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APPENDICES

Appendix A Calculation of Methane Conversion and Product Selectivity

A.1 %CH₄ Conversion

1. Definition of CH₄ Conversion

$$\% CH_4 \text{ Conversion} = \frac{\text{Total mol of } CH_{4,in} - \text{Total mol of } CH_{4,out}}{\text{Total mol of } CH_{4,in}} \times 100 \quad (1)$$

2. *Total mol of CH_{4,in}* can be replaced by the term *Total C* since the source of total carbon are derived from only methane and *Total C in CH_{4,out}* can be replaced by *C in unreacted CH₄*.

Therefore, the eq. (1) becomes;

$$\% CH_4 \text{ Conversion} = \frac{\text{Total C} - \text{C in unreacted } CH_4}{\text{Total C}} \times 100 \quad (2)$$

3. Carbon balance with no coke formation:

Total C = Total C_{in} = Total C_{out} (obtained from all carbon in outlet stream)

Hence, *Total C – C in unreacted CH₄* in eq. (2) can be replaced by *Total C in all Products*

Thus eq. (2) becomes;

$$\% CH_4 \text{ Conversion} = \frac{\text{Total C in all Products}}{\text{Total C}} \times 100 \quad (3)$$

Example; for the reaction in case of the blank tube

1. The resulting peak area from online GC of all chemicals in the exhaust stream listed below:

Table A1 Peak area of exhaust stream

FID					TCD
CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
11808.5	0	38.8	1126.4	79.7	932.7

2. Change area to mol by multiplying with response factor of each substance.

Table A2 Response factor (obtained from Calibration Data)

Substance	Response factor(mol/area)
CH ₄	4.5969E-10
C ₂ H ₆	3.4581E-10
C ₂ H ₄	3.4151E-10
CH ₃ Br	1.0000E-09
CH ₂ Br ₂	5.0000E-10
CO	5.6853E-10

Table A3 Mol of each chemical species in the exhaust stream

Mol					
CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
5.42827E-06	0	1.32506E-08	1.1264E-06	3.985E-08	5.30274E-07

3. Total mol of C was calculated by

$$= \text{mol of } C_{CH_4} + 2(\text{mol of } C_{C_2H_6}) + 2(\text{mol of } C_{C_2H_4}) + \text{mol of } C_{CH_3Br} + \text{mol of } C_{CH_2Br_2} + \text{mol of } C_{CO}$$
C in product was calculated by

$$= 2(\text{mol of } C_{C_2H_6}) + 2(\text{mol of } C_{C_2H_4}) + \text{mol of } C_{CH_3Br} + \text{mol of } C_{CH_2Br_2} + \text{mol of } C_{CO}$$

Accordingly, the methane conversion calculated from eq. (3) was shown in the below table

Table A4 Methane conversion

Total mol of C	C in product	% CH ₄ Conversion
7.15129E-06	1.72302E-06	24.0939

A.2 %CH₃Br Selectivity

$$\% \text{CH}_3\text{Br Selectivity} = \frac{\text{mol of CH}_3\text{Br}}{\text{Total mol of Product}} \times 100$$

Example;

1. Mol of CH₃Br (shown in Table A3)
= 1.1264E-06 mol
2. Total mol of Product was calculated by
= mol of C₂H₆ + mol of C₂H₄ + mol of C_{CH₃Br} + mol of C_{CH₂Br₂} + mol of C_{CO}

Table A5 Total mol of Product

Mol					Total mol of product
C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO	
0	1.32506E-08	1.1264E-06	3.985E-08	5.30274E-07	1.70977E-06

$$\% \text{CH}_3\text{Br Selectivity} = \frac{1.1264\text{E}-06}{1.70977\text{E}-06} \times 100 = 65.88 \%$$

A.3 %CO Selectivity

*The conceptual calculation of %CO Selectivity is the same as %CH₃Br Selectivity.

Appendix B Calculation of Catalyst Composition

The Rh/SiO₂ catalyst was prepared by incipient wetness impregnation method which means support containing the same pore volume as the volume of the solution that was added.

Example 2 g of 0.5 wt% Rh/SiO₂ catalyst
 = 2 g (Rh₂O₃ + SiO₂) of 0.5 % (Rh⁰ w/w to Rh₂O₃ + SiO₂)

Rh₂O₃ form must be involved in this case due to small amount of catalyst prepared.

Step 1: wt. Rh⁰

wt. of Rh ⁰	0.0100	g
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Step 2: wt. Rh₂O₃

MW of Rh ⁰	102.9100	g/mol
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MW of Rh ₂ O ₃	253.8200	g/mol
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wt. of Rh ₂ O ₃	0.0123	g
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Step 3: wt SiO₂

wt. of Support (SiO ₂)	1.9877	g
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Step 4: wt RhCl₃*3H₂O

MW of RhCl ₃ *3H ₂ O	263.3103	g/mol
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g of RhCl ₃ *3H ₂ O	0.0256	g
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(assay ≥ 99.9% trace metal basis)

Corrected weight	0.0258	g
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Step 5: Lattice water

Lattice water	0.0052	ml
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(negligible)

Step 6: Volume of SiO₂

From BET surface analysis, total pore volume of SiO₂ is

2.1606 ml/g

Volume of SiO ₂	4.2925	ml
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Step 7: Water required

Impregnation volume (100% pore volume)

Water required for RhCl ₃ *3H ₂ O	4.2925	ml
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Table B1 The ingredients of prepared catalyst

Loading (wt%)	SiO ₂ (g)	RhCl ₃ *3H ₂ O (g)	Required water (ml)
0.1	2	0.0052	4.3160
0.3	2	0.0258	4.2925
0.5	2	0.0155	4.3052

Appendix C Calibration Data and Feed Flow Calibration

The response factors of methane (CH₄) ethane (C₂H₆) ethylene (C₂H₄), and carbon monoxide (CO) were determined by using the Single Point External Standard assuming analyte response to be linear over a range of concentrations. This method requires a known amount of analytes and record the peak area. The peak area of each substrate was calculated from average areas. The volume of each online injection equals to 2.5 ml which subsequently converted to mol bases on an ideal gas. Then calculate a response factor using an equation below.

$$\text{Response Factor} = \frac{\text{mol}}{\text{area}}$$

Table C1 The response factors calculated from the Single Point External Standard

No./ Retention time	Methane	Ethane	Ethylene	Carbon monoxide
	3.49	4.46	5.02	9.34
1	22225.7	295.6	299.1	180.7
2	22315.9	294.3	297.9	180.5
3	22222.5	297.2	301.1	178.2
4		296.3	300.2	180.4
Area	22254.7	295.85	299.575	179.95
Volume(l)	0.00025	0.0000025	0.0000025	0.0000025
Mol	1.023E-05	1.023E-07	1.023E-07	1.023E-07
Response factor (mol/area)	4.5969E-10	3.4581E-10	3.4151E-10	5.6853E-10

For methyl bromide (CH₃Br) and dibromomethane (CH₂Br₂), the response factors were determined by using the Multiple Point External Standard. The samples used in this method cover the expected analyte concentration range. Use a line fitting algorithm such as point to point, linear least squares, or quadratic least squares to produce a calibration curve. The response factor used for calculation the

products amount were derived from the the reciprocal of slope of calibration curve as shown in Figure C1 and Figure C2.

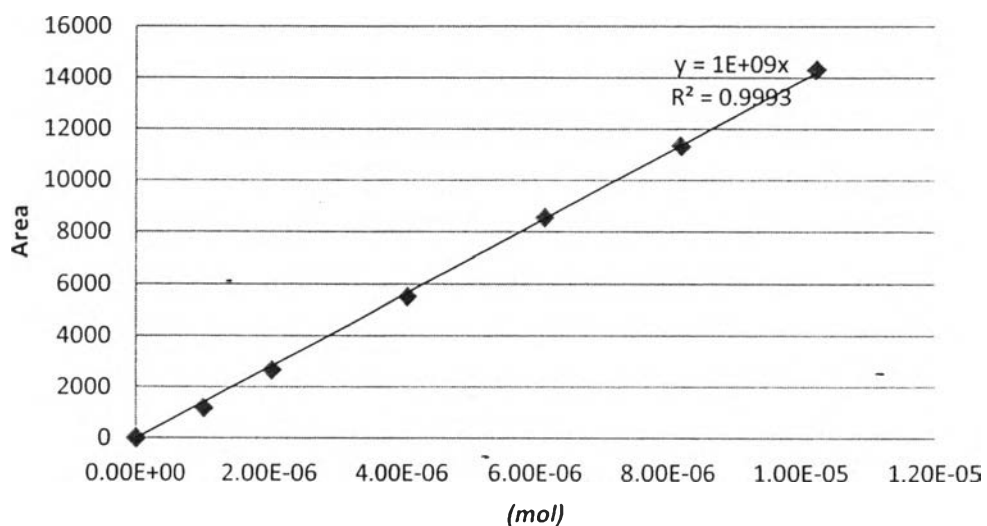


Figure C1 Response factors from GC FID as a function of injection volume of methyl bromide.

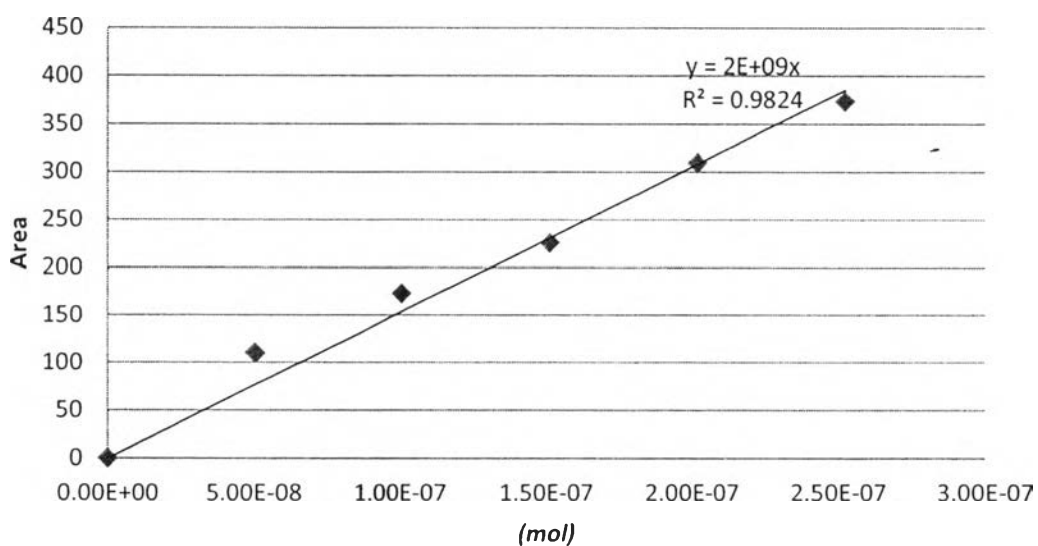


Figure C2 Response factors from GC FID as a function of injection volume of dibromomethanes.

Table C2 The response factors calculated from the Multiple Point External Standard

Chemicals	Retention time	Slope (area/mol)	Response factor (mol/area)
Methyl bromide	12.50	1E+09	1E-09
Dibromomethanes	21.03	2E+09	5E-10

Appendix D Raw Data of Reaction Results

The reaction results as a raw data of GC FID and TCD peak area and calculated data are shown below.

Table D1 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 660 °C, and 2 g of SiO₂

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	13352.9	0	9.8	315.5	0	2398.6	21.55	0	0.20	18.75	0	81.05
64	1.1	12641.3	0	9.8	683.6	13.5	2316.2	25.74	0	0.17	34.00	0.34	65.50
98	1.6	12912.3	0	9.2	816.1	19.6	2113.3	25.52	0	0.15	40.19	0.48	59.17
132	2.2	12341.9	0	9.9	791.8	18.1	2152.1	26.36	0	0.17	39.05	0.45	60.34
166	2.8	12123.2	0	11.9	889.5	27.2	2217.6	28.04	0	0.19	41.03	0.63	58.16
200	3.3	12120.9	q	11.7	930.4	29.5	2106.8	27.85	0	0.19	43.34	0.69	55.79
234	3.9	11968	0	12.5	986.8	31.8	2079.2	28.50	0	0.20	45.08	0.73	54.00
268	4.5	12667.5	0	9.7	991	30.3	1723	25.49	0	0.17	49.82	0.76	49.25
302	5.0	11898.2	0	11.5	1017.7	33.5	2020	28.60	0	0.18	46.54	0.77	52.52
336	5.6	12240.6	0	10.2	964.3	32.9	1897.8	26.86	0	0.17	46.74	0.80	52.30
370	6.2	12114.9	0	9.8	962.2	30	1807.6	26.54	0	0.17	47.91	0.75	51.17
404	6.7	12070.2	0	7.8	993	31.9	1700.1	26.31	0	0.13	50.20	0.81	48.86
438	7.3	12511.4	0	6.6	891.5	26.3	1733.5	24.78	0	0.12	47.11	0.69	52.08

Table D2 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 660 °C, and 2 g of Al₂O₃

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	12722	0	15.1	40.7	0	2226.8	18.38	0	0.39	3.10	0	96.50
64	1.1	12509.5	0	18.3	35.6	0	2667.6	21.39	0	0.40	2.28	0	97.31
98	1.6	12979.8	0	14.1	29.2	0	2488.9	19.59	0	0.33	2.02	0	97.65
132	2.2	12406.8	0	22	20.3	0	2713.5	21.67	0	0.48	1.29	0	98.23
166	2.8	12599.9	0	20.1	18	0	2700	21.29	0	0.44	1.15	0	98.41
200	3.3	12544.3	0	19	14	0	2830.3	22.10	0	0.40	0.86	0	98.74
234	3.9	12683.9	0	23.1	0	0	2994.5	22.76	0	0.46	0.00	0	99.54
268	4.5	13425.4	0	16.1	12.8	0	2685.9	20.08	0	0.36	0.83	0	98.82
302	5.0	13824.5	0	20.1	0	0	2621.9	19.14	0	0.46	0	0	99.54
336	5.6	13872.3	0	17.4	0	0	2606.3	18.98	0	0.40	0	0	99.60
370	6.2	13428	0	21.9	0	0	2776.4	20.52	0	0.47	0	0	99.53
404	6.7	13838.2	0	19.8	0	0	2761.7	19.93	0	0.43	0	0	99.57
438	7.3	13763	0	20.4	0	0	2693.1	19.63	0	0.45	0	0	99.55

Table D3 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 660 °C, and 2 g of ZSM-5

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	14386.4	0	19.3	0	0	1974.9	14.66	0	0.58	0	0	99.42
64	1.1	12348.1	0	33.8	18.4	0	3382.7	25.71	0	0.59	0.94	0	98.47
98	1.6	13795.2	0	32.2	20.5	0	2950.4	21.33	0	0.64	1.20	0	98.16
132	2.2	12625.8	0	40.6	17	0	3009.8	23.23	0	0.80	0.98	0	98.23
166	2.8	13556.6	0	33.3	22.4	0	2344.1	18.11	0	0.83	1.64	0	97.53
200	3.3	12695.6	0	50.3	20.8	0	2823.8	22.15	0	1.05	1.27	0	97.69
234	3.9	12772.8	0	56.3	21.7	0	2633.1	20.96	0	1.25	1.41	0	97.34
268	4.5	13157.8	0	53.8	20.4	0	2500	19.64	0	1.26	1.40	0	97.34
302	5.0	13557.1	0	55.8	23.5	0	2389.3	18.56	0	1.36	1.68	0	96.96
336	5.6	12495.9	0	62.2	24	0	2606.5	21.23	0	1.39	1.57	0	97.04
370	6.2	12385.9	0	60.8	24.2	0	2617.3	21.44	0	1.35	1.58	0	97.07
404	6.7	12512.8	0	60.9	27.4	0	2573.4	21.03	0	1.38	1.81	0	96.81
438	7.3	12802.3	0	61.3	28.7	0	2463.2	20.00	0	1.44	1.98	0	96.58

Table D4 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 660 °C, and 2 g of Activated carbon

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	12457.4	0	0	0	0	1827.3	15.36	0	0	0	0	100
64	1.1	11355.6	0	0	0	0	2190.6	19.26	0	0	0	0	100
98	1.6	11136.7	0	0	0	0	2268.3	20.12	0	0	0	0	100
132	2.2	10878.8	0	0	0	0	2186.7	19.91	0	0	0	0	100
166	2.8	11676.7	0	0	0	0	2029.4	17.69	0	0	0	0	100
200	3.3	11412.2	0	0	0	0	2089.6	18.46	0	0	0	0	100
234	3.9	11547.1	0	0	0	0	2004.9	17.68	0	0	0	0	100
268	4.5	10876.1	0	0	0	0	2439	21.71	0	0	0	0	100
302	5.0	10961.2	0	0	0	0	2278.5	20.45	0	0	0	0	100
336	5.6	10990.6	0	0	0	0	2286.4	20.46	0	0	0	0	100
370	6.2	10255.4	0	0	0	0	2417.1	22.57	0	0	0	0	100
404	6.7	10559.8	0	0	0	0	2357.3	21.64	0	0	0	0	100
438	7.3	10869.7	0	0	0	0	2277.3	20.58	0	0	0	0	100

Table D5 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	14407.6	0	0	49.6	0	0	0.74	0	0	100	0	0
64	1.1	13885.5	0	0	19	0	0	0.30	0	0	100	0	0
98	1.6	14061.1	0	0	13.3	0	0	0.21	0	0	100	0	0
132	2.2	14630.7	0	0	10.2	0	0	0.15	0	0	100	0	0
166	2.8	14026.6	0	0	12.7	0	0	0.20	0	0	100	0	0
200	3.3	13343.5	0	0	17.2	0	0	0.28	0	0	100	0	0
234	3.9	13498.8	0	0	10.3	0	0	0.17	0	0	100	0	0
268	4.5	13613.5	0	0	12.2	0	0	0.19	0	0	100	0	0
302	5.0	13349.7	0	0	11.6	0	0	0.19	0	0	100	0	0
336	5.6	13486.7	0	0	11.5	0	0	0.19	0	0	100	0	0
370	6.2	13032.5	0	0	14	0	0	0.23	0	0	100	0	0
404	6.7	13108.6	0	0	17.6	0	0	0.29	0	0	100	0	0
438	7.3	13166.3	0	0	13.9	0	0	0.23	0	0	100	0	0

Table D6 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 500 °C

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	14349.3	0	0	359.7	27.3	124.3	6.31	0	0	81.01	3.07	15.92
64	1.1	14214.5	0	0	269.3	22.8	88.6	4.82	0	0	81.34	3.44	15.21
98	1.6	14619.4	0	0	302.9	23.4	101.2	5.25	0	0	81.40	3.14	15.46
132	2.2	13774.4	0	0	346.6	31.3	135.9	6.49	0	0	78.86	3.56	17.58
166	2.8	13423.9	0	0	395.1	36	159.1	7.54	0	0	78.46	3.57	17.96
200	3.3	13223.2	0	0	410.8	42	177.5	8.06	0	0	77.11	3.94	18.94
234	3.9	13117.2	0	0	419.9	44.2	185.2	8.32	0	0	76.72	4.04	19.24
268	4.5	12820.1	0	0	380.2	38.7	158.3	7.67	0	0	77.66	3.95	18.38
302	5.0	13035.8	0	0	359.5	37.3	141.7	7.11	0	0	78.37	4.07	17.56
336	5.6	13355.4	0	0	346.5	33.3	134.2	6.68	0	0	78.85	3.79	17.36
370	6.2	13135	0	0	367.4	34.8	141.8	7.16	0	0	78.94	3.74	17.32
404	6.7	13073.8	0	0	383	39.1	148.4	7.49	0	0	78.66	4.02	17.33
438	7.3	13222.2	0	0	373.2	38.1	147	7.26	0	0	78.43	4.00	17.56

Table D7 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 600 °C

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	13706.7	0	11.2	1023.3	95.4	391.9	17.12	0	0.29	78.86	3.68	17.17
64	1.1	13409.2	0	7.7	938.9	87.1	372.2	16.29	0	0.22	78.46	3.64	17.68
98	1.6	13782.2	0	6.8	839.9	69.3	343.3	14.50	0	0.22	78.35	3.23	18.21
132	2.2	13618.2	0	0	847.1	70.4	336.3	14.64	0	0	78.91	3.28	17.81
166	2.8	13906	0	0	837.5	67.7	319.1	14.14	0	0	79.55	3.22	17.23
200	3.3	14116.1	0	0	786.5	60.7	296.5	13.18	0	0	79.81	3.08	17.11
234	3.9	14138.1	0	0	812.3	61.7	316.3	13.60	0	0	79.41	3.02	17.58
268	4.5	14311.7	0	0	785.7	58.9	285.2	12.93	0	0	80.40	3.01	16.59
302	5.0	13499.4	0	0	868.7	68.9	336.5	14.99	0	0	79.37	3.15	17.48
336	5.6	13625.1	0	0	848.1	66.2	354.4	14.74	0	0	78.33	3.06	18.61
370	6.2	13743.4	0	0	820.6	64.9	311.1	14.02	0	0	79.68	3.15	17.17
404	6.7	13773	0	0	798.7	62	297.4	13.63	0	0	79.97	3.10	16.93
438	7.3	12714.8	0	0	843.4	71.8	353.1	15.60	0	0	78.09	3.32	18.59

Table D8 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 660 °C

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	11808.5	0	38.8	1126.4	79.7	932.7	24.09	0	0.77	65.88	2.33	31.01
64	1.1	11710.8	0	30	1518.3	145.9	886	28.21	0	0.49	72.12	3.47	23.93
98	1.6	11834.5	0	30.5	1447	133.2	840.9	27.00	0	0.52	72.27	3.33	23.88
132	2.2	11760.8	0	29.4	1387	124.7	794.6	26.22	0	0.53	72.57	3.26	23.64
166	2.8	11779.8	0	26.4	1362.4	116.5	791.2	25.86	0	0.48	72.49	3.10	23.93
200	3.3	12135.3	0	35.2	1284.4	104.8	801.8	24.57	0	0.67	71.17	2.90	25.26
234	3.9	11844.6	0	33.7	1292.3	103.8	810.4	25.13	0	0.63	71.14	2.86	25.36
268	4.5	11872.4	0	30.7	1294.2	100.6	790.5	24.96	0	0.58	71.72	2.79	24.91
302	5.0	12019.5	0	35.5	1240.9	93.8	810.5	24.29	0	0.69	70.48	2.66	26.17
336	5.6	12012.6	0	34.2	1298.4	97.3	834.7	25.04	0	0.64	70.82	2.65	25.89
370	6.2	12213.1	0	33.4	1256.1	92.9	811.8	24.14	0	0.64	70.75	2.62	25.99
404	6.7	12340.8	0	45.4	1261.8	92.6	879.2	24.48	0	0.85	69.20	2.54	27.41
438	7.3	11796.8	0	40.9	1143.4	84.8	805.3	23.56	0	0.84	68.98	2.56	27.62

Table D9 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 700 °C

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	13577.5	16.1	217	1080.7	43.1	1085.7	23.23	0.31	4.11	59.90	1.19	34.21
64	1.1	12464.8	0	94.6	1361.9	84.6	1178.8	27.26	0	1.53	64.52	2.00	31.75
98	1.6	11991.3	0	78.1	1264.9	70.5	1067.6	26.30	0	1.38	65.30	1.82	31.34
132	2.2	11969.3	0	66.5	1368.2	71.8	1038.7	27.11	0	1.12	67.71	1.78	29.22
166	2.8	12360.9	0	76.8	1223.8	62.2	1016.8	24.97	0	1.41	65.72	1.67	31.05
200	3.3	12649	0	78.4	1224.5	57.6	1018.5	24.55	0	1.44	65.76	1.55	31.10
234	3.9	12081	0	69.2	1256	59.5	1046.8	25.83	0	1.24	65.84	1.56	31.20
268	4.5	12324.9	0	67.2	1297.4	59.2	980.8	25.47	0	1.20	67.91	1.55	29.19
302	5.0	11633.5	0	70.9	1221	62.8	1158.6	26.88	0	1.25	62.99	1.62	33.98
336	5.6	11799.8	0	72.2	1175.7	61.1	1080	25.69	0	1.33	63.63	1.65	33.23
370	6.2	11872.8	0	67.7	1244.5	61.1	1095.4	26.32	0	1.20	64.69	1.59	32.37
404	6.7	11953.7	0	62.7	1242.3	61.4	961	25.36	0	1.16	67.39	1.67	29.64
438	7.3	12138.7	0	70.1	1203.6	59	1029.9	25.12	0	1.30	65.22	1.60	31.73

Table D10 The results of the reaction with 20 ml/min of CH₄, 10 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C, and 2 g of 0.5 wt% Rh/SiO₂-calcined at 450 °C 6 h

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	14280	0	0	0	0	0	0	0	0	0	0	0
64	1.1	14373.5	0	0	0	0	0	0	0	0	0	0	0
98	1.6	14293.2	0	0	0	0	0	0	0	0	0	0	0
132	2.2	15425.9	0	0	0	0	0	0	0	0	0	0	0
166	2.8	14940.5	0	0	0	0	0	0	0	0	0	0	0
200	3.3	15099.5	0	0	0	0	0	0	0	0	0	0	0
234	3.9	14621	0	0	0	0	0	0	0	0	0	0	0
268	4.5	14783.2	0	0	0	0	0	0	0	0	0	0	0
302	5.0	15310.6	0	0	0	0	0	0	0	0	0	0	0
336	5.6	15593.7	0	0	0	0	0	0	0	0	0	0	0
370	6.2	15549.6	0	0	0	0	0	0	0	0	0	0	0

Table D11 The results of the reaction with 20 ml/min of CH₄, 3.5 ml/min of O₂, 7.5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C, and 2 g of 0.5 wt% Rh/SiO₂-calcined at 450 °C 6 h

TOS		FID					TCD	%CH ₄	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO	Conversion	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	12113.2	0	0	98.9	0	71.1	2.44	0	0	70.99	0	29.01
64	1.1	12357.6	0	0	113.4	0	45.3	2.39	0	0	81.49	0	18.51
98	1.6	11700.9	0	0	103.1	0	39.1	2.28	0	0	82.26	0	17.74
132	2.2	11889.5	0	0	109.8	0	45.2	2.42	0	0	81.03	0	18.97
166	2.8	11850.3	0	0	109.5	0	44.4	2.41	0	0	81.27	0	18.73
200	3.3	11935.8	0	0	107.7	0	47.9	2.40	0	0	79.82	0	20.18
234	3.9	11848.5	0	0	101.8	0	35.5	2.19	0	0	83.45	0	16.55
268	4.5	11979.8	0	0	108.6	0	44	2.37	0	0	81.28	0	18.72
302	5.0	12245.3	0	0	113	0	43.5	2.39	0	0	82.04	0	17.96
336	5.6	12440.2	0	0	123.1	0	44.9	2.53	0	0	82.82	0	17.18
370	6.2	11616.7	0	0	125.9	0	48.9	2.80	0	0	81.91	0	18.09

Table D12 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C , and 2 g of 0.5 wt% Rh/SiO₂-calcined at 450 °C 6 h

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	14919.2	0	0	249.6	0	80.3	4.13	0	0	84.54	0.00	15.46
64	1.1	14256.4	0	0	248.7	8.4	92.7	4.46	0	0	81.38	1.37	17.25
98	1.6	14624.3	0	0	277.8	12	72.4	4.61	0	0	85.49	1.85	12.67
132	2.2	14258.6	0	0	267.2	13	54.7	4.44	0	0	87.66	2.13	10.20
166	2.8	14157.2	0	0	266.1	13.2	52.7	4.44	0	0	87.92	2.18	9.90
200	3.3	14509	0	0	245.3	13.1	35.8	3.92	0	0	90.12	2.41	7.48
234	3.9	14593.9	0	0	256.6	13	38.3	4.07	0	0	90.07	2.28	7.64
268	4.5	14749.6	0	0	252.5	13.1	29.7	3.91	0	0	91.51	2.37	6.12
302	5.0	14777.2	0	0	246	12.9	22.1	3.75	0	0	92.83	2.43	4.74
336	5.6	13857.2	0	0	245.8	12.8	24.8	4.01	0	0	92.30	2.40	5.29
370	6.2	13999.4	0	0	245.6	12.4	20.6	3.93	0	0	93.20	2.35	4.44

Table D13 The results of the reaction with 20 ml/min of CH₄, 6 ml/min of O₂, 4 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C , and 2 g of 0.5 wt% Rh/SiO₂-calcined at 450 °C 6 h

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	11798	0	0	120.6	0	177.3	3.92	0	0	54.47	0	45.53
64	1.1	11445.2	0	0	151.9	0	106.6	3.88	0	0	71.48	0	28.52
98	1.6	11298.5	0	0	145.4	0	106.9	3.82	0	0	70.52	0	29.48
132	2.2	11487.1	0	0	153	0	119.3	4.01	0	0	69.29	0	30.71
166	2.8	11775.9	0	0	154.1	0	103.9	3.79	0	0	72.29	0	27.71
200	3.3	12131.8	0	0	158.5	0	96.6	3.69	0	0	74.27	0	25.73
234	3.9	12220.4	0	0	167.9	0	111.1	3.95	0	0	72.66	0	27.34
268	4.5	11701.5	0	0	167.7	0	112.1	4.12	0	0	72.46	0	27.54
302	5.0	11727.8	0	0	168.6	0	105.8	4.07	0	0	73.70	0	26.30
336	5.6	11376.2	0	0	168.1	0	109.9	4.22	0	0	72.90	0	27.10
370	6.2	11590.7	0	0	175.6	6.5	107.8	4.31	0	0	73.12	1.35	25.52

Table D14 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C , and 2 g of 0.3 wt% Rh/SiO₂-calcined at 450 °C 6 h

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	11844.5	0	0	122.7	27.4	81.1	3.24	0	0	67.23	7.51	25.26
64	1.1	13557.6	0	0	148.9	0	131.3	3.46	0	0	66.61	0	33.39
98	1.6	11609	0	0	126.2	0	59.5	2.91	0	0	78.86	0	21.14
132	2.2	12479	0	0	129.7	0	50.8	2.69	0	0	81.79	0	18.21
166	2.8	12276.5	0	0	115.7	0	34.4	2.34	0	0	85.54	0	14.46
200	3.3	13041.5	0	0	122.5	0	0	2.00	0	0	100	0	0
234	3.9	12864.1	0	0	119.6	0	0	1.98	0	0	100	0	0
268	4.5	12132.3	0	0	100.5	0	0	1.77	0	0	100	0	0
302	5.0	12140.6	0	0	102.5	0	0	1.80	0	0	100	0	0
336	5.6	11711	0	0	99.6	0	0	1.82	0	0	100	0	0
370	6.2	11592	0	0	103.9	0	0	1.91	0	0	100	0	0

Table D15 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C , and 2 g of 0.3 wt% Rh/SiO₂-calcined at 900 °C 10 h

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	12510.7	0	0	38.5	0	0	0.66	0	0	100	0	0
64	1.1	12193.2	0	0	53.7	0	0	0.95	0	0	100	0	0
98	1.6	11643.9	0	0	43.6	0	0	0.81	0	0	100	0	0
132	2.2	11884	0	0	35.7	0	0	0.65	0	0	100	0	0
166	2.8	11605.1	0	0	44.5	0	0	0.83	0	0	100	0	0
200	3.3	11827.1	0	0	40.5	0	0	0.74	0	0	100	0	0
234	3.9	12069	0	0	43.3	0	0	0.77	0	0	100	0	0
268	4.5	11584.8	0	0	49.7	0	0	0.92	0	0	100	0	0
302	5.0	11840.5	0	0	42.4	0	0	0.77	0	0	100	0	0
336	5.6	12069	0	0	51.9	0	0	0.93	0	0	100	0	0
370	6.2	11863	0	0	48.9	0	0	0.89	0	0	100	0	0

Table D16 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C , and 2 g of 0.5 wt% Rh/SiO₂-calcined at 900 °C 10 h

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	12697.1	0	0	35	0	50.7	1.08	0	0	54.84	0	45.16
64	1.1	12016.6	0	0	44.3	0	0	0.80	0	0	100	0	0
98	1.6	11789.8	0	0	35.9	0	0	0.66	0	0	100	0	0
132	2.2	11564.5	0	0	38.7	0	0	0.72	0	0	100	0	0
166	2.8	11757.2	0	0	41	0	0	0.75	0	0	100	0	0
200	3.3	11958.6	0	0	50.2	0	0	0.90	0	0	100	0	0
234	3.9	12146	0	0	45.3	0	0	0.80	0	0	100	0	0
268	4.5	12367.4	0	0	53	0	0	0.92	0	0	100	0	0
302	5.0	12521.7	0	0	45.8	0	0	0.79	0	0	100	0	0
336	5.6	12569.5	0	0	49.1	0	0	0.84	0	0	100	0	0
370	6.2	12024.1	0	0	57.8	0	0	1.03	0	0	100	0	0

Table D17 The results of the reaction with 20 ml/min of CH₄, 5 ml/min of O₂, 5 ml/min of N₂, 6.5 ml/h of 48 wt% HBr/H₂O, reaction temperature 400 °C , and 2 g of SiO₂

TOS		FID					TCD	%CH ₄ Conversion	% Selectivity				
min	h	CH ₄	C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO		C ₂ H ₆	C ₂ H ₄	CH ₃ Br	CH ₂ Br ₂	CO
30	0.5	15083.4	0	0	194.7	7.8	25.6	2.98	0	0	91.34	1.83	6.83
64	1.1	14822.8	0	0	161.7	7	41.1	2.69	0	0	85.75	1.86	12.39
98	1.6	14377.4	0	0	126.4	0	0	1.88	0	0	100	0	0
132	2.2	13460.6	0	0	102.3	0	0	1.63	0	0	100	0	0
166	2.8	13803.8	0	0	99.5	0	0	1.54	0	0	100	0	0
200	3.3	13922.4	0	0	103.4	0	0	1.59	0	0	100	0	0
234	3.9	14279	0	0	94.9	0	0	1.43	0	0	100	0	0
268	4.5	13355.4	0	0	93	0	0	1.49	0	0	100	0	0
302	5.0	13660.7	0	0	97	0	0	1.52	0	0	100	0	0
336	5.6	13766.7	0	0	93.2	0	0	1.45	0	0	100	0	0
370	6.2	13844.6	0	0	94.7	0	0	1.47	0	0	100	0	0

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