposites: effect of compatibilizing agents on clay dispersion. European Polymer Journal, 39, 945-950.
- Hotta, S. and Paul, D.R. (2004) Nanocomposites formed from linear low density polyethylene and organoclays. Polymer, 45, 7639-7654.
- Capek, I. (2004) Dispersion of polymer ionomers: I Advances in Colloid and Interface Science., 112, 1-29.
- Shah, R.K., Hunter, D.L. and Paul, D.R. (2005) Nanocomposites from poly (ethylene-co-methacrylic acid) ionomers: effect of surfactant structure on morphology and properties. Polymer, 46, 2646-2662.
- Usuki, A., Kato, M., Okada, A. and Kurauchi, T. (1997) Synthesis of polypropylene-clay hybrid. Journal of Applied Polymer Science, 63, 137-139.
- Kawasumi, M., Hasegawa, N., Kato, M., Usuki, A., and Okada, A. (1997) Preparation and mechanical properties of polypropylene-clay hybrids. Macromolecules, 30, 6333-6338.
- Choy, J.-H., Kwak, S.-Y., Han, Y.-S. and Kim, B.-W. (1997) New organo-montmorillonite complexes with hydrophobic and hydrophilic functions. Materials Letters, 33, 143-147.
- Thajjaroen, W. (2000) Preparation and mechanical properties of NR/clay nanocomposite. M.S. Thesis, The Petroleum and Petrochemical College, Chulalongkorn University, Bangkok, Thailand.

- Williams-Daryn, S. and Thomas, R. K. (2002) The intercalation of a vermiculite by cationic surfactants and its subsequent swelling with organic solvents. Journal of Colloid and Interface Science, 255, 303–311.
- Tang, Y., Hu, Y., Song, L., Zong, R., Gui, Z., Chen, Z. and Fan, W. (2003) Preparation and thermal stability of polypropylene/montmorillonite nanocomposites. Polymer Degradation and Stability, 82, 127-131.
- Lee, Y.J. and Lee, K.H. (2004) Characterization of organobentonite used for polymer nanocomposites. Materials Chemistry and Physics, 85, 410-415.
- Ding, C., Jia, D., He, H., Guo, B. and Hong, H. (2004) How organo-montmorillonite truly affects the structure and properties of polypropylene. Polymer Testing, 24, 94-100.
- Parija, S., Nayak, S.K., Verma, S.K. and Tripathy, S.S. (2004) Studies on physico-mechanical properties and thermal characteristics of polypropylene/layered silicate nanocomposites. Polymer Composites, 25, 646-652.
- Ramos Filho, F.G., Melo, T.A., Rabello, M.S., and Silva, S.M. (2005) Thermal stability of nanocomposites based on polypropylene and bentonite. Polymer Degradation and Stability, 89, 383-392.
- Lertwimolnun, W. and Vergnes, B. (2005) Influence of compatibilizer and processing conditions on the dispersion of nanoclay in a polypropylene matrix. Polymer, 46, 3462–3471.
- Yang, Y., Zhu, Z.-K., Yin, J., Wang, X.-Y. and Qi, Z.-E. (1999) Preparation and properties of hybrids of organo-soluble polyimide and montmorillonite with various chemical surface modification methods. Polymer, 40, 4407–4414.
- Alexandre, M. and Dubois, P. (2000) Polymer-layered silicate nanocomposites: preparation, properties and uses of a new class of materials. Materials Science and Engineering, 28, 1-63.
- Hasegawa, N., Kawasumi, M., Kato, M., Usuki, A. and Okada, A. (1998) Preparation and mechanical properties of polypropylene–clay hybrids using a maleic anhydride-modified polypropylene oligomer. Journal of Applied Polymer Science, 67, 87-92.

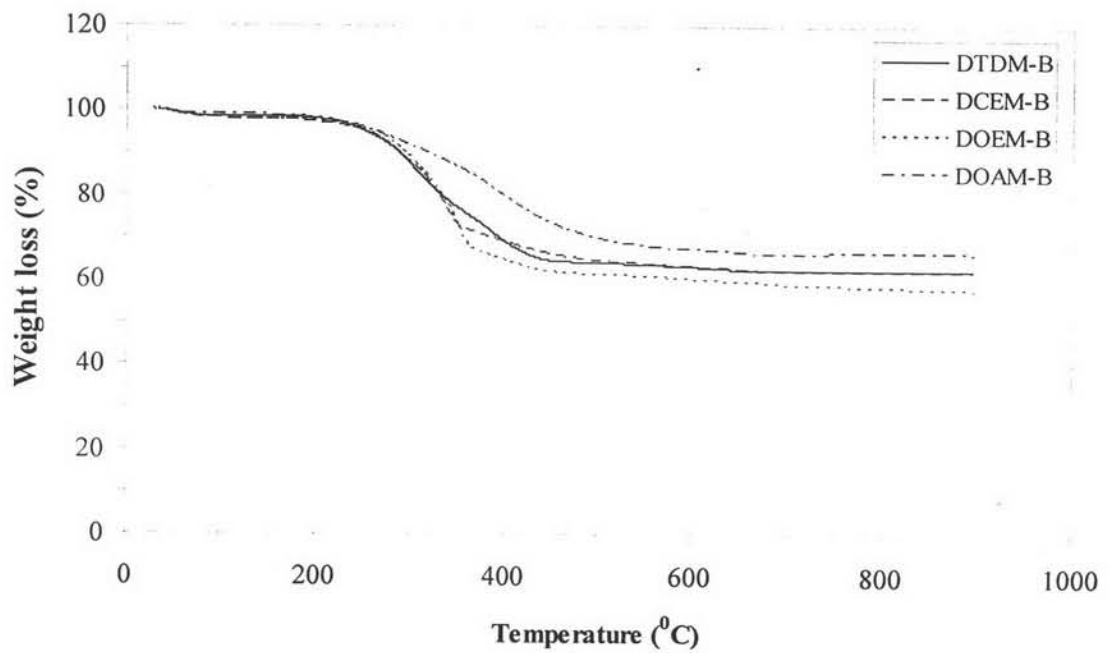
Vaia, R.A., Teukolsky, R.K., and Giannelis, E.P. (1994) Interlayer structure and molecular environment of alkylammonium layered silicates. Chem. Mater., 6, 1017-1022.

APPENDICES

Appendix A Thermal Behavior of Organomodified Clay

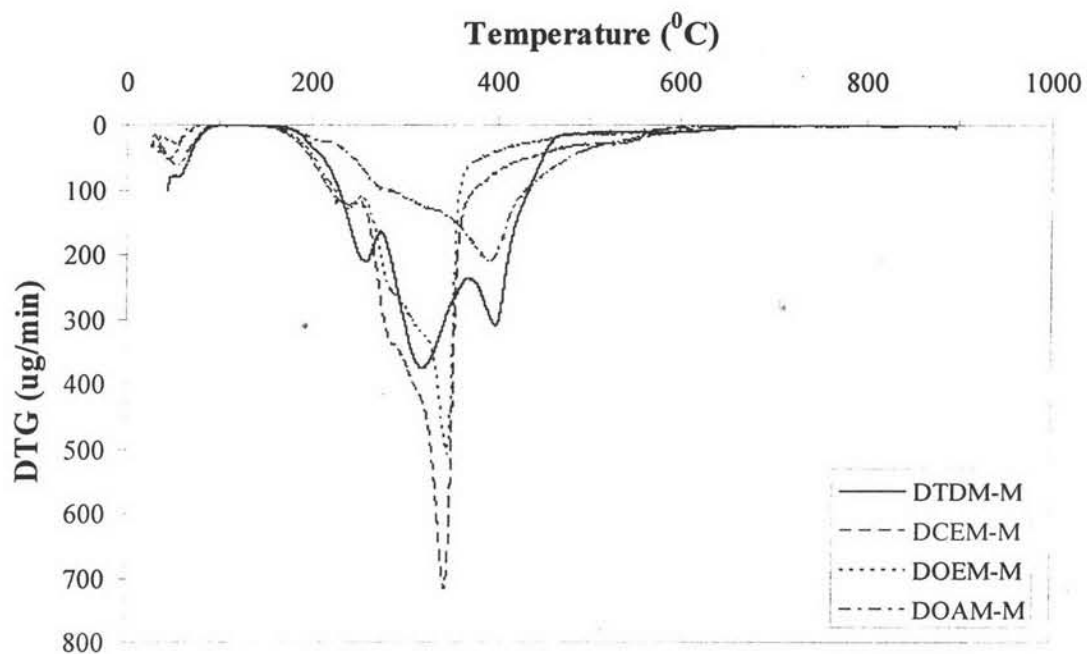


(A)

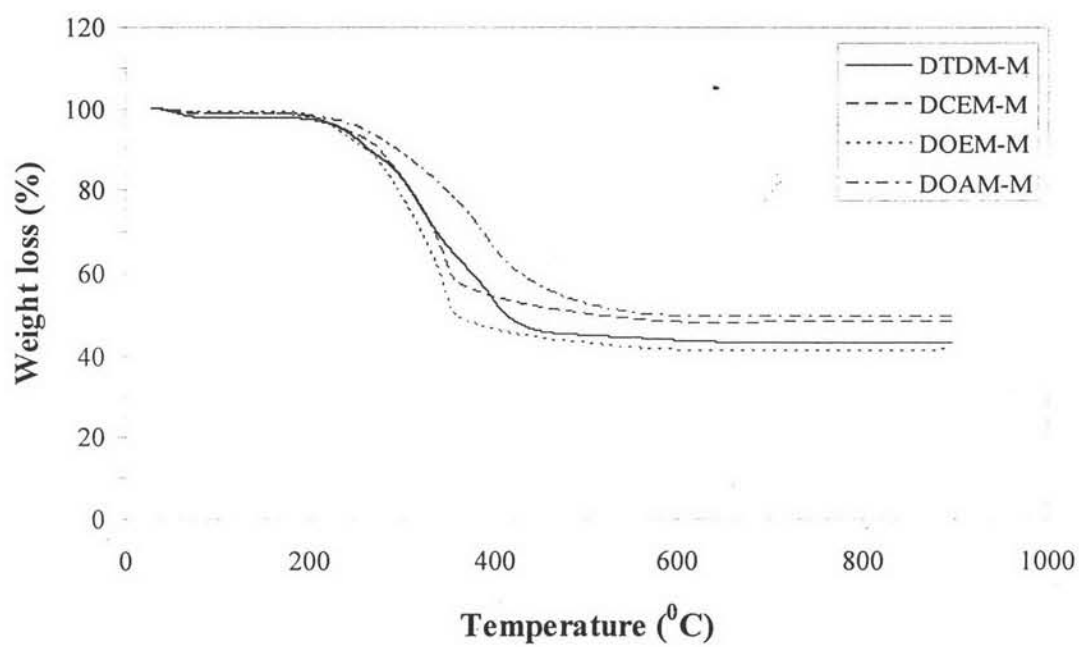


(B)

Figure A1 TG-DTA curves of organomodified bentonite (A) differential weight loss curves (DTG) (B) weight losses of the samples.



(A)



(B)

Figure A2 TG-DTA curves of organomodified montmorillonite (A) differential weight loss curves (DTG) (B) weight losses of the samples.

Appendix B Experimental Data

Table B1 Young' s modulus (MPa) of PP, PP/ Surlyn and nanocomposites

Composition	1	2	3	4	5	Avg.	S.D.
PP	1523.37	1565.25	1906.74	1421.61	1720.11	1627.41	189.51
PP/Surlyn	1569.81	1451.38	1642.56	2183.33	1510.70	1671.56	294.73
PP/Surlyn/ DTDM-B	2075.86	1566.82	1792.36	1731.74	2076.26	1848.63	223.46
PP/Surlyn/DCEM-B	1630.62	1458.95	1980.04	2030.98	1733.04	1764.73	239.65
PP/Surlyn/DOEM-B	2073.88	2069.87	1874.56	1564.83	1912.79	1899.19	207.51
PP/Surlyn/DOAM-B	1554.73	1659.16	1922.32	1560.24	1727.91	1684.87	151.13
PP/Surlyn/DTDM-M	2116.77	1912.70	1488.82	2053.11	1697.21	1853.72	259.85
PP/Surlyn/DCEM-M	1897.96	1454.90	1573.08	2144.19	2055.16	1825.06	300.26
PP/Surlyn/DOEM-M	2205.61	1900.45	1964.39	1949.79	2045.24	2013.10	119.54
PP/Surlyn/DOAM-M	1905.25	1548.57	2004.76	1729.75	1301.86	1698.04	281.65

Table B2 Tensile strength (MPa) of PP, PP/ Surlyn and nanocomposites

Composition	1	2	3	4	5	Avg.	S.D.
PP	33.64	33.21	33.23	33.07	33.24	33.22	0.10
PP/Surlyn	32.44	33.01	32.03	32.15	32.43	32.41	0.37
PP/Surlyn/ DTDM-B	33.43	33.55	33.33	33.39	33.73	33.49	0.16
PP/Surlyn/DCEM-B	31.51	31.48	31.55	31.79	31.74	31.62	0.14
PP/Surlyn/DOEM-B	32.37	32.42	32.30	32.37	31.99	32.29	0.17
PP/Surlyn/DOAM-B	31.27	31.40	31.42	31.25	31.21	31.31	0.09
PP/Surlyn/DTDM-M	32.39	32.09	32.66	32.89	32.43	32.49	0.30
PP/Surlyn/DCEM-M	31.97	32.17	31.72	32.28	32.37	32.10	0.26
PP/Surlyn/DOEM-M	32.21	32.32	32.44	32.12	32.18	32.25	0.13
PP/Surlyn/DOAM-M	32.12	32.23	31.92	32.49	32.41	32.23	0.23

Table B3 Strain at break (%) of PP, PP/ Surlyn and nanocomposites

Composition	1	2	3	4	5	Avg.	S.D.
PP	113.18	135.18	118.36	135.27	152.00	130.80	15.44
PP/Surlyn	242.55	185.55	240.09	200.09	322.64	238.18	53.33
PP/Surlyn/ DTDM-B	56.83	69.76	67.86	15.30	104.09	62.77	31.90
PP/Surlyn/DCEM-B	100.36	189.73	137.91	35.78	215.34	135.82	71.59
PP/Surlyn/DOEM-B	18.52	66.01	30.16	79.37	71.93	53.19	27.08
PP/Surlyn/DOAM-B	319.18	161.00	223.73	305.82	242.46	250.44	64.36
PP/Surlyn/DTDM-M	92.27	42.71	95.00	92.18	100.27	84.49	23.58
PP/Surlyn/DCEM-M	118.10	192.73	152.46	139.82	136.27	147.87	27.93
PP/Surlyn/DOEM-M	74.38	133.91	76.83	87.41	103.34	94.57	24.26
PP/Surlyn/DOAM-M	91.51	89.56	118.52	87.83	88.05	95.09	13.18

Table B4 Yield strength (MPa) of PP, PP/ Surlyn and nanocomposites

Composition	1	2	3	4	5	Avg.	S.D.
PP	10.50	10.68	9.06	11.32	9.67	10.25	0.89
PP/Surlyn	10.82	11.55	9.38	10.09	10.30	10.43	0.81
PP/Surlyn/ DTDM-B	9.72	10.48	10.56	10.86	9.45	10.21	0.60
PP/Surlyn/DCEM-B	10.22	11.41	9.33	9.51	9.97	10.09	0.82
PP/Surlyn/DOEM-B	9.29	9.27	10.54	12.24	10.92	10.45	1.24
PP/Surlyn/DOAM-B	10.51	9.87	8.61	10.44	10.40	9.97	0.80
PP/Surlyn/DTDM-M	10.35	8.44	12.25	9.90	10.24	10.25	1.37
PP/Surlyn/DCEM-M	8.39	13.23	10.63	9.61	9.76	10.32	1.81
PP/Surlyn/DOEM-M	9.04	8.40	8.94	9.40	9.45	9.05	0.42
PP/Surlyn/DOAM-M	9.43	12.44	9.26	10.29	16.10	11.50	2.87

Table B5 Notch Izod impact strength (kJ/m^2) of PP, PP/ Surlyn and nanocomposites

Composition	1	2	3	4	5	Avg.	S.D.
PP	3.0	2.8	2.8	2.9	3.1	2.92	0.13
PP/Surlyn	2.8	2.8	3.2	3.0	3.1	2.98	0.18
PP/Surlyn/ DTDM-B	3.0	3.2	3.2	2.7	2.9	3.00	0.20
PP/Surlyn/DCEM-B	2.3	2.4	2.3	2.0	2.4	2.28	0.16
PP/Surlyn/DOEM-B	2.0	2.3	2.2	2.5	2.5	2.30	0.20
PP/Surlyn/DOAM-B	2.5	2.6	2.7	2.6	2.4	2.56	0.11
PP/Surlyn/DTDM-M	2.7	2.7	2.6	2.8	2.9	2.74	0.11
PP/Surlyn/DCEM-M	2.5	2.7	2.2	2.2	2.5	2.42	0.20
PP/Surlyn/DOEM-M	2.6	2.6	2.3	2.3	2.5	2.46	0.15
PP/Surlyn/DOAM-M	2.7	2.8	2.5	2.5	2.6	2.62	0.13

Table B6 The degradation temperature (°C) of nanocomposites

Composition	1	2	3	mean
PP/Surlyn/ DTDM-B	460.1	459.3	458.7	459.4
PP/Surlyn/DCEM-B	461.3	461.9	460.1	461.1
PP/Surlyn/DOEM-B	460.7	461.1	459.0	460.3
PP/Surlyn/DOAM-B	459.1	460.3	458.2	459.2
PP/Surlyn/DTDM-M	456.6	454.8	455.1	455.5
PP/Surlyn/DCEM-M	457.8	459.3	458.6	458.6
PP/Surlyn/DOEM-M	458.0	457.2	458.5	457.9
PP/Surlyn/DOAM-M	458.6	456.7	457.4	457.5

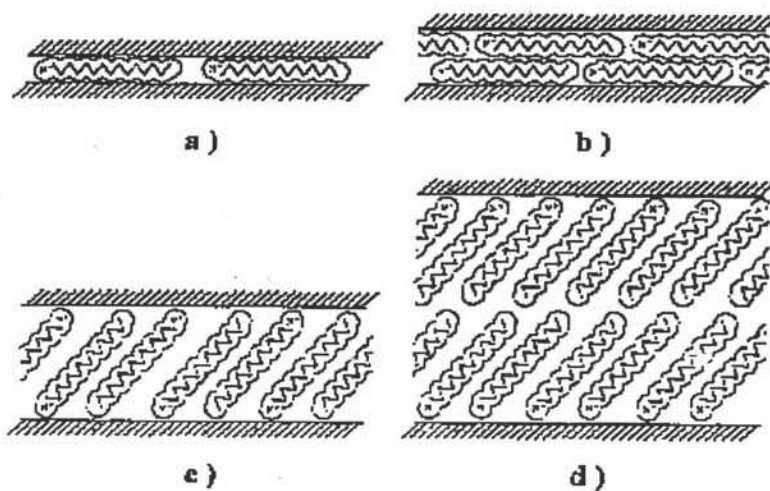
Appendix C Interlayer Structure of Alkylammonium Layered Silicates

Figure C1 Alkyl chain aggregation in mica-type silicates: (a) lateral monolayer, (b) lateral bilayer, (c) paraffin-type monolayer, (d) paraffin-type bilayer (Richard *et al.*, 1994).

Appendix D Cation Exchange Capacities (CEC) Values of Clay Minerals

Cation exchange capacities (CEC) of clay minerals as shown in Table D1 are determined by exchanging their surface cation with methylene blue. Dried clay minerals 2.0 g were dispersed in distilled water 300 ml and stirred until a uniform dispersion was obtained. The suspension was adjusted pH value in a range of 2.5-3.8 by sulfuric acid and stirred for 15 min. Methylene blue solution (0.01 normal) was added slowly by buret into the clay slurry, while it was maintained a constant stirring. A drop of the suspended solution was placed on a filter paper to determine the end point. The end point was indicated by a formation of light blue halo around the drop. After the end point was reached, the suspension was stirred for 2 min and retested for the end point.

Methylene blue index (MBI)

$$MBI = \frac{E * V * 100}{W}$$

where:

- MBI = methylene blue index for the clay (meq/100g of clay),
 E = concentration of methylene blue (meq/ml),
 V = amount of methylene blue required for titration (ml), and
 W = weight of dried nanoclay (g)

Table D1 Cation exchange capacity (CEC) values of clay minerals

Clay minerals	Na-MMT (Kunipia F [®])	Na-BN (Mac-Gel [®] wn-02)
CEC (meq/100g)	115.50	52.55

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