

## REFERENCES

- Bampenrat, A., Meeyoo, V., Kitiyanan, B., Rangsuvigit, P., and Rirksomboon, T. (2010) Naphthalene steam reforming over Mn-doped CeO<sub>2</sub>-ZrO<sub>2</sub> supported nickel catalysts. Applied Catalysis A: General, 373(1-2), 154-159.
- Bayrakdar, E., Gürkaynak Altınçekiç, T., and Öksüzömer, M.A.F. (2013) Effects of PVP on the preparation of nanosized Al<sub>2</sub>O<sub>3</sub> supported Ni catalysts by polyol method for catalytic partial oxidation of methane. Fuel Processing Technology, 110, 167-175.
- Biao, L., Jian-guo, G., Qi, W., and Qing-jie, Z. (2005) Preparation of Nanometer Cobalt Particles by Polyol Reduction Process and Mechanism Research. Materials Transactions, 46(8), 1865-1867.
- Carroll, K.J., Reveles, J.U., Shultz, M.D., Khanna, S.N., and Carpenter, E.E. (2011) Preparation of Elemental Cu and Ni Nanoparticles by the Polyol Method: An Experimental and Theoretical Approach. The Journal of Physical Chemistry C, 115(6), 2656-2664.
- Chen, J., Wu, Q., Zhang, J., and Zhang, J. (2008) Effect of preparation methods on structure and performance of Ni/Ce<sub>0.75</sub>Zr<sub>0.25</sub>O<sub>2</sub> catalysts for CH<sub>4</sub>-CO<sub>2</sub> reforming. Fuel, 87, 2901-2907
- Dong, W., Roh, H., Jun, H., Park, S., and Oh, Y. (2002) Methane reforming over Ni/Ce-ZrO<sub>2</sub> catalysts: effect of nickel content. Applied Catalysis A: General, 226, 63-72.
- Elmasides, C. (2001) Mechanistic Study of Partial Oxidation of Methane to Synthesis Gas over Modified Ru/TiO<sub>2</sub> Catalyst. Journal of Catalysis, 203(2), 477-486.
- Enger BC, Lødeng R., and Holmen A. (2008) A review of catalytic partial oxidation of methane to synthesis gas with emphasis on reaction mechanisms over transition metal catalysts. Applied Catalysis A: General, 346, 1-27.
- Eriksson, S., Rojas, S., Boutonnet, M., and Fierro, J.L.G. (2007) Effect of Ce-doping on Rh/ZrO<sub>2</sub> catalysts for partial oxidation of methane. Applied Catalysis A: General, 326(1), 8-16.

- García, V., Fernández, J.J., Ruiz, W., Mondragón, F., and Moreno, A. (2009) Effect of MgO addition on the basicity of Ni/ZrO<sub>2</sub> and on its catalytic activity in carbon dioxide reforming of methane, *Catalysis Communications*, 11, 240–246.
- Haag, S., Burgard, M., and Ernst, B. (2007) Beneficial effects of the use of a nickel membrane reactor for the dry reforming of methane: comparison with thermodynamic predictions. *Journal of Catalysis* 252 (2007) 190–204.
- Hao, Z., Zhu, Q., Jiang, Z., Hou, B., and Li, H. (2009) Characterization of aerogel Ni/Al<sub>2</sub>O<sub>3</sub> catalysts and investigation on their stability for CH<sub>4</sub>–CO<sub>2</sub> reforming in a fluidized bed, *Fuel Processing Technology*, 90, 113–121.
- Hegarty, M., Connor, A., and Ross, J. (1998) Syngas production from natural gas using ZrO<sub>2</sub>-supported metals. *Catalysis Today*, 42, 225–232.
- Horn, R., Williams, K.A., Degnstein, N.J., and Schmidt, L.D. (2006) Syngas by catalytic partial oxidation of methane on rhodium: mechanistic conclusions from spatially resolved measurements and numerical simulations. *Journal of Catalysis*, 242(1), 92-102.
- Hu, R.-r., Yan, C.-f., Zheng, X.-x., Liu, H., and Zhou, Z.-y. (2013) Carbon deposition on Ni/ZrO<sub>2</sub>–CeO<sub>2</sub> catalyst during steam reforming of acetic acid. *International Journal of Hydrogen Energy*, 38(14), 6033–6038.
- Ikeda, T., Xiong, A., Yoshinaga, T., Maeda, K., Domen, K., and Teranishi, T. (2013) Polyol Synthesis of Size-Controlled Rh Nanoparticles and Their Application to Photocatalytic Overall Water Splitting under Visible Light. *The Journal of Physical Chemistry*, 117, 2467-2473.
- Larimi, A.S. and Alavi, S.M. (2012) Ceria-Zirconia supported Ni catalysts for partial oxidation of methane to synthesis gas. *Fuel*, 102, 366-371.
- Li, B., Li, H., Weng, W.-Z., Zhang, Q., Huang, C.-J., and Wan, H.-L. (2013) Synthesis gas production from partial oxidation of methane over highly dispersed Pd/SiO<sub>2</sub> catalyst. *Fuel*, 103, 1032-1038.
- Li, X., Zhang, Y., and Smith, K.J. (2004) Metal–support interaction effects on the growth of filamentous carbon over Co/SiO<sub>2</sub> catalysts. *Applied Catalysis*, 264, 81–91.

- Li. Z., Hu. X., Zhang. L., Liu., S. and Lu. G. (2012). Steam reforming of acetic acid over Ni/ZrO<sub>2</sub> catalysts: Effect of nickel loading and particle size on product distribution and coke formation. Applied Catalysis A: General, 417-418, 281-289.
- Meshkani, F., Rezaei, M., and Andache, M. (2013) Investigation of the catalytic performance of Ni/MgO catalysts in partial oxidation, dry reforming and combined reforming of methane. Journal of Industrial and Engineering Chemistry.
- Montoya, J.A., Romero-Pascual, E., Gimon, C., Del Angle, P., and Monzon, A. (2000) Methane Reforming with CO<sub>2</sub> over Ni/ZrO<sub>2</sub>-CeO<sub>2</sub> Catalysts Prepared by Sol-Gel. Catalyst Today, 63, 71.
- Naeem, M.A., Al-Fatesh, A.S., Abasaeed, A.E., and Fakieha, A.H. (2014) Activities of Ni-based nano catalysts for CO<sub>2</sub>-CH<sub>4</sub> reforming prepared by polyol process. Fuel Processing Technology, 122, 141-152.
- Nematollahi, B., Rezaei, M., and Khajenoori, M. (2011) Combined dry reforming and partial oxidation of methane to synthesis gas on noble metal catalysts. International Journal of Hydrogen Energy, 36(4), 2969-2978.
- Ningthoujam, R., Gajbhiye, N., and Sharma, S. (2009) Reduction mechanism of Ni<sup>2+</sup> into Ni nanoparticles prepared from different precursors: Magnetic studies. Indian Academy of Sciences, 72(3), 577-586.
- Oh, H.-S., Oh, J.-G., Hong, Y.-G., and Kim, H. (2007) Investigation of carbon-supported Pt nanocatalyst preparation by the polyol process for fuel cell applications. Electrochimica Acta, 52(25), 7278-7285.
- Ozdemir, H., Oksuzomer, M.A., and Gurkaynak, M.A. (2010) Preparation and characterization of Ni based catalysts for the catalytic partial oxidation of methane: effect of support basicity on H<sub>2</sub>/CO ratio and carbon deposition. International Journal of Hydrogen Energy, 35, 12147-12160.
- Passos, F.B., de Oliveira, E.R., Mattos, L.V., and Noronha, F.B. (2005) Partial oxidation of methane to synthesis gas on Pt/Ce<sub>x</sub>Zr<sub>1-x</sub>O<sub>2</sub> catalysts: the effect of the support reducibility and of the metal dispersion on the stability of the catalysts. Catalysis Today, 101(1), 23-30.

- Pengpanich, S., Meeyoo, V., and Rirksomboon, T. (2004) Methane partial oxidation over Ni/CeO<sub>2</sub>-ZrO<sub>2</sub> mixed oxide solid solution catalysts. *Catalysis Today*, 93-95, 95-105.
- Pengpanich, S., Meeyoo, V., Rirksomboon, T., and Bunyakiat, K. (2002) Catalytic oxidation of methane over CeO<sub>2</sub>-ZrO<sub>2</sub> mixed oxide solid solution catalysts prepared via urea hydrolysis. *Applied Catalysis A: General*, 234, 221-233.
- Pengpanich, S., Meeyoo, V., Rirksomboon, T., and Schwank, J. (2006) Hydrogen production from partial oxidation of iso-octane over Ni/Ce0.75Zr0.25O2 and Ni/ $\beta$ -Al<sub>2</sub>O<sub>3</sub> catalysts. *Applied Catalysis A: General*, 302(1), 133-139.
- Pengpanich, S., Meeyoo, V., Rirksomboon, T., and Schwank, J. (2007) The Effect of Nb Loading on Catalytic Properties of Ni/Ce0.75Zr0.25O2 Catalyst for Methane Partial Oxidation. *Journal of Natural Gas Chemistry*, 16(3), 227-234.
- Pengpanich, S., Meeyoo, V., Rirksomboon, T., and Schwank, J. (2008) iso-Octane partial oxidation over Ni-Sn/Ce0.75Zr0.25O2 catalysts. *Catalysis Today*, 136(3-4), 214-221.
- Requies, J., Cabrero, M.A., Barrio, V.L., Güemez, M.B., Cambra, J.F., Arias, P.L., Pérez-Alonso, F.J., Ojeda, M., Peña, M.A., and Fierro, J.L.G. (2005) Partial oxidation of methane to syngas over Ni/MgO and Ni/La<sub>2</sub>O<sub>3</sub> catalysts. *Applied Catalysis A: General*, 289(2), 214-223.
- Ross, J., Keulen, A., Hegarty, M., and Seshan, K. (1996) The catalytic conversion of natural gas to useful products. *Catalysis Today*, 30, 193-199.
- Ruckenstein, E. and Wang, H.Y. (2000) Partial oxidation of methane to synthesis gas over MgO-supported Rh catalysts: the effect of precursor of MgO. *Applied Catalysis A: General*, 198, 33-41.
- Roh, H., Jun, K., Dong, W., and Chang, J. (2002) Highly active and stable Ni/Ce-ZrO<sub>2</sub> catalyst for H<sub>2</sub> production from methane. *Journal of Molecular Catalysis A: Chemical*, 181, 137-142.
- Sarkar, B., Tiwari, R., Singha, R.K., Suman, S., Ghosh, S., Acharyya, S.S., Mantri, K., Konathala, L.N.S., Pendem, C., and Bal, R. (2012) Reforming of methane with CO<sub>2</sub> over Ni nanoparticle supported on mesoporous ZSM-5. *Catalysis Today*, 198(1), 209-214.

- Shishido, T., Sokenobu, M., Morioka, H., Konda, M., Wang, Y., Takaki, K., and Takehira, K., (2002) Partial Oxidation of Methane over Ni/Mg-Al Oxide Catalysts Prepared by Solid Phase Crystallization Method from Mg-Al Hydrotalcite-like Precursors. *Applied Catalysis A: General*, 223, 235.
- Silva, F.D.A., Ruiz, J.A.C., de Souza, K.R., Bueno, J.M.C., Mattos, L.V., Noronha, F. B., and Hori, C. E. (2009) Partial oxidation of methane on Pt catalysts: Effect of the presence of ceria-zirconia mixed oxide and of metal content. *Applied Catalysis A: General*, 364(1-2), 122-129.
- Tanaka, H., Kaino, R., Nakagawa, Y., and Tomishige, K. (2010) Comparative study of Rh/MgO modified with Fe, Co or Ni for the catalytic partial oxidation of methane at short contact time. Part II: Catalytic performance and bed temperature profile. *Applied Catalysis A: General*, 378(2), 187-194.
- Tanaka, H., Kaino, R., Okumura, K., Kizuka, T., Nakagawa, Y., and Tomishige, K. (2010) Comparative study of Rh/MgO modified with Fe, Co or Ni for the catalytic partial oxidation of methane at short contact time. Part I: Characterization of catalysts. *Applied Catalysis A: General*, 378(2), 175-186.
- Tang, W., Hu, Z., Wang, M., Stucky, G.D., Metiu, H., and McFarland, E.W. (2010) Methane complete and partial oxidation catalyzed by Pt-doped CeO<sub>2</sub>. *Journal of Catalysis*, 273(2), 125-137.
- Thaicharoensutcharittham, S., Meeyoo, V., Kitiyanan, B., Rangsuvigit, P., and Rirksomboon, T. (2009) Catalytic combustion of methane over NiO/Ce<sub>0.75</sub>Zr<sub>0.25</sub>O<sub>2</sub> catalyst. *Catalysis Communications*, 10(5), 673-677.
- Thaicharoensutcharittham, S., Meeyoo, V., Kitiyanan, B., Rangsuvigit, P., and Rirksomboon, T. (2011) Hydrogen production by steam reforming of acetic acid over Ni-based catalysts. *Catalysis Today*, 164(1), 257-261.
- Thammachart, M., Meeyoo, V., Rirksomboon, T., and Osuwan S. (2001) Catalytic activity of CeO<sub>2</sub>-ZrO<sub>2</sub> mixed oxide catalysts prepared via sol-gel technique: CO oxidation. *Catalysis Today*, 68, 53-61.

- Tientong, J., Garcia, S., Thurber, C.R., and Golden, T.D. (2014) Synthesis of Nickel and Nickel Hydroxide Nanopowders by Simplified Chemical Reduction. Journal of Nanotechnology, 2014, 1-6.
- Trimm, D.L. (1999) Catalysts for the control of coking during steam reforming. Catalysis Today, 49(1-3), 3-10.
- Utaka, T., Al-Drees, S. A., Ueda, J., Iwasa, Y., Takeguchi, T., Kikuchi, R., and Eguchi, K. (2003) Partial oxidation of methane over Ni catalysts based on hexaaluminate- or perovskite-type oxides. Applied Catalysis A: General, 247(1), 125-131.
- Vella, L.D. and Specchia, S. (2011) Alumina-supported nickel catalysts for catalytic partial oxidation of methane in short-contact time reactors. Catalysis Today, 176(1), 340-346.
- Vij, K. (2013) Chapter I Nickel Nanoparticles-synthesis, characterization and catalytic applications in organic synthesis : A review. Shodhganga 2013. 7 May 2014. <[http://shodhganga.inflibnet.ac.in/bitstream/10603/6499/6/06\\_chapter%201.pdf](http://shodhganga.inflibnet.ac.in/bitstream/10603/6499/6/06_chapter%201.pdf)>
- Xia, W.-S., Hou, Y.-H., Chang, G., Weng, W.-Z., Han, G.-B., and Wan, H.-L. (2012) Partial oxidation of methane into syngas ( $H_2 + CO$ ) over effective high-dispersed  $Ni/SiO_2$  catalysts synthesized by a sol-gel method. International Journal of Hydrogen Energy, 37(10), 8343-8353.
- Yu, C.-l., Zhou, X.-c., Weng, W.-z., Hu, J.-b., Xi-rong, C., and Wei, L.-f. (2012) Effects of alkaline-earth strontium on the performance of  $Co/Al_2O_3$  catalyst for methane partial oxidation. Journal of Fuel Chemistry and Technology, 40(10), 1222-1229.
- Zhu, T. and Flytzani-Stephanopoulos, M. (2001) Catalytic Partial Oxidation of Methane to Synthesis gas over  $Ni-CeO_2$ . Applied Catalysis A: General, 208, 403.

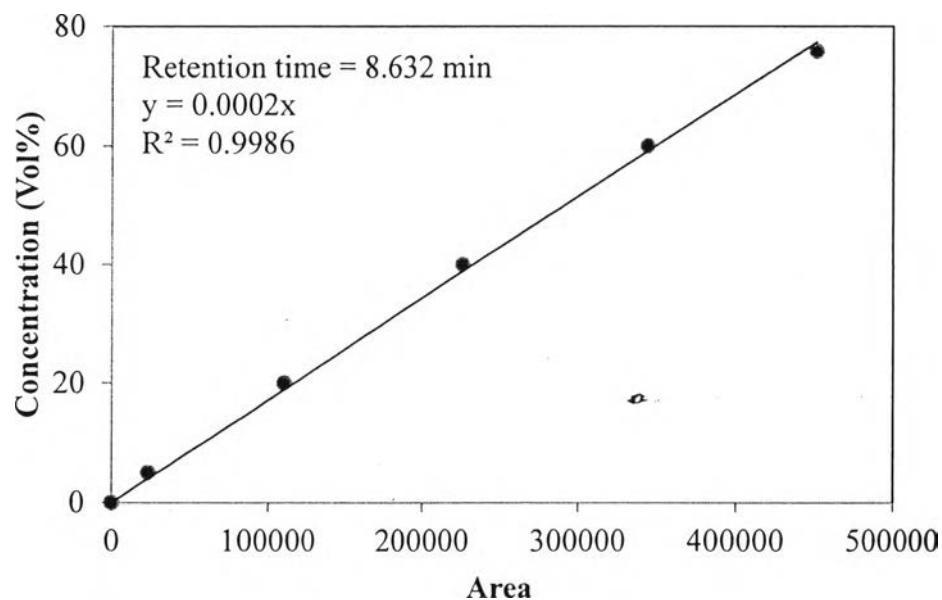
## APPENDICES

### Appendix A Experimental Data of Gas Calibration of GC-8A

Condition: Detector Current 80 mA

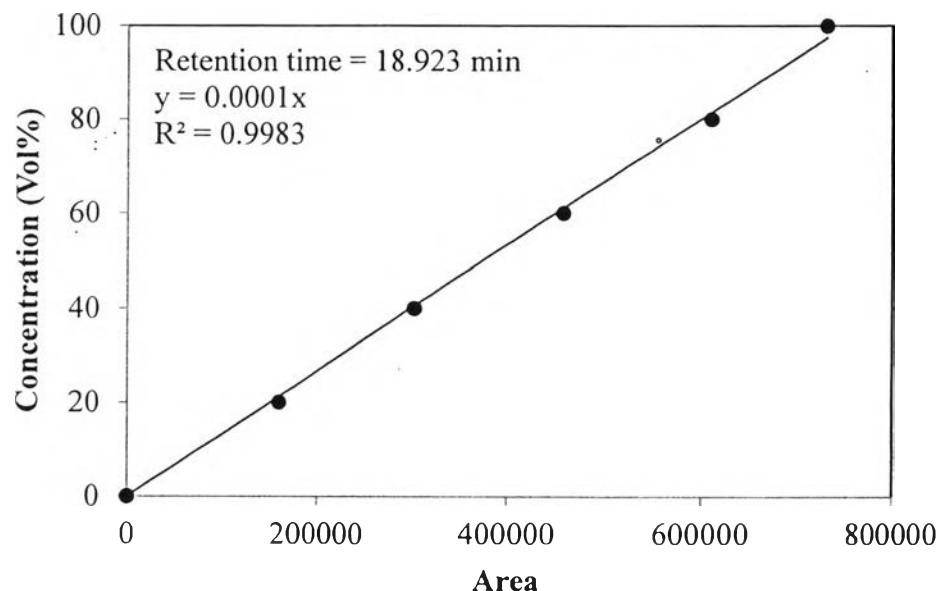
Temperature	°C	Column	Model
Column	50	P/W HAYESEP DB	HP model 5890 SERIES 2
Detector	100		
TCD-T	100		
Pressure	kPa		
Carrier Pressure (Primary)	600		
Carrier Pressure (1)	550		
TCD-Ref	10		

#### 1. Methane



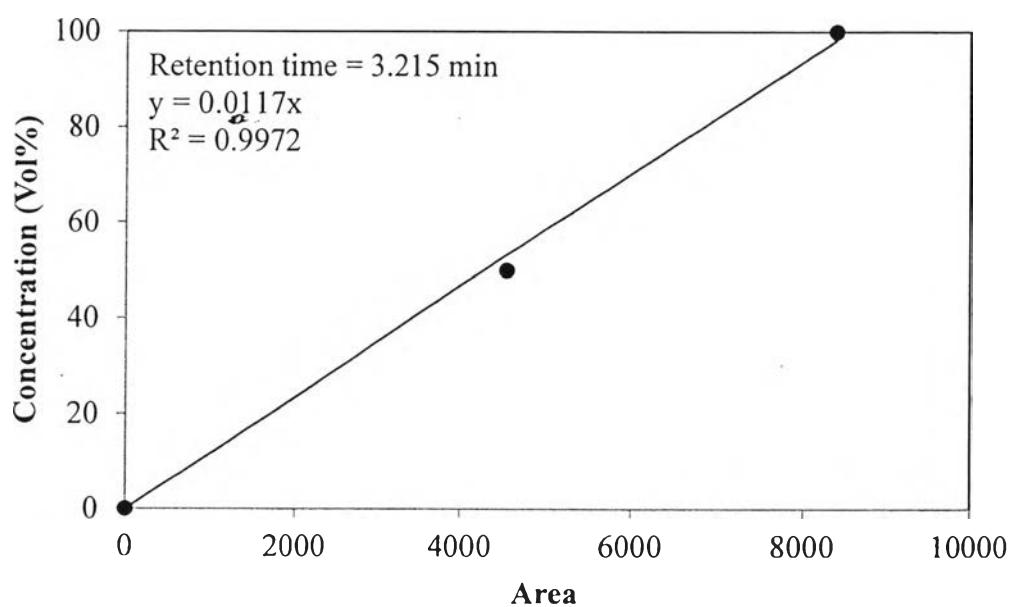
**Figure A1** Relationship between area and concentration of methane.

## 2. Carbon dioxide



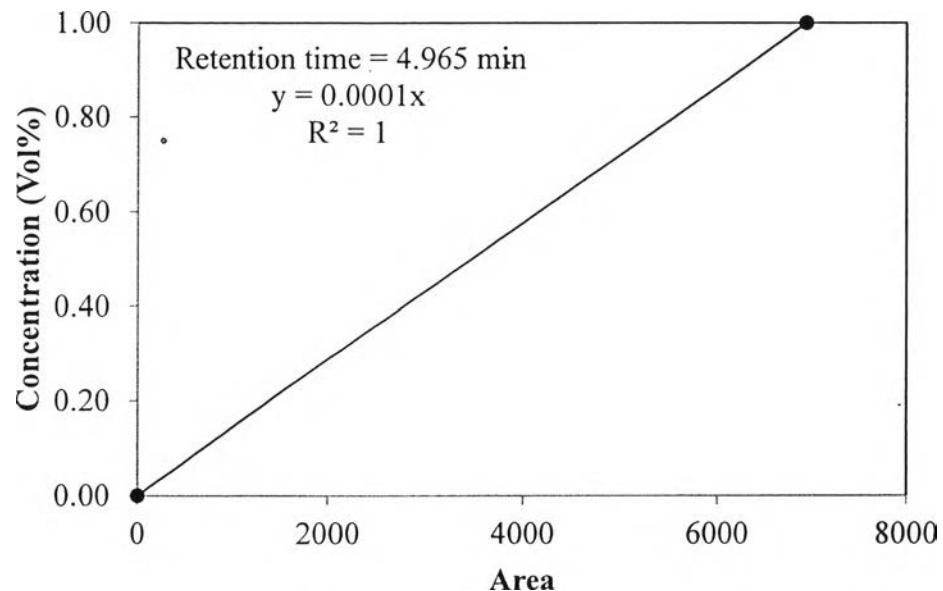
**Figure A2** Relationship between area and concentration of carbon dioxide.

## 3. Hydrogen



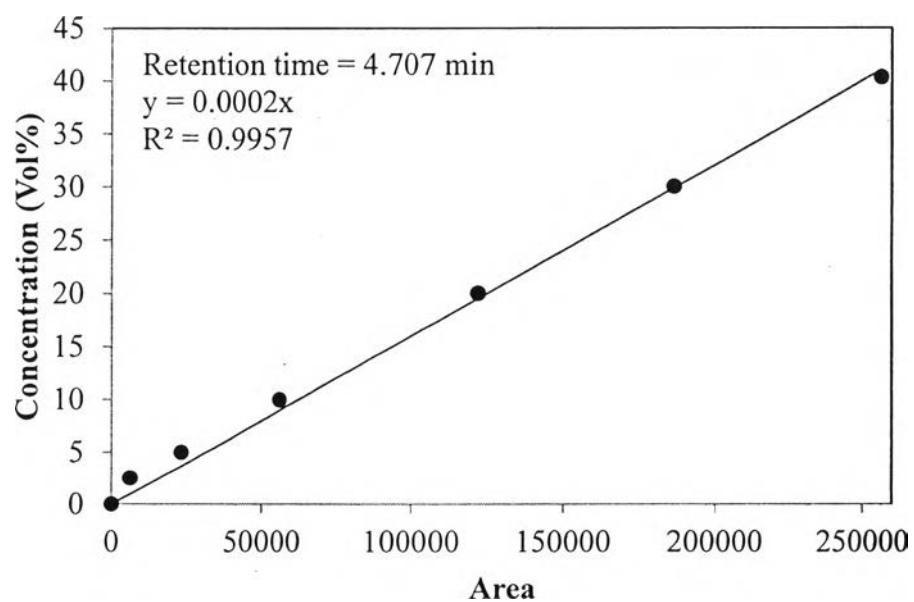
**Figure A3** Relationship between area and concentration of hydrogen.

#### 4. Carbon monoxide



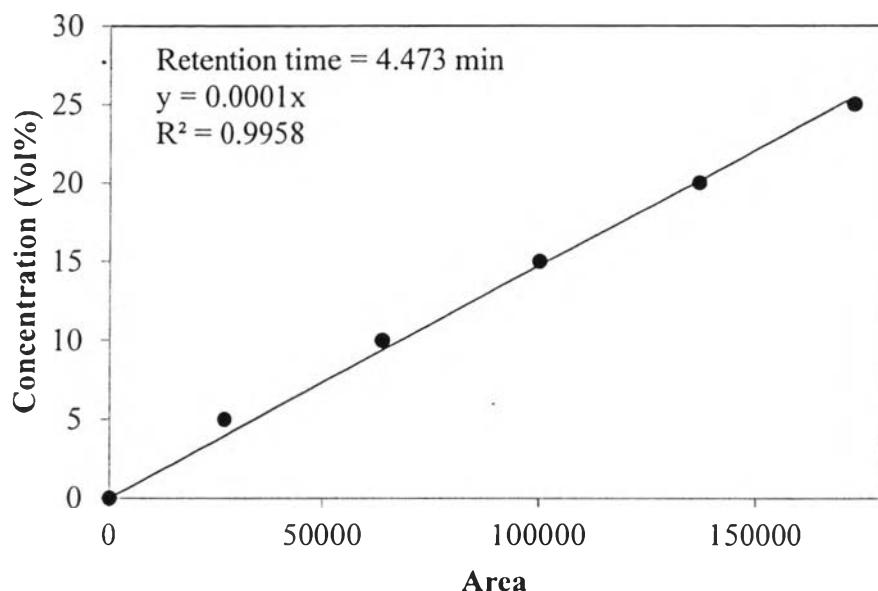
**Figure A4** Relationship between area and concentration of carbon monoxide.

#### 5. Oxygen



**Figure A5** Relationship between area and concentration of oxygen.

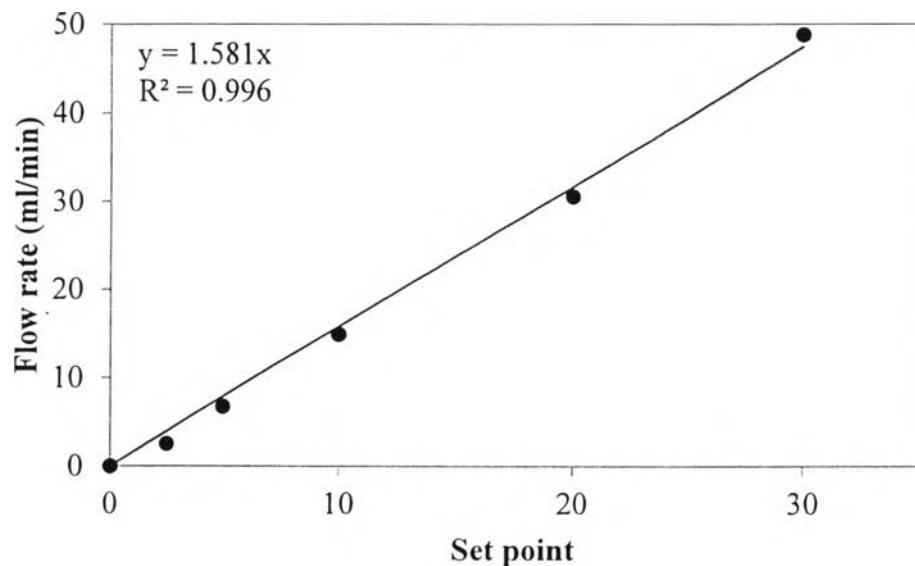
## 6. Nitrogen



**Figure A6** Relationship between area and concentration of nitrogen.

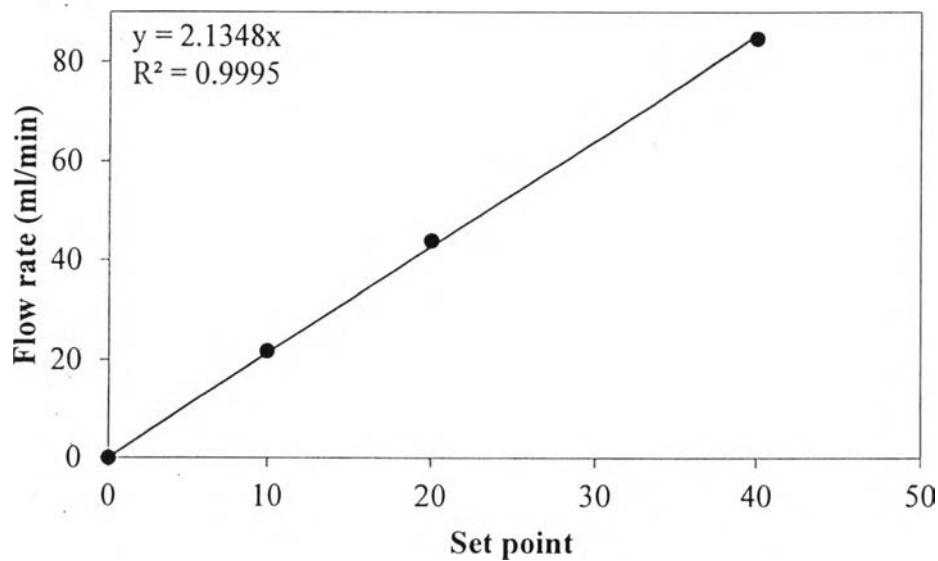
## Appendix B Calibration of Brooks 5850E Mass Flow Controllers

### 1. Methane



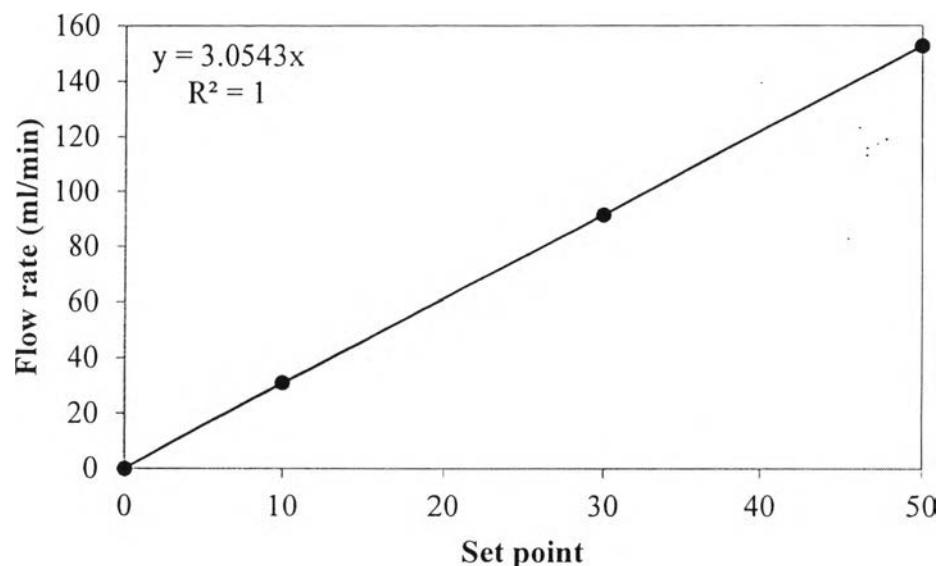
**Figure B1** Relationship between SP and flow rate of methane.

### 2. Air



**Figure B2** Relationship between SP and flow rate of air.

### 3. Helium



**Figure B3** Relationship between SP and flow rate of helium.

### Appendix C Experimental Data of Catalytic Activity Tests for MPO

**Table C1** Catalytic activity test of 5Ni/CZOi catalyst ( $\text{CH}_4/\text{O}_2$  ratio = 2:1)

Temperature (°C)	X <sub>CH<sub>4</sub></sub> (%)	X <sub>O<sub>2</sub></sub> (%)	S <sub>H<sub>2</sub></sub> (%)	S <sub>CO</sub> (%)	Y <sub>H<sub>2</sub></sub> (%)	Y <sub>CO</sub> (%)
401	0.15	1.64	0.00	0.00	0.00	0.00
452	1.21	9.05	0.00	0.00	0.00	0.00
503	6.01	37.26	0.00	0.00	0.00	0.00
559	9.32	52.41	0.00	0.00	0.00	0.00
602	63.77	100.00	87.24	77.67	27.54	35.02
658	64.36	100.00	88.10	82.17	29.05	38.25
705	65.74	100.00	89.50	85.03	31.97	41.95
753	66.79	100.00	90.36	88.48	34.08	43.03
809	67.75	100.00	91.59	90.43	37.10	45.79

**Table C2** Catalytic activity test of 5Ni/CZOp catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

Temperature (°C)	X <sub>CH<sub>4</sub></sub> (%)	X <sub>O<sub>2</sub></sub> (%)	S <sub>H<sub>2</sub></sub> (%)	S <sub>CO</sub> (%)	Y <sub>H<sub>2</sub></sub> (%)	Y <sub>CO</sub> (%)
400	2.77	2.42	0.00	0.00	0.00	0.00
450	3.99	10.62	0.00	0.00	0.00	0.00
501	10.72	40.40	0.00	0.00	0.00	0.00
553	17.18	53.19	0.00	0.00	0.00	0.00
603	67.02	100.00	89.94	83.43	33.40	46.01
653	68.55	100.00	89.78	86.09	33.85	46.58
705	69.40	100.00	90.81	88.22	36.32	46.95
753	70.46	100.00	91.29	90.79	37.90	47.02
808	71.41	100.00	91.57	91.33	39.06	47.45

**Table C3** Catalytic activity test of 10Ni/CZOi catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

Temperature (°C)	X <sub>CH<sub>4</sub></sub> (%)	X <sub>O<sub>2</sub></sub> (%)	S <sub>H<sub>2</sub></sub> (%)	S <sub>CO</sub> (%)	Y <sub>H<sub>2</sub></sub> (%)	Y <sub>CO</sub> (%)
402	0.31	2.66	0.00	0.00	0.00	0.00
453	2.26	9.91	0.00	0.00	0.00	0.00
504	8.10	38.04	0.00	0.00	0.00	0.00
557	11.00	52.95	0.00	0.00	0.00	0.00
603	65.24	100.00	87.03	76.29	27.87	39.73
657	66.49	100.00	88.64	79.03	30.88	41.18
708	67.83	100.00	90.04	83.69	33.99	43.27
757	68.45	100.00	91.37	85.14	36.99	45.02
811	68.78	100.00	92.76	88.04	40.39	46.79

**Table C4** Catalytic activity test of 10Ni/CZOp catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

Temperature (°C)	X <sub>CH<sub>4</sub></sub> (%)	X <sub>O<sub>2</sub></sub> (%)	S <sub>H<sub>2</sub></sub> (%)	S <sub>CO</sub> (%)	Y <sub>H<sub>2</sub></sub> (%)	Y <sub>CO</sub> (%)
400	5.39	3.21	0.00	0.00	0.00	0.00
452	6.24	10.23	0.00	0.00	0.00	0.00
503	12.60	41.97	0.00	0.00	0.00	0.00
554	18.65	53.98	0.00	0.00	0.00	0.00
604	68.33	100.00	89.59	82.66	33.40	46.01
654	69.07	100.00	89.97	86.31	34.48	48.24
706	70.45	100.00	91.16	89.30	37.62	48.94
756	72.55	100.00	91.48	91.28	39.48	50.09
809	72.83	100.00	92.03	91.60	40.94	50.45

**Table C5** Catalytic activity test of 15Ni/CZOi catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

<b>Temperature (°C)</b>	<b>X<sub>CH<sub>4</sub></sub> (%)</b>	<b>X<sub>O<sub>2</sub></sub> (%)</b>	<b>S<sub>H<sub>2</sub></sub> (%)</b>	<b>S<sub>CO</sub> (%)</b>	<b>Y<sub>H<sub>2</sub></sub> (%)</b>	<b>Y<sub>CO</sub> (%)</b>
402	.6600	5.63	0.00	0.00	0.00	0.00
455	4.64	18.05	0.00	0.00	0.00	0.00
507	12.51	46.96	0.00	0.00	0.00	0.00
560	18.91	76.17	0.00	0.00	0.00	0.00
608	77.58	100.00	90.34	72.76	39.54	47.93
659	80.75	100.00	91.25	77.32	43.34	53.12
709	84.72	100.00	90.94	78.95	44.67	55.47
760	85.77	100.00	91.64	81.99	47.10	59.17
812	85.83	100.00	90.77	83.60	44.82	57.29

**Table C6** Catalytic activity test of 15Ni/CZOp catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

Temperature (°C)	X <sub>CH<sub>4</sub></sub> (%)	X <sub>O<sub>2</sub></sub> (%)	S <sub>H<sub>2</sub></sub> (%)	S <sub>CO</sub> (%)	Y <sub>H<sub>2</sub></sub> (%)	Y <sub>CO</sub> (%)
401	7.53	14.64	0.00	0.00	0.00	0.00
453	8.90	23.20	0.00	0.00	0.00	0.00
504	16.39	48.79	0.00	0.00	0.00	0.00
555	23.13	82.41	0.00	0.00	0.00	0.00
604	80.40	100.00	90.37	75.60	41.05	53.33
655	83.20	100.00	91.41	79.27	45.07	57.62
705	87.03	100.00	92.15	82.16	49.27	57.95
757	87.88	100.00	92.05	85.80	49.44	62.41
809	89.16	100.00	92.00	86.18	50.01	63.22

**Table C7** Catalytic activity test of 25Ni/CZOi catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

Temperature (°C)	X <sub>CH<sub>4</sub></sub> (%)	X <sub>O<sub>2</sub></sub> (%)	S <sub>H<sub>2</sub></sub> (%)	S <sub>CO</sub> (%)	Y <sub>H<sub>2</sub></sub> (%)	Y <sub>CO</sub> (%)
405	7.53	4.86	0.00	0.00	0.00	0.00
457	8.90	14.14	0.00	0.00	0.00	0.00
509	16.39	43.29	0.00	0.00	0.00	0.00
562	23.13	64.62	0.00	0.00	0.00	0.00
609	80.40	100.00	87.99	70.83	33.35	41.54
660	83.20	100.00	88.88	74.46	34.99	44.77
710	87.03	100.00	88.29	78.56	36.20	45.47
763	87.88	100.00	88.84	79.73	37.77	46.54
814	89.16	100.00	89.61	82.95	39.59	48.96

**Table C8** Catalytic activity test of 25Ni/CZOp catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

Temperature (°C)	X <sub>CH<sub>4</sub></sub> (%)	X <sub>O<sub>2</sub></sub> (%)	S <sub>H<sub>2</sub></sub> (%)	S <sub>CO</sub> (%)	Y <sub>H<sub>2</sub></sub> (%)	Y <sub>CO</sub> (%)
402	7.53	4.55	0.00	0.00	0.00	0.00
454	8.90	15.66	0.00	0.00	0.00	0.00
505	16.39	36.12	0.00	0.00	0.00	0.00
556	23.13	67.66	0.00	0.00	0.00	0.00
605	80.40	100.00	89.47	77.81	38.34	50.33
656	83.20	100.00	89.21	79.30	38.86	51.06
705	87.03	100.00	89.56	81.16	41.31	54.11
758	87.88	100.00	90.71	85.00	44.68	57.57
808	89.16	100.00	90.66	87.40	45.43	58.36

**Table C9** Catalytic activity test of 15Ni/CZOi catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 4:1)

Temperature (°C)	X <sub>CH<sub>4</sub></sub> (%)	X <sub>O<sub>2</sub></sub> (%)	S <sub>H<sub>2</sub></sub> (%)	S <sub>CO</sub> (%)	Y <sub>H<sub>2</sub></sub> (%)	Y <sub>CO</sub> (%)
405	0.32	5.11	0.00	0.00	0.00	0.00
455	2.14	12.50	0.00	0.00	0.00	0.00
507	6.15	35.36	0.00	0.00	0.00	0.00
556	10.63	63.02	0.00	0.00	0.00	0.00
606	53.37	100.00	83.57	83.57	19.27	25.61
655	57.22	100.00	88.02	88.02	25.72	30.45
707	61.47	100.00	89.07	89.07	29.20	34.44
757	66.17	100.00	88.68	88.68	30.79	37.01
808	67.49	100.00	87.28	87.28	29.20	36.41

**Table C10** Catalytic activity test of 15Ni/CZOp catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 4:1)

Temperature (°C)	X <sub>CH4</sub> (%)	X <sub>O2</sub> (%)	S <sub>H2</sub> (%)	S <sub>CO</sub> (%)	Y <sub>H2</sub> (%)	Y <sub>CO</sub> (%)
401	2.48	12.44	0.00	0.00	0.00	0.00
453	6.63	21.19	0.00	0.00	0.00	0.00
504	13.57	44.70	0.00	0.00	0.00	0.00
555	19.57	62.44	0.00	0.00	0.00	0.00
604	69.01	100.00	86.35	90.61	28.49	33.05
655	70.99	100.00	89.23	94.82	34.04	37.95
705	73.24	100.00	89.27	94.23	35.18.	39.06
757	75.12	100.00	88.67	93.81	34.93	39.80
809	75.87	100.00	88.21	93.45	34.44	40.41

## Appendix D Experimental Data of Stability Tests for MPO

**Table D1** Stability test of 15Ni/CZOi catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

Time (hr)	X <sub>CH4</sub> (%)	X <sub>O2</sub> (%)	S <sub>H2</sub> (%)	S <sub>CO</sub> (%)
1	85.84	100.00	92.05	81.16
2	85.60	100.00	92.14	81.24
3	85.30	100.00	91.89	81.10
4	85.84	100.00	91.86	81.31
5	85.86	100.00	92.08	80.93
6	85.58	100.00	91.58	80.73

**Table D2** Stability test of 15Ni/CZOp catalyst (CH<sub>4</sub>/O<sub>2</sub> ratio = 2:1)

Time (hr)	X <sub>CH4</sub> (%)	X <sub>O2</sub> (%)	S <sub>H2</sub> (%)	S <sub>CO</sub> (%)
1	87.67	100.00	92.16	85.81
2	87.52	100.00	92.18	86.72
3	87.80	100.00	92.22	86.07
4	87.81	100.00	92.16	85.98
5	87.76	100.00	92.07	85.49
6	87.68	100.00	92.18	84.48

**Table D3** Stability test of 15Ni/CZOi catalyst ( $\text{CH}_4/\text{O}_2$  ratio = 4:1)

Time (hr)	$X_{\text{CH}_4}$ (%)	$X_{\text{O}_2}$ (%)	$S_{\text{H}_2}$ (%)	$S_{\text{CO}}$ (%)
1	64.96	100.00	89.67	89.02
2	64.94	100.00	89.43	88.42
3	64.35	100.00	89.36	87.56
4	64.10	100.00	88.63	86.85
5	64.32	100.00	89.67	85.03
6	62.77	100.00	89.43	84.17

**Table D4** Stability test of 15Ni/CZOp catalyst ( $\text{CH}_4/\text{O}_2$  ratio = 4:1)

Time (hr)	$X_{\text{CH}_4}$ (%)	$X_{\text{O}_2}$ (%)	$S_{\text{H}_2}$ (%)	$S_{\text{CO}}$ (%)
1	75.12	100.00	89.63	93.81
2	75.12	100.00	89.58	93.78
3	75.12	100.00	89.49	93.62
4	74.87	100.00	89.51	91.46
5	74.83	100.00	89.31	89.80
6	74.78	100.00	89.30	88.97

## Appendix E Calculation of Metal Surface Area

The calculation of the metal surface area is based on the following relation:

$$\text{MSS} = (\text{Vm Na}) / (\text{Sf Sd})$$

MSS = metal surface area ( $\text{m}^2/\text{g}$  of sample)

Vm = gas adsorbed at monolayer (moles/gram of sample)

Na = Avogadro number (molecule/mole of gas) =  $6.02 \times 10^{23}$  molecule/mole

Sf = stoichiometric factor of the reaction (number of molecule of adsorbate/surface metal atom) = 0.5

Sd = Metal surface density (number of metal atoms/ $\text{m}^2$ ) =  $0.154 \times 10^{20}$  atoms/ $\text{m}^2$

For example

Metal Surface Area of 5Ni/CZOi

$$\begin{aligned}\text{MSS} &= [(3.63223 \times 10^{-5}) \times (6.02 \times 10^{23})] / [0.5 \times (0.154 \times 10^{20})] \\ &= 2.84 \text{ m}^2/\text{g}\end{aligned}$$

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**Presentation:**

1. Junsai, K.; Rirksomboon, T.; and Meeyoo, V. (2015) Catalytic Performance of Ni-based Ceria-Zirconia Catalysts for Hydrogen Production: Effect of Ni loading method. Paper presented at The 6<sup>th</sup> Research Symposium on Petrochemical and Meterials Technology and the 19<sup>th</sup> PPC Symposium on the Petroleum, Petrochemicals, and Polymers, Bangkok, Thailand.