การทำไบโอคีเซลให้บริสุทธิ์โดยใช้สารคูคซับ

นางสาวจิรพรรณ เจริญสินวรกุล

วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต สาขาวิชาปี โตรเคมีและวิทยาศาสตร์พอลิเมอร์ คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย ปีการศึกษา 2549 ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย

PURIFICATION OF BIODIESEL USING ADSORBENTS

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A Thesis Submitted in Partial Fulfillment of the Requirements

for the Degree of Master of Science Program in Petrochemistry and Polymer Science

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วัตถุประสงค์ของงานวิจัยนี้ เพื่อลดปริมาณกรดไขมันอิสระในไบโอดีเซลที่สังเคราะห์จากน้ำมัน ปาล์มดิบที่ผ่านกระบวนการโดยมีเบสเป็นตัวเร่งปฏิกิริยา ใบโอดีเซลได้จากกระบวนการทรานส์เอสเทอริ ฟิเค-ชันของน้ำมันปาล์มดิบกับเมทานอลที่ใช้โซเคียมไฮครอกไซค์เป็นตัวเร่งปฏิกิริยา สภาวะที่เหมาะสม สำหรับผลิตไบโอดีเซลใช้อัตราส่วนระหว่างเมทานอลต่อน้ำมันเป็น 16:1 โซเดียมไฮครอกไซค์ เปอร์เซ็นต์โดยน้ำหนักและที่อุณหภูมิ 50 องศาเซลเซียสเป็นเวลา 1.50 ชั่วโมง การใช้ที่สภาวะนี้จะได้ ไบโอดีเซล 88-96 เปอร์เซ็นต์ ใบโอดีเซลที่ได้มีกรคไขมันอิสระ 0.184 เปอร์เซ็นต์ จากนั้นทำให้บริสุทธิ์ โดยสารดูดซับ ได้แก่ แมกนีซอล ถ่านกัมมันต์ อะลูมินาชนิดเบส โมเลคิวร์ลาซีฟ(4Å) คินเบนโทไนต์ และดินเบนโทในต์ที่ผ่านการทรีตด้วยโซเดียมไฮครอกไซด์ ผลการทคลองแสคงให้เห็นว่าถ่านกัมมันต์ใน รูปผงและแมกนี้ซอลสามารถลดปริมาณกรดใขมันอิสระเหลือเพียง 0.014 และ 0.016 เปอร์เซ็นต์ตามลำดับ คังนั้น จึงใช้ถ่านกับมันต์ในรูปผงเป็นสารคูคซับที่มีข้อคีในการศึกษานี้

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The purpose of this research is to reduce the free fatty acids (FFAs) in biodiesel synthesized from crude palm oil via a base-catalyzed process. The biodiesel was obtained by the transesterification of crude palm oil with methanol using sodium hydroxide as a catalyst. The optimum conditions for biodiesel production were obtained using a 16:1 molar ratio of methanol:oil, 1.4%wt of sodium hydroxide and reaction temperature of 50°C for 1.50 h. Using these conditions, the 88-96% yield of biodiesel was obtained. The biodiesel with 0.184% of FFAs was further purified by adsorbents including magnesol, activated charcoal, basic alumina, molecular sieve (4Å), bentonite clay and NaOH-treated bentonite clay. The results showed that the activated charcoal and magnesol could reduced FFAs to 0.014% and 0.016%, respectively. Therefore, the activated charcoal was used as an advantage adsorbent in this study.

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LIST OF ABBREVIATIONS

AOAC = Association of Official Analytical Chemists/Official

Methods of Analysis of AOAC

ASTM = American Society for Testing and Materials

°C = Degree Celsius

¹³C-NMR = Carbon-13 Nuclear Magnetic Resonance

CN = Cetane number

CO = Carbon monoxide

 CO_2 = Carbon dioxide

CPO = Crude palm oil

 δ = Delta

DF = Diesel fuel

EN = European Standards

FAME = Fatty acid methyl ester

FFA = Free fatty acid

g = Gram

¹H-NMR = Proton Nuclear Magnetic Resonance

h = Hour

K = Degree Kelvin

KTOE = Kilo Ton of Oil Equivalent

mg = Milligram

min = Minute

ml = Milliliter

 NO_x = Nitrogen oxides

PAHs = Polycyclic aromatic hydrocarbon compounds

rpm = Round per minute

sq. m. = Square meter, m^2

TLC = Thin Layer Chromatography

TLC-FID = Thin Layer Chromatography-Flame Ionization Detector

% wt = Percent by weight

% yield = Percent yield

DEFINITIONS

Cetane number = A measure of the ignition quality of a diesel fuel, as determined in a standard single cylinder test engine, which measures ignition

in a standard single cylinder test engine, which measures ignition delay, compared to primary reference fuels. The higher the

Cetane Number, the easier a high-speed, direct-injection engine

will start, and the less "white smoking" and "diesel knock" after

start-up.

Cloud point = The start of crystal formation in the oil. As the temperature is lowered, the crystals continue to grow until the oil is no longer

fluid but a solid.

Flash point = The lowest temperature at which evaporation of a substance

produces sufficient vapor to form an ignitable mixture with air.

Magnesol = It is a registered trademark of The Dallas Group of America, Inc.

for a synthetic, amorphous, hydrous form of magnesium silicate.

With a porous internal structure and enormous activated surface,

Magnesol® products are ideal for use as adsorbents, anti-static

flow agents, anti-caking agents, catalyst supports, carriers, and

reinforcing fillers.

PAHs = Polynuclear aromatic hydrocarbons (PAHs) are hydrocarbon

compounds with multiple benzene rings. PAHs are typical

components of asphalts, fuels, oils and greases. Because of, they

are insoluble in water but dissolve readily in fats and oils.

Pour point = The lowest temperture at which it will pour or flow under

prescibed conditions. It is a rough indication of the lowest

temperature at which oil is readily pumpable. Also, the pour point

can be defined as the minimum temperature of a liquid,

particularly a lubricant, after which, on decreasing the

temperature, the liquid creases to flow.

Transesterification = It is used to describe the direct conversion of triacylglycerols

lipids by alcohols to alkyl esters without first isolating the free

fatty acids (FFAs).