



## REFERENCES

- Abuluwafa, H., R. I. L. Guthrie, et al. (1996). "The effect of oxygen concentration on the oxidation of low-carbon steel in the temperature range 1000 to 1250°C." Oxidation of Metals 46(5): 423-440.
- Abuluwafa, H. T. (1996). Characterization of Oxide (Scale) Growth of Low Carbon Steel During Reheating. Doctor of Philosophy, McGill University.
- Addach, H., P. Berçot, et al. (2005). "Hydrogen permeation in iron at different temperatures." Materials Letters 59(11): 1347-1351.
- Bruzzone, P., R. M. Carranza, et al. (1999). "A pressure modulation method to study surface effects in hydrogen permeation through iron base alloys." Electrochimica Acta 44(24): 4443-4452.
- Bruzzone, P., R. M. Carranza, et al. (1999). "Hydrogen transport through  $\alpha$ -iron studies using a current modulation method." International Journal of Hydrogen Energy 24(11): 1093-1099.
- Cheng, Y. F. and F. R. Steward (2004). "Corrosion of carbon steels in high-temperature water studied by electrochemical techniques." Corrosion Science 46(10): 16.
- Cornell, R. M. and U. Schwertmann (2003). The iron oxides: structure, properties, reactions, occurrences, and uses, Wiley-VCH.
- Davies, M. H., M. T. Simnad, et al. (1951). "Transactions of the Metallurgical Society of AIME." Journal of Metals 3.
- Davis, J. R., J. D. Destefani, et al. (1987). Metals Handbook. Metals Park, Ohio, USA, Corrosion ASM International.
- Fontana, M. G. (1988). Corrosion Engineering, McGraw-Hill.
- Gaskell, D. R. (1981). Introduction to metallurgical thermodynamics, Taylor & Francis.
- Gulbransen, E. A. and R. Ruka (1952). "Role of Crystal Orientation in the Oxidation of Iron." Journal of The Electrochemical Society 99(9): 360-368.
- Hansen, M., K. Anderko, et al. (1965). Constitution of binary alloys, McGraw-Hill.
- Kofstad, P. (1988). High Temperature Corrosion, Elsiver Applied Science.

- Lecourt, S. (1996). Pickling. The book of steel. G. Béranger, G. Henry and G. Sanz. Andover, UK, Intercept: 584-595.
- Leelasangsai, C. (2009). Measurement of the hydrogen diffusion through various steels with and without oxide films. M.Sc, Chulalongkorn University.
- Lister, D. H., N. Arbeau, et al. (1994). "Erosion and Cavitation in the CANDU Primary Heat Transport System." Atomic Energy Control Board Report RSP-009.
- Lister, D. H., J. Slade, et al. (1997). The accelerated corrosion of CANDU outlet feeders - Observations, possible mechanisms and potential remedies. CAN/CNS Annual Conference. Toronto, Ontario, Canada.
- McKeen, K., M. Lalonde, et al. (2007). Hydrogen Effusion Probe development and installation at the Point Lepreau Nuclear Generating Station. 28th Annual Canadian Nuclear Society Conference. Canada.
- Piggott, M. R. and A. C. Siarkowski (1972). "Hydrogen diffusion through oxide films on steel." Journal of the Iron and Steel Institute: 4.
- Potter, E. C. and G. M. W. Mann (1962). Oxidation of Mild Steel in High Temperature Aqueous Systems. First International Congress on Metallic Corrosion. Butterworth, London.
- Potter, E. C. and G. M. W. Mann (1963). Mechanism of magnetite growth on low-carbon steel in steam and aqueous solutions up to 550 degrees C. 2nd International congress on metallic corrosion. Houston.
- Pyun, S.-I. and R. A. Oriani (1989). "The permeation of hydrogen through the passivating films on iron and nickel." Corrosion Science 29(5): 485-496.
- SCHOMBERG, K. and H. J. GRABKE (1996). Hydrogen permeation through oxide and passive films on iron. Düsseldorf, ALLEMAGNE, Stahleisen.
- Smithells, C. J. and C. E. Ransley (1935). "The Diffusion of Gases through Metals." Proceedings of the Royal Society of London. Series A, Mathematical and Physical Sciences 150(869): 172-197.
- Stone, J. M. (1981). Deuterium permeation and surface effects.
- Tomlinson, L. (1981). "Mechanism of corrosion of carbon and low alloy ferritic steels by high temperature water." Corrosion-NACE 39(10): 6.

Tomlinson, L. and N. J. Cory (1989). "Hydrogen emission during the steam oxidation of ferritic steels: Kinetics and mechanism." Corrosion Science 29(8): 939-965.

## **APPENDICES**

### **Appendix A Cavity Volume and Amount of Oxygen in Each Test Section**

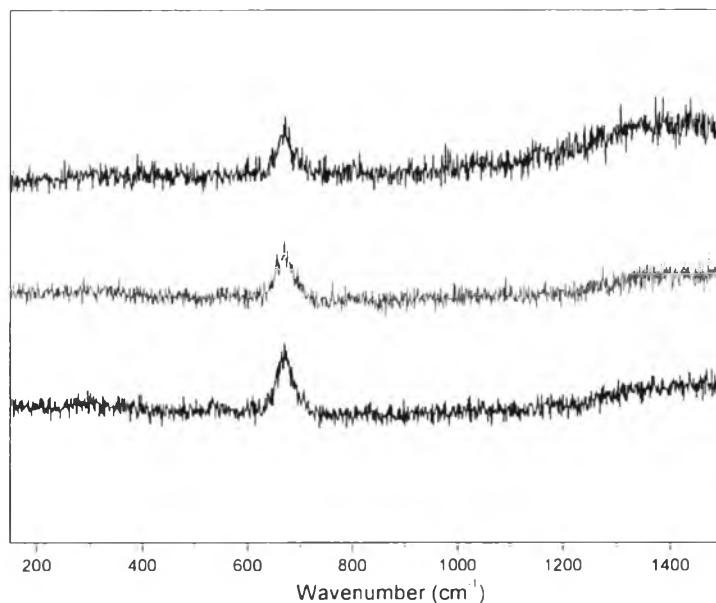
The amount of oxygen in each test section was calculated from volume of the cavity in the test section using the ideal gas law.

**Table A1** Cavity volume and amount of oxygen in experimental set 1

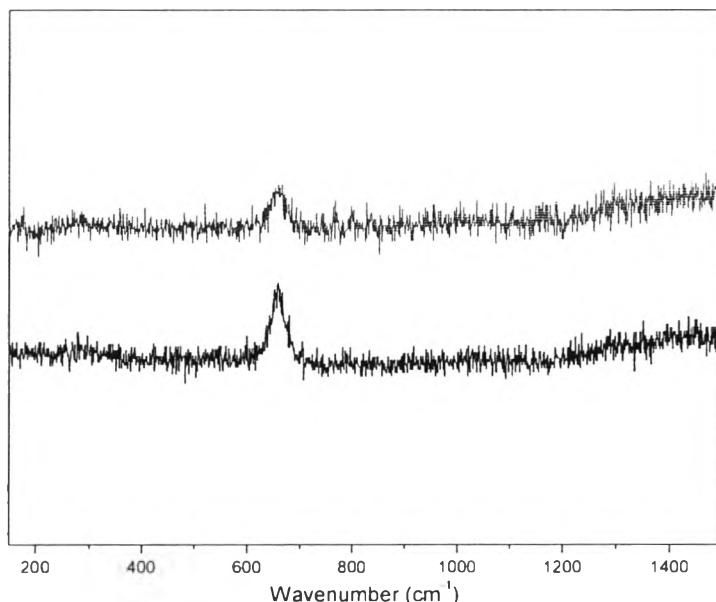
Exp.	RUN #	TAG	Sample	Material	cavity Vol (m3)	Amount of O2 in cavity			
						n = PV/RT T at install (K) 298.15	P at install (Pa) 101325	R (m3*Pa/mol*K) 8.314472	O2 in Air 0.206836
						mol AIR	mol O2	weight O2 (g)	
A	1 400C, 4days Labelled 3.0	P5	Membrane+Wire	CS	1.30015E-05	0.000531423	0.000109917	0.003517354	
		P6	Membrane+Wire	SS	1.31944E-05	0.000539308	0.000111548	0.003569547	
		P7	Membrane+Wire	NI	1.32228E-05	0.000540468	0.000111788	0.003577226	
	2 400C, 7days Labelled 2.0	P5	Membrane+Wire	CS	1.30015E-05	0.000531423	0.000109917	0.003517354	
		P6	Membrane+Wire	SS	1.31974E-05	0.000539429	0.000111573	0.003570346	
		P7	Membrane+Wire	NI	1.32241E-05	0.000540521	0.000111799	0.003577574	
	3 400C, 1day Labelled 1.0	P5	Membrane+Wire	CS	1.30015E-05	0.000531423	0.000109917	0.003517354	
		P6	Membrane+Wire	SS	1.32096E-05	0.00053993	0.000111677	0.003573664	
		P7	Membrane+Wire	NI	1.32315E-05	0.000540824	0.000111862	0.003579583	
	4 90C, 14days Labelled 4.0	P5	Membrane+Wire	CS	1.33293E-05	0.000544821	0.000112689	0.003606034	
		P6	Membrane+Wire	SS	1.34166E-05	0.000548389	0.000113427	0.003629648	
		P7	-	-	0	0	0	0	0
	5 90C, 7days Labelled 5.0	P5	Membrane+Wire	CS	1.33293E-05	0.000544821	0.000112689	0.003606034	
		P6	Membrane+Wire	SS	-2.02204E-05	-0.00082649	-0.000170948	-0.00547033	
		P7	-	-	0	0	0	0	0
	6 90C, 1day Labelled 6.0	P5	Membrane+Wire	CS	1.33293E-05	0.000544821	0.000112689	0.003606034	
		P6	Membrane+Wire	SS	1.3473E-05	0.000550697	0.000113904	0.003644925	
		P7	-	-	0	0	0	0	0

**Table A2** Cavity volume and amount of oxygen in experimental set 2

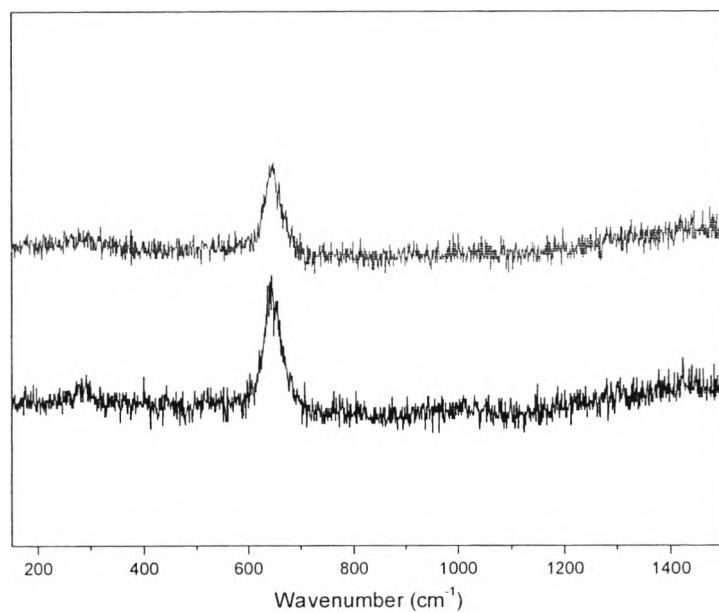
Exp.	RUN #	TAG	Sample	Material	cavity Vol	Amount of O2 in cavity			
					(m3)	n = PV/RT T at install (K) 298.15	P at install (Pa) 101325	R (m3*Pa/mol*K) 8.314472	O2 in Air 0.206836
						mol AIR	mol O2	weight O2 (g)	
A-2	7  400C, 7days  Labelled 7.0	P5	-	Empty tube	1.32694E-05	0.000542373	0.000112182	0.003589831	
		P6	Membrane+Wire	CS	1.46768E-05	0.000599899	0.000124081	0.003970581	
		P7	Wire	CS	1.25363E-05	0.000512408	0.000105984	0.0033915	
	Test section  Preparation  +	P5	-	-	0	0	0	0	
		P6	-	-	0	0	0	0	
		P7	EXTRA WIRE	CS	1.29128E-05	0.000527797	0.000109167	0.003493356	
	WIRE TEST  8  400C, 1day  labelled 14.0	P5	WIRE-03	CS	1.28542E-05	0.000525404	0.000108673	0.00347752	
		P5	WIRE-01	CS	1.28542E-05	0.000525404	0.000108673	0.00347752	
		P7	WIRE-02	CS	1.29128E-05	0.000527797	0.000109167	0.003493356	
	9  400C, 5 hr  labelled14.FP	P5	(nitrogen filled)	Nitrogen	1.40089E-05	0.0005726	0.000118434	0.003789899	
		P6	Membrane+WIRE04	CS	1.27031E-05	0.000519226	0.000107395	0.003436631	
		P7	WIRE-05	CS	1.29128E-05	0.000527797	0.000109167	0.003493356	

**Appendix B Raman Spectra of Each Wire in Table A1 and Table A2**

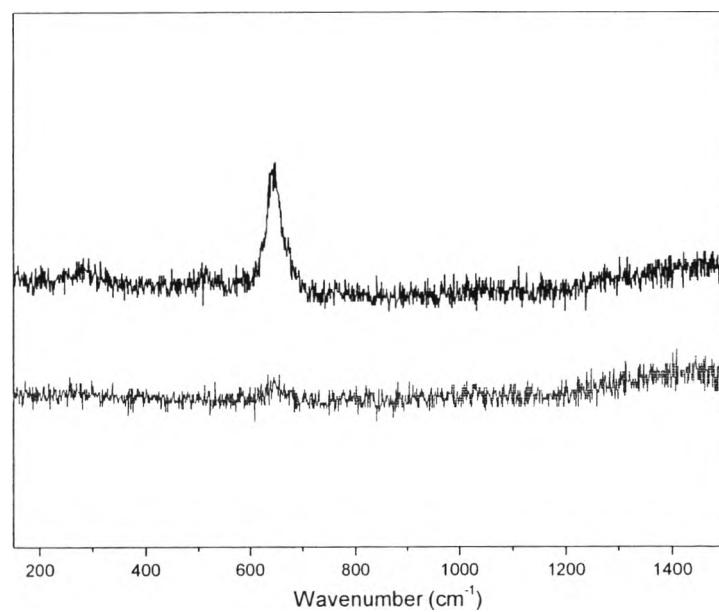
**Figure B1** Raman spectra of wire in run 7.



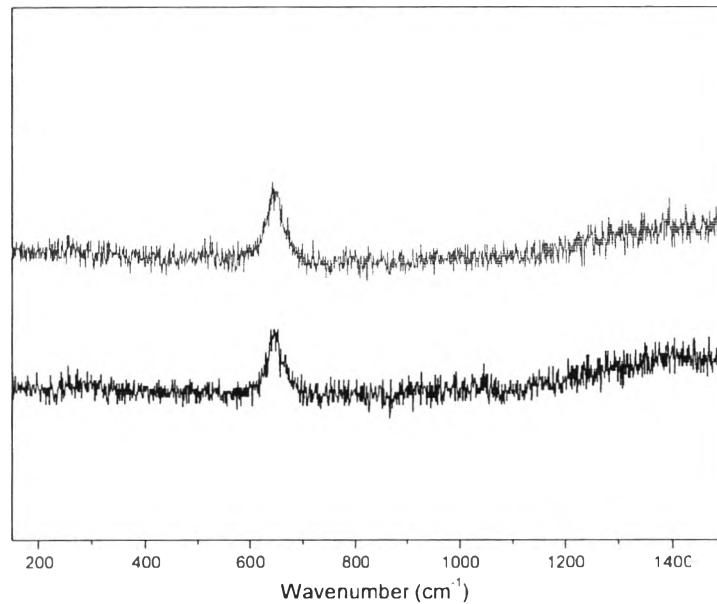
**Figure B2** Raman spectra of WIRE-01.



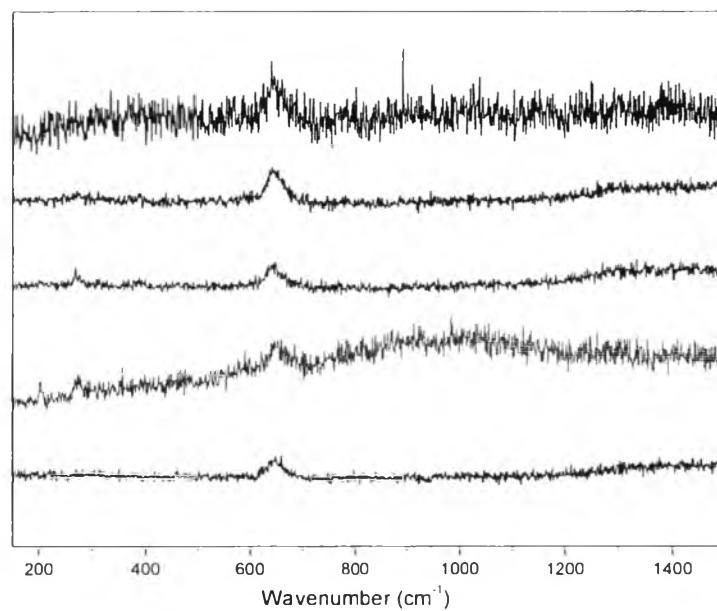
**Figure B3** Raman spectra of WIRE-02.



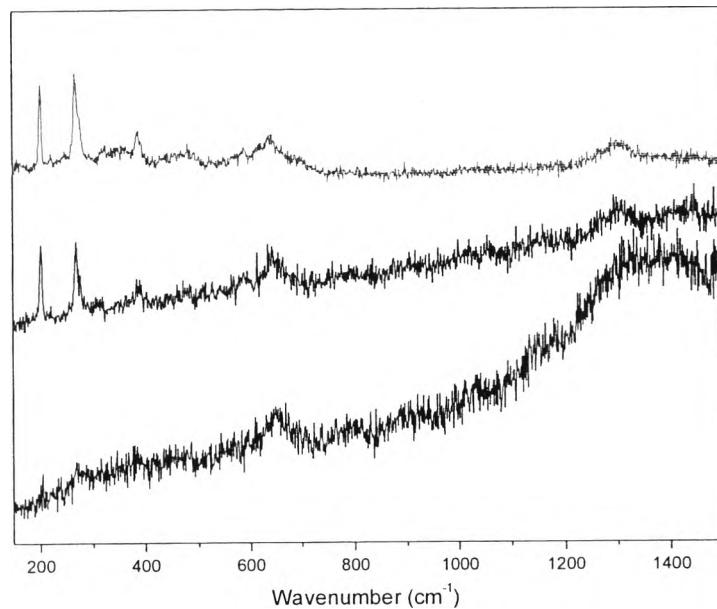
**Figure B4** Raman spectra of WIRE-03.



**Figure B5** Raman spectra of WIRE-05.



**Figure B6** Raman spectra of WIRE-07.



**Figure B7** Raman spectra of WIRE-11.

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1. Weerakul, S.; Rirksomboon, T.; and Steward, F. R. (2012, April 24) Kinetics of oxide formation on various steel surfaces in the presence of oxygen-nitrogen mixtures. Proceedings of the 3<sup>rd</sup> Research Symposium on Petrochemical and Materials Technology and the 18<sup>th</sup> PPC Symposium on Petroleum, Petrochemicals, and Polymers, Bangkok, Thailand.

