ADSORPTION AND DISPERSION OF TALCUM POWDER IN DIFFERENT SURFACTANT SOLUTIONS



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ABSTRACT

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Sliders used to write and read data from the disk, require extremely low levels of impurities in the manufacturing process. The most common impurity found on the surface of sliders is identified as talcum powder. The dispersion stability is expected that related to prevent redeposition on the surface of slider in cleaning agent. The objective of this work was to investigate the surfactant adsorption and dispersion of talcum powders in different surfactant solutions. Surfactant solutions can enhance the dispersion stability of talcum powder as compared with in absence surfactant. The dispersion stability of talcum powder increased with increasing surfactant concentration. Solution pH at low surfactant concentration affected on dispersion of talcum powder but at high surfactant concentration slightly affected on dispersion. SDS surfactant adsorbed with the highest adsorption density, leading to increase negative electrical zeta potentials that exhibited the highest dispersion stability. Adsorption of CTAB surfactant changed the negative charge surfaces to positive charge surfaces with increase positive electrical zeta potentials that can stabilize talcum powder to disperse. AE7 surfactant adsorbed on talcum powder with the lowest adsorption but the steric effect of surfactant adsorbed on surface can enhance the dispersion stability of talcum powder.

บทคัดย่อ

อารีย์ ปิ่นปิติ : การดูคซับและการกระจายตัวของอนุภาคทัลคัมในสารละลายสารลดแรง ตึงผิวที่แตกต่างกัน (Adsorption and Dispersion of Talcum Powder in Different Surfactant Solutions) อ. ที่ปรึกษา : ศ. คร. สูเมช ชวเคช และ คร. แพนภัทร เตชางาม 71 หน้า

หัวอ่านคืออุปกรณ์ที่ใช้ในการอ่านหรือเขียนข้อมูลจากฮาร์คคิสก์ซึ่งในกระบวนการผลิต หัวอ่านนั้นต้องการปริมาณสิ่งเจือปนดำ โดยส่วนใหญ่อนุภาคที่ถูกตรวจพบบนพื้นผิวของหัวอ่าน คืออนุภาคทัลคัม การกระจายตัวของอนุภาคคาคว่าเกี่ยวข้องกับการป้องกันการติคลงบนพื้นผิวใน สารทำความสะอาค คังนั้นจุดประสงค์ของงานวิจัยนี้คือ การศึกษาการดูดซับและการกระจายตัวของอนุภาคทัลคัมในสารละลายด่วยลงผิวที่แตกต่างกัน จากการศึกษาพบว่าสารลดแรงตึงผิว ช่วยเพิ่มความสามารถในการกระจายตัวของอนุภาคทัลคัลเมื่อเปรียบเทียบกับสารละลายที่ไม่มีสารลดแรงตึงผิว การกระจายตัวของอนุภาคเพิ่มขึ้นเมื่อความความเข้มข้นของสารละลายตัว แต่ความเป็นกรค เบสมีผลต่อการกระจายตัว แต่ความเป็นกรค เบสมีผลต่อการกระจายตัว ที่ความเข้มข้นสูง โซเดียมโคเคคซิลซัลเฟต (SDS) ดูดซับบนพื้นผิวมี ค่าสูงสุดทำให้เพิ่มความต่างศักย์ที่เป็นลบบนพื้นผิว ซึ่งทำให้เพิ่มการกระจายตัวของอนุภาคได้สูง ที่สุด ส่วนการดูดซับของเฮกซะเคคซิลไตรเมธิลแอมโมเนียมโบรไมด์ (CTAB)นั้น ทำให้เปลี่ยน พื้นผิวของอนุภาคที่เป็นลบให้กลายเป็นบวกและเพิ่มความต่างศักย์ที่เป็นบวกบนพื้นผิว ส่งผลทำให้สารลดแรงดึงผลช่วยในการกระจายตัวของอนุภาคได้ สำหรับแอลกอฮอล์อีทอกซีเลท (AE7) นั้นคูดซับบนพิ้นผิวของอนุภาคมีค่าต่ำสุด แต่ความเกะกะของสารลดแรงตึงผิวที่ดูดซับบนพื้นผิว ส่งผลทำให้เกิดการกระจายตัวของอนุภาคทีลคัม

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ABBREVIATIONS

AE7 Alcohol ethoxylate

CTAB Cetyltrimethy ammonium bromide

SDS Sodium dodecyl sulfate

NaOH Sodium hydroxide

HBr Hydrobromic acid

CMC Critical micelle concentration