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**APPENDIX A**  
**PROPERTY DATA BANK**

The following tables is listing of the component currently available in database system. The data of component receive from reference data(Ried R.C and et al., 1987) and Sinnott R.K., 1983). Tables consist of five groups of compounds as follows:

**1. HYDROCARBON**

**1.1. PARAFINS**

Number	Formula	Compound name
1	CH4	Methane
2	C2H6	Ethane
3	C3H8	Propane
4	C4H10	Isobutane
5	C4H10	N-butane
6	C5H12	N-pentane
7	C6H14	N-hexane
8	C7H16	N-heptane
9	C8H18	N-octane
10	C9H20	N-nonane
11	C9H22	N-decane

## 1.2. AKYL PARAFINS

Number	Formula	Compound name
1	C5H12	2,2-dimethylpropane
2	C6H14	2-methylpentane
3	C6H14	3-methylpentane
4	C6H14	2,2-dimethylbutane
5	C6H14	2,3-dimethylbutane
6	C7H16	2-methylhexane
7	C7H16	3-methylhexane
8	C7H16	3-ethylpentane
9	C7H16	2,2-dimethylpentane
10	C7H16	2,4-dimethylpentane
11	C7H16	3,3-dimethylpentane
12	C7H16	2,2,3-trimethylbutane
13	C8H18	2-methylheptane
14	C8H18	3-methylheptane
15	C8H18	4-methylheptane
16	C8H18	2,2-dimethylhexane
17	C8H18	2-methyl-3-ethylpentane
18	C8H18	3-methyl-3-ethylpentane
19	C8H18	2,2,3-trimethylpentane
20	C8H18	2,2,4-trimethylpentane

### 1.3 CYCLOPARAFINS

Number	Formula	Compound name
1	C2H6	Cyclopropane
2	C4H8	Cyclobutane
3	C5H10	Cyclopentane
4	C6H12	Cyclohexane
5	C6H12	Methylcyclopentane
6	C7H14	Methylcyclohexane
7	C7H14	Ethylcyclopentane
8	C7H14	1,1-dimethylcyclopentane
9	C7H14	Cycloheptane
10	C8H16	Ethylcyclohexane

### 1.4 AROMATICS

Number	Formula	Compound name
1	C6H6	Benzene
2	C7H8	Toluene
3	C8H10	Ethylbenzene
4	C8H10	o-Xylene
5	C8H10	m-Xylene
6	C8H10	p-Xylene
7	C8H8	Styrene
8	C9H12	N-propylbenzene
9	C9H12	1-methyl-2-ethylbenzene
10	C9H12	1-methyl-3-ethylbenzene
11	C9H12	1-methyl-4-ethylbenzene
12	C10H8	Naphthalene

## 1.5 OLEFINS

Number	Formula	Compound name
1	C <sub>2</sub> H <sub>4</sub>	Ethylene
2	C <sub>3</sub> H <sub>6</sub>	Propene
3	C <sub>3</sub> H <sub>4</sub>	Propadiene
4	C <sub>4</sub> H <sub>6</sub>	1,2-Butadiene
5	C <sub>4</sub> H <sub>6</sub>	1,3-Butadiene
6	C <sub>4</sub> H <sub>8</sub>	1-Butene
7	C <sub>4</sub> H <sub>8</sub>	Isobutylene
8	C <sub>5</sub> H <sub>8</sub>	1,2-Pentadiene
9	C <sub>5</sub> H <sub>8</sub>	3-methyl-1,2-butadiene
10	C <sub>5</sub> H <sub>10</sub>	1-Pentene
11	C <sub>6</sub> H <sub>12</sub>	1-Hexene
12	C <sub>6</sub> H <sub>12</sub>	cis-2-Hexene
13	C <sub>6</sub> H <sub>12</sub>	trans-2-Hexene
14	C <sub>7</sub> H <sub>14</sub>	1-Heptene
15	C <sub>8</sub> H <sub>16</sub>	1-Octene
16	C <sub>10</sub> H <sub>20</sub>	1-Decene

## 1.6 ACETYLENE

Number	Formula	Compound name
1	C <sub>2</sub> H <sub>2</sub>	Acetylene
2	C <sub>2</sub> H <sub>4</sub>	Ethylene
3	C <sub>4</sub> H <sub>4</sub>	Vinylacetylene
4	C <sub>4</sub> H <sub>6</sub>	1-Butyne
5	C <sub>4</sub> H <sub>6</sub>	2-Butyne
6	C <sub>5</sub> H <sub>8</sub>	1-Pentyne

## 2. INORGANICS COMPOUND

Number	Formula	Compound name
1	N <sub>2</sub>	Nitrogen
2	CO <sub>2</sub>	Carbondioxide
3	H <sub>2</sub> S	Hydrogen sulfide
4	H <sub>2</sub> O	Water
5	H <sub>2</sub>	Hydrogen
6	CO	Carbonmonoxide
7	Ar	Argon
8	NH <sub>3</sub>	Ammonia
9	O <sub>2</sub>	Oxygen
10	COS	Carbonyl sulfide
11	He	Helium
12	CS <sub>2</sub>	Carbon disulfide
13	Cl <sub>2</sub>	Chlorine
14	F <sub>2</sub>	Fluorine
15	NO	Nitric oxide
16	NO <sub>2</sub>	Nitrogen dioxide
17	N <sub>2</sub> O	Nitrous oxide
18	O <sub>3</sub> S	Sulfur trioxide
19	Br <sub>2</sub>	Bromine
20	I <sub>2</sub>	Iodine
21	SO <sub>2</sub>	Sulfur dioxide
22	BCl <sub>3</sub>	Boron trichloride
23	D <sub>2</sub>	Deuterium
24	Ne	Neon
25	O <sub>3</sub>	Ozone
26	Xe	Xenon
27	HCN	Hydrogen cyanide

### 3. ALCOHOLS

Number	Formula	Compound name
1	CH <sub>4</sub> O	Methanol
2	C <sub>2</sub> H <sub>6</sub> O	Ethanol
3	C <sub>3</sub> H <sub>8</sub> O	1-Propanol
4	C <sub>3</sub> H <sub>8</sub> O	Isopropyl alcohol
5	C <sub>4</sub> H <sub>10</sub> O	N-Butanol
6	C <sub>4</sub> H <sub>10</sub> O	2-Butanol
7	C <sub>4</sub> H <sub>10</sub> O	i-Butanol
8	C <sub>5</sub> H <sub>12</sub> O	1-Pentanol
9	C <sub>6</sub> H <sub>14</sub> O	1-Hexanol
10	C <sub>7</sub> H <sub>16</sub> O	1-Heptanol
11	C <sub>8</sub> H <sub>18</sub> O	1-Octanol

### 4. HALOGENATED HYDROCARBON

Number	Formula	Compound name
1	CCl <sub>4</sub>	Carbontetrachloride
2	CHCl <sub>3</sub>	Chloroform
3	C <sub>2</sub> H <sub>5</sub> Cl	Ethyl chloride
4	C <sub>3</sub> H <sub>7</sub> Cl	Propyl chloride
5	C <sub>4</sub> H <sub>9</sub> Cl	1-Chlorobutane
6	C <sub>4</sub> H <sub>9</sub> Cl	2-Chlorobutane
7	C <sub>6</sub> H <sub>5</sub> Cl	Chlorobenzene
8	CF <sub>4</sub>	Carbon tetrafluoride
9	C <sub>2</sub> H <sub>3</sub> Cl	Vinyl chloride
10	CH <sub>3</sub> F	Methyl fluoride
11	CH <sub>2</sub> F <sub>2</sub>	Difluoromethane
12	C <sub>2</sub> F <sub>4</sub>	Perfluoroethane
13	C <sub>6</sub> H <sub>5</sub> F	Fluorobenzene

## 5. MISCELLANEOUS COMPOUNDS

Number	Formula	Compound name
1	CH <sub>5</sub> N	Methyl amine
2	C <sub>2</sub> H <sub>7</sub> N	Ethyl amine
3	C <sub>3</sub> H <sub>9</sub> N	N-Propyl amine
4	C <sub>3</sub> H <sub>9</sub> N	Isopropyl amine
5	C <sub>4</sub> H <sub>9</sub> N	Pyrrolidine
6	C <sub>5</sub> H <sub>5</sub> N	Pyridine
7	C <sub>6</sub> H <sub>7</sub> N	Aniline
8	C <sub>3</sub> H <sub>6</sub> O	Acetone
9	C <sub>2</sub> H <sub>2</sub> O	Ketone
10	CH <sub>2</sub> O	Formaldehyde
11	C <sub>2</sub> H <sub>4</sub> O	Acetaldehyde
12	C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	Methyl formate
13	C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	Ethyl formate
14	HCl	Hydrogen chloride
15	HF	Hydrogen fluoride
16	HBr	Hydrogen bromide
17	HI	Hydrogen iodide
18	C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	Acetic acid
19	C <sub>7</sub> H <sub>6</sub> O <sub>2</sub>	Benzoic acid
20	C <sub>2</sub> H <sub>3</sub> N	Acetonitrile
21	C <sub>3</sub> H <sub>3</sub> N	Acrylonitrile
22	C <sub>3</sub> H <sub>5</sub> N	Propionitrile
23	CH <sub>3</sub> NO <sub>2</sub>	Nitromethane
24	C <sub>6</sub> H <sub>6</sub> O	Phenol

## VITA

Thanit Swasdisevi was born on July 28, 1964 in Nakornpanom, Thailand. He received his Bachelor Degree of Chemical Engineering from Prince of Songkla university in 1991. He has studied Chemical engineering at Chulalongkorn university since 1993.

