อิทธิพลของตัวเร่งปฏิกิริยาต่อเวลาการขึ้นรูปโพลิเอสเตอร์เอสเอ็มชื่

นาย สุชล มั่นรักเรียน



วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต สาขาวิชาปิโตรเคมี

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# Effect of Catalysts on Cure Time of Polyester SMC Processing

Mr. Suchol Manrukrian

A Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of Master of Science

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| Thesis Title       | Effect of Catalysts on Cure Time of Polyester       |  |  |
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|                    | SMC Processing                                      |  |  |
| Ву                 | Mr. Suchol Manrukrian                               |  |  |
| Department         | Petrochemistry                                      |  |  |
| Thesis Advisors    | Associate Professor Suda Kiatkamjornwong, Ph.D.;    |  |  |
|                    | Wing Commander Somsak Naviroj, Ph.D.                |  |  |
| Acccepted          | by the Graduate School, Chulalongkorn University in |  |  |
| Partial Fulfillmen | t of the Requirements for the Master's Degree.      |  |  |
| /                  | Dean of Graduate School                             |  |  |
| (Professor         | Thavorn Vajrabhaya, Ph.D.)                          |  |  |
| Thesis Committee   |   |  |  |
|                    | Father Audl. Chairman                               |  |  |
| (Associate         | Professor Pattarapan Prasassarakich, Ph.D.)         |  |  |
|                    | ade kiathan f. Thesis Adivisor                      |  |  |
| (Associate         | Professor Suda Kiatkamjornwong, Ph.D.)              |  |  |
|                    | Smok Thesis Co-Adivisor                             |  |  |
| (Wing Comm         | nander Somsak Naviroj, Ph.D.)                       |  |  |
|                    | Swasawan: Member                                    |  |  |
| (Chachawar         | Surasawadi, Ph.D.)                                  |  |  |
|                    | Payson South L. Member                              |  |  |
| (Assistant         | Professor Paipan Santisuk, M.S.)                    |  |  |

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การวิจัยนี้ได้ศึกษาตัวเร่งปฏิกิริยา 3 ชนิค คือ เทอเชียรี-บิลทิลเปอร์เบนโซเอต (ที่บีพีบี), 1.1-ไค-(เทอเชียรี-บิลทีลเบอร์ออกซี) ไซโคลเฮกเซน (คีทีบีซี), เบนโซอิล เบอร์ออกไซค์ (บีพีโอ) และตัวเร่งปฏิกิริยาผสมระหว่างทีบีพีบีกับคีทีบีซี เพื่อศึกษาอิทธิพลของตัวแปรต่าง ๆ ได้แก่ ชนิคและปริมาณ ของตัวเร่งปฏิกิริยาต่ออายุการเก็บรักษาของแผ่นเอสเอ็มชี และลักษณะการบ่มในแม่พิมพ์ที่เวลา 1 และ 2 นาที่ การทคลองใช้ที่บีที่บีเป็นตัวเร่งปฏิกิริยา สามารถเก็บแผ่นเอสเอ็มซี่ได้นานถึง 45 วัน โดยมีระยะเวลา เก็บรักษาที่เหมาะต่อการนำไปขึ้นรูปที่ประมาณ 17-30 วัน ความเข้มข้นที่เหมาะสม 2 ส่วนในร้อย ที่อุณหภูมิ แม่พิมพ์ 150 ช. การใช้เวลาบุ่ม 1 นาที่ วัสดุจะมีความแช็งแรงเชิงกลต่ำกว่าเมื่อใช้เวลาบุ่ม 2 นาที่ แต่ สามารถทำให้วัสคุมีความแข็งแรงสูงขึ้นได้ โดยการผสมพารา–เบนโชควินโนน (พีบีคิว) ความเข้มข้น 0.025 ส่วนในร้อย ในทานองเคียวกับการใช้คีทีบีซีเป็นตัวเร่งปฏิกิริยา แผ่นเอสเอ็มซีจะมีอายุการเก็บรักษา ได้นานประมาณ 30 วัน การใช้ตัวเร่งปฏิกิริยาผสมระหว่างที่บีพีบีกับดีที่บีซี จะเร่งให้วัสดุเกิดการบ่มได้เร็ว ขึ้นเล็กน้อย ส่วนการใช้บีพีโอเป็นตัวเร่งปฏิกิริยาให้ข้อคีคือ สามารถใช้อุณหภูมิขึ้นรูปต่ำเพี่ยง 120 ํช. แต่มี ซ้อเสียที่ของผสมเอสเอ็มซีมีความหนืดสูงขึ้นเร็วมาก ความเข้มข้นที่เหมาะสมของบี้พีโอประมาณ 1 ส่วนใน ร้อย และสามารถเก็บแผ่นเอสเอ็มซีได้นานเพียง 10 วันเท่านั้น วัสดุที่บุ่มเพียง 1 นุาที่มีความแข็งแรงต่ำ กว่าการใช้เวลาบ่ม 2 นาทีเพียงเล็กน้อย แต่การใช้บีพีโอเป็นตัวเร่งปฏิกิริยา วัสดุที่ได้มีความแข็งแรง เชิงกลต่ำกว่าการใช้ที่บีพีบีและดีที่บีซีเป็นตัวเร่งปฏิกิริยาประมาณร้อยละ 10 ตังนั้น ตัวเร่งปฏิกิริยาที่ เหมาะสมสำหรับเอสเอ็มซีของโพลิเอสเตอร์ในงานวิจัยนี้คือ ที่บีพีบีและคีที่บีซี โคยมีการผสมพีบีคิวความ เข้มขัน 0.025 ส่วนในร้อย ในองค์ประกอบของเอสเอ็มซี่ด้วย ส่วนบีพีโอมีความไม่เหมาะสมที่จะนำมาใช้ เป็นตัวเร่งปฏิกิริยาเพียงลำพังในโพลิเอสเอร์เอสเอ็มชี

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|-------------------|-------------------------------|
|                   | ปิโตรเคมี                     |
| ขีการ <b>ส</b> ถน | 2534                          |

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| ลายมือชื่ออาจารย์ที่ปรึกษารวม              |

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KEY WORDS: SMC/DUAL CATALYST/CURE TIME/UNSATURATED POLYESTER/COMPOSITES.

SUCHOL MANRUKRIAN: EFFECT OF CATALYSTS ON CURE TIME OF POLYESTER SMC PROCESSING. THESIS ADVISORS: ASSOC. PROF.SUDA KIATKAMJORNWONG, Ph.D.; Wg.Cdr.SOMSAK NAVIROJ, Ph.D., 156 PP. ISBN 974-581-166-2

In this research, three types of catalysts: t-butyl perbenzoate (TBPB), 1,1-di-(t-butyl peroxy) cyclohexane (DTBC), and benzoyl peroxide (BPO), and the effect of added parabenzoquinone (PBQ) were studied. The effect of types and concentration of catalysts on storage life of the SMC, mechanical properties, and curing characteristics in the mold of the SMC at 1 and 2 minutes were investigated. When the TBPB catalyst was used, the storage time of SMC could be as long as 45 days. Between 17-30 days the hardness of SMC was suitable for molding at 150°C and 2 minutes curing time. The shorter curing time (1 minute) gave lower strength to the composite but with an addition of PBQ of 0.025 phr, the strength was enhanced. So did the SMC with the DTBC catalyst and the storage life of SMC was about 30 days. The dual catalysts system of the DTBC catalyst in conjunction with the TBPB catalyst provided a faster cure than did the TBPB catalyst alone. By using the BPO catalyst, it was found that the molding temperature was reduced to 120°C but the viscosity of paste was increased rapidly and the sheet could be stored for only 10 days. After a 1 minute of cure, the strength of composite became somewhat lower than that of a 2 minutes curing time, and was also lower than those by TBPB and DTBC catalysts by 10%. Therefore, the most suitable catalysts for the SMC in this research were TBPB and DTBC, in the presence of 0.025 phr of PBQ in the SMC component, the BPO cannot be used as the sole catalyst in the SMC.

| ภาควิชาสี   | หสาขาวิชาปิโตรเคมี-โพลิเมอร์ | ř. |
|-------------|------------------------------|----|
| สาขาวิชา    | ปิโตรเคมี                    |    |
| ปีการศึกษา. | 2534                         |    |



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