

REFERENCES

- Adams, R., Carson, C., Ward, J., Tannenbaum, R., and Koros, W. (2010) Metal organic framework mixed matrix membranes for gas separations. *Microporous and Mesoporous Materials*, 131(1-3), 13-20.
- Aroon, M.A., Ismail, A.F., Matsuura, T., and Montazer-Rahmati, M.M. (2010) Performance studies of mixed matrix membranes for gas separation: A review. *Separation and Purification Technology*, 75(0), 229-242.
- Basu, S., Cano-Odena, A., and Vankelecom, I.F.J. (2011) MOF-containing mixed-matrix membranes for CO₂/CH₄ and CO₂/N₂ binary gas mixture separations. *Separation and Purification Technology*, 81(1), 31–40.
- Bouma, R.H.B., Checchetti, A., Chidichimo, G., and Drioli, E. (1997) Permeation through a heterogeneous membrane: the effect of the dispersed phase. *Journal of Membrane Science*, 128(2), 141–149.
- Bux, H., Liang, F., Li, Y., Cravillon, J., Wiebcke, M., and Caro, J. (2009) Zeolithic Imidazolate Framework Membrane with Molecular Sieving Properties by Microwave-Assisted Solvothermal Synthesis, *Journal of the American Chemical Society*, 131(0), 16000-16001.
- Chiew, Y.C. and Glandt, E.D. (1983) The effect of structure on the conductivity of a dispersion. *Journal of Colloid and Interface Science*. 94(1), 90–104.
- Chung, T.S., Jiang, L.Y., Li, Y., and Kulprathipanja, S. (2007) Mixed matrix membranes (MMMs) comprising organic polymers with dispersed inorganic fillers for gas separation. *Progress in Polymer Science*, 32(0), 483-507.
- Dai, Y., Johnson, J.R., Karvan, O., Sholl, D.S., and Koros, W.J. (2012) Ultem®/ZIF-8 mixed matrix hollow fiber membranes for CO₂/N₂ separations. *Journal of Membrane Science*, 401-402, 76-82.
- Goh, P.S., Ismail, A.F., Sanip, S.M., Ng, B.C., and Aziz, M. (2011) Recent advances of inorganic fillers in mixed matrix membrane for gas separation. *Separation and Purification Technology*, 81(3), 243-264.
- Gonzo, E.E., Parentis, M.L., and Gottifredi, J.C. (2006) Estimating models for predicting effective permeability of mixed matrix membranes. *Journal of Membrane Science*, 277(1-2), 46–54.

- Gutiérrez-Sevillano, J.J., Vicent-Luna, J.M., Dubbeldam, D., and Calero, S. (2013) Molecular mechanisms for adsorption in Cu-BTC metal organic framework. *Journal of Physical Chemistry*, 117(0), 11357-11366.
- Hassan, M.H., Way, J.D., Thoen, P.M., and Dillon, A.C. (1995) Single component and mixed gas transport in a silica hollow fiber membrane. *Journal of Membrane Science*, 104(1-2), 27-42.
- Hu, J., Cai, H., Ren, H., Wei, Y., Xu, Z., Liu, H., and Hu, Y. (2010) Mixed-matrix membrane hollow fibers of Cu(3)(BTC)(2) MOF and polyimide for gas separation and adsorption. *Industrial & Engineering Chemistry Research*, 49(24), 12605–12612.
- Hunger, K., Schmeling, N., Jeazet, H.T., Janiak, C., Staudt, C., and Kleinermanns, K. (2012) Investigation of cross-linked and additive containing polymer materials for membranes with improved performance in pervaporation and gas separation. *Journal of Membranes*, 2(0), 727-763.
- Keskin, S. and Sholl, D.S. (2010) Selecting metal organic frameworks as enabling materials in mixed matrix membranes for high efficiency natural gas purification. *Energy and Environmental Science*, 3(0), 343-351.
- Kim, S., Pechar, T.W., and Marand, E. (2006) Poly(imide siloxane) and carbon nanotube mixed matrix- membranes for gas separation. *Separation and Purification Technology*, 192(0), 330-339.
- Koros, W.J. and Chern, R.T. (1987) *Handbook of Separation Process Technology: Separation of Gaseous Mixtures using Polymer Membranes*. New York: John Wiley & Sons.
- Kurdi, J. and Tremblay, A.Y. (1998) Preparation of defect-free asymmetric membranes for gas separations. *Journal of Applied Polymer Science*, 73(0), 1471-1482.
- Leiknes, T. (1999) *Theory of Transport in Membranes*. Norway: Trondheim.
- Lewis, T. and Nielsen, L. (1970) Dynamic mechanical properties of particulate-filled composites. *Journal of Applied Polymer Science*, 14(6), 1449–1471.
- Li, J.-R., Ma, Y., McCarthy, M.C., Sculley, J., Yu, J., Jeong, H.-K., Balbuena, P.B., and Zhou, H.-C. (2011) Carbon dioxide capture-related gas adsorption and

- separation in metal-organic frameworks. *Coordination Chemistry Reviews*, 255(15-16), 1791-1823.
- Liu, C., McCulloch, B., Wilson, S.T., Benin, A.I., and Schott, M.E. (2009) U.S. Patent 7637983.
- Mahajan, R. and Koros, W.J. (2004) Mixed matrix membrane materials with glassy polymers. *Polymer Engineering & Science*, 42(7), 1420–1431.
- Moore, T.T., Mahajan, R., Vu, D.Q., and Koros, W.J. (2004) Hybrid membrane materials comprising organic polymers with rigid dispersed phases. *American Institute of Chemical Engineers Journal*, 50(2), 311–321.
- Nielsen, L. (1973) Thermal conductivity of particulate-filled polymers. *Journal of Applied Polymer Science*, 17(12), 3819–3820.
- Nik, O.G., Chen, X.Y., and Kaliaguine, S. (2012) Functionalized metal organic framework-polyimide mixed matrix membranes for CO₂/CH₄ separation. *Journal of Membrane Science*, 413–414(0), 48–61.
- Ordonez, M.J.C., Balkus, K.J., Ferraris, J.P., and Musselman, I.H. (2010) Molecular sieving realized with ZIF-8/Matrimid mixed-matrix membranes. *Journal of Membrane Science*, 361(0), 28–37.
- Othman, M.R. (2009) Permeability and separability of methane and carbon dioxide across meso-porous Mg-Al hydrotalcite and activated carbon media. *Chemical Engineering Science*, 64(5), 925–929.
- Perez, E.V., Balkus, K.J., Ferraris, J.P., and Musselman, I.H. (2009) Mixed-matrix membranes containing MOF-5 for gas separations. *Journal of Membrane Science*, 328(1-2), 165–173.
- Ploegmakers, J., Japip, S., and Nijmeijer, K. (2013) Mixed matrix membranes containing MOFs for ethylene/ethane separation Part A: Membrane preparation and characterization. *Journal of Membrane Science*, 428, 445–453.
- Song, Q., Nataraj, S.K., Roussenova, M.V., Tan, J.C., Hughes, D.J., Li, W., Bourgoin, P., Alam, M.A., Cheetham, A.K., Al-Muhtaseb, S.A., and Sivaniah, E. (2012) Zeolitic imidazolate framework (ZIF-8) based polymer nanocomposite membranes for gas separation. *Energy & Environmental Science*, 5(8), 8359–8369.

- Singha-in, P., Rirksomboon, T., and Kulprathipanja, S. (2008) Mixed matrix membranes for CO₂/CH₄ separation: Effects of various zeolites incorporated into cellulose acetate. The Petroleum and Petrochemical College, Chulalongkorn University, Bangkok, Thailand.
- Suntiworawut, T., Rirksomboon, T., and Kulprathipanja, S. (2009) Mixed matrix membranes for gas separation: Effects of various glycols incorporated into activated carbon and Ultem. The Petroleum and Petrochemical College, Chulalongkorn University, Bangkok, Thailand.
- Verweij, H. (2012) Inorganic membranes. Current Opinion in Chemical Engineering, 1(2), 156-162.
- Vu, D.Q., Koros, W.J., and Miller, S.J. (2003) Mixed matrix membranes using carbon molecular sieves: I. Preparation and experimental results. Journal of Membrane Science, 211(2), 311-334.
- White, J.C., Dutta, P.K., Shqau, K., and Verweij, H. (2010) Synthesis of ultrathin zeolite Y membranes and their application for separation of carbon dioxide and nitrogen gases. American Chemical Society, 26(12), 10287-10293.
- Yang, Q. and Zong, C. (2006) Molecular simulation of carbon dioxide/methane/hydrogen mixture adsorption in metal-organic frameworks. Journal of Physical Chemistry, 110, 17776-17783.
- Yehia, H., Pisklak, T.J., Ferraris, J.P., Balkus, K.J., and Musselman, I.H. (2004) Methane facilitated transport using copper (II) biphenyl dicarboxylate-triethylenediamine poly(3-acetoxyethylthiophene) mixed matrix membranes. Polymer Preprints, 45(1), 35-36.
- Zhang, Y., Musselman, I.H., Ferraris, J.P., and Balkus, K.J.Jr. (2008) Gas permeability properties of Matrimid® membranes containing the metal-organic framework Cu-BPY-HFS. Journal of Membrane Science, 313 (1-2), 170-181.
- Zhang, Y., Sunarso, J., Liu, S., and Wang, R. (2013) Current status and development of membranes for CO₂/CH₄ separation: A review. International Journal of Greenhouse Gas Control, 12(0), 84-107.
- Zornoza, B., Tellez, C., Coronas, J., Gascon, J., and Kapteijn, F. (2013) Metal organic framework based mixed matrix membranes: An increasingly

important field of research with a large application potential. Microporous and Mesoporous Materials, 166, 67-78.

APPENDICES

Appendix A The Experimental Flow Rate of Methane (CH_4), Carbon Dioxide (CO_2), and Nitrogen (N_2) of Dense Membrane and Mixed Matrix Membranes in Performance at 50 psi and 100 psi

Table A1 Pure Ultem membrane

Gas	Pressure (psi)	Flow rate (cm^3/s)	Permeance (GPU)	Average of Permeance (GPU)	STDEV of Permanence
N_2	50	1.81E-05	0.00199	0.00199	2.88E-07
		1.81E-05	0.00199		
		1.81E-05	0.00199		
	100	3.66E-05	0.00201	0.00201	3.40E-07
		3.66E-05	0.00201		
		3.66E-05	0.00201		
CH_4	50	6.79E-06	-0.00074	0.00074	4.05E-08
		6.79E-06	0.00074		
		6.79E-06	0.00075		
	100	1.36E-05	0.00075	0.00075	4.67E-08
		1.36E-05	0.00074		
		1.36E-05	0.00075		
N_2	100	3.66E-05	0.00201	0.00201	1.22E-06
		3.66E-05	0.00201		
		3.66E-05	0.00201		
CO_2	50	1.88E-04	0.02067	0.02065	1.80E-05
		1.88E-04	0.02064		
		1.88E-04	0.02064		
	100	3.76E-04	- 0.02062	0.02063	1.79E-05
		3.76E-04	0.02062		
		3.77E-04	0.02066		
N_2	100	3.66E-05	0.00201	0.00201	8.99E-07
		3.66E-05	0.00201		
		3.66E-05	0.00201		

Table A2 10 wt% MOF-199 MMMs

Gas	Pressure (psi)	Flow rate (cm ³ /s)	Permeance (GPU)	Average of Permeance (GPU)	STDEV of Permeance
N ₂	50	2.41E-05	0.00264	0.00264	5.87E-07
		2.41E-05	0.00264		
		2.41E-05	0.00264		
	100	5.83E-05	0.00320	0.00266	4.65E-04
		4.36E-05	0.00239		
		4.36E-05	0.00239		
CH ₄	50	1.03E-05	0.00113	0.00112	9.23E-08
		1.03E-05	0.00112		
		1.03E-05	0.00112		
	100	2.05E-05	0.00113	0.00113	2.82E-07
		2.05E-05	0.00113		
		2.05E-05	0.00113		
N ₂	100	5.84E-05	0.00320	0.00266	4.66E-04
		4.36E-05	0.00239		
		4.36E-05	0.00239		
CO ₂	50	3.76E-04	0.04125	0.04125	6.20E-05
		3.75E-04	0.04119		
		3.77E-04	0.04131		
	100	7.46E-04	0.04094	0.04082	2.10E-04
		7.40E-04	0.04058		
		7.46E-04	0.04094		
N ₂	100	5.82E-05	0.00319	0.00266	4.63E-04
		4.36E-05	0.00239		
		4.36E-05	0.00239		

Table A3 20 wt% MOF-199 MMMs

Gas	Pressure (psi)	Flow rate (cm ³ /s)	Permeance (GPU)	Average of Permeance (GPU)	STDEV of Permeance
N ₂	50	2.53E-05	0.00278	0.00271	5.57E-05
		2.44E-05	0.00268		
		2.44E-05	0.00268		
	100	5.13E-05	0.00282	0.00273	1.46E-04
		5.13E-05	0.00282		
		4.67E-05	0.00256		
CH ₄	50	1.14E-05	0.00125	0.00124	3.75E-06
		1.13E-05	0.00124		
		1.13E-05	0.00124		
	100	2.27E-05	0.00125	0.00125	2.62E-07
		2.27E-05	0.00125		
		2.27E-05	0.00125		
N ₂	100	5.13E-05	0.00281	0.00273	1.47E-04
		5.13E-05	0.00281		
		4.67E-05	0.00256		
CO ₂	50	4.56E-04	0.05006	0.05021	1.92E-04
		4.57E-04	0.05015		
		4.60E-04	0.05042		
	100	9.09E-04	0.04987	0.04994	2.78E-04
		9.06E-04	0.04969		
		9.16E-04	0.05024		
N ₂	100	5.13E-05	0.00282	0.00273	1.46E-04
		5.13E-05	0.00281		
		4.67E-05	0.00256		

Table A4 30 wt% MOF-199 MMMs

Gas	Pressure (psi)	Flow rate (cm ³ /s)	Permeance (GPU)	Average of Permeance (GPU)	STDEV of Permeance
N ₂	50	2.79E-05	0.00306	0.00306	6.82E-07
		2.79E-05	0.00306		
		2.79E-05	0.00306		
	100	6.29E-05	0.00345	0.00308	3.27E-04
		5.33E-05	0.00293		
		5.20E-05	0.00285		
CH ₄	50	1.24E-05	0.00136	0.00136	2.72E-07
		1.24E-05	0.00137		
		1.24E-05	0.00136		
	100	2.49E-05	0.00136	0.00137	4.15E-07
		2.49E-05	0.00136		
		2.49E-05	0.00137		
N ₂	100	4.95E-05	0.00271	0.00272	2.15E-06
		4.95E-05	0.00272		
		4.96E-05	0.00272		
CO ₂	50	5.34E-04	0.05861	0.05857	7.22E-05
		5.33E-04	0.05849		
		5.34E-04	0.05861		
	100	1.06E-03	0.05812	0.05771	3.72E-04
		1.05E-03	0.05739		
		1.05E-03	0.05763		
N ₂	100	4.96E-05	0.00272	0.00272	1.64E-06
		4.95E-05	0.00272		
		4.95E-05	0.00272		

Table A5 10 wt% ZIF-8 MMMs

Gas	Pressure (psi)	Flow rate (cm ³ /s)	Permeance (GPU)	Average of Permeance (GPU)	STDEV of Permeance
N ₂	50	2.98E-05	0.00327	0.00327	3.25E-06
		2.98E-05	0.00327		
		2.98E-05	0.00327		
	100	6.01E-05	0.00330	0.00329	2.74E-06
		6.00E-05	0.00329		
		6.00E-05	0.00329		
CH ₄	50	1.15E-05	0.00127	0.00126	2.94E-07
		1.15E-05	0.00126		
		1.15E-05	0.00127		
	100	2.32E-05	0.00127	0.00127	8.08E-06
		2.29E-05	0.00126		
		2.32E-05	0.00127		
N ₂	100	6.00E-05	0.00329	0.00329	2.41E-06
		6.00E-05	0.00329		
		6.00E-05	0.00329		
CO ₂	50	3.35E-04	0.03677	0.03674	2.84E-05
		3.35E-04	0.03672		
		3.35E-04	0.03672		
	100	6.56E-04	0.03600	0.03581	1.87E-04
		6.53E-04	0.03581		
		6.49E-04	0.03562		
N ₂	100	6.00E-05	0.00329	0.00329	3.97E-06
		6.00E-05	0.00329		
		5.99E-05	0.00328		

Table A6 20 wt% ZIF-8 MMMs

Gas	Pressure (psi)	Flow rate (cm ³ /s)	Permeance (GPU)	Average of Permeance (GPU)	STDEV of Permeance
N ₂	50	3.60E-05	0.00395	0.00395	4.73E-06
		3.60E-05	0.00395		
		3.59E-05	0.00394		
	100	7.27E-05	0.00399	0.00398	4.81E-06
		7.25E-05	0.00398		
		7.25E-05	0.00398		
CH ₄	50	1.48E-05	0.00163	0.00162	4.84E-07
		1.48E-05	0.00162		
		1.48E-05	0.00162		
	100	2.95E-05	0.00162	0.00162	1.52E-06
		2.94E-05	0.00162		
		2.94E-05	0.00161		
N ₂	100	7.24E-05	0.00397	0.00398	8.11E-06
		7.25E-05	0.00398		
		7.27E-05	0.00399		
CO ₂	50	4.58E-04	0.05024	0.05018	5.30E-05
		4.57E-04	0.05015		
		4.57E-04	0.05015		
	100	8.90E-04	0.04881	0.04847	3.43E-04
		8.83E-04	0.04846		
		8.77E-04	0.04812		
N ₂	100	7.27E-05	0.00399	0.00398	1.06E-05
		7.24E-05	0.00397		
		7.23E-05	0.00397		

Appendix B The Experimental Gas Selectivity of Dense Membrane and Mixed Matrix Membranes in Performance at 50 psi and 100 psi

Table B1 Gas selectivity determined from gas permeance of Ultem membrane and MOF-MMMs at pressures of 50 psi and 100 psi

Membrane	MOF Loading (wt%)	CO ₂ /CH ₄ Selectivity		CO ₂ /N ₂ Selectivity	
		50 psi	100 psi	50 psi	100 psi
Pure Ultem	0	27.719	27.696	10.396	10.266
MOF-199 MMMs	10	36.394	36.104	15.867	15.614
	20	40.361	40.025	18.501	18.280
	30	42.910	42.276	19.145	18.758
ZIF-8 MMMs	10	29.042	28.261	11.232	10.872
	20	30.887	30.008	12.707	12.172

Appendix C The Gas Permeance Predicted by Maxwell Model for MOF-MMMs at Pressure of 50 psi

Table C1 Comparison of gas permeances for MOF-MMMs at pressure of 50 psi based on Maxwell model and experimental data.

MOFs	Gas	Loading		P_c (GPU)	P_d (GPU)	P_{eff}	
		wt. fraction	vol. fraction			Maxwell's model	Experiment
MOF-199	CO_2	10	0.168	0.02065	44	0.03314	0.04240
		20	0.312	0.02065	44	0.04869	0.05021
		30	0.437	0.02065	44	0.06862	0.05857
	CH_4	10	0.168	0.00074	15	0.00119	0.00116
		20	0.312	0.00074	15	0.00175	0.05802
		30	0.437	0.00074	15	0.00246	0.06583
	N_2	10	0.168	0.00199	19	0.00320	0.00267
		20	0.312	0.00199	19	0.00470	0.00271
		30	0.437	0.00199	19	0.00662	0.07364
ZIF-8	CO_2	10	0.168	0.02065	44	0.03314	0.08145
		20	0.312	0.02065	44	0.04869	0.05018
		30	0.437	0.02065	44	-	-
	CH_4	10	0.168	0.00074	15	0.00119	0.08926
		20	0.312	0.00074	15	0.00175	0.09707
		30	0.437	0.00074	15	-	-
	N_2	10	0.168	0.00199	19	0.00320	0.00327
		20	0.312	0.00199	19	0.00470	0.00395
		30	0.437	0.00199	19	-	-

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Presentations:

1. Ketsuwan, T.; Rirksomboon, T.; Kulprathipanja, S.; and Bowen, T.C. (2014, April 22) Solid-Polymer Mixed Matrix Membranes for CO₂/CH₄ Separation: Metal Organic Frameworks and Polyetherimide. Paper presented at The 5th Research Symposium on Petrochemical and Materials Technology and The 20th PPC Symposium on Petroleum, Petrochemicals, and Polymers, Bangkok, Thailand.
2. Ketsuwan, T.; Rirksomboon, T.; Kulprathipanja, S.; and Bowen, T.C. (2014, May 7-8) Enhanced Selectivity of Metal Organic Frameworks-Polyetherimide Mixed Matrix Membranes for CO₂/CH₄ Separation. Paper presented at ICERE: International Conference on Environment and Renewable Energy, Paris, France.