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APPENDICES

Appendix A Calculations and Experimental Data

Gas Permeability

Gas permeation experiments were investigated by Illinois model 8000. The sample films were cut into circular shape with 120 mm in diameter according to ASTM D3985-02. The thickness of the films was 2-3 mm. The films were placed in a desiccator over NaCl and kept for not less than 3 days prior to test.

Gas transmission rate (GTR)

$$GTR = \frac{\Delta M}{A \cdot \Delta t \cdot \Delta p}$$

Permeability (Q)

$$Q = \frac{\Delta M \cdot L}{A \cdot \Delta t \cdot \Delta p}$$

$$\therefore Q = GTR \times L$$

where:

$\frac{\Delta M}{\Delta t}$ = amount of gas passing through film in unit time (cm³/s)

A = area (m²)

Δp = the differential partial pressure of the permeat gas across the film (bar)

L = film thickness (mm)

The gas permeability rate, G , in units of $\text{cm}^3/(\text{m}^2 \cdot \text{day} \cdot \text{bar})$ is calculated from,

$$G = \frac{7.76 \times 10^7 \times V}{78.5K \times 29N}$$

where:

V = volume of the evacuation chamber

K = absolute temperature (degrees Kelvin)

N = the slope of the graph which is determined by dividing the time (s) by the scale divisions (mm)

if the evacuation chamber volume, V , is 0.4370 cm^3 then this expression simplifies to,

$$G = \frac{1.49 \times 10^7}{KN} \text{ cm}^3 / (\text{m}^2 \cdot \text{day} \cdot \text{bar})$$

$$G = \frac{1.49 \times 10^7 \times 0.9896}{KN} \text{ cm}^3 / (\text{m}^2 \cdot \text{day} \cdot \text{atm})$$

Table A1 Oxygen Gas Transmission Rate of PLA sheet

Oxygen Gas Transmission Rate (cc/m ² .d)		
Test	1 st Sample	2 nd Sample
1	138	125
2	139	125
3	139	125
4	139	125
5	140	124
6	139	124
7	139	124
8	139	124
9	139	124
10	139	125
Average	139	125
SD	0.5	0.5
Thickness of Sample (μm)	204	210

Oxygen Gas Transmission Rate of PLA sheet = 132 (cc/m².d)

Table A2 Oxygen Gas Transmission Rate of PLA/5%wt PCL sheet

Oxygen Gas Transmission Rate (cc/m ² .d)		
Test	1 st Sample	2 nd Sample
1	119	105
2	118	103
3	118	104
4	119	105
5	120	105
6	120	106
7	120	106
8	119	105
9	119	105
10	119	105
Average	119	105
SD	0.7	0.9
Thickness of Sample (μm)	201	212

Oxygen Gas Transmission Rate of PLA/5%wt PCL sheet = 112 (cc/m².d)

Table A3 Oxygen Gas Transmission Rate of PLA nanocomposite sheet

Oxygen Gas Transmission Rate (cc/m ² .d)		
Test	1 st Sample	2 nd Sample
1	106	104
2	106	104
3	106	104
4	107	104
5	107	105
6	107	105
7	107	105
8	107	105
9	106	105
10	106	104
Average	119	105
SD	0.7	0.9
Thickness of Sample (μm)	209	214

Oxygen Gas Transmission Rate of PLA nanocomposite sheet = 106 (cc/m².d)

The molar ratio of dodecylamine and TEOS

Example : D/T=20/200 in organoclay 1 g; MW of dodecylamine = 185.35

MW of TEOS = 208.33

Density of TEOS = 0.933 g/mol

CEC of bentonite = 55 mmol/100g of clay

Composition	mmol	g	ml
Organoclay	0.55	1	-
Dodecylamine	20	2.04	-
TEOS	200	-	24.56

Appendix B Types of Adsorption Isotherm and Hysteresis Loop

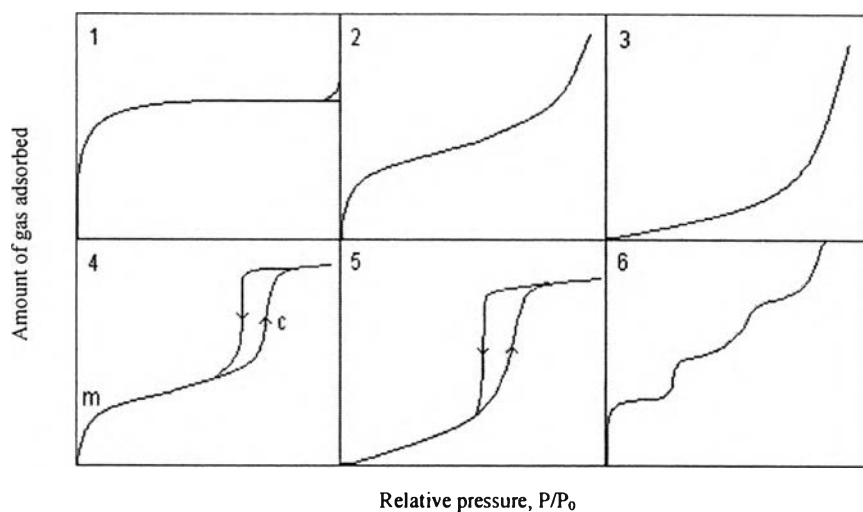


Figure B1 Types of adsorption isotherm according to BDDT classification.

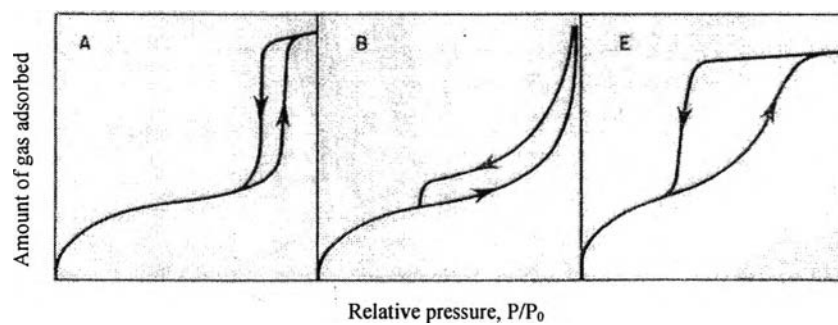


Figure B2 Types of hysteresis loop according to De Boer classification.

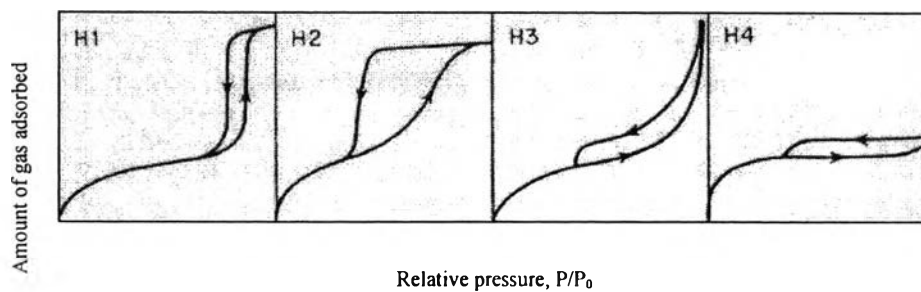
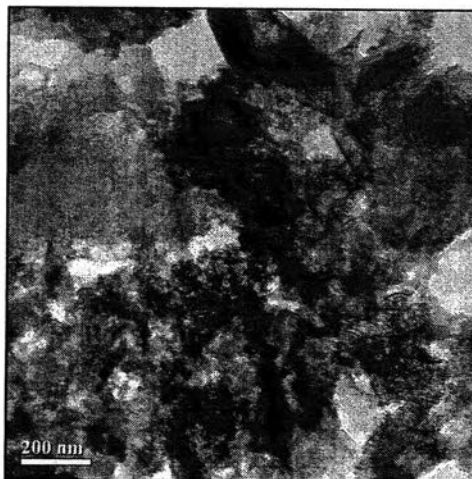
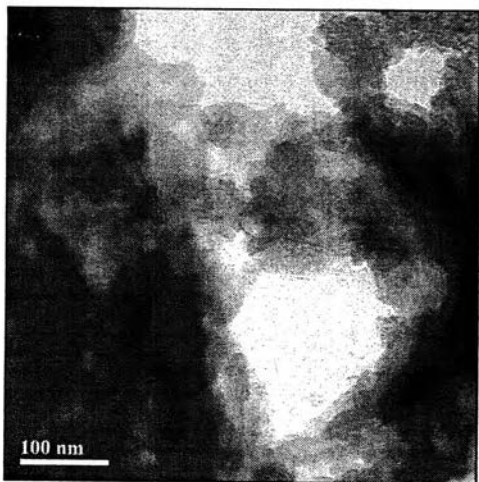
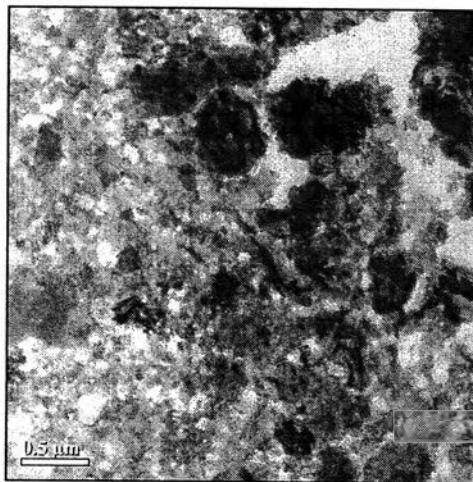
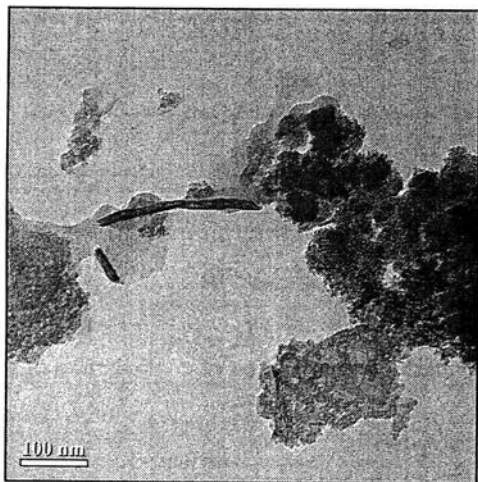


Figure B3 Types of hysteresis loop according to IUPAC classification.

Appendix C Supplementary Results**Figure C1** TEM images of magnetic PCHs

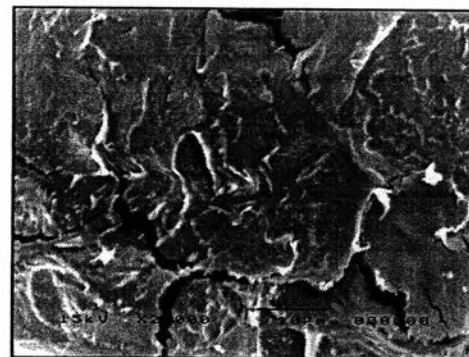
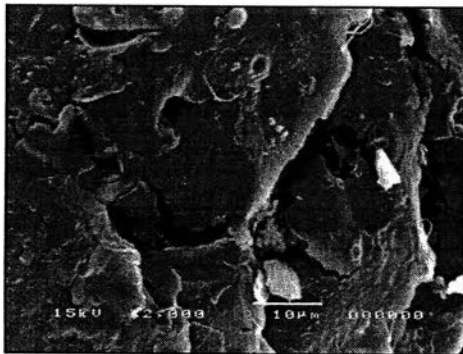
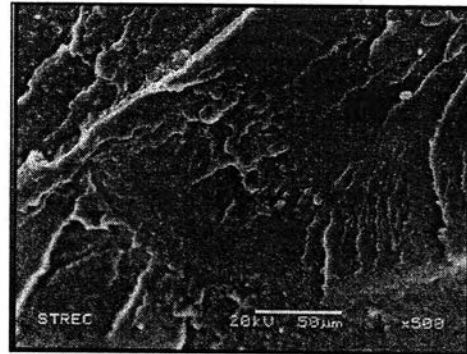
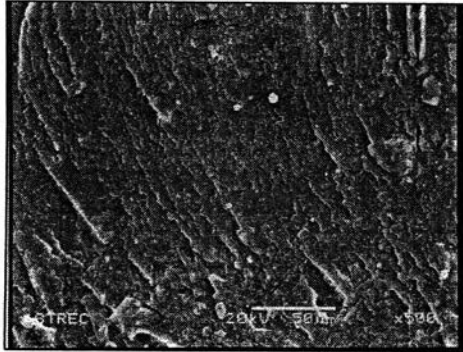


Figure C2 SEM images of PLA nanocomposites

CIRRICULUM VITAE

Name: Ms. Arunsri Mattayan

Date of Birth: November 15, 1984

Nationality: Thai

University Education:

2003–2007 Bachelor Degree of Science in Polymer Science, Faculty of Science, Prince of Songkla University, Songkla, Thailand.

Proceedings:

1. Aksonnum, T., Mattayan, A., Nithithanakul, M., Magaraphan, R., and Manuspiya, H. (2008, August 19-20) Nanoclay/Polypropylene Composite for Ethylene and Carbon Dioxide Scavenging Films. Proceedings of the Thai-Japan Joint Symposium on Advances in Materials Science and Environmental Technology 2008, Bangkok, Thailand.
2. Mattayan, A., Magaraphan, R., and Manuspiya, H. (2008, November 6-8) Synthesis of Surface Modified Porous Clay Heterostructure(PCH) for Magnetic Sensor in Film Packaging. Proceedings of the NanoThailand Symposium 2008, Bangkok, Thailand.
3. Mattayan, A., Magaraphan, R., and Manuspiya, H. (2009, February 25-27) Polylactide/Magnetic Modified Mesoporous Clay Nanocomposites for Food Packaging. Proceedings of the GPEC Conference 2009, Orlando, Florida, USA.
4. Mattayan, A., Magaraphan, R., and Manuspiya, H. (2009, April 22) Induced Magnetic Properties to Surface-Modified Mesoporous Clay Heterostructures for Sensor Application in Food Packaging. Proceedings of the 15th PPC Symposium on Petroleum, Petrochems, and Polymers, Bangkok, Thailand.

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